Minerals and Gemstones

300 of the Earth's natural treasures

Dr. David C. Cook & Dr. Wendy L. Kirk

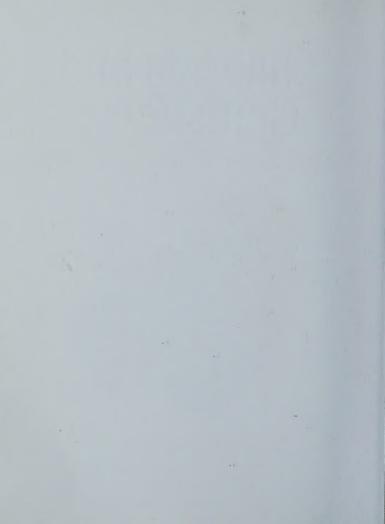






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 $\frac{Thunder Bay}{P \cdot R \cdot E \cdot S \cdot S}$

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Introduction

Minerals are mostly naturally occurring, inorganic, crystalline solids found in varying quantities in the Earth and beyond. Gems are usually minerals that are prized for beauty and strength; they have an eye catching colour, lustre or clarity coupled with a durability to make them items of lasting value. These definitions can be somewhat blurred at the edges; this has been reflected in our choice of some of the entries. To help identify minerals we list their properties after each entry.

COLOUR

The most immediately noticeable property, colour is only occasionally useful if the mineral has a distinctive, inherent colour such as yellow sulphur or blue azurite. For minerals whose colour varies with impurity content, it is much less useful. For example pure quartz is colourless, but may be of almost any hue through to black.



Lazurite has a distinct blue colour and sky-blue streak which is helpful in identification.

STREAK

Streak is the colour of a mineral in a finely powdered form. This is usually demonstrated by scratching across unglazed porcelain, crushing a sample or scratching the surface with a knife. The streak tends to remain the same for minerals which appear to be coloured differently in larger masses. It is therefore a more consistent indicator of a mineral. Streak is not useful for most silicate minerals as they are usually white and often too hard to powder easily.

Quartz has a typical crystal form of six-sided prismatic crystals with pointed terminations (as seen in the picture here).

LUSTRE

Lustre describes the nature of light from the surface of a mineral. A *metallic* lustre is shown by opaque minerals such as metals and many sulphides; if imperfect, it is called *submetallic*. A *nonmetallic* lustre is a catch-all term for all the rest and is shown by transparent and translucent minerals. It includes:



Adamantine highly reflective like diamond Vitreous glassy, as in quartz Resinous like resin, as in amber and opal Pearly like a pearl, due to alignment of platy minerals under the surface, as in talc and mother-of-pearl Silky like silk, due to an underlying fibrous structure, as in satin spar, a variety of gypsum Greasy produced by an irregular surface Earthy or dull matt surface shown by minerals having no lustre



A diamond has few weaknesses and many strengths: it is the hardest mineral known and is the highest on Moh's scale of hardness.

OPACITY OR TRANSPARENCY

It is usual to indicate if a mineral is transparent, translucent or opaque, although this is not a diagnostic property. A mineral may be inherently opaque or be so because it contains many small fragments; translucency (partial transparency) may be a function of the specimen thickness or the number of internal flaws.

HARDNESS

The hardness of a mineral has been defined as its resistance to abrasion or scratching, A practical system for measurement was devised by the Austrian mineralogist Friedrich Mohs in 1812. A set of ten common minerals was chosen of different hardnesses such that each one will scratch the surface of all softer minerals. These were then given the number 1 to 10 in increasing order of hardness. Most literature gives hardness to the nearest half unit, for example as 3.5, as a working approximation.

Mohs' scale of hardness

Tests may be roughly carried out using a fingernail (hardness 2-2.5), a copper coin (hardness 3) or a steel knife (hardness 5.5-6.5); minerals of hardness over 6 will scratch glass.

Number 1	Mineral Talc	Number 6	Mineral Orthoclase
2	Gypsum	7	Quartz
3	Calcite	8	Topaz
4	Fluorite	9	Corundum
5	Apatite	10	Diamond

SPECIFIC GRAVITY

This is the weight of a mineral compared to an equal volume of water and can be taken as equivalent to density. Thus the density of water is taken as 1 g/cm^3 (i.e. one cubic centimetre weighs one gram, or one litre weighs one kilogram), and its specific gravity is *defined* as 1 (note: it has no units).



In contrast to the diamond, Talc is the softest mineral, it is the lowest on Moh's scale of hardness.

HABIT AND FORM

The form of the mineral is given first in the properties box and describes the shape adopted by crystals. Some shapes may be well-defined closed geometric forms such as a cube, octahedron or dodecahedron, or describe an open form such as a prism. Some terms are used to describe the appearance of aggregates of crystals. Common terms used to describe single crystals or aggregates include:

Acicular needle-shaped Bladed flattened like a knife blade Botryoidal like a bunch of grapes Dendritic branching like a tree or moss Fibrous fine strands Lamellar forming distinctly flat sheets Massive crystalline aggregates with no distinct form Radiating radial arrangement of needles or fibres Reniform kidney-shaped Tabular showing broad, flat surfaces



An aggregate of natrolite crystals display an attractive radiating, acicular habit. 10



An aggregate of small cubic crystals of sal ammoniac, which is not often in a visually appealing habit.

CRYSTAL SYSTEM

Crystals are made up of atoms or molecules arranged in a regular threedimensional repeated pattern. Each unit which can be seen to repeat in order to build this structure is called a *unit cell*. Crystallography is governed by geometric possibilities (for example a cube can be repeated but a sphere or dodecahedron cannot), rather like the equivalent two-dimensional property of tessellation. Only seven possible patterns are recognised in unit cells. These are referred to as the *crystal system* adopted by that particular mineral. Each crystal system constrains the shape that crystals can adopt. Some crystal shapes can be characteristic of a crystal system such as the cube and the octahedron in the cubic system, but many require specialised knowledge and measurement to be diagnostic. Crystal systems are summarised below.

Unit cell shape
all three sides equal in length; all angles 90°
two sides equal in length; all angles 90°
all sides different lengths; all angles 90°
two angles 90°
no angles at 90°
prism of regular hexagonal cross-section
prism of regular triangular cross-section



Pseudohexagonal muscovite crystals apparently made up of stacked thin sheets – illustrating the mineral's perfect cleavage in one direction.

CLASSIFICATION

The minerals in this book are divided into groups following a conventional system based on their chemical composition. Classification is firstly on minerals in the same chemical grouping or those having similar properties. The silicates, being the largest group are further subdivided according to structure. All silicates are based on the 'silicate tetrahedron' – a silicon atom bonded to four oxygen atoms which are arranged as if at the corners of a tetrahedron. They are then classified according to how these tetrahedra are joined and arranged.

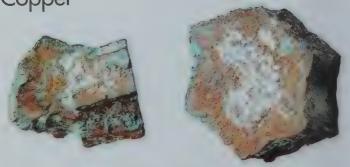
Classification

1	Native elements comprise single elements which occur uncombined in nature.
2	Sulphides contain the S ²⁻ group and are classed with arsenides, antimonides and tellurides.
3	Oxides contain the O ²⁻ group.
4	Carbonates, nitrates and borates
5	Sulphates
6	Phosphates
7	Nesosilicates contain isolated SiO ⁴⁻ tetrahedra.
8	Sorosilicates have two tetrahedra joined as Si ₂ O ₇ ⁶⁻ groups.
9	Cyclosilicates contain three, four or six tetrahedra joined as a ring.
10	Inosilicates have the tetrahedra joined into indefinite chains, usually as single or double chains.
11	Phyllosilcates comprise indefinite two-dimensional arrays of tetradedra joined at three corners in hexagonal arrangements.
12	Tectosilicates are joined at all four corners into indefinite three- dimensional frameworks containing voids or 'cages'.



Transparent crystals of hyalophane showing some brown iron oxide staining.





Copper has a characteristic reddish colour darkened by a coating of black copper oxide. Further weathering produces a covering of green copper carbonate. Native copper is quite rare, but the metal is easily obtained from ores. Copper was one of the earliest metals known, and bronze, a hard alloy of copper and tin, has been used since c. 3000 Bc. Native copper is found in altered copper deposits, cracks in basaltic lavas, and cemented sandstones and conglomerates. The southern shore of Lake Superior (USA) is the best location to find copper, allegedly the source of a piece weighing 381 tonnes (420 tons); others localities include Mansfeld (Germany), Bisbee (Arizona), Tsumeb (Namibia) and Burra Burra (Australia).

Cu	
Colour:	light rose-red on fresh surface
Lustre; opacity:	metallic; opaque, translucent green when very thin
Streak:	copper-red
Hardness:	2.5–3
Specific gravity:	8.93
Cleavage; fracture:	absent; hackly, conchoidal
Habit:	rarely hexahedral, tetrahedral, dodecahedral crystals; wiry, arborescent, massive
Crystal system:	cubic

Silver



Silver has been prized, like gold, as one of the 'noble' metals. Jewellery has been made of silver for millennia (c. 4000 BC, the ancient Egyptians were using silver beads). Sterling silver is the term for metal containing at least 92.5 per cent silver. Its high reflectivity makes silver plating an excellent coating for mirrors, utensils and ornaments. Silver's exceptionally high electrical conductivity is utilized in highquality electronics. Silver occurs in the oxidized zone of hydrothermal sulphide veins associated with other silver-bearing minerals. Good localities to find native silver include Kongsberg (Norway), Freiberg (Germany), Jáchymov (Czech Republic), the Comstock Lode (Nevada, USA) and Cobalt (Canada).

Ag	
Colour:	silver-white
Lustre; opacity:	metallic; opaque, highly reflective (95 per cent)
Streak:	silver-white
Hardness:	2.5-3
Specific gravity:	10.5
Cleavage; fracture:	none
Habit:	crystals very rare; wiry or scaly
Crystal system:	cubic





G old was the first metal known to humans, and has long been highly valued. It is chemically very unreactive and one of the 'noble' metals, used in coinage and jewellery for thousands of years. Gold is graded according to its purity, from pure 24 carat to 9 carat, which contains 37.5 per cent gold. Gold can be made harder and paler by alloying with silver, platinum, zinc or nickel. The largest nuggets include one of 153kg (337lb), found in Chile, and one of 93kg (205lb), found in Hill End (Australia). Gold is found in igneous rocks, often associated with quartz veins, and as placer deposits. The main mining areas are in Wittwatersrand (South Africa), California and Alaska (USA), Australia, South America and Siberia.

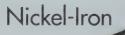
Au	
Colour:	characteristic yellow
Lustre; opacity:	metallic; opaque
Streak:	yellow
Hardness:	2.5–3
Specific gravity:	19.3
Cleavage; fracture:	none; hackly, malleable
Habit:	crystals octahedral, dodecahedral, hexahedral; grains, nuggets, compact, dendritic
Crystal system:	cubic

Mercury



Mercury, also known as quicksilver, is the only metal that is liquid at room temperature, having a freezing point of -39°C (-38°F). Such mobility is the reason It was named after the mythical messenger of the gods. Mercury occurs as small droplets in cinnabar deposits and in some volcanic rocks. It has a high coefficient of thermal expansion, which makes it useful in thermometers. Mercury forms amalgams with other metals, and these are used in gold extraction, tooth fillings and chemical processes. Locations where mercury is found include Almaden (Spain), Monte Amiata (Italy), Idrija (Slovenia), Moschellandsberg (Germany) and Juan Cavelica (Peru). Almaden is the oldest and largest mercury mine in the world.

Hq Colour tin-white Lustre; opacity: metallic; opaque Streak: not applicable at room temperature Harriness' liquid at room temperature Specific gravity: 13.6 Cleavage; fracture: not applicable at room temperature Habit: droplets: rhombohedra (at -39°C/-38°F) Crystal system: trigonal (at -39°C/-38°F)



Nickel-iron is a term for native iron, as there is invariably some nickel content. Terrestrial nickel-iron is formed in basalts in contact with carbonate rocks. Iron of low nickel content (2–3 per cent), in masses up to 23 tonnes (25 tons), is mined at Ovifak and Disko Island (Greenland). Nickel-iron in the form of meteorites is rare, but of great interest. Characteristic of these meteorites is the 'Widmanstätten pattern' of intergrowths of two different phases, brought out by chemical etching: a dark, low-nickel phase (kamacite) and a lighter one of high nickel content (taenite). The most famous meteorite, with pieces found at Canyon Diablo (Arizona, USA), fell c. 20,000–40,000 years ago, creating a crater 1.6km (1 mile) wide.

Ni–Fe	
Colour:	grey to black
Lustre; opacity:	metallic; opaque
Streak:	grey
Hardness:	4–5
Specific gravity:	7.0-7.8
Cleavage; fracture:	perfect in rare iron crystals; hackly
Habit:	never distinct crystals; massive or disseminated grains
Crystal system:	isometric

Platinum

Platinum is a rare and valuable metal, used in laboratory equipment, electrical couplings and jewellery. It is invariably mixed with impurities such as iron, iridium, palladium, rhodium, nickel and/or osmium. Platinum and the related metal palladium are used in chemistry as powerful and versatile catalysts. It was discovered by the Spanish during their conquest of South America in the sixteenth century. Named *platina del Pinto* (Pinto silver) after the Rio Pinto in Colombia, it is also found at Nizhni Tagil and Norilsk (Russia), Ontario (Canada) and Bushveld (South Africa). It occurs in igneous rocks, associated with ilmenite, magnetite and chromite, and as a placer deposit. Like gold, it does not dissolve in any acids except aqua regia.

Pt Colour: steel-grey, silvery-white Lustre; opacity: metallic; opaque Streak: steel-grey, silvery-white Hardness: 4-4.5 Specific gravity: 14-19 (21.5 when pure) Cleavage; fracture: absent; hackly Habit: rarely crystals; grains, nuggets, irregular lumps Crystal system: cubic Other: ductile, malleable



A rsenic is rare as a native element and obtained commercially as a by-product from the smelting of sulphide ores. It is well known as a poison, a property put to positive use in pesticides, preservatives and pharmaceuticals for treating parasitic illnesses. Arsenic has been added to copper to strengthen it since c. 2000 BC. The name is derived from the Greek *arsenikos*, for 'masculine' or 'brave'. Native arsenic is found in hydrothermal veins associated with silver, cobalt and nickel ores and in some igneous and metamorphic rocks. It is found in Erzgebirge and the Harz Mountains (Germany), Gikos (Russia), Sterling Hill (New Jersey, USA), Jáchymov and Pribram (Czech Republic) and St Marle-aux-Mines (France).

As	
Colour:	light grey tarnishing quickly to dark grey
Lustre; opacity:	metallic; opaque
Streak:	light grey
Hardness:	3.5
Specific gravity:	5.7
Cleavage; fracture:	perfect; uneven, brittle
Habit:	crystals rare, pseudo-cubic; granular, massive, concentrically layered nodules or stalactites
Crystal system:	trigonal

Antimony



Antimony is a metalloid related to arsenic, with some metallic and non-metallic characters. Of its three various forms, the metallic one is more stable and is bright, silvery and hard. It is used in alloys to increase hardness and lower the melting point. Antimony compounds are toxic and have been used as antiparasitic agents in medicine. The word *antimony* is its old Greek name. Native antimony is formed by the alteration of sulphides, such as stibnite, in hydrothermal veins often accompanying silver and arsenic. It is found at St Andreasberg (Germany), Sala (Sweden), Coimbra (Portugal), Pribram (Czech Republic), New Brunswick (Canada), Kern County (California, USA), Sardinia (Italy) and Sarawak (Borneo).

Sb

Colour:	light grey
Lustre; opacity:	metallic; opaque
Streak:	greys
Hardness:	3–3.5
Specific gravity:	6.6–6.7
Cleavage; fracture:	perfect; uneven, very brittle
Habit:	rare crystals as rhombohedra or coarse plates; usually massive and reniform
Crystal system:	trigonal

Bismuth

Native bismuth is rare and is found in hydrothermal veins associated with gold, Silver, cobalt, tin, nickel and lead. Bismuth is related to arsenic and antimony. It has a low melting point (271°C/520°F) and is used in low-melting alloys for electrical fuses and fire protection devices. Native bismuth is an ore, although the metal is nostly obtained as a by-product from the smelting of bismuth-rich ores. Many specimens of bismuth on sale are crystallized masses, artificially produced but nevertheless in impressive geometrical squared patterns with an attractive iridescence. Bismuth is found at Kongsberg (Norway), Erzgebirge (Germany), Cornwall (England), Ontario (Canada) and Oruro and Tasna (Bolivia).

Bi Colour: lead-grey, pink tarnish Lustre; opacity: metallic; opaque Streak: silver-white, shiny Hardness: 2 - 2.5Specific gravity: 9.7-9.8 Cleavage; fracture: perfect; uneven, sectile Habit: rare crystals; massive, granular, branching forms Crystal system: trigonal

Graphite



Graphite is a form of carbon made of stacked sheets of hexagonally arranged atoms that can quite easily slide past each other, whereas diamond (another carbon form) has a rigid three-dimensional lattice structure. Buckminsterfullerene (C_{60}), a third form, is a recently synthesized molecule found in nature in minute traces. Graphite is used as a lubricant, in crucibles, as a moderator in nuclear reactors and in pencils (as pencil 'lead'). It has a greasy feel and, unusually for a non-metal, conducts electricity. Graphite is found in metamorphosed sediments, basic igneous rocks, pegmatites and quartz veins. Graphite deposits are in Sri Lanka, Madagascar, Russia, South Korea, Mexico, the Czech Republic and Italy.

black
dull metallic; opaque
black, shiny
1-2
2.1-2.2
perfect, producing thin flexible sheets
rare six-sided flat crystals; scales, foliated or earthy masses, compact lumps
hexagonal

Diamond

Diamond is the most famous gemstone and the hardest mineral on Earth. Lowquality diamonds are used in industry for cutting and drilling equipment. Bort is a variety of diamond that has a rounded, fibrous, radiate structure, and carbonado is black and microcrystalline. A diamond is formed at very high temperatures and pressures deep within the Earth's mantle (c. 80 km/50 miles), and is then brought to the surface through kimberlite pipes. Diamonds are found in ultramafic rocks, especially kimberlite breccias, and as placer deposits. The main mining areas include the Argyle pipe (Western Australia), Kimberley (South Africa), Golconda (India), Diamontina (Minas Gerais, Brazil) and Yakutia (Russia).

С	
Colour:	colourless, yellowish to yellow, brown, black, blue, green or red, pink, champagne-tan, cognac-brown
Lustre; opacity:	adamantine, greasy; transparent to opaque
Streak:	none, too hard to abrade
Hardness:	10
Specific gravity:	3.52
Cleavage; fracture:	perfect; conchoidal; uneven
Habit:	octahedral, dodecahedral, cubic crystals
Crystal system:	cubic

Diamond: Gem Varieties



Diamond's unequalled hardness, brilliant lustre and considerable 'fire' make it the most highly prized of gems. Jewellers usually favour the 'brilliant' cut, which gives maximum internal reflection of light and thus minimum loss through the back of the stone, enhancing the natural fire. Diamonds are measured in carats, derived from the weight of a carob seed pod, which is now a standard 0.2g. The Cullinan diamond, originally from South Africa, was of 3106 carats and cut into 104 gemstones, the largest of which was of 531 carats. The Koh-i-Noor diamond (109 carats) is set in the crown of British Queens. The largest deep blue diamond, and some would say the most beautiful, is the Hope diamond (44.5 carats).

С	
Colour:	colourless, yellowish to yellow, brown, black, blue, green or red, pink, champagne-tan, cognac-brown
Lustre; opacity:	adamantine, greasy; transparent to opaque
Streak:	none, too hard to abrade
Hardness:	10
Specific gravity:	3.52
Cleavage; fracture:	perfect; conchoidal; uneven
Habit:	octahedral, dodecahedral, cubic crystals
Crystal system:	cubic



Sulphur has a characteristic yellow colour, is very brittle and will often crumble if roughly handled. Also called brimstone, it usually forms in volcanoes as a sublimate around vents and fumaroles. Sulphur can also be found precipitated in hot springs, in shales associated with gypsum and bitumen, and in the cap rock of salt domes. Sulphur is used in gunpowder, matches, sulphuric acid production and is extremely important as the vulcanizing agent for rubber. It has a low melting point (113°C/235°F) and burns to give sulphur dioxide. Sulphur is mined in Louisiana and Texas (USA), Sicily, Japan, Indonesia and the South American Andes. The best examples of crystalline sulphur come from Sicily and the Romagna (Italy).

S	
Colour:	yellow, brownish or greenish yellow, orange, white
Lustre; opacity:	resinous, greasy; transparent to translucent
Streak:	none
Hardness:	1.5–2.5
Specific gravity :	2.05-2.08
Cleavage; fracture:	imperfect, fair; irregular, uneven, conchoidal
Habit:	massive, encrusting, powdery and stalactitic
Crystal system:	orthorhombic

Chalcocite



Chalcocite occurs in two distinct forms, forming or thorhombic crystals below 103°C (217°F) and hexagonal crystals above this. Chalcocite is also called chalcosine or copper glance. It is found in hydrothermal veins and in the reduced zones of copper deposits. It is an important copper ore, often produced by the alteration of chalcopyrite, with which it is often associated, and which, in turn, may be altered to malachite, azurite and covellite. Good examples of pseudohexagonal crystals come from the Transvaal (South Africa), Bristol (Connecticut, USA) and Redruth (Cornwall, England). Large deposits are mined in Butte (Montana, USA), Bisbee (Arizona, USA), Chuquicamata (Chile), Tsumeb (Namibia), Peru and Russia.

Cu₂S Colour: black or lead grey, often with greenish or bluish tarnish Lustre; opacity: metallic; opaque Streak: greyish black, sometimes shiny Hardness: 2.5 - 3Specific gravity: 5.5-5.8 Cleavage; fracture: indistinct; conchoidal, uneven Habit: rarely as tabular, pseudohexagonal, striated crystals; mostly dull grey, granular aggregates Crystal system: orthorhombic, hexagonal above 103°C (217°F)

SULPHIDES

Bornite



Bornite, named after the Austrian mineralogist Ignatius von Born (1742–1791), Bis known as peacock ore because of the beautiful iridescent tarnish shown by many specimens. Too much weathering, however, produces an unattractive black tarnish. Bornite is a dense mineral that settles in magmas and so is found concentrated in igneous rocks, especially mafic rocks. It also occurs in pegmatites, hydrothermal veins and as a secondary mineral in copper deposits. A well-known deposit is found in the copper shales of Mansfeld (Germany). Good crystals come from Cornwall (England), Tsumeb (Namibia) and Butte (Montana, USA). Large quantities are mined in the USA, Mexico, Peru, Chile, Australia and Zambia.

Cu ₅ FeS ₄	
Colour:	copper-red or bronze when fresh, tarnishes to an iridescent blue, red and/or purplish surface film
Lustre; opacity:	metallic; opaque
Streak:	grey-black
Hardness:	3
Specific gravity:	5.1
Cleavage; fracture:	poor/indistinct; conchoidal, indistinct
Habit:	crystalline as cubes, dodecahedra or octahedra; commonly as compact granular masses
Crystal system:	orthorhombic

Acanthite-Argentite

A canthite occurs in the monoclinic system below 179°C (354°F); above this, cubic argentite is the stable form. As it is usually deposited at high temperatures, acanthite often occurs as cubic pseudomorphs after argentite. Acanthite's metallic lustre quickly blackens and is best seen on a freshly exposed surface. It is named from the Greek for 'arrow'; argentite is named from the Latin *argentum*, meaning 'silver'. It is found in hydrothermal veins with other silver minerals or disseminated in galena deposits. It can also occur in cemented parts of lead and zinc deposits. Acanthite is the main ore for silver and is mined extensively in Mexico, Peru and Honduras. Good crystals come from Kongsberg (Norway) and Freiberg (Germany).

Ag ₂ S	
Colour:	shiny lead grey, black on surface
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	2-2.5
Specific gravity:	7.2-7.4
Cleavage; fracture:	poor; conchoidal, uneven
Habit:	distorted, pseudo-cubic crystals in groups; dendritic aggregates, masses and encrustations
Crystal system:	monoclinic (acanthite) – cubic (argentite)

SULPHIDES Sphalerite

Sphalerite or zinc blende is an important zinc ore, also providing cadmium, gallium and indium as by-products. Zinc is mostly used as sheets for galvanizing iron, and alloyed with copper to make brass. Sphalerite can be mistaken for galena, hence its name, from the Greek for 'treacherous'. The colour of sphalerite tends to darken as the iron content increases. A content of about 10 per cent imparts a black colour; any higher, and the ore may be called marmatite. Sphalerite forms in pegmatites and hydrothermal veins accompanying galena, acanthite, barite and chalcopyrite. It is found at Alston Moor (England), the Tri-State area (USA), Broken Hill (NSW, Australia), Kapnik (Hungary), Santander (Spain) and Sullivan (Canada).

ZnS	
Colour:	usually brown or black, also yellow or reddish, rarely colourless
Lustre; opacity:	resinous, greasy; transparent to translucent to opaque
Streak:	brownish white
Hardness:	3.5-4
Specific gravity:	3.9–4.2
Cleavage; fracture:	perfect; conchoidal, brittle
Habit:	dodecahedral and octahedral crystals common; massive, compact, botryoidal or fibrous
Crystal system:	isometric

SULPHIDES



Chalcopyrite, or copper pyrites, is the most important copper ore. By-products from copper extraction include silver and gold. It can be distinguished from pyrite by the fact that it does not produce sparks when hit by a hammer and also because it crumbles when cut with a knife due to poor cleavage. It can be distinguished from pyrrhotite because it is non-magnetic, and from gold because it is brittle. Chalcopyrite occurs mainly in hydrothermal veins associated with cassiterite, galena, pyrite, quartz, calcite and/or sphalerite, and in metamorphosed volcanic rocks. It is found in the Copper Belt of Zambia, Rio Tinto (Spain), Katanga (Congo), Cyprus, the Urals (Russia), and Montana, Arizona and Utah (USA).

CuFeS₂

Colour:	brass yellow, honey yellow
Lustre; opacity:	metallic; opaque
Streak:	greenish black
Hardness:	3.5
Specific gravity:	4.1-4.3
Cleavage; fracture:	indistinct; conchoidal, uneven
Habit:	pseudo-tetrahedral crystals, uncommon; usually massive and compact, sometimes reniform or mammilated
Crystal system:	tetragonal

SULPHIDES

Tetrahedrite-Tennanite



retrahedrite and tennanite form a continuous series. The replacement of sulphur for tellurium in goldfieldite and of silver for iron in freibergite also occurs. It can act as a geothermometer because the silver content rises as the temperature of formation falls. Most mineral specimens are antimony-rich tetrahedrite. The group have been called fahlerz minerals, and are copper ores that have also been used for silver, mercury and antimony recovery. They occur in hydrothermal veins associated with copper, lead, zinc and silver minerals. Tetrahedrite-tennannite is found in Botés and Kapnik (Romania), Boliden (Sweden), Pribram (Czech Republic), Tsumeb (Namibia) and Butte (Montana, USA).

$(Cu,Fe)_{12}Sb_4S_{13} - (Cu,Fe)_{12}As_4S_{13}$		
Colour:	steel grey to brown, occasionally twinned	
Lustre; opacity:	metallic; opaque	
Streak:	dark grey	
Hardness:	3-4.5	
Specific gravity:	4.6-5.2	
Cleavage; fracture:	none; conchoidal, uneven	
Habit:	crystals often modified tetrahedral; massive, granular, compact	
Crystal system:	isometric	
Other:	melt easily on heating; soluble in nitric acid	

Stannite



C tannite, named from the Latin *stannum* for tin, is also known as tin pyrites, Doljvianite or bell metal ore. It has been worked as a tin ore in Cornwall and forms part of silver deposits in Bolivia. It is a fairly rare mineral found in hydrothermal veins and pegmatites associated with cassiterite, wolframite, pyrite and arsenopyrite. It may contain other metals such as silver, cadmium and indium, and is called zincian stannite when zinc-rich. Stannite is found at Etna Mine (South Dakota, USA), Seward Peninsula (Alaska, USA), Llallagua (Bolivia), Cinovec (Czech Republic), Zeehan (Tasmania, Australia) and Wheal Rock (Cornwall, England), the last being the type locality for the mineral.

Cu ₂ FeSnS ₄	
Colour:	greyish black, steel grey; often has an iridescent olive green or blue surface
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	3.5-4
Specific gravity:	4.3-4.5
Cleavage; fracture:	imperfect; conchoidal, uneven
Habit:	crystals rare, tetrahedral or more often pseudooctahedral due to twinning; massive, granular, disseminated
Crystal system:	tetragonal



Wurtzite is dimorphic with sphalerite, the more common form of zinc sulphide. The zinc can be replaced partially by iron, giving a darker colour. A rarer trigonal polymorph exists, called matraite. Crystals are usually quite small, but can be delightful shapes of somewhat elongated six-sided pyramids on a flat hexagonal base. Wurtzite is found in hydrothermal veins associated with sphalerite, pyrite, chalcopyrite, barite and marcasite. It is named after the French chemist Charles Wurtz (1817–1884). Locations include Thomaston Dam (Connecticut, USA), Butte (Montana, USA), Frisco (Utah, USA), Cornwall (England), Pribram (Czech Republic), Baia Sprie (Romania) and Oruro and Potosi (Bolivia).

ZnS	
Colour:	light to dark brown
Lustre; opacity:	adamantine, greasy; translucent to opaque
Streak:	light brown
Hardness:	3.5-4
Specific gravity:	3.98–4.08
Cleavage; fracture:	good; uneven
Habit:	rare crystals usually hemimorphic pyramids, sometimes hexagonal tabular crystals or short prisms; usually concentrically banded crusts, fibrous or columnar
Crystal system:	hexagonal

Greenockite



G reenockite is a cadmium ore named after Lord Greenock, on whose land near Glasgow it was first found. Other cadmium ore minerals are cadmoselite (CdSe) and otavite (CdCO₃), and all are quite rare. Commercial production of cadmium, however, is mostly as a by-product in the smelting of zinc sulphide ores. Cadmium is used in rechargeable batteries, alloys and pigments. Greenockite is a secondary mineral formed on the surface of cadmium-rich sphalerites. It is brightly coloured when fresh, but becomes dull on weathering. Tiny crystals are found at Llallagua (Bolivia), Bishopstown (Scotland) and Paterson (New Jersey, USA); encrustations are found at Pribram (Czech Republic), Joplin (Missouri, USA) and Sardinia (Italy).

CdS

Colour:	yellow, orange, red
Lustre; opacity:	adamantine, resinous; translucent
Streak:	orange yellow
Hardness:	3-3.5
Specific gravity:	4.9-5
Cleavage; fracture:	distinct, imperfect
Habit:	prismatic, hexagonal crystals rare, sometimes twinned; usually powdery encrustations or coatings
Crystal system:	hexagonal

Enargite



Enargite is a minor ore of copper and arsenic named from the Greek enarges, Immeaning 'obvious', because of its notably distinct cleavage. It is a dimorph of the mineral luzonite and is often mixed with the analogous antimony mineral stibioluzonite (Cu₃SbS₄). Twinning can give rise to attractive star shapes called 'trillings'. Enargite occurs in hydrothermal veins associated with chalcopyrite, tetrahedrite, bornite, pyrite, barite, quartz and covellite. It is found at Bor (Serbia), Chuquicamata (Chile), Bingham and Tintic (Utah, USA), Butte (Montana, USA), Luzon Island (Phillipines), Morococha, Quiruvilca and Cerro de Pasco (Peru), and Freiberg (Germany). Prisms up to 8cm (3in) have been found at Tsumeb (Namibia).

Cu ₃ AsS ₄	
Colour:	grey to black, can have violet or rose-brown internal reflections
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	3 .
Specific gravity:	4.4-4.5
Cleavage; fracture:	perfect; uneven
Habit:	crystals rare tabular, blocky or prismatic often pseudohexagonal and striated ; usually as aggregates or granular masses
Crystal system:	orthorhombic

Galena



G alena is a common sulphide and is the main ore for lead, used and named by the ancient Greeks. Lead is used in batteries, glass, solder and radiation shields. Up to 1 per cent silver is found in galena, which makes it an important silver ore. It is a natural semiconductor and was one of the favourite 'crystals' used as crude diodes in crystal radio sets. Galena occurs in hydrothermal veins associated with chalcopyrite, pyrite, sphalerite barite, calcite, fluorite and quartz. It is mined in the Tri-State area of Missouri, Oklahoma and Kansas (USA), Broken Hill (Australia) and Santa Eulalia (Mexico); crystals are found at Pribram (Czech Republic), Isle of Man (UK) and Sardinia (Italy).

PbS	
Colour:	light lead grey, dark lead grey
Lustre; opacity:	metallic; opaque
Streak:	greyish black
Hardness:	2.5
Specific gravity:	7.2-7.6
Cleavage; fracture:	perfect; soft no fracture
Habit:	well-formed crystals; massive, granular
Crystal system:	cubic

Cinnabar



Cinnabar is a toxic, dense red mineral formerly used as a pigment known as vermilion, the use of which began in ancient times. Free mercury can be produced by heating cinnabar above 580°C (1076°F). Mercury compounds are used in fine chemicals and paints, and mercury itself in thermometers and scientific instruments. Cinnabar is found in a variety of rocks often associated with volcanic activity, such as hot springs. It also occurs in hydrothermal veins and as placer deposits. The three most important locations are Almadén (Spain), Monte Amiato (Italy) and Idrija (Slovenia). Other locations include Nikotawa (Russia), Hunan Province (China) and the Altai Mountains of Central Asia.

HgS	
Colour:	bright scarlet to brick red
Lustre; opacity:	adamantine; translucent
Streak:	bright red
Hardness:	2-2.5
Specific gravity:	8.1
Cleavage; fracture:	perfect; uneven, splintery
Habit:	rare rhombohedral or thick tabular crystals; earthy films; massive or granular
Crystal system:	trigonal



Pyrrhotite is distinguishable from pyrite in being magnetic, hence the alternative name magnetic pyrites. It is somewhat deficient in iron, there being approximately 11 sulphur atoms per 10 iron atoms. The mineral corresponding exactly to the formula FeS, which has been found only in meteorites, is called troilite. Named after the Greek *pyrrotes*, for 'red coloured', pyrrhotite is not a major iron ore, but nickel-rich pyrrhotite deposits are used to extract nickel, cobalt and platinum. It is found in mafic and ultramafic igneous rocks, hydrothermal veins and some high-grade metamorphic rocks. Good crystals come from Trepca (Serbia), Kysbanya (Romania), New York (USA) and Freiberg (Germany).

FeS

Colour:	bronze, bronze red, dark brown
Lustre; opacity:	metallic; opaque
Streak:	grey black
Hardness:	3.5–4
Specific gravity:	4.58–4.65
Cleavage; fracture:	imperfect; uneven
Habit:	prismatic or tabular crystals; massive, granular
Crystal system:	monoclinic (hexagonal when low in sulphur, close to FeS)

Miargyrite



Miargyrite, common among silver ore deposits, is named from the Greek for ismaller' and 'silver', alluding to its silver content, which is lower than the mineral pyrargyrite. Easily confused with other silver minerals, miargyrite may be distinguished by its unusual deep red streak. Crystals usually grow to only about lcm (½in), but the red internal reflections seen in this mineral make it appealing. It is found in hydrothermal veins associated with pyrargyrite, silver, galena, sphalerite, quartz, calcite and barite. Miargyrite is found at Baia Sprie (Romania), Pribram and Trebsko (Czech Republic), Randsberg (California, USA), Owyhee (Idaho, USA), Copiapo and Tarapaca (Chile), Potosi and Huanchaka (Peru), and Freiberg (Germany).

grey to black with dark red internal reflections
metallic; opaque
cherry red
2.5
5.25
imperfect/fair; subconchoidal
well-formed crystals, often coarse plates, blades, equant or wedge-shaped; granular, massive aggregates, disseminated
monoclinic

Nickeline

Nickeline, or niccolite, is the first mineral from which nickel was extracted, but is now only a minor ore. The names of both mineral and element come from the German *kupfernickel*, or devil's nickel, because it was impossible to extract copper from the mineral, despite its copper-like appearance. It occurs in hydrothermal veins or is disseminated in basic igneous rocks such as gabbros, and is associated invariably with sulphides of silver, nickel and cobalt. Specimens are often coated with a pale to dark green film of annabergite. Nickeline is found at Eisleben and Freiberg (Germany), Schladming (Austria), Cobalt and Eldorado (Canada), Natsume (Japan), Bou Azzer (Morocco), Anarak (Iran) and La Rioja (Argentina).

NiAs Colour: copper-red Lustre; opacity: metallic: opaque Streak: brownish black Hardness: 5.5 7 78-7 8 Specific gravity: Cleavage; fracture: imperfect; uneven to conchoidal Habit: crystals stocky, tabular or pyramidal, rare; columnar; massive; reniform Crystal system: hexagonal

Millerite

Millerite is a widespread but uncommon minor ore of nickel, also found in ironnickel meteorites, albeit in minute quantities. Named after the English mineralogist W.H. Miller (1801–1880), it is also known as hair pyrites after its fine acicular crystals, or capillary pyrite after its occurrence as fine hollow crystals. It is a hydrothermal deposit found in cavities in limestones and dolomites, on barite and as an alteration product of nickel minerals. Millerite is found at Ramsbeck and Kamsdorf (Germany), Kotalahti (Finland), Keokuk (Iowa, USA), St Louis (Missouri, USA), Temagami (Ontario, Canada), Thompson (Manitoba, Canada), Onllwyn (Wales) and Kambalda and Leinster (Australia).

NiS

Colour:	brassy yellow, iridescent on tarnishing
Lustre; opacity:	metallic; opaque
Streak:	greenish black
Hardness:	3–3.5
Specific gravity:	5.2–5.6
Cleavage; fracture:	perfect; uneven
Habit:	acicular crystals, often in tufts or felted masses; rarely granular or massive
Crystal system:	trigonal

Covellite



Covellite is a rare mineral, also called indigo copper, occasionally found as flat, hexagonal crystals up to 10cm (4in) large and attractive rosettes of platy crystals. Covellite crystals are indigo blue and show pleasing iridescent yellow and red flashes. Strongly tarnished samples are coloured purple or black. It was discovered on Vesuvius by the Italian mineralogist N. Covelli (1790–1829). Covellite usually occurs in the oxidized zones of copper deposits and in volcanic sublimates, associated with chalcopyrite, chalcocite, bornite and pyrite. Covellite is found at the Calabona mine (Sardinia) and Vesuvius (Italy), Salzberg (Austria), Butte (Montana, USA), Kennicott (Alaska, USA), Bor (Serbia) and Bou Azzer (Morocco).

CuS

Colour:	indigo blue, strongly iridescent
Lustre; opacity:	metallic; opaque
Streak:	black grey
Hardness:	1.5–2
Specific gravity:	4.6-4.76
Cleavage:	perfect
Habit:	foliated, platy crystal aggregates; rarely hexagonal, flattened crystals; compact masses
Crystal system:	hexagonal

Pyrite



Pyrite, also called iron pyrite or simply pyrites, is also well known as fool's gold. It has often been mistaken for gold due to its yellow metallic lustre. Unlike gold, however, pyrite is hard, brittle and often unmistakably crystalline. Pyrite is used as a source of sulphur, especially for sulphuric acid manufacture. It occurs in igneous rocks, and is found in hydrothermal veins associated with sphalerite, galena, quartz, copper sulphides and gold, the latter two sometimes being commercially extracted from pyrite-rich ores. Pyrite also occurs in some metamorphic rocks and as pseudomorphs infilling the shapes of fossils. Excellent crystals are to be found in many locations. Pyritized fossils are found in Germany, England and Italy.

FeS ₂	
Colour:	pale brass yellow
Lustre; opacity:	metallic; opaque
Streak:	greenish black
Hardness:	6.5
Specific gravity:	5–5.02
Cleavage; fracture:	poor; uneven to conchoidal
Habit:	striated cubes, octahedra or pyritohedra, sometimes as 'iron cross' twins; compact granular aggregates, nodules, concretions and stalactitic forms, pseudomorphs after fossils
Crystal system:	cubic

Pyrite Varieties



Pyrite is very popular as large cubes and distorted dodecahedra called pyritohedra. Large crystals, especially cubes in metamorphic rocks, can make intriguing specimens with a bright, golden cube embedded incongruously in a contrasting granular or schistose matrix. Discs of sectioned nodules showing a radiating internal structure make attractive ornamental pieces. Pyrite used in jewellery is sometimes called marcasite by jewellers, especially when used as faceted stones set in silver. Pyrite is difficult to cut for jewellery because of its brittleness. It has been used by the ancient Greeks to adorn earrings and pins, and for jewellery in Victorian times. The Incas are said to have used pyrite tablets as mirrors.

FeS ₂	
Colour:	pale brass yellow
Lustre; opacity:	metallic; opaque
Streak:	greenish black
Hardness:	6.5
Specific gravity:	5–5.02
Cleavage; fracture:	poor; uneven to conchoidal
Habit:	striated cubes, octahedra or pyritohedra, sometimes as 'iron cross' twins; compact granular aggregates, nodules, concretions and stalactitic forms, pseudomorphs after fossils
Crystal system:	cubic

Stibnite



Stibnite, also known as antimonite, is the major ore of antimony. The name is derived from the old Greek word for antimony, *stibi*. Widely distributed but in small deposits, stibnite is mostly found in hydrothermal veins, but can occur in association with hot springs, associated with sulphides of silver, lead and mercury, pyrite, galena and quartz. Stibnite can form striking arrays of roughly aligned or radiating elongated prisms. The largest deposits of stibnite are in Hunan province (China); others locations include Shikoku Island (Japan) and Baia Sprie and Kapnik (Romania). Antimony is used in alloys to harden other metals, especially lead in storage batteries. It is increasingly used in semiconductors.

Sb ₂ S ₃	
Colour:	lead grey, bluish lead grey, steel grey, black
Lustre; opacity:	metallic; opaque
Streak:	blackish grey
Hardness:	2
Specific gravity:	4.63
Cleavage; fracture:	perfect; conchoidal
Habit:	granular, prismatic crystals, striated surface or cleavage face
Crystal system:	orthorhombic



Cobaltite is a major ore of cobalt, the name of which comes from the German *kobold*, meaning 'underground spirit' or 'goblin', in reference to the difficulty of smelting it. Crystal forms of cobaltite are similar to those of pyrite, but the two are easily distinguished by colour. Cobaltite is often weathered to give crusts of pink to bright purple erythrite, called cobalt bloom by miners (evidence of underlying cobalt minerals). It is found in hydrothermal veins or contactmetamorphosed rocks. Excellent crystals come from Tunaberg (Sweden), Skutterud (Norway), Cornwall (England), Española, Cobalt and Sudbury (Ontario, Canada), Broken Hill and Torrington (NSW, Australia), and Bou Azzer (Morocco).

CoAsS	
Colour:	reddish silver white, violet steel grey, black
Lustre; opacity:	metallic; opaque
Streak:	greyish black
Hardness:	5.5
Specific gravity:	6.33
Cleavage; fracture:	good; uneven
Habit:	cubes, octahedra or pyritohedra, usually striated; granular or compact masses, disseminated
Crystal system	orthorhombic

Bismuthinite



Bismuthinite is similar in appearance and properties to stibnite, but can be distinguished by its inability to melt in a match flame. Sprays of steel grey prismatic crystals resemble those of stibnite. It is the major ore of bismuth, but the free metal is mostly obtained as a by-product of lead and copper smelting. Substitution of bismuth for lead and copper gives the mineral aikinite (CuPbBiS₃), which forms a series with bismuthinite. It occurs in hydrothermal veins associated with tin, silver and cobalt minerals. Bismuthinite is found at Haddam (Connecticut, USA), Liallagua and Tasno (Bolivia), Cerro de Pasco (Peru), Cornwall (England) and Mount Biggenden (Australia).

grey, silver white, tin white
metallic; opaque
grey
2
6.8-7.2
perfect; uneven
prismatic, acicular crystals, finely striated; granular and
compact aggregates
orthorhombic

Sylvanite



Sylvanite is one of the few ores of gold, other than native gold itself, and also of silver and tellurium. Gold, normally very unreactive, has a particular affinity for tellurium. Sylvanite is named after Transylvania, where the mineral was first discovered. It is rare and not commercially mined. Tellurium is obtained from the anode slime in copper refining. Although less than 0.1 per cent is usually added to steels, more than half of the tellurium production is used in this way. Sylvanite occurs in hydrothermal veins, associated with calaverite (AuTe₂) and petite (Ag₃AuTe₂). It is found at Baia de Aries (Romania), Kalgoorie (Australia), Bereznyakov and Yaman-Kasy, Ural Mountains (Russia), and Cripple Creek (Colorado, USA).

AgAuTe₄

Colour:	yellowish silver white, white
Lustre; opacity:	metallic; opaque
Streak:	steel grey
Hardness:	1.5–2
Specific gravity:	7.9–8.3
Cleavage; fracture:	perfect; uneven
Habit:	stubby, prismatic or arborescent crystals; branching encrustations resembling script; granular or bladed masses
Crystal system:	monoclinic

Hauerite



H auerite is a rare form of manganese disulphide named after the Austrian geologists J. and F. von Hauer. The finest specimens are large octahedral crystals sometimes modified by cubic faces, from the Destricella mine (Raddusa, Sicily). Specimens often comprise hauerite with associated rambergite (MnS). Hauerite occurs in sulphur-rich clay deposits and altered lavas associated with sulphur, realgar, gypsum, aragonite and calcite, and in the ferromanganese deposits in the Pacific Ocean. It is found at Kalinka and Banská Stiavnica (Slovakia), Bohemia (Czech Republic), Jezyorko and Grzybow (Poland), Raddusa (Italy), Yazovsk and Podgornensk (Ural Mountains, Russia) and in salt domes in Texas (USA).

MnS ₂	
Colour:	brownish grey to brownish black, reddish tints
Lustre; opacity:	metallic; opaque
Streak:	reddish brown
Hardness:	4
Specific gravity:	3.46
Cleavage; fracture:	perfect; uneven
Habit:	octahedral and cubo-octahedral crystals up to 5 cm (2 in), often fractured unevenly; rounded aggregates
Crystal system:	isometric



Ullmannite is a rare form of nickel antimony sulphide of the cobaltite group, named after the German chemist J. Ullmann (1771–1821). It is closely related to gersdorfite (NiAsS) and will almost invariably contain some arsenic. Ullmannite also forms a series with willyamite, (Co,Ni)SbS. Crystals grow to about 3cm (1½in) and resemble those seen in pyrite. Ullmannite occurs in hydrothermal veins associated with minerals such as skutterudite, galena, nickeline, pyrrhotite and tetrahedrite. It is found at Siegerland, Harzgerode and Lobenstein (Germany), Waldenstein and Lolling (Austria), Fourstones (Northumberland, England), Durham (England), Broken Hill (NSW, Australia) and Sarrabus (Sardinia, Italy).

NiSbS

Colour:	steel grey, silver white, tin white
Lustre; opacity:	metallic; opaque
Streak:	greyish black
Hardness:	5-5.5
Specific gravity:	6.65
Cleavage; fracture:	good; uneven
Habit:	crystals as cubes, octahedra, dodecahedra and tetrahedra, massive, granular
Crystal system:	cubic

Marcasite



Marcasite is a polymorph of pyrite, and is used, like pyrite, for sulphuric acid production, not as an iron ore. It is called spear pyrites after the arrow- or spear-head shape of twinned crystals; it is also known as white iron pyrite, as it has a paler colour than pyrite. Marcasite was formerly used in jewellery, but is now likely to be pyrite. Marcasite forms in hydrothermal veins, often in lead and zinc ores, is precipitated in sedimentary rocks and can be found replacing fossils. Marcasite is found at Carlsbad and Rammelsberg (Germany), Karlovy Vary (Czech Republic), Derbyshire (England) and the Tri-State mining area (USA). Nodules with a radiating internal structure are found in the chalk of Southeast England.

FeS ₂	
Colour:	steel grey, silver white
Lustre; opacity:	metallic; opaque
Streak:	greyish black
Hardness:	5.5
Specific gravity:	6.65
Cleavage; fracture:	good; uneven
Habit:	crystals as flat prisms, occasionally as rosettes, often twinned in shapes described as cockscombs and spearheads; massive, granular, crusty aggregates, concretions
Crystal system:	cubic



A rsenopyrite is common in sulphide deposits and is the major ore of arsenic, and is often also rich in silver, gold, cobalt and tin. Arsenic, however, is obtained commercially as a by-product during the refining of sulphide ores. Arsenopyrite has a silvery colour that distinguishes it from pyrite and marcasite, although it can tarnish on exposure. Mispickel and arsenical pyrites are alternative names. It occurs in sulphide deposits, in hydrothermal veins and in some metamorphic rocks. Good crystals are found at Roxbury (Connecticut, USA), Leadville (Colorado, USA) and Valle Anzasca and Val Sugana (Italy). Large deposits are at Boliden (Sweden), Freiberg (Germany), Deloro (Canada), Sulitjelma (Norway) and Cornwall (England).

FeAsS

Colour:	tin white, light steel grey, can have a pink tint
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	5
Specific gravity:	6.07
Cleavage; fracture:	distinct; uneven
Habit:	crystals as elongated, striated prisms and cruciform twins,
	granular masses
Crystal system:	monoclinic

Glaucodot



G laucodot contains up to 25 per cent cobalt, but is not a commercially important ore. It is regarded by some as a member of the arsenopyrite-cobaltite (FeAsS-CoAsS) series, but these crystallize in different systems. The name comes from the Greek for 'blue', after its use in a dark blue glass called smalt. It occurs in hydrothermal veins associated with sulphides and in metamorphosed lavas. Weathering and alteration often produce a bloom of erythrite. Good crystals come from Hakansbö (Sweden) and Huasco (Tarapaca, Chile). Other localities include Cobalt (Ontario, Canada), Oravita (Romania), Skutterud (Norway), Sumpter (Oregon, USA), Franconia (New Hampshire, USA) and Alston (Cumbria, England).

(Cu,Fe)AsS

Colour:	greyish or reddish silver-white
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	5
Specific gravity:	5.9-6.01
Cleavage; fracture:	perfect; uneven, fragile
Habit:	prismatic or elongate crystals; granular or radiating fibrous masses
Crystal ystem:	orthorhombic

Skutterudite

Skutterudite, also known as smaltite, is one of the major ores of cobalt. The cobalt is invariably partially replaced by other elements, such as copper, zinc, silver and particularly nickel. When nickel forms the major component, it is referred to as nickel-skutterudite. It forms fine crystals resembling those of pyrite, but with a silvery colour, and is often altered to crimson red erythrite. Skutterudite is formed in hydrothermal veins associated with arsenopyrite, calcite and sulphides of cobalt and nickel. It is named after Skutterud in Norway, where it occurs as fine crystals; it is also found at Cobalt (Canada), Huelva (Spain), Schneeberg (Germany) and Bou Azzer (Morocco). Deposits are mined in Germany, Austria and the Czech Republic.

(Co,Ni)As₃ Colour: white, light steel grey, sometimes with a an iridescent film Lustre; opacity: metallic; opaque Streak: black Hardness: 5.5-6 Specific gravity: 6.1-6.9 Cleavage; fracture: distinct; conchoidal Habit: crystals are cubic, octahedral or pyritohedral; compact granular masses Crystal system: cubic



Molybdenite forms two-dimensional sheets, and these, like micas and graphite, give rise to flaky crystals. It is an excellent high-temperature dry lubricant and the main ore of molybdenum. Similar in appearance to graphite, it has a higher density and a more metallic bluish appearance. The name is derived from the Greek molybdos, for lead having been mistaken for galena. It mostly occurs in granitic rocks, in pegmatites and pneumatolytic veins, and in contact-metamorphosed rocks. Good crystals are found at Edison (New Jersey, USA), Climax (Colorado, USA), Arendal (Norway), Temikaming (Quebec, Canada) and Kingsgate (NSW, Australia), the latter providing crystals 70 x 5mm (2½ x ½in).

MoS ₂	
Colour:	bluish-grey
Lustre; opacity:	metallic; opaque
Streak:	bright blue-grey
Hardness:	1-1.5
Specific gravity:	4.7
Cleavage; fracture:	perfect
Habit:	tabular hexagonal crystals; usually as bladed, foliated or interwoven masses
Crystal system:	hexagonal

Proustite



Proustite forms beautiful translucent cinnabar-red crystals, but exposure to light and air causes them to darken, becoming semi-opaque with a grey streak. Rarely for a sulphide mineral, it is not metallic or opaque. Resembling pyrargyrite, it was identified only by analysis by the French chemist J. Proust (1755–1826). Proustite is also known as ruby silver ore or light red silver ore. It is found in oxidized zones of hydrothermal veins associated with other silver and sulphide minerals. Proustite crystals of up to 15cm (6in) have been found at Chañarcillo (Chile); other locations are Jáchymov and Pribram (Czech Republic), Chihuahua and Zacatas (Mexico), Annaberg and Wittichen (Germany), and Sarrabus (Sardinia).

Ag ₃ AsS ₃	
Colour:	dark red, darkening on exposure to light and air
Lustre; opacity:	adamantine; semi-transparent to translucent
Streak:	brick red, becoming grey in altered specimens
Hardness:	2–2.5
Specific gravity:	5.57
Cleavage; fracture:	good, rhombohedral; conchoida
Habit:	crystals rare, as rhombohedra or scalenohedra, usually striated and distorted; usually massive
Crystal system:	trigonal



Pyrargyrite is related to proustite, with antimony replacing arsenic, both in the hexagonal system. To distinguish it from proustite, pyrargyrite has been called dark red silver ore, but is also referred to as ruby silver. Pyrargyrite crystals can be striking and attractive, but darken fairly quickly on exposure to light and become opaque, posing problems with storage and display. It is formed in hydrothermal veins associated with other silver minerals and by alteration of argentite and silver-rich galena. Good hexagonal crystals occur at St Andreasberg and Freiberg (Germany), and Colquechaca (Bolivia); other localities are Sarrabus (Sardinia), Pribram (Czech Republic), Zacetas (Mexico) and Chañarcillo (Chile).

Ag ₃ SbS ₃	
Colour:	black with dark red tints, darkening on exposure to light
Lustre; opacity:	adamantine, submetallic; translucent, becoming opaque on exposure to light
Streak:	purple-red
Hardness:	2.5-3
Specific gravity:	5.85
Cleavage; fracture:	good, rhombohedral; uneven to conchoidal
Habit:	crystals hemimorphic being prisms with varying terminations; granular aggregates, compact masses, disseminated grains
Crystal system:	trigonal

Stephanite



Named after Archduke Victor Stephan (1817–1867), Mining Director for Austria, stephanite is also called black silver ore or brittle silver ore. It is a major ore of silver, alongside acanthite in importance. Good crystals are rare, but can grow to about 6cm (2½in) and display a brilliant metallic lustre. Twinning often occurs to give pseudohexagonal crystals. Stephanite is found in hydrothermal veins associated with silver, proustite, pyrargyrite, polybasite, tetrahedrite and acanthite. Excellent crystals have been found at St Andreasberg and Freiberg (Germany), and at Zacatecas and Arizpe (Mexico). It is mined at Cobalt in Ontario and Elsa in Yukon (Canada) and is an important ore at the famous Comstock Lode (Nevada, USA).

Ag ₅ SbS ₄	
Colour:	iron-black
Lustre; opacity:	metallic; opaque
Streak:	black, shiny
Hardness:	2-2.5
Specific gravity:	6.3
Cleavage; fracture:	poor; uneven
Habit:	crystals rare, prismatic to tabular; granular aggregates, massive, disseminated
Crystal system:	orthorhombic

Polybasite



 $\label{eq:polybasite} \begin{array}{l} \textbf{P} olybasite is a minor silver ore and yet another that has been called ruby silver. It forms a series with pearceite, <math>(Ag,Cu)_{16}As_2S_{11}$, although the antimony-rich polybasite is much more common. Good crystals are rare, but can grow to about 6cm (2½in) and have a reddish tinge due to internal reflections. Fine crystals are found at Husky Mine (Yukon, Canada) and rosettes of hexagonal plates at Guanajuata (Mexico). Polybasite occurs in hydrothermal veins usually accompanied by quartz, calcite, barite or other silver minerals. Other good localities include St Andreasberg and Freiberg (Germany), and Pribram (Czech Republic). It is mined at Chañarcillo (Chile), Sarrabus (Sardinia) and Colorado (USA).

(19/04/16522511	
Colour:	iron black, rarely cherry red
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	2-3
Specific gravity:	6-6.2
Cleavage; fracture:	poor; uneven
Habit:	bevelled, tabular, pseudohexagonal crystals; bladed, in rosettes, granular compact aggregates, crusts
Crystal system:	monoclinic

(Ag,Cu)₁₆Sb₂S₁₁



Bournonite, also known as endellionite, forms crystals that have a good metallic liustre and interesting shapes. Twinning often produces shapes resembling worn cog wheels, hence the name wheel ore. Such twinned crystals are called 'trillings' and frequently are striated along the 'teeth' of the cog. Some crystals can develop a greyish tarnish, so some care with storage is advised. Bournonite occurs in hydrothermal veins, associated with galena, tetrahedrite, pyrite and especially silver. Excellent crystals are found at Harz (Germany), St Endellion and Lanreath (Cornwall, England), Pribram (Czech Republic), Kapnik-Bnya (Hungary), Sarrabus (Sardinia), Broken Hill (NSW, Australia) and Park City (Utah, USA).

PbCuSbS₃

5	
Colour:	dark grey to black
Lustre; opacity:	metallic; opaque
Streak:	steel grey
Hardness:	2.5–3
Specific gravity:	5.7–5.9
Cleavage; fracture:	good; fragile
Habit:	crystals as stubby, tabular, multi-faceted prisms; granular aggregates, disseminated grains
Crystal system:	orthorhombic

Hutchinsonite



H utchinsonite is an ore of the metal thallium, others being crooksite (Cu₇TiSe₄), thalcusite (Cu₃FeTl₂S₄) and lorandite (TlAsS₂). It was named to honour Arthur Hutchinson (1866–1937), Professor of Mineralogy at Cambridge University. Thallium is obtained as a by-product of the smelting of thallium-rich zinc and lead ores (which are likely to contain hutchinsonite). Thallium is very toxic, but is used in some specialist glass manufacture. Hutchinsonite occurs in hydrothermal deposits associated with orpiment, realgar, pyrite, sphalerite and galena. It is found at the Lengenbach Quarry (Binnental, Switzerland), Segen Gottes (Black Forest, Germany), Quiluvrilca (Peru) and Toya-Takarada (Hokkaido, Japan).

(
Colour:	scarlet-vermilion to deep cherry red, strong red internal reflections
Lustre; opacity:	submetallic; subtranslucent to opaque
Streak:	red
Hardness:	1.5-2
Specific gravity:	4.6 .
Cleavage; fracture:	good; conchoidal, brittle
Habit:	acicular or prismatic crystals; granular masses
Crystal system:	orthorhombic
the second s	

(Tl,Pb)₂As₅S₉

Boulangerite



Bc. Boulangerite is a minor ore of lead named after the French mining engineer Bc. Boulanger (1810–1849). It can form thin, acicular crystals up to 2cm (½in) long, resembling fibres. Boulangerite and jamesonite have been called feather ores or plumosite, after the feathery habit of some varieties. Thin fibres of boulangerite are flexible, unlike the brittle jamesonite. Disseminated boulangerite can easily be overlooked, being mistaken for stray hairs. Boulangerite occurs in hydrothermal veins in lead, zinc and antimony deposits. It is found at Pribram (Czech Republic), Molières (France), Bottino (Italy), Noche Buena Mine (Zacatecas, Mexico), Boliden (Sweden), Claustal (Germany) and Stevens County (Washington, USA).

Pb ₅ Sb ₄ S ₁₁	
Colour:	grey
Lustre; opacity:	dull, metallic; opaque
Streak:	black
Hardness:	2.5–3
Specific gravity:	5.8-6.2
Cleavage; fracture:	good; brittle but flexible as thin needles
Habit:	acicular crystals; fibrous masses or tufts, disseminated
Crystal system:	monoclinic
Other:	melts easily

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Realgar



Realgar, named from the Arabic *rahj al jahr* for 'mine powder', is often found with yellow orpiment (As_2S_3), to which it alters on exposure to light and air. Its colour and transparency contrast with the grey, opaque nature of most other sulphides. Realgar has been used in fireworks (to give a white colour) and paints. Like sulphur, which occurs as rings of eight atoms, it is in the form of alternating sulphur and arsenic atoms As_4S_4 . Realgar mostly occurs in veins with orpiment, cinnabar or stibnite, but may be found in limestones, clays, volcanic rocks and hot springs. Crystals are rare, but are found at Nagyag (Romania), Binnental (Switzerland), Matra (Corsica), Manhattan (Nevada, USA) and King County (Washington, USA).

AsS

Colour:	orange-red to red
Lustre; opacity:	resinous; semi-transparent
Streak:	yellow-orange
Hardness:	1.5-2
Specific gravity:	3.5–3.6
Cleavage; fracture:	perfect; conchoidal
Habit:	rare small, stubby, prismatic crystals; compact aggregates; films
Crystal system:	monoclinic



Orpiment is a rare mineral often found in association with realgar. Its name comes from the Latin *auripigmentum*, meaning 'golden pigment'. On standing, orpiment will slowly turn into a powder, a process accelerated by light. Orpiment was traded as a pigment throughout the Roman Empire and used as a medicine in China. Orpiment is deposited around hot springs and volcanic fumaroles, associated with realgar and, sometimes, cinnabar. It also occurs in metamorphosed dolerites and in hydrothermal veins. The major deposits of orpiment are in Zashuran (Iran), Kurdistan (Turkey), Georgia and Manhattan (Nevada, USA). Large crystals have been found at Shimen (Hunan, China) and Quiruvica (La Libertad, Peru).

As ₂ S ₃	
Colour:	orange-yellow to yellow
Lustre; opacity:	greasy, pearly on fracture surfaces; translucent to transparent
Streak:	yellow
Hardness:	1.52
Specific gravity:	3.48
Cleavage; fracture:	perfect; flaky
Habit:	crystals rare, as small, flat prisms, occasionally fibrous; crusts or bladed masses, earthy, powdery
Crystal system:	monoclinic

Rammelsbergite



Rammelsbergite, named after the German chemist Karl Rammelsberg (1813–1899), is rare and difficult to distinguish from other sulphides, often being mistaken for gersdorfite (NiAsS). It occurs in hydrothermal veins together with nickel and cobalt minerals. Associated minerals include skutterudite, lollingtonite, safflorite, nickeline, bismuth and silver. Rammelsberg is found at Franklin and Sterling Hill (New Jersey, USA), Mohawk Mine (Michigan, USA), Coniston (Cumbria, England), St Marie-aux-Mines (France), Legnica (Poland), Löllington-Hüttenberg (Austria) and the type locality, Schneeberg (Saxony, Germany). Large crystals come from Bou Azzer (Morocco).

NiAs ₂	
Colour:	tin white with a faint pinkish hue, darker tarnish
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	5.5
Specific gravity:	7.1 "
Cleavage; fracture:	indistinct; uneven
Habit:	small crystals as imperfect prisms; granular, massive, radiating aggregates
Crystal system:	orthorhombic

Jordanite



J ordanite is a rare mineral, named after the German mineralogist H. Jordan (1808–1887). It forms a series with antimony-rich geocronite, $Pb_{14}(Sb,As)_6S_{23}$. It occurs in hydrothermal veins and metamorphosed lead-arsenic deposits in dolomite, and may be associated with galena, sphalerite, pyrite and anatase. The most important occurrence is at Lengenbach (Binnental, Switzerland). In the Seravezza marble in Tuscany, Italy, jordanite occurs in cavities, as lustrous, lead-grey hexagonal crystals up to 1cm ($\frac{1}{2}$ in). Other localities include Săcărîmb (Romania), Wiesloch (Germany), Zuni Mine (Colorado, USA) and Sinking Valley (Pennsylvania, USA). It has also been found in 'black smoker' chimneys on ocean floors.

1014(~3,50)6523	
lead-grey, commonly tarnished and iridescent	
metallic; opaque	
black	
3	
6.4	
perfect; conchoidal, brittle	
crystals deeply striated prismatic, and tabular, pseudohexagonal twins; granular, globular and botryoidal aggregates	
monoclinic	

$Pb_{14}(As,Sb)_6S_{23}$

Jamesonite



T amesonite, also known as antimony glance or grey antimony, was named for the J Scottish mineralogist Robert Jameson (1774–1854). It is characterized by hairlike fibres, which may be thick and felted, or may occur as individual 'hairs', and is sometimes also known as feather ore. Jamesonite may be distinguished from the acicular mineral boulangerite because it has brittle rather than flexible crystals and from acicular yellow millerite by colour. It occurs in hydrothermal lead-silver-zinc veins. It may be associated with pyrite, sphalerite, arsenopyrite, siderite, dolomite, calcite, rhodochrosite or quartz. Notable localities are Cornwall (England), Herja Mine (Maramures, Romania) and Zacatecas (Mexico).

Pb ₄ FeSb ₆ S ₁₄	
Colour:	lead grey, iridescent tarnish
Lustre; opacity:	metallic or silky; opaque
Streak:	grey-black
Hardness:	2.5
Specific gravity:	5.63–5.78
Cleavage; fracture:	perfect basal (perpendicular to length); uneven
Habit:	very elongated acicular or fibrous crystals, occasionally prismatic; usually compact or felted masses, may be radial or plumose
Crystal system:	monoclinic

Cosalite



Cosalite is a rare mineral named after the Cosalá Mine (Sinaloa, Mexico). It occurs in hydrothermal deposits in lead sulphide ores, in contact skarns and in pegmatites. Associated minerals include chalcopyrite, pyrite, sphalerite, skutterudite, bismuth, tremolite, diopside, epidote and quartz. It forms hairlike crystals (found at Carrock Fell Mine, in Cumbria, England) or elongated prisms (as at Limestone Quarry in Saxony, Germany). Other good localities are Kara Oba (Kazakhstan), Braichyroen (Snowdon, Wales), Cobalt (Ontario, Canada) and Hecla Mine, Dundas (Tasmania, Australia). Kudriavite, (Cd,Pb)Bi₂S₄, is a closely related mineral found around fumaroles (e.g. Kudriavy Volcano, Kuril Islands, Russia).

Pb2Bi2S5	
Colour:	lead grey or steel grey to silver-white
Lustre; opacity:	metallic; opaque
Streak:	black
Hardness:	2.5-3
Specific gravity:	6.86–6.99
Cleavage; fracture:	very rare; uneven
Habit:	long prisms, vertically striated, hairlike; radial and granular aggregates
Crystal system:	orthorhombic

HALOGENIDES



Halite, or rock salt, typically forms through the evaporation of enclosed bodies of sea water. It may form a thick, stratified deposit, or domes where the lowdensity salt has risen through overlying sediments, as along the Gulf Coast (USA). Halite also forms as a volcanic sublimate or as a cave deposit. It is associated with other evaporites, such as sylvite (KCI), gypsum, anhydrite and dolomite. Rock salt is used as common table salt, and its taste is characteristic. It is an important raw material in the chlor-alkali industry, drilling muds, aluminium purification and many other industries. Well-known deposits include those at Stassfurt (Germany) and in Texas and New Mexico (USA). Good crystals are found at Wieliczka (Galicia, Poland).

NaCl	
Colour:	white when pure, greyish, pinkish, bluish, violet, orange
Lustre; opacity:	vitreous, greasy; transparent to translucent
Streak:	white
Hardness:	2.5
Specific gravity:	2.1–2.2
Cleavage; fracture:	perfect cubic; conchoidal
Habit:	crystals cubic or as octahedra, skeletal 'hopper' crystals; granular, massive, rarely stalactitic or fibrous
Crystal system:	cubic



Villiaumite is a rare mineral named after the French explorer Maxime Villiaume, who collected the mineral from cavities in nepheline syenites at Rouma (Islands of Los, Guinea). It is strongly coloured, with some shades of red being unique in minerals. It is found in nepheline syenites and their pegmatites, in which alkali metals such as sodium, lithium and potassium are concentrated. Associated minerals include nepheline, aegerine, sodalite and zeolites. Villiaumite is soluble in water and displays weak red fluoresecence under shortwave ultraviolet light. Most specimens come from the Kola Peninsula (Russia); good crystals are found at Mont Saint-Hilaire (Quebec, Canada) and Windhoek (Namibia).

carmine red, lavender pink to light orange
vitreous; transparent
white, pinkish
2.5
2.79
perfect in three directions forming cubes; conchoidal
rarely cubic or octahedral crystals; granular aggregates, massive
cubic
2 2 7

HALOGENIDES

Chlorargyrite



Chlorargyrite, a silver chloride mineral, was named after the Latin *argentum* (silver) and the Greek *chloro* (pale green). Also called cerargyrite or horn silver, it forms a complete solid-solution with bromargyrite (AgBr). It is pale when fresh, but darkens to brownish-purple on exposure to light. Chlorargyrite dissolves in ammonia, but not in nitric acid. It forms as a secondary mineral in the weathered and enriched zones of silver deposits, especially in arid regions, and can form very rich albeit small silver ore deposits. The type locality is Saxony (Germany); it occurs elsewhere at Lake Valley District (New Mexico, USA), Atonopah (Nevada, USA), Broken Hill (NSW, Australia), Freiburg (Saxony, Germany) and Atacama (Chile).

AgCl	
Colour:	colourless when fresh, pale grey, yellowish; can darken
Lustre; opacity:	resinous, waxy, adamantine; transparent to translucent
Streak:	white, shining
Hardness:	1.5–2.5
Specific gravity:	5.55 (6.5 in bromargyrite)
Cleavage; fracture:	absent; uneven to subconchoidal, sectile, ductile
Habit:	crystals rare, as cubes, or modified by octahedra; usually massive or in crusts, columnar
Crystal system:	cubic

lodargyrite



I odargyrite, or iodyrite, was named after its chemical composition (from the Greek *iodes* meaning 'violet') and the Greek *argyros* for 'silver'. A very rare secondary mineral, it forms in the oxidized parts of silver deposits. The type locality is Albarradón Mine (Zacatecas, Mexico), but the best place for specimens is Broken Hill (NSW, Australia), the world's largest silver-lead-zinc deposit. Iodargyrite is also important for other halides, such as the very rare iodides marshite (copper), miersite (silver and copper), perroudite (a mercury silver halide), and bright yellow crystals of bromargyrite (a silver halide). Other locations include Atacama (Chile), Dzhezkazgan (Kazakhstan) and Nevada (USA).

Agl	
Colour:	colourless, becomes yellowish on exposure to light
Lustre; opacity:	greasy to adamantine, pearly on cleavage surfaces; transparent
Streak:	white or yellow, shiny
Hardness:	1-1.5
Specific gravity:	5.7
Cleavage; fracture:	perfect basal; conchoidal, sectile and flexible
Habit:	crystals prismatic, platy, barrel-shaped; scales, powdered
Crystal system:	hexagonal

HALOGENIDES

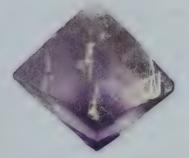
Fluorite



Fluorite, or fluorspar, is a popular mineral, which can fluoresce as a result of impurities such as yttrium. Some specimens phosphoresce; others thermoluminesce (glow when heated, as at Franklin in New Jersey, USA, and Mont Saint-Hilarie in Quebec, Canada) or triboluminesce (glow when crushed or struck). Fluorite forms in hydrothermal veins as a gangue mineral associated with sulphides (e.g. galena or sphalerite), barite and quartz, and as an accessory in granite and granite pegmatites. It is an important industrial mineral, used as a flux in iron smelting, as a source of fluorine, as special optical lenses, and as a rare gemstone. The main mining areas are Canada, USA, Russia, Mexico and Italy.

CaF ₂	
Colour:	colourless, white, yellow, green, purple and blue
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	4
Specific gravity:	3.18
Cleavage; fracture:	perfect octahedral; subconchoidal to uneven
Habit:	crystals common, as cubes, octahedra or rarely dodecahedra, nodular, granular aggregates, earthy masses
Crystal system:	cubic
the second se	

Fluorite: Gem Varieties



Fluorite shows a wide range of colours. The most famous variety is Blue John, a purple and yellow banded variety from Castleton (Derbyshire, England), which has been used for ornaments and jewellery. The Ancient Greeks and the Ancient Egyptians, and the Chinese have used it for decorative purposes for over 300 years. Rather soft for general use as a gemstone and too well cleaved to be easily cut, it can nonetheless be brightly polished and cut into cabochons, protected by rock crystal (quartz). A deposit of colourful highly silicated fluorite discovered quite recently in Utah has been given a number of fanciful names, such as bertandite and Picasso stone. More properly called opalized fluorite, it makes attractive cabochon gems.

CaF ₂	
Colour:	colourless, white, yellow, green, purple and blue
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	4
Specific gravity:	3.18
Cleavage; fracture:	perfect octahedral; subconchoidal to uneven
Habit:	crystals common, as cubes, octahedra or rarely dodecahedra; nodular, granular aggregates, earthy masses
Crystal system:	cubic

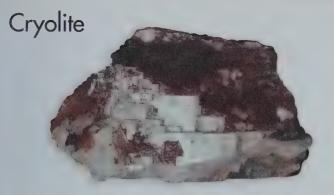
HALOGENIDES

Sal-ammoniac



Sal-ammoniac forms in volcanic regions around fumaroles, with solid white crystals forming directly from the bluish ammonium chloride vapour as a sublimate (i.e. there is no liquid phase). It has a pungent, cool and saline taste. Salammoniac forms on Vesuvius (Campania), Etna (Sicily), and other southern Italian volcanoes, Mont Pelée (Martinique), Parícutin (Mexico) and Kilauea (Hawaii). Associated minerals include sulphur, realgar and orpiment. Crystals must be collected quickly, as they will dissolve in the first rain shower. Sal-ammoniac formation is also associated with burning coal seams, as at Duttweller (Saarland, Germany), and, unusually, with guano on Cicna and Guanape Islands (Peru).

NH ₄ Cl	
Colour:	colourless, white, yellow, reddish or brown
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	1-2
Specific gravity:	1.53
Cleavage; fracture:	poor; conchoidal to earthy
Habit:	crystals as cubes, octahedra or dodecahedra, also skeletal or dendritic; generally efflorescent or encrusting
Crystal system:	cubic



Cryolite is an uncommon mineral of very restricted distribution. It possesses the unusual property that, if submerged in water, it becomes almost invisible, since it has a refractive index of 1.33, close to that of water. It is soluble in sulphuric acid, producing fumes of hydrofluoric acid. Cryolite occurs only in pegmatites, probably formed as a precipitate from fluoride-rich solutions. The most notable occurrence is in a granitic pegmatite at lvigtut (Greenland), where it is associated with topaz, siderite, galena, microcline, fluorite and other unusual fluorides. It is also found in a topaz mine at Miask (Urals, Russia). Artificial cryolite has now replaced the natural mineral as a flux in the extraction of aluminium from bauxite.

Na ₃ AlF ₆	
Colour:	usually colourless to white, or grey, reddish, brownish
Lustre; opacity:	vitreous to greasy, pearly; transparent
Streak:	white
Hardness:	2.5-3
Specific gravity:	2.95
Cleavage; fracture:	absent, although has three partings; uneven, brittle
Habit:	pseudo-cubic crystals, commonly twinned; granular aggregates often arranged in a parquet-like pattern
Crystal system:	monoclinic

HALOGENIDES

Carnallite



Carnallite is a rare mineral that sometimes fluoresces, and will colour a flame Violet due to its potassium content. It has a bitter, salty taste, will readily dissolve in water (emitting a creaking sound), and must be kept in sealed containers. Carnallite is one of the last minerals to precipitate in an evaporating basin, in particular conditions that do not exist today. It occurs with other potassium and magnesium evaporites such as sylvite, kainite and kieserite. It is a valuable source of potash (for fertilizers) and magnesium. The most famous locality is the Stassfurt potash deposit (Germany); it also occurs in Carlsbad (New Mexico, USA), Paradox Basin (Colorado and Utah, USA) and the Perm Basin (Russia).

KMgCl₃.6H₂O

Colour:	colourless, milky white, yellow, pink, brown, rarely blue
Lustre; opacity:	vitreous to greasy; transparent to translucent
Streak:	white
Hardness:	2.5
Specific gravity:	1.6 (very light)
Cleavage; fracture:	absent; conchoidal
Habit:	pseudo-hexagonal pyramidal crystals, rare; granular and fibrous aggregates forming layers
Crystal system:	orthorhombic

Atacamite

A tacamite is named from the Atacama Desert in Chile, which is one of the most arid regions in the world. Dark green itself, atacamite is often associated with other coloured minerals such as green malachite, red cuprite, blue-green chrysocolla, and also gypsum. It generally forms through the oxidation of copper minerals in arid, salty conditions, and is found in many places in Chile. It occurs as a fumarolic deposit on Vesuvius (Italy) and Etna (Sicily, Italy), and through the weathering of sulphides formed as black smoker deposits around volcanic vents on the ocean floor. It also forms through the alteration of bronze and copper antiquities, and is found in the slag heaps at Laurion (Attica, Greece).

Cu₂Cl(OH)₃

vitreous to adamantine; transparent to translucent
apple green
3–3.5
3.76
good; conchoidal, brittle
crystals acicular, striated, or tabular; massive, as fibrous aggregates or granular
orthorhombic

HALOGENIDES

Cotunnite



Cotunnite forms as a volcanic sublimate, its type locality being Monte Somma-Vesuvius (Italy). It was named in honour of Domenico Cotugno (Cottunius) (1736–1822), an Italian physician and professor of anatomy from Naples (Italy). It was formed during the 1975–76 fissure eruption of Tolbachik volcano (Kamchatka Peninsula, Russia), where it is associated with halite, silver, gold, tenorite, and rare minerals such as burnsite and ponomarevite. It also forms by the alteration of galena in saline environments, and as an alteration product of lead-bearing slag or other archaeological material after immersion in sea water. Hence, like diaboleite, cotunnite occurs in the ancient slag heaps of Laurion (Greece).

PbCl ₂	
Colour:	colourless, pale green, pale yellow, white
Lustre; opacity:	adamantine, silky, pearly; transparent to translucent
Streak:	white
Hardness:	2.5
Specific gravity:	5.8 ~
Cleavage; fracture:	perfect; conchoidal to uneven
Habit:	prismatic or acicular crystals, or skeletal; in aggregates of radiating sprays, massive, granular, crusts, pseudomorphs
Crystal system:	orthorhombic

Boleite



Boleite is an unusual indigo blue colour, and makes a most attractive collector's Bitem, although crystals are rarely cut as gems. It was named after its type locality, Boleó (Baja California, Mexico). Although tetragonal, it is always twinned, and as a result appears as cubes, with corners sometimes cut by octahedral faces. A secondary mineral, it forms from the alteration of sulphide deposits by chlorinebearing aqueous solutions. It may be associated with atacamite, anglesite, cerussite or gypsum. Noted localities include Broken Hill (NSW, Australia), Mammoth District (Arizona, USA) and the Mendip Hills (Somerset, England). It also forms in Laurion (Greece), where smelter slag has been immersed in the sea.

Colour:	light indigo blue, azure-blue, dark blue
Lustre; opacity:	weakly vitreous to pearly on cleavage surfaces; translucent to transparent
Streak:	light green, light blue
Hardness:	3.5
Specific gravity:	4.8–5.1
Cleavage; fracture:	perfect; uneven
Habit:	normally as pseudo-cubic twinned crystals
Crystal system:	tetragonal

Pb9Cu8Ag3Cl21(OH)16.H2O

HALOGENIDES

Diaboleite



Diaboleite comes from the Greek *dia* ('difference') and boleite (the mineral); it should not be confused with pseudoboleite, $(Pb_5Cu_4Cl_{10}(OH)_8.2H_2O)$, meaning false boleite. Diaboleite is a very rare secondary mineral, its type locality being the Mendip Hills (Somerset, England). The other main locality is in the copper porphyry deposits of Mammoth Mine (Arizona, USA), associated mainly with cerussite, wulfenite, quartz and hemimorphite, and also with boleite, linarite and pseudoboleite. It occurs in the 2000-year-old mineral slags of Laurion (Greece), many of which are now in the sea. This has led to the production of many new minerals, including laurionite, the first slag mineral to be described (in 1887).

- SZCaciz(OII)	·4
Colour:	dark blue
Lustre; opacity:	adamantine; transparent to translucent
Streak:	blue
Hardness:	2.5
Specific gravity:	5.48 (extremely heavy for a translucent mineral)
Cleavage; fracture:	perfect; conchoidal
Habit:	very small tabular crystals; also as platy aggregates, grains, or as encrustations
Crystal system:	tetragonal

Pb2CuCl2(OH)4

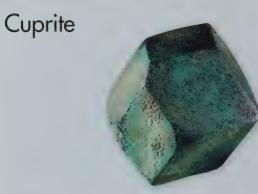
Zincite



Although zinc oxide itself is colourless, zincite is almost invariably coloured, due to manganese and iron impurities. The more attractive deep red crystals have earned it the name ruby zinc (at Franklin, where it is mined); the red colour is due to the presence of manganese dioxide or hematite. It was one of the first minerals described in an American mineralogy journal. Zincite is found as a primary mineral in metamorphosed ore deposits and as a secondary mineral in weathered zinc ore deposits. It is associated with franklinite, willemite and calcite at Sterling Hill and Franklin (New Jersey, USA): other locations are Tsumeb (Namibia), Kapushi, (Katanga, Congo) and in the ash from Mt St Helens (Washington, USA).

ZnO	
Colour:	yellow, orange, red, brown, rarely green or colourless
Lustre; opacity:	subadamantine; transparent, more often translucent
Streak:	yellow–orange
Hardness:	4
Specific gravity:	5.68
Cleavage; fracture:	perfect; irregular
Habit:	hemimorphic crystals, hexagonal prisms terminated differently at either end; granular, massive and foliated in veins
Crystal system:	hexagonal

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Cuprite is known as red copper oxide, which distinguishes it from CuO, or black copper oxide. Cuprite oxidizes slowly in air, tarnishing the surface. It is an important copper ore, mainly being found in dull massive bodies. Fine, fibrous crystals are called chalcotricite; ruby copper is a gem-quality crystal found at Santa Rita (New Mexico, USA) and Onganyo (Namibia). It occurs as a secondary mineral in the oxidized zones of copper ore deposits, accompanied typically by malachite, azurite, chalcocite and native copper. Cuprite often occurs with oxides of iron as an earthy red-brown material known as tile ore. It is found at Tsumeb (Namibia), Cornwall (England), Chessy (France) and Bisbee (Arizona, USA).

red to dark red
submetallic, adamantine, earthy; translucent
brownish-red
3.5-4
5.8–6.1 **
poor; irregular, conchoidal
crystals usually octahedral, sometimes dodecahedral or cubes rarely needles; mostly massive, granular or earthy
cubic

Perovskite



Perovskite is mined not for titanium, but for rare earths in it such as niobium and cerium. Named after the Russian mineralogist L. Perovski (1792–1856), perovskite has a structure well known among crystallographers adopted by superconducting ceramics and high-pressure minerals in the Earth's mantle. It occurs in silica-poor igneous rocks and metamophic rocks. Associations include rare earth minerals such as loparite ((Na,Ce)TiO₃), as well as pyrochlore, ilmenite, leucite and titanite. Good crystals have been obtained from Alno (Sweden), Magnet Cove (Arkansas, USA) and the Urals (Russia); other locations include Val Malenco and Vesuvius (Italy), Bagagem (Brazil), Kaiserstuhl (Germany) and Quebec (Canada).

CaTiO₃

Colour:	red-brown, grey-black, yellow
Lustre; opacity:	adamantine, submetallic; translucent
Streak:	pale yellow
Hardness:	5.5
Specific gravity:	4.0
Cleavage; fracture:	none; subconchoidal to indistinct
Habit:	crystals pseudocubic, often striated parallel to edges; granular aggregates and reniform masses
Crystal system:	orthorhombic

Tenorite



Tenorite, or black copper oxide, is a rare mineral named after the Italian botanist Michel Tenore (1781–1861). Earthy specimens containing tenorite are called melaconite. It is formed in the oxidized zones of copper deposits, associated with azurite, malachite, chalcocite, cuprite and limonite, and occasionally found in volcanic sublimates. The bladed crystals are collectable specimens, and botryoidal tenorite provides a contrasting dull, grey setting as a matrix for bright blue-green chrysocolla. Tenorite occurs at Val d'Ossola (Italy), Cornwall (England), Leadhills (Scotland), Rio Tinto (Spain), the Urals and Kamchatka (Russia), Bisbee (Arizona, USA), Chuquicamata (Chile) and Tsumeb (Namibia).

steel-grey to black
metallic, dull; opaque
black, greenish
3-4
5.8–6.4
indistinct; conchoidal to irregular
crystals usually elongated plates, often striated and serrated edges; scaly and earthy aggregates; encrustations
monoclinic

Gahnite



G ahnite, known as automolite or zinc-spinel, is named after its discoverer, J. Gahn (1745–1818), the Swedish chemist who discovered manganese. It occurs in granite pegmatites and metamorphic rocks and as a placer deposit. It is associated with galena, sphalerite and magnetite. Gahnite is usually found as small crystals in the type localities Falun (Sweden), Silberberg (Bavaria, Germany) and Tiriolo (Calabria, Italy), but larger crystals have been obtained from mines in Franklin and Sterling Hill (New Jersey, USA). Other deposits are in Connecticut (USA), Minas Gerais (Brazil), Smilovne (Bulgaria) and Victoria Range (New Zealand). A lead-bearing variety called limaite is found at Ponto de Lima (Portugal).

ZnAl ₂ O ₄	
Colour:	green to bluish green
Lustre; opacity:	vitreous, greasy; translucent, opaque on edges
Streak:	grey
Hardness:	7.5–8
Specific gravity:	4.6
Cleavage; fracture:	indistinct; conchoidal
Habit:	crystals; granular aggregates, grains
Crystal system:	cubic, but twinning is ubiquitous and crystals may appear trigonal

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Spinel



The term 'spinel' is also used as a general term for minerals of formula AB₂O₄ such as gahnite, hercynite (FeAl₂O₄) and galaxite (MnAl₂O₄). Spinel has been mistaken for ruby and sapphire; the so-called Timur Ruby in the British Crown jewels is a ruby spinel. Spinel is found in contact-metamorphosed rocks, in Igneous rocks and as a placer deposit. Synthetic spinels have been produced as artificial gemstones since 1910. The word spinel comes from the Latin *spina*, for 'little thorn', after the sharpness of its crystals. The best gem-quality spinels come from gravels in Sri Lanka, Myanmar and Madagascar. Good crystals are found at Vesuvius and Lazio (Italy), Orange County (New York, USA) and Sterling Hill (New Jersey, USA).

MgAl ₂ O ₄	
Colour:	colourless pure; red, blue, green, black
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white to grey or brown
Hardness:	7.5–8
Specific gravity:	3.5–4.1
Cleavage; fracture:	none, octahedral parting; conchoidals
Habit:	octahedral crystals; granular aggregates; grains
Crystal system:	cubic

Magnetite



Magnetite, or magnetic oxide of iron, is the essential constituent of lodestone, known since ancient times. It is a major iron ore and is mined in vast quantities for the iron and steel industries; slag is further worked to recover vanadium and phosphorus. Swedish magnetite, containing some silicates, is used to make a very hard silicon steel. Magnetite is widespread among many igneous and metamorphic rocks, especially mafic and ultramafic rocks in high temperature mineral veins. It is also found in river and marine sediments and in dune deposits. Good crystals come from Val Malenco and Val de Vizze (Italy), Binnental (Swizerland) and Pfitschtal (Austria). Lodestone is famously found at Magnet Cove (Arkansas, USA).

Fe ₃ O ₄	
Colour:	black
Lustre; opacity:	lustre, submetallic, dull; opaque
Streak:	black
Hardness:	5.5-6.5
Specific gravity:	5.2
Cleavage; fracture:	none, octahedral parting; conchoidal
Habit:	crystals octahedral, sometimes dodecahedral; granular or massive
Crystal system:	cubic

Chromite



Chromite is chemically related to magnetite, chromium atoms replacing those of iron; it is also magnetic, but weaker than magnetite. It is the only commercial ore of chromium, which is used in steels, especially stainless steel. Chromium salts are used in electroplating, leather tanning, fireproofing fabrics and in paints. Chromite is used as a refractory for lining furnaces in the ceramic industry. The range in specific gravity is a reflection of its contamination with magnesiochromite ($MgCr_2O_4$), spinel ($MgAl_2O_4$) and related minerals. Chromite occurs in ultrabasic rocks and serpentinites, and as a placer deposit. Large deposits are mined in South Africa, Russia, Albania, Turkey, Zimbabwe and the Philippines.

FeCr ₂ O ₄	
Colour:	brownish black to black
Lustre; opacity:	metallic to submetallic; opaque
Streak:	dark brown
Hardness:	5.5
Specific gravity:	4.1–5.1
Cleavage; fracture:	none; uneven, conchoidal
Habit:	crystals octahedral, rare; usually massive, granular
Crystal system:	cubic
Other:	weakly magnetic; insoluble in acids

Franklinite



Franklinite is a spinel group mineral named after its type locality of Franklin. White calcite, green willemite and red zincite often accompany franklinite, and specimens displaying combinations of these are very collectable. Franklinite occurs in beds and veins in high temperature metamorphosed dolomites, associated with calcite, zincite, willemite, rhodonite, magnetite and garnet. It is also found in some manganese and iron deposits. The most famous and most mined location is that at Franklin and Sterling Hill (New Jersey, USA), which has produced crystals up to 30cm (12in) large. Other locations are Långban (Sweden), Hranicna (Czech Republic), Atasui (Khazakstan), Sayan (Siberia, Russia) and Western Australia.

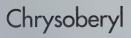
ZnFe ₂ O ₄	
Colour:	black
Lustre; opacity:	metallic, with reddish internal reflections; opaque
Streak:	red-brown
Hardness:	5.5-6.5
Specific gravity:	5.1–5.2
Cleavage; fracture:	imperfect; conchoidal
Habit:	octahedral, sometimes dodecahedral crystals; massive aggregates
Crystal system:	cubic

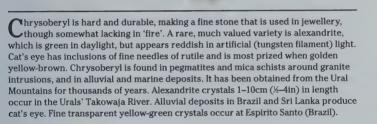
Minium



Minium, red oxide of lead or simply red lead, has been used as a pigment, especially in anti-rusting paints, although lead's toxicity has lowered demand. Its unusual name comes from the river Minius in northwest Spain. It is formed by the alteration of galena and cerrussite, under extreme oxidizing conditions, and its presence may function as an indicator of the degree of oxidation. Some specimens from Broken Hill (NSW, Australia) are particularly good, but are said to have been the result of a mine fire. Its powdery form can disguise its high density. Minium is found at Badenweiller and Horhausen (Germany), Monteponi (Sardinia, Italy), Bolaños and Zimapan (Mexico), Leadhills (Scotland) and Broken Hill (Australia).

light red to red-brown
dull to greasy; opaque
orange-yellow
2-3
8.2
perfect; earthy
rare, scaly crystals; usually powdered or massive aggregates
tetragonal





BeAl ₂ O ₄	
Colour:	colourless, grey, brown, green, yellow
Lustre; opacity:	subadamantine, silky; transparent to translucent
Streak:	white
Hardness:	8.5
Specific gravity:	3.7
Cleavage; fracture:	good; conchoidal
Habit:	prismatic, tabular crystals, often pseudohexagonal twins; grains and pebbles in alluvial deposits
Crystal system:	orthorhombic

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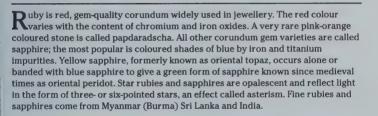
Corundum



Extremely hard (equal to silicon carbide and second only to diamond), corundum lis a gem mineral (ruby and sapphire) and is used as an abrasive in the form of emery for cutting, grinding and drilling. Grey-coloured masses are often forms that have been altered by hydrothermal solutions to margarite or zoisite. Corundum occurs in igneous rocks, in contact metamorphosed shales and bauxite, in pegmatites and metamorphosed limestones and as a placer deposit. Emery deposits are granular masses, often with magnetite, hematite and spinel. Good localities are Mogok (Myanmar), Sri Lanka, Madagascar, Glebe Hill (South Africa), Ardnamurchan (Scotland), Naxos (Greece) and the central Urals (Russia).

Colour:	colourless, grey or brown granular mass; can be red, blue yellow, green, purple or colourless gems
Lustre; opacity:	adamantine; transparent to semi-opaque
Streak:	white
Hardness:	9
Specific gravity:	4
Cleavage; fracture:	none; uneven to conchoidal
Habit:	rough prisms or barrel shaped crystals bounded by steep pyramids; massive, granular
Crystal system:	trigonal

Corundum: Gem Varieties



Al ₂ O ₃	
Colour:	colourless, grey or brown granular mass; can be red, blue, yellow, green, purple or colourless gems
Lustre; opacity:	adamantine; transparent to semi-opaque
Streak:	white
Hardness:	9
Specific gravity:	4
Cleavage; fracture:	none; uneven to conchoidal
Habit:	rough prisms or barrel shaped crystals bounded by steep pyramids; massive, granular
Crystal system:	trigonal

Arsenolite



A rsenolite or white arsenic, produced as a by-product of sulphide smelting, is widely used for preparing arsenic compounds, but quite rare in nature. Usually dull in appearance, crystal specimens – e.g. from White Caps (Nevada, USA) and St Etienne (France) – can be quite beautiful under magnification. A dimorphic form of arsenolite is the monoclinic clauditite. Arsenolite is toxic and should be handled with care. Arsenolite is formed by alteration of arsenic minerals, including by mine fires. Associated minerals are realgar, orpiment and erythrite. It is found at Annaberg (Germany), Jáchymov (Czech Republic), Cornwall (England), Laurion (Greece), St Marie-aux-Mines (France) and Smolnik (Slovakia).

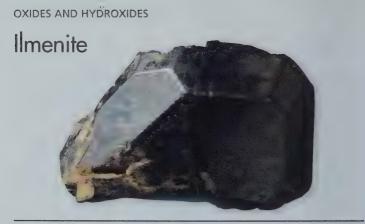
As ₂ O ₃	
Colour:	white, pale blue, pale pink or yellow if contaminated with realgar or orpimiment
Lustre; opacity:	vitreous
Streak:	white
Hardness:	1.5
Specific gravity:	3.7
Cleavage; fracture:	perfect; conchoidal
Habit:	rare octahedral crystals; encrustations, earthy
Crystal system:	cubic
and the second s	

Senarmontite



Senarmontite is dimorphic with the orthorhombic valentinite – also called antimony bloom, with which it is commonly associated. Senarmontite is named after the French mineralogist H. de Sénarmont (1808–1862), whereas valentinite is more intriguingly named after the sixteenth-century alchemist B. Valentinus. Commercially obtained from other antimony-bearing minerals, senarmontite is used in paints, plastics, medicines and especially as a flame-retardant for PVC in aircraft and motor vehicles. It occurs in oxidized antimony-bearing hydrothermal deposits. Large crystals are found at Hamimate Mine (Sensa, Algeria); other locations are Pernek (Slovakia), Mopung Hills (Nevada, USA) and Cornwall (England).

Sb ₂ O ₃	
Colour:	colourless to grey
Lustre; opacity:	resinous; translucent to transparent
Streak:	white
Hardness:	2
Specific gravity:	5.3
Cleavage; fracture:	conchoidal; irregular
Habit:	well-formed octahedral crystals; granular, encrusting, massive
Crystal system:	cubic



Imenite is a major ore of titanium. Pure ilmenite is black with a metallic lustre, but its properties can be altered by incorporated giekielite (MgTiO₃) and pyrophanite (MnTiO₃). Resistant to weathering, it often appears in sands such as at Menaccan Sands (Cornwall, England), where menaccanite was an early reported form of the mineral. The streak is useful to distinguish it from hematite and, unlike magnetite, it is non-magnetic. It is found in igneous rocks, where it settles in magma intrusions with other dense minerals such as hematite and magnetite. Good crystals are found at Kragerö (Norway), Val Devero (Italy), St Gotthard (Switzerland) and Orange County (New York, USA), and in sands at Travancore (India).

FeTiO ₃	
Colour:	iron black
Lustre; opacity:	metallic to submetallic; opaque
Streak:	black to brownish black
Hardness:	5-6
Specific gravity:	4.5-5
Cleavage; fracture:	none, basal partings; conchoidal
Habit:	Crystals flat tabular rhombohedra; massive, compact; grains disseminated in igneous rocks
Crystal system:	trigonal



Hematite, or red oxide of iron, is named after the Greek for 'blood'. On oxidation of iron-rich fluids, it precipitates, to give rocks a rusty appearance. Red ochre is an earthy clay-rich mixture containing hematite. Rounded opaque bodies, such as kidney ore, have been carved into figures that take a high polish. Rare crystalline forms are petal-shaped aggregates called 'iron roses' and shiny specular hematite. Powdered, it is an abrasive for polishing as jewellers' rouge. It occurs in oxidized igneous rocks and veins, and as a sedimentary cement. Huge quantities are mined at Lake Superior (USA), Quebec (Canada), Brazil and Australia. Beautiful crystals occur at Rio Marino (Elba, Italy), Bahia (Brazil) and Cumbria (England).

Fe ₂ O ₃	
Colour:	steel–grey to iron–black, can be iridescent; dull to bright red when massive or earthy
Lustre; opacity:	metallic, dull, earthy; opaque
Streak:	dark red to red-brown
Hardness:	5-6
Specific gravity :	4.9–5.3
Cleavage; fracture:	none; conchoidal
Habit:	crystals tabular or rhombohedral, sometimes with striated or curved faces; massive, laminated or earthy
Crystal system:	trigonal

Quartz



Quartz is the most abundant mineral on the Earth's surface. It forms beautiful orystals as hexagonal prisms, weighing up to 130kg (287lb). Quartz is used in glass-making, ceramics, refractories and abrasives. It produces electricity when strained (piezoelectric) or heated (pyroelectric), used in electrical sensors. Quartz is an essential mineral of many acid igneous and metamorphic rocks, and occurs in most clastic sediments. Some sandstones and their metamorphic equivalent, both called quartzites, are almost entirely quartz. It often occurs veins and fissures; these provide the best crystals. Good specimens occur in the Alps in Switzerland and Austria, at Carrara (Italy), Bourg d'Oisans (France) and the Urals (Russia).

SiO ₂	
Colour:	colourless or white; tinted many shades
Lustre; opacity:	vitreous; transparent to opaque
Streak:	white
Hardness:	7
Specific gravity:	2.65
Cleavage; fracture:	absent; conchoidal, splintery
Habit:	usually six-sided prismatic crystals terminated by six faces, the prisms faces often horizontally striated; massive, compact, drusy
Crystal system:	trigonal

Quartz: Gem Varieties



Clear, colourless crystals are called rock crystal. The opacity of milky quartz is caused by small, gas or liquid bubbles. Colours are produced by iron hydrates in yellow citrine, ferric oxide in violet amethyst, and titanium or manganese in pink rose quartz. Brown or smoky quartz, found as huge crystals in Brazil, can be produced artificially by irradiating rock crystal. Quartz cat's eye, tiger's eye and hawk's eye have inclusions giving a wavy, striped effect. Rutilated quartz contains rutile needles intersecting at 60' angles. Minute reflective scales in aventurine quartz give a green or brown spangled appearance. Many varieties are used in jewellery and specimens of amethyst are widely collected.

colourless or white; tinted many shades
vitreous; transparent to opaque
white
7
2.65
absent; conchoidal, splintery
usually six-sided prismatic crystals terminated by six faces, the prisms faces often horizontally striated; massive, compact, drusy
trigonal

Cristobalite

Cristobalite is a much rarer form of silicon dioxide than quartz and rarer than tridymite. It has two forms: tetragonal cristobalite, which is stable up to 270°C (518°F), and cubic cristobalite, which is stable above 1470°C (2678°F). Between these temperatures, both forms can exist. Beautiful specimens found at Cerro San Cristobal (Portugal) comprise octahedral cristobalite embedded in transparent but dark, glassy obsidian. Cristobalite occurs in intermediate igneous rocks and as recrystallized opals. Associated minerals are opal, chalcedony and tridymite. Other locations are Monte Dore (France), Glass and Sugarloaf mountains (California, USA), Crater Lake (Oregon, USA), Eiffel (Germany), Sarospatak (Hungary) and Tokatoka (New Zealand).

white to yellowish vitreous; translucent to transparent white
white
Winte
6.5
2.2
absent; conchoidal
rare crystals; microcrystalline as small balls, fibres (called lussatite), crusts
tetragonal

Tridymite



Tridymite is stable between 870°C (1598°F) and 1470°C (2678°F) and is a hightemperature polymorph of quartz. Although in a metastable state at normal temperatures, it is widespread and possibly underestimated in abundance due to the difficulty in identification. Tridymite will change into quartz, but the process is very slow. Tridymite is prepared synthetically for use in heat-resistant porcelain and as refractory material in furnaces. Tridymite is found as a sublimate in acid volcanic rocks and in contact metamorphosed sandstones. Tridymite is found in the Eiffel district (Germany), Cerro San Cristobal (Portugal), Pomona (California, USA), Mule Springs (Oregon, USA) and Kamomoto (Japan). It has been found in story meteorites.

SiO ₂	
Colour:	colourless, white
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	white
Hardness:	6.5-7
Specific gravity:	2.27
Cleavage; fracture:	absent; conchoidal
Habit:	small pseudo-hexagonal blades; spherical aggregates
Crystal system:	monoclinic

Chalcedony is made of microscopic quartz crystals and is mostly banded, which imparts much of its appeal as an ornamental stone. Flints and cherts are impure forms, the former being well known as nodules in chalk. It is named from the Ancient Greek town of Kalchedon in Asia Minor. Chalcedony is usually produced by precipitation from aqueous solutions, often associated with hot springs or volcanoes. It can be produced by dehydration of opal. Chalcedony is found as the variety agate in Rio Grande de Sul (Brazil) and Idar-Oberstein (Germany), as carnellan in Brazil, Uruguay and California (USA), as chrysoprase in the Urals (Russia), California and Queensland (Australia), and flint in Southern England.

SiO2	
Colour:	variable, white through shades of brown, red and grey to black
Lustre; opacity:	waxy to vitreous; translucent
Streak:	white
Hardness:	6.5
Specific gravity :	2.6
Cleavage; fracture:	absent; uneven, splintery, conchoidal
Habit:	no crystals; botryoidal or stalactitic, often banded; in fissures and veins, massive or nodular
Crystal system:	none, microcrystalline

Chalcedony: Gem Varieties



Chalcedony, like quartz, has a number of attractive forms coloured by various impurities, but shows different patterns of banding. Agate has curved, often semi-concentric bands. Moss agate is a pale translucent stone with dendritic moss- or tree-like inclusions of iron oxides. The translucent carnelian is orange-red and jasper is a more massive, sometimes striped opaque stone of varying colour, but often red. Chrysoprase is a fine translucent, apple-green; heliotrope, or bloodstone, is dark green and opaque with red spots; and plasma has yellow spots. Onyx has white and black or brown bands, with straight bands rather than curved. Sard is a variety of onyx, and sardonyx has white bands like onyx and red-brown bands like sard.

SiO ₂	
Colour:	variable, white through shades of brown, red and grey to black
Lustre; opacity:	waxy to vitreous; translucent
Streak:	white
Hardness:	6.5
Specific gravity:	2.6
Cleavage; fracture:	absent; uneven, splintery, conchoidal
Habit:	no crystals; botryoidal or stalactitic, often banded; in fissures and veins, massive or nodular
Crystal system:	none, microcrystalline

Opal



Opal, named from the Sanskrit for 'precious stone', is amorphous hydrated silica. It is made of tiny spheroids, giving internal reflections of light and a play of colours, or 'fire'. Precious opal is beautifully iridescent. Red fire opal and black opal are the most sought-after varieties; the latter has been more expensive than diamond. Opal is produced by the low-temperature weathering and alteration of silica-rich rocks, especially around geysers and hot springs. The skeletons of diatoms, sponges and radiolaria are composed of opal. Opals have been obtained from White Cliffs (NSW, Australia) since the nineteenth century. Mines in the Czech Republic have been worked since Roman times. An artificial form is called Slocum stone.

white when pure; can be yellow, red, blue, brown or others waxy or resinous; transparent to opaque
Waxy or resinguis: transparent to opague
way of resilious, transparent to opaque
white
5.5-6.5
1.8–2.3
none; conchoidal, uneven
massive, often nodular, stalactitic, cavity-filling, concretions or crusts
none, amorphous

Pyrolusite



Pyrolusite is an important ore of manganese, a metal used widely in steel-making. It is also used in glass-making for removing coloration due to impurities of iron oxides, hence the name from the Greek words for 'fire' and 'wash'. Pyrolusite occurs with manganite and psilomelane, in a dark earthy ore called wad. Found on the surface of rocks in joints, it forms dendritic markings, often picturesque or fanciful. Pyrolusite is formed by chemical precipitation in lakes or oxidation of manganese ores. Large nodules are found at the bottom of seas and oceans. Mined localities include Platten (Bavaria, Germany), Epleny (Hungary), Cornwall (England), Nikepol (Ukraine), Deccan (India), Minas Gerais (Brazil), Cuba and South Africa.

MnO₂ Colour: grey or grey-black metallic; opaque Lustre: opacity: Streak: black to bluish-black 6-6.5 Hardness: **Specific gravity:** 4.7-5.1 Cleavage; fracture: poor; uneven rare prismatic crystals; usually concretionary aggregates and Habit: earthy masses **Crystal system:** tetragonal



Cassiterite, named after the Greek for 'tin', is its major ore. Tin is widely used for metal plating in food canning and for alloys, famously in bronze (one of the earliest metals known to the ancients). Twinned crystals known as knee twins are popularly collected. Radiating, fibrous concretions are known as wood tin. Cassiterite is found in hydrothermal veins and pegmatites associated with minerals such as fluorite, scheelite, topaz and tourmaline. Placer deposits are called stream tin. Rarely occurring in workable quantities in the USA, it is found at Erzgebirge, Altenberg and Zinnwald (Germany), Cornwall (England), Brittany (France), Bolivia, Sumatra, Yunnan (China) and Elba and Lake Como (Italy).

SnO ₂		
Colour:	brown to black	
Lustre; opacity:	adamantine; opaque	
Streak:	white to yellow	
Hardness:	7	
Specific gravity:	6.8-7.1	
Cleavage; fracture:	imperfect; conchoidal	
Habit:	crystals as prisms, bipyramids and needles	
Crystal system:	tetragonal	

Rutile

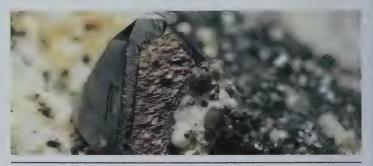


Rutile is a form of titanium dioxide often embedded in other minerals as needlelike inclusions of metallic appearance. Such inclusions have been named maiden hair, cat's eye and star forms. Quartz crystals with embedded rutile needles intersecting at angles of 60' are called sagenite or rutilated quartz. Rutile is durable and cuts well, having a high 'fire', but its colour is usually unattractive. Twinning is common, giving elbow-shaped crystals. Pure powdered titanium dioxide is brilliant white and used as a pigment in white paints. Rutile is found in igneous intrusions, pegmatites and metamorphic rocks and placer deposits. Good crystals are found at Binnental (Switzerland) and in Grove Mountain (Georgia, USA).

TiO₂

-	
Colour:	yellow, red, brown, black
Lustre; opacity:	metallic; translucent to opaque
Streak:	yellow browns
Hardness:	6-6.5
Specific gravity :	4.3
Cleavage; fracture:	perfect; uneven, conchoidal
Habit:	crystals usually elongate prisms, often striated; grains, fibrous aggregates and inclusions
Crystal system:	tetragonal

Anatase



A natase is named from the Greek for 'elongated', after the sharp crystals it often forms. It is a polymorph of titanium dioxide and is formed, like brookite, at lower temperatures than rutile, to which it can be converted by heating. Anatase mostly forms as transparent to opaque 1–3mm (up to ½in) isolated crystals embedded in a matrix of accessory minerals. It is easily distinguished from rutile by crystal form and specific gravity. It is found as a secondary mineral in titanium-bearing rocks, in igneous rocks, in metamorphic rocks and as a placer deposit. Associated minerals are brookite, albite, quartz and titanite. Excellent specimens are found at Binnental and Tavetschal (Switzerland) and Val Devero and Vatellina (Italy).

yellow to brown, deep blue, black
adamantine; translucent to opaque
white to light brown
5.5-6
3.8–3.9 ~
perfect; conchoidal
usually sharp, bipyramidal crystals, sometimes tabular
tetrahedral

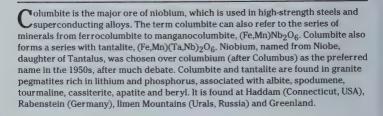
Brookite



Named after the English mineralogist H. Bruke (1771–1857), brookite is the third polymorph of titanium dioxide. Like anatase, it forms at lower temperatures than rutile. The larger crystals, especially heart-shaped twins, are popular. Brookite is found in veins and fissures in altered igneous and metamorphic rocks, such as granites and gniesses. Associated minerals include anatase, albite, quartz and rutile. It also occurs as a placer deposit. Fine crystals are found at Bourg d'Oisans (Dauphiné, France), Grisons and Uri cantons (Switzerland) and Tyrol and Untersulzbachtal (Austria). The site at Prenteg (Wales) produced some of the finest specimens in the nineteenth century.

TiO ₂	
Colour:	brown to black
Lustre; opacity:	adamantine, submetallic; transparent to translucent
Streak:	yellow, yellow-brown
Hardness:	5.5–6
Specific gravity:	3.9–4.2
Cleavage; fracture:	imperfect; subconchoidal
Habit:	tabular or lamellar crystals, sometimes striated
Crystal system:	orthorhombic

Columbite



FeNb ₂ O ₆	
Colour:	brown or black
Lustre; opacity:	metallic; nearly opaque, transparent in thin lamellae
Streak:	black to dark red
Hardness:	6
Specific gravity:	5.1
Cleavage; fracture:	good; subconchoidal
Habit:	crystals as complex, stubby prisms, aggregates of thin, tabular crystals; granular, massive
Crystal system:	orthorhombic

Furgusonite



Furgusonite, named after the Scottish politician and landowner Robert Furguson (1767–1840), is an ore of yttrium found widespread but in small amounts. Specimens are usually radioactive and often metamict. Yttrium is used to give red colours on TV screens, in X-ray filters, in superconductors and in alloys. Invariably substituted by a number of the other rare-earth elements, the term furgusonite-Y is used for the yttrium-rich mineral. Fergusonite occurs in rare earth-bearing granite pegmatites and as a placer deposit. The type locality where it was first discovered is Qeqertaussaq Island (Greenland). Major mining areas include Arendal (Norway), Blum mine (Ilmen Mountains, Russia) and Ytterby (Sweden).

YNbO₄

Colour:	black, brown, grey, yellow
Lustre; opacity:	submetallic; translucent to opaque
Streak:	brown
Hardness:	5.5–6
Specific gravity:	4.3–5.8
Cleavage; fracture:	indistinct; subconchoidal
Habit:	crystals as prismatic to acicular dipyramids, usually powdery; granular, massive
Crystal system:	tetragonal

Brucite



Brucite, an ore of magnesium, is used to produce magnesia (MgO), magnesium compounds and refractories. Brucite can be quite rich in manganese, weathering to dark brown. Brucite occurs in marbles by alteration of periclase, in hydrothermal veins in metamorphic limestones and chlorite schists, and in serpentinized dunites. Associated minerals include calcite, aragonite, dolomite, magnesite, talc and chrysotile. It is found at Castle Point (Hoboken, New Jersey, USA), Gabbs (Nevada, USA), Asbestos (Quebec, Canada), Vesuvius and Sardinia (Italy), Shetland Islands and Isle of Muck (Scotland), and the Ethyl Mine (Mutorashanga, Zimbabwe). It was first described by Archibald Bruce (1777–1818), an American mineralogist.

Mg(OH) ₂	
Colour:	white, shades of grey, blue and green
Lustre; opacity:	waxy, pearly on cleavage faces; transparent
Streak:	white
Hardness:	2–2.5
Specific gravity:	2.4
Cleavage; fracture:	perfect; uneven, separates into flexible plates
Habit:	tabular crystals in platy or foliate masses, rosettes, sometimes fibrous; granular, massive
Crystal system:	hexagonal

Goethite



G oethite is a major constituent of limonite and ochres, and an important ore mineral. Named after the famous German poet J. Goethe (1749–1832), it is also known as iron hydroxide, needle ironstone or acicular iron ore. Black crystals are attractive but rare. Sprays of acicular crystals are found at Pfibram. Goethite often forms an attractive matrix in specimens of other minerals such as vanadite. Its perfect cleavage gives it a soft and greasy feel. Goethite is formed by oxidation of iron-rich minerals, usually pyrite, siderite and magnetite. Large amounts are mined at Pribram (Czech Republic) and in Cuba, Alsace-Lorraine (France), Westphalia (Germany), Lake Superior (USA) and Labrador (Canada).

FeO(OH)

Colour:	brown to black
Lustre; opacity:	adamantine, submetallic, silky; translucent to opaque
Streak:	yellow-brown
Hardness:	5-5.5
Specific gravity:	4.3
Cleavage; fracture:	perfect; uneven to hackly
Habit:	rare prisms with vertical striations; usually stalactitic, massive or earthy aggregates of radiating fibrous forms
Crystal system:	orthorhombic

Diaspore



Diaspore is most widely found in bauxite, the major ore of aluminium. Other minerals constituting bauxite are closely related, especially boehmite (AlO(OH)) and gibbsite (Al(OH)₃). Bauxite is formed by extreme weathering of aluminosilicate rocks in tropical regions. Purer forms of diaspore are rare relative to the abundance of bauxite and good crystals rarer still. Other important associations are with corundum and margarite among emery deposits and with chlorite and chloritoid in metamorphic rocks. The name from the Greek 'to disperse' alludes to its easy disintegration in a flame. Crystals are found in Chester (Massachusetts, USA), Naxos (Greece), Campolongo (Switzerland) and Mramorskoi (Urals, Russia).

AIO(OH)

Colour:	white; greenish, grey or pink as aggregates
Lustre; Opacity:	vitreous or pearly; transparent to translucent
Streak:	white
Hardness:	6.5-7
Specific Gravity :	3.3-3.5
Cleavage; Fracture	perfect; conchoidal
Habit:	crystals tabular or acicular; foliated or stalactitic aggregates
Crystal System:	orthorhombic

Limonite



Named from the Latin *limus* for 'mud', limonite has been variously called brown ironstone, brown iron and brown haematite. It is a term used to describe a rock made of an ill-defined mixture of mostly amorphous hydrated iron oxides, microcrystalline goethite (FeOOH) and lepidocrocite (FeOOH). Not strictly a mineral, but not quite seen as a rock, limonite is usually included among minerals. It is used as a pigment as yellow ochre and in modelling clay. It is quite hard, but very fragile, easily disintegrating into grains and powders. It is formed by surface oxidation of iron deposits or is left after the dissolution of iron-rich rocks in tropical regions. Limonite is often found as a cubic polymorph after pyrite.

Colour:	yellow-brown to black
Lustre; opacity:	subvitreous, dull, earthy; opaque
Streak:	yellow brown
Hardness:	5-5.5
Specific gravity:	ca. 4
Cleavage; fracture:	absent; conchoidal, splintery
Habit:	botryoidal, stalactitic, oolitic earthy or porous masses; loose yellow-brown crusts to dark, iridescent bodies when compact
Crystal system:	none, amorphous

hydrated iron oxides



Manganite, or brown ore of manganese, is a minor ore of manganese. It usually forms rather dull bodies similar to other manganese minerals, but when crystalline it is easier to identify and is quite collectable. Such specimens include lustrous feathery crystals (up to 8mm/Xin), found in the Kalahari manganese field. Pyrolusite, which can resemble manganite, is softer and has a bluish streak. It is formed by the alteration of other manganese minerals in low-temperature hydrothermal veins, associated with calcite and barite, and in hot springs, associated with psilomelane and pyrolusite. Crystals up to 7–8cm (2%–3in) large are found in Harz (Germany), Como (Italy) and Negaunee and Marquette (Michigan, USA).

MnO(OH)	
Colour:	black, opaque
Lustre; opacity:	metallic; opaque, red, translucent in thin plates
Streak:	dark brown
Hardness:	4
Specific gravity:	4.3-4.4
Cleavage; fracture:	perfect; uneven
Habit:	elongate prisms with deep striations lengthwise, usually in bundles; microcrystalline, granular, radially fibrous, oolitic
Crystal system:	monoclinic

Pyrochlore



 $\label{eq:product} \begin{array}{l} \textbf{P}_{\text{(Na,Ca)}_2\text{Nb}_2\text{O}_6(\text{OH},\text{F}), \text{ to microlite, } (Na,Ca)_2\text{Ta}_2\text{O}_6(\text{OH},\text{F}). Pyrochlore turns green on heating and is named from the Greek for 'fire' and 'green'; microlite is named from the Greek for 'small stone'. This compositionally diverse mineral is a source of a number of rare earth elements and uranium, and is usually radioactive. It is found in pegmatites and carbonatites, associated with apatite, nepheline, zircon, biotite and forsterite. It is rare and found at Mbeya (Tanzania), Fen (Norway), Kaiserstuhl (Germany), Oka (Canada), Newry (Maine), Haddam (Connecticut, USA), Betanima (Madagascar), Minas Gerais (Brazil) and Varutrask (Sweden). \end{array}$

(Na, Ca)2(ND, Ta, T)206(01, T, O)	
Colour:	usually brown; greenish, reddish
Lustre; Opacity:	greasy; translucent to opaque
Streak:	yellowish brown
Hardness:	5-5.5
Specific Gravity :	4.3-4.5
Cleavage; Fracture:	distinct, octahedral; conchoidal, uneven
Habit:	crystals usually octahedral, disseminated; granular
Crystal System:	cubic

(Na,Ca)₂(Nb,Ta,Ti)₂O₆(OH,F,O)

Thorianite



Thorianite forms a series with uraninite (UO_2) and contains other elements substituting for thorium; a ThO₂ content of about 70 per cent would be typical. It is an ore of thorium, which is used in nuclear fuel elements, refractory materials and incandescent gas mantles. Like all thorium and uranium minerals, thorianite contains radiogenic elements such as lead, which have been formed by radioactive decay. It is found in pegmatites, carbonatites and serpentinites, and as a placer deposit. Thorianite is found at Taolañaro (Madagascar), Transvaal (South Africa), Kola Peninsula (Siberia, Russia) and Easton (Pennsylvania, USA). Thorium is named after the old Scandinavian god of war, Thor.

ThO ₂		
Colour:	brownish black, greenish, yetlow	
Lustre; opacity:	metallic; opaque	
Streak:	greenish-grey	
Hardness:	6.5-7	
Specific gravity:	9.7–9.8	
Cleavage; fracture:	imperfect; subconchoidal, brittle	
Habit:	cubic crystals, often as interpenetrant twins; granular	
Crystal system:	cubic	

Uraninite



Ores containing uraninite, known as pitchblende, are major ores not only of uranium, but also of radium. Uranium is a major source of energy via nuclear fission. Uraninite is radioactive and found in pegmatites, in sandstones and conglomerates, and in hydrothermal veins in sulphide ores. Pitchblende from Yáchymov (Czech Republic) was used by the Curies in the discovery of polonium and radium in 1898. Radium and uranium are also sources of radioactivity for various uses in industry, science and medicine. Large crystals are found at Wilberforce (Canada). Mining areas include Great Bear Lake (Canada), Shinkolobwe (Zaire), the Colorado Plateau (USA) and Witwatersrand (South Africa).

UO ₂	
Colour:	brown, grey, black
Lustre; opacity:	greasy, submetallic; opaque
Streak:	black
Hardness:	5.5-6
Specific gravity:	4.3-4.5
Cleavage; fracture:	difficult; conchoidal
Habit:	rare modified cubes; usually dense botryoidal, reniform, massive, colloform; sometimes dendritic
Crystal system:	cubic

Billietite



Billietite is a radioactive mineral found as rare but attractive amber yellow crystals. It is an uncommon alteration product of uraninite, named after the Belgian crystallographer Valère Billiet (1903–1945). Billietite is similar to becquerelite (Ca(UO₂)₆O₄(OH)₆.6-8H₂O), and related to compreignacite (K₂(UO₂)₆O₄(OH)₆.6-8H₂O). Associated minerals are barite, calcite, chalcopyrite, hematite and other rare species such as uranophane, fourmarierite, metabernite, rutherfordine and becquerelite. Billietite is found in Shaba Province (Congo), the La Crouzille, Margnac and Rabéjac mines (France), Menzenschwand (Black Forest, Germany) and the Delta mine (Utah, USA).

Colour:	yellow-brown
Lustre; opacity:	adamantine; transparent to translucent
Streak:	yellow
Hardness:	not determined
Specific gravity:	5.3
Cleavage; fracture:	perfect; brittle
Habit:	crystals pseudohexagonal, tabular
Crystal system:	orthorhombic

Ba(UO₂)₆O₄(OH)₆.6-8H₂O

Plattnerite



Plattnerite is an uncommon dense mineral composed of lead oxide. It mostly occurs as masses, but specimens comprising drusy crusts showing many shiny crystals are fairly common. Specimens of plattnerite with contrasting minerals such as calcite and wulfenite are popular. Plattnerite is named after the German metallurgist K. Plattner (1800–1858). It occurs in oxidized, weathered hydrothermal lead deposits, typically in arid climates, associated with cerussite, smithsonite, hemimorphite, calcite and quartz. It is found in Leadhills and Wanlockhead (Scotland), Anarak and Anjireh (Iran), Tsumeb (Namibia), Idaho, Arizona, Nevada and New Mexico (USA), and Durango and Chihuahua (Mexico).

PbO₂

Colour:	black
Lustre; opacity:	adamantine to submetallic; opaque
Streak:	dark brown
Hardness:	5.5
Specific gravity:	9.6
Cleavage; fracture:	good; conchoidal to uneven
Habit:	small, prismatic crystals, sometimes acicular; nodular or botryoidal, fibrous or concentrically zoned, massive
Crystal system:	tetragonal

Betafite

Betafite is named after the type locality of Betafo in Madagascar. The hardness, specific gravity and other properties vary widely as composition changes. It is an ore of uranium and consequently radioactive; also used to obtain niobium, tantalum, other rare earth elements and associated thorium. Although black in colour, the surface is invariably altered (metamict), making it appear greenish or yellowish. It is found in pegmatites and adjacent metamorphosed limestones. Crystals up to 100kg (2201b) have been reported. Associated minerals are quartz, feldspars, biotite, zircon and rare earth minerals. Betafite is found at Tangen (Norway), Sludianka (Baikal, Russia), Bancroft (Ontario, Canada) and Val d'Ossola (Italy).

Colour:	black, with a tint of yellow, brown or green
Lustre; opacity:	waxy, greasy to adamantine; translucent to opaque
Streak:	yellowish white
Hardness:	3-5.5
Specific gravity:	3.7-4.9
Cleavage; fracture:	none; conchoidal to uneven
Habit:	octahedral and dodecahedral crystals, sometimes elongated, granular masses common; crusts
Crystal system:	cubic

(Ca,Na,U)₂(Ti,Nb,Ta)₂O₆(OH)

Psilomelane



Psilomelane is a term given to a collection of poorly defined hydrated bariumbearing manganese ores, the main component probably being romanechite (BaMn₅O₁₀.H₂O). The term wad is used for earthy mixtures of psilomelane, pyrolusite and others. Psilomelane is found in weathed ores and as a replacement deposit in limestones and dolomites. It also forms desert varnish, a dark coating of rocks in dry regions. Romanechite was authenticated at Romanèche (France) and occurs as crystals at Schneeberg (Saxony, Germany) and Oberwolfach (Black Forest, Germany). Psilomelane occurs at Cornwall (England), Virginia, Nevada and New Mexico (USA), Pilbara (Western Australia) and Chihuahua (Mexico).

No fixed formula

Colour:	iron black to steel grey	
Lustre; opacity:	dull; opaque	
Streak:	brownish black	
Hardness:	5–5.5 (6 for romanechite)	
Specific gravity:	4.4-4.5	
Cleavage; fracture:	absent; conchoidal to uneven	
Habit:	rare euhedral crystals (romanechite); massive, fibrous, botryoidal, stalactitic, concretionary, earthy, powdery	
Crystal system:	monoclinic (romanechite)	

Nitratine



The nitrates as a whole readily dissolve in water, so nitratine (or soda nitre) should be kept in an airtight container with a desiccant. It has a bitter, cooling taste, and colours a flame yellow. It can be distinguished from nitre (salt-petre, KNO3) by the flame test, as nitre turns a flame violet. Nitratine is used in the chemical industry, for fertilizers and in fireworks. It occurs as a bedded deposit in playa lakes in arid regions, associated with halite, gypsum and other evaporites. It may also precipitate from nitrate-bearing ground water. Economically valuable deposits of billions of tons occur in regions adjacent to the type locality of Tarapacá (Chile). It also occurs in arid regions of the Persian Gulf.

NaNO ₃	
Colour:	colourless, white, sometimes greyish, yellowish brown
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	1.5-2
Specific gravity:	2.2-2.3
Cleavage; fracture:	perfect, rhombohedral; conchoidal, sectile
Habit:	crystals rare; granular or massive encrustations, stalactitic
Crystal system:	trigonal

Calcite



Calcite is a major rock-forming mineral, the principal component of limestone and marble, and an important constituent of igneous carbonatite. It effervesces strongly in cold, dilute hydrochloric acid. Crystals show a wide variety of forms, such as nail-head spar (flat-topped rhombohedra) or dog-tooth spar (sharply pointed). Fine sand-covered crystals come from Fontainebleau (France); calcite with hematite inclusions from Chihuahua (Mexico); and fluorescent calcite from Franklin (New Jersey, USA). An image appears double when viewed through transparent rhombs of Iceland spar, a property known as 'double refraction'. This was used to produce nicol prisms for petrological microscopes.

CaCO₃

Colour:	usually colourless or white; may be brown, red, green or black
Lustre; opacity:	vitreous, dull, pearly; transparent to translucent
Streak:	white to greyish
Hardness:	3
Specific gravity:	2.71
Cleavage; fracture:	perfect, rhombohedral; subconchoidal
Habit:	wide variety of crystal forms; stalactitic, granular, massive, fibrous or many other habits
Crystal system:	trigonal

Magnesite



Magnesite, or bitter spar, is much less common than calcite, and will only Mdissolve with effervescence in hydrochloric acid on warming. It forms when limestone is altered by magnesian solutions, often accompanied by the formation of dolomite, and during the hydrothermal metamorphism of ultramafic rocks to talc schists and serpentinite. An important ore of magnesium, it is heated to produce magnesium oxide (MgO), used in the manufacture of cements and refractory bricks. Powdered magnesia is used in the rubber, paper and pharmaceutical industries. Large deposits are found in Styria (Austria), Manchuria (China), Silesia (Poland), Madras (India) and Euboea (Greece).

MgCO ₃	
Colour:	white or colourless, also greyish or yellowish brown
Lustre; opacity:	vitreous or dull when compact; transparent to translucent
Streak:	white
Hardness:	3.5-4.5
Specific gravity:	3.0–3.2
Cleavage; fracture:	perfect rhombohedral; conchoidal fracture
Habit:	crystals rare and usually rhombohedral; massive, lamellar, granular or fibrous
Crystal system:	trigonal

Siderite



Siderite is a common carbonate, forming series with magnesite, rhodochrosite and smithsonite. It dissolves slowly with effervescence in cold hydrochloric acid, and becomes magnetic on heating. It usually forms in sedimentary deposits, such as in coalfields, as nodules and beds of impure iron carbonate called clay ironstone, and as oolitic ironstone. It is also found in metamorphic iron formations (as at Biwabik, Minnesota, USA). Magnesite-siderite series carbonates occur in carbonatites at Nkombwa (Zambia) and Newania (Rajasthan, India). It also forms in hydrothermal veins associated with barite, fluorite and galena. Large crystals occur at Mont Saint-Hilaire (Quebec, Canada) and Mosojillacta (Colavi, Bolivia).

FeCO₂ Colour: shades of brown, tan, greenish grey vitreous, silky, pearly on cleavage; translucent Lustre; opacity: Streak: white Hardness: 3.5-4.5 Specific gravity: 3.96 when pure Cleavage; fracture: perfect rhombohedral; uneven to conchoidal Habit¹ crystals rhombohedral, often with curved faces made up of overlapping scales; massive, granular, concretionary, botryoidal Crystal system: trigonal

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Smithsonite



Smithsonite is very attractive and popular, and can fluoresce pale green or blue in Jultraviolet light. A secondary mineral, it forms in the weathered zone of zinc deposits and may replace adjacent carbonate rocks. Associated minerals include malachite, azurite, aurichalcite, anglesite and hemimorphite. Pale pink twinned crystals are found in Ojuela Mine (Durango, Mexico); pale green botryoidal smithsonite at Broken Hill (NSW, Australia); and blue-green botryoidal specimens from the Kelly Mine (New Mexico, USA). Cream or yellow globular turkey-fat ore is found in Rush Mine (Arkansas, USA). Other good specimens come from Laurion (Greece), Tsumeb (Namibia) and Chessy (France).

ZnCO ₃		
Colour:	white, grey, yellow, bro	wn, shades of green, pink, purple
Lustre; opacity:	vitreous or pearly; trans	slucent
Streak:	white	
Hardness:	4-4.5	
Specific gravity:	4.4-4.5	
Cleavage; fracture:	perfect rhombohedral;	subconchoidal to uneven
Habit:	crystals rare, rhombohedral, with curved faces; massive, botryoidal, reniform, stalactitic, granular or encrustations	
Crystal system:	trigonal	

Rhodochrosite



Rhodochrosite is a beautiful rose colour, but darkens on exposure through dilute hydrochloric acid, and can luminesce light pink in long-wave ultraviolet light. Its prime commercial source is the USA. Crystals can be cut as gems, and the banded rock is very attractive when polished; the type known as Ina Rose comes from the oldest mines in Argentina. Rhodochrosite occurs as a metasomatic deposit, as a primary mineral in hydrothermal veins, associated with sulphides and manganesebearing minerals, and more rarely in pegmatites. It is found in Freiburg (Germany), Las Cabesses (France), Pasto Bueno (Peru) and Colorado (USA).

MnCO ₃	
Colour:	pale to deep rose pink, yellowish grey, brownish or orange
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	white
Hardness:	3.5-4
Specific gravity:	3.5–3.7
Cleavage; fracture:	perfect rhombohedral; uneven
Habit:	crystals rhombohedral, prismatic, scalenohedral or tabular, often curved faces; massive, granular, stalactitic, globular or botryoidal
Crystal system:	trigonal

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Dolomite



olomite is a major rock-forming mineral. It typically forms rhombohedral crystals, the faces of which may be curved, and can form a saddle-shape. Unlike calcite, it bubbles only weakly in dilute hydrochloric acid. Dolomite is the major constituent of the sedimentary rock called dolomite, most of which has formed from the passage of magnesium-bearing solutions through limestone. It also occurs in marbles, associated with talc, tremolite, diopside or wollastonite, in hydrothermal veins with metallic ores, and in igneous carbonatites. Dolomite has many uses, e.g. in the manufacture of cement, and as a source of magnesium oxide. Very good crystals are found at Eugui (Navarra Province, Spain) and at Touissite (Morocco).

$CalVig(CO_3)_2$	
Colour:	colourless, white to cream, pale pink, grey or brown
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	white
Hardness:	3.5–4
Specific gravity:	2.85
Cleavage; fracture:	perfect rhombohedral; subconchoidal
Habit:	crystals rhombohedral, often composed of overlapping scales; massive or granular
Crystal system:	trigonal

Ankerite



A nkerite (brown-spar) is a common mineral that forms a series with dolomite, from which it can be distinguished by colour, which darkens on heating. It effervesces in hydrochloric acid and may fluoresce in long-wave ultraviolet light. Ankerite forms in sedimentary rocks through hydrothermal alteration and low temperature metasomatism; as a gangue mineral in sulphide veins; and in low-grade metamorphic ironstones and sedimentary banded iron formations. It can be associated with dolomite, siderite, quartz, copper sulphides and occasionally gold. The type locality is Styria (Austria). Sharply pointed 'dog-tooth' crystals occur in Chihuahua (Mexico) and crystals up to 5cm (2in) long at Mlynky (Slovakia).

CaFe(CO₃)₂

Colour:	white, grey, yellow, yellow-brown; weathers dark brown
Lustre; opacity:	vitreous to pearly; translucent
Streak:	white
Hardness:	3.5-4
Specific gravity:	2.9–3.1
Cleavage; fracture:	perfect rhombohedral; subconchoidal
Habit:	crystals rhombohedral; massive or granular
Crystal system:	trigonal

Witherite

Witherite forms a series with strontianite (SrCO₃), from which it is distinguished by a flame test; witherite colours the flame green rather than crimson. It dissolves with effervescence in dilute hydrochloric acid; it sometimes fluoresces light blue, and may phosphoresce. It forms in hydrothermal veins, often by the alteration of barite. Associated minerals include barite, fluorite and galena. It also forms as an anoxic sediment, the barium being sourced through hot springs. An ore of barium, it is used in speciality glasses, in the production of rat poison, and was formerly used for refining sugar. Good crystals are found at Alston Moor (Cumbria) and at Hexham (Northumberland, England).

BaCO ₃	
Colour:	colourless, white, grey, yellow, green or brown
Lustre; opacity:	vitreous or resinous; transparent to translucent
Streak:	white
Hardness:	3–3.5
Specific gravity:	4.29
Cleavage; fracture:	one distinct; uneven
Habit:	twinned crystals of pseudo-hexagonal form, commonly striated; massive, botryoidal, granular, fibrous, columnar
Crystal system:	orthorhombic

Aragonite



A ragonite is less widespread than its polymorph calcite. It also dissolves with effervescence in dilute hydrochloric acid, but lacks the excellent rhombohedral cleavage of calcite. Aragonite crystals are often twinned, giving a hexagonal appearance. It is the main constituent of shells of many recent and fossil organisms. Although precipitating from warm marine waters, it inverts to calcite over time. It occurs as sinter from hot springs, or as dripstone in caves, and is the constituent of pearls in oysters. It occurs in amygdales in basalt, and is the stable polymorph in high-pressure metamorphic rocks. Aragonite is an ornamental stone: onyx is a popular banded variety, and flos-ferri is a coral-like form.

CaCO ₃	
Colour:	colourless to white, or yellowish
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	3.5-4
Specific gravity:	2.95
Cleavage; fracture:	distinct; subconchoidal
Habit:	crystals prismatic, elongated, often in radiating groups; coral- like form known as flos-ferri; columnar, stalactitic or encrusting
Crystal system:	orthorhombic

Strontianite

Strontianite was named after Strontian in Scotland, and is the mineral from which Sir Humphrey Davy separated strontium in 1807. It is common in carbonatites (e.g. Kola Peninsula; Russia); forms in hydrothermal veins in limestone, marl and chalk; occurs rarely in sulphide-rich veins associated with galena, sphalerite and chalcopyrite; and can form concretions in limestones and marls, occasionally replacing celestite. Fine crystals occur at Strontian (Scotland) and Oberdorf (Styria, Austria). It used to be mined from veins in limestones at Drensteinfurt (Westphalia, Germany). It is a principal ore of strontium, used for special glass for televisions and VDUs; strontium salts are used for the red colour in fireworks.

SrCO ₃	
Colour:	white, colourless, yellow, greenish or brownish
Lustre; opacity:	vitreous to resinous; transparent to translucent
Streak:	white
Hardness:	3.5
Specific gravity:	3.78 .
Cleavage; fracture:	perfect in one direction; subconchoidal to uneven
Habit:	crystals prismatic, pseudohexagonal twins, or radiating aggregates; massive, fibrous, granular or concretionary
Crystal system:	orthorhombic

Cerussite



Cerussite is a popular mineral due to its high sparkle and its yellow fluorescence. It often forms chevron twins, star-shaped pseudo-hexagonal twins, or complex twins of snowflake appearance. It forms in the oxidation zones of lead veins, associated with galena, sphalerite, pyromorphite, smithsonite, anglesite or goethite. Cerussite is toxic, but was used as a cosmetic by Queen Elizabeth I. It was formerly mined at the Llanfurnach silver-lead mine in Wales and large deposits were worked in southern Kazakhstan. It is mined today at Anguran Mine (Zanjan, Iran). Beautiful 'snowflake' twins occur at Mt Isa Mine (Queensland, Australia) and at Tsumeb (Namibia), where they are known as Jack Straw.

PbCO ₃	
Colour:	colourless, white or greyish
Lustre; opacity:	adamantine, vitreous, resinous; transparent to translucent
Streak:	white
Hardness:	3–3.5
Specific gravity:	6.56 (high for a transparent mineral)
Cleavage; fracture:	two directions, distinct; conchoidal or uneven, very brittle
Habit:	crystals common, often tabular but may be acicular; massive, granular, compact or stalactitic
Crystal system:	orthorhombic

Malachite



Malachite is a highly regarded semi-precious mineral, mined for about 6000 years. Its light and dark green banding make it extremely popular for carvings and jewellery, and it was used both in the Winter Palace in Leningrad and the Taj Mahal, India. Malachite is a common secondary mineral, formed by the weathering of copper deposits. Associated minerals are often very colourful: striking blue azurite and chrysocolla, black mottramite and red limonite. Good crystals are rare, but occur at Chessy (France) and at Onganja mine (Namibia). It has been mined in the Urals (Russia), individual blocks weighing many kilograms, but the most famous ore deposit is in Katanga Province (Democratic Republic of Congo).

Cu₂CO₃(OH)₂

Colour:	bright green and commonly banded
Lustre; opacity:	vitreous to silky; translucent to opaque
Streak:	pale green
Hardness:	3.5-4
Specific gravity:	3.9–4
Cleavage; fracture:	perfect; subconchoidal to uneven
Habit:	acicular crystals or tabular pseudomorphs after azurite, twinning common; botryoidal with concretionary structure
Crystal system:	monoclinic



A zurite is a striking azure blue mineral, popular with collectors, but rarely faceted. It was used as a pigment during the Middle Ages and Renaissance (blue verditer). It forms as a secondary mineral in the oxidized zone of copper deposits associated with carbonate rocks, occurring with malachite, chrysocolla, cerussite or smithsonite. It forms better crystals than malachite, which sometimes pseudomorphs it through hydration of the azurite. There are more than a hundred different crystal forms; good crystals are found at Chessy (France), hence the synonym chessylite, Sardinia (Italy), Broken Hill (NSW, Australia), Tsumeb (Namibia), Laurion (Greece) and Oujda (Morocco).

Cu₃(CO₃)₂(OH)₂

intense azure blue, paler when earthy, darker in crystals
vitreous, adamantine or dull; transparent to opaque
pale blue
3.5–4
3.78
perfect, prismatic; conchoidal
crystals complex, often tabular or short prismatic, twinning common; massive, nodular or earthy
monoclinic

Hydrozincite



H ydrozincite, or zinc bloom, forms in the oxidation zones of zinc deposits through the alteration of sphalerite, zincite and other zinc minerals. It usually occurs as masses or crusts with internal fibrous structures, frequently with smithsonite. It readily dissolves in hydrochloric acid and often fluoresces strongly. When visited at night and with ultraviolet lights, the Trotter mine dump at Franklin (New Jersey, USA) makes a most unusual sight, with willemite fluorescing bright green, calcite bright red, and hydrozincite blue-white. Hydrozincite has been mined as a zinc ore, such as in the Picos de Europa near Santander (Spain), Nevada (USA) and currently at Mae Sod (Thailand) and Skorpion Mine (Namibia).

Zn₅(CO₃)₂(OH)₂

Colour:	usually white to grey, may be yellowish or brownish
Lustre; opacity:	crystals pearly, silky or dull; transparent to translucent
Streak:	white, dull to shining
Hardness:	2-2.5
Specific gravity:	4
Cleavage; fracture:	perfect; uneven
Habit:	crystals small and very rare, sharply pointed; usually massive, earthy, botryoidal, encrusting or stalactitic
Crystal system:	monoclinic

Rosasite



D osasite is a very attractive rare bluish green mineral discovered in 1908 at Rosas Mine (Sardinia, Italy). A rare secondary mineral, it occurs in the oxidation zone of zinc-copper deposits, typically formed by the action of zinc-bearing solutions on primary copper minerals, but it can also be of post-mine origin. It may be associated with colourful minerals, such as limonite, hydrozincite, malachite, aurichalcite, smithsonite, cerussite or hemimorphite. Single crystals can be found at the Summit Mine (Montana); elsewhere in the USA it is found in California and New Mexico. Rosasite occurs in Kisil Espe (Turkestan), Ojuela Mine (Durango, Mexico), and Tsumeb (Namibia). It is a minor ore of copper.

(Cu,Zn) ₂ CO ₃ (OH) ₂	
Colour:	green to bluish green or ský blue
Lustre; opacity:	silky, vitreous or dull; translucent to opaque
Streak:	paler than colour
Hardness:	4.5
Specific gravity:	4.0-4.2
Cleavage; fracture:	in two directions at right angles; fibrous, splintery
Habit:	commonly radiating tufts of fibrous crystals; as crusts, or botyroidal masses or nodules
Crystal system:	monoclinic

Aurichalcite



urichalcite is a popular mineral characterized by delicate acicular crystals, and great care must be taken when handling. It was named either after the Greek word oreichalchos, which means 'mountain copper', or possibly after aurichalcum which means 'yellow copper ore'. Aurichalcite effervesces in cold hydrochloric and nitric acids, and in ammonia. It forms in the oxidized zones of zinc-copper deposits, and is occasionally used as an ore. It can be associated with red limonite and other colourful minerals such as azurite, smithsonite and malachite. Localities include Rosas Mine (Sardinia, Italy), Chessy (Rhône, France), Laurion (Attica, Greece), Tsumeb (Namibia) and notably Ojuela Mine (Mapimi, Durango, Mexico).

$(2n, Cu)_5(CO_3)$	2(OH) ₆
Colour:	pale green to greenish blue and sky blue
Lustre; opacity:	silky or pearly; transparent to translucent
Streak:	pale green or blue
Hardness:	1–2
Specific gravity:	3.6–3.9 (above average for non-metallic minerals)
Cleavage; fracture:	perfect; uneven or fibrous, fragile
Habit:	delicate acicular crystals, often striated, radiating tufts; as feathery encrustations or columnar, laminated, granular
Crystal system:	orthorhombic

100 1

Phosgenite



Phosgenite is a rare lead mineral of high lustre, which fluoresces yellow. It dissolves with effervescence in dilute nitric acid. It forms in the oxidation zone of hydrothermal lead ore deposits, especially in the presence of sea water. It has also been found in slag heaps from the ancient lead-silver mines at Laurion (Attica, Greece), which are now in the sea. Phosgenite was used as a white pigment in ancient Egyptian cosmetics, but it was probably synthesized from smelted lead oxides, as there was no access to the natural mineral. Enormous crystals occur at Monteponi and Montevecchio (Sardinia, Italy), in association with cerussite and anglesite; other notable localities are Tsumeb (Namibia), and Dundas (Tasmania).

Pb₂(CO₃)Cl₂

Colour:	colourless, white, pale brown or pale yellowish brown
Lustre; opacity:	adamantine; transparent to translucent
Streak:	white
Hardness:	2.5-3
Specific gravity:	6.13
Cleavage; fracture:	two distinct, prismatic; conchoidal, sectile and flexible
Habit:	crystals usually prismatic, may be tabular, weakly striated; also massive or granular
Crystal system:	tetragonal

Trona



Trona is deposited from saline lakes in arid regions, associated with halite, natron, gypsum or thenardite. It has an alkaline taste, is soluble and fluoresces white or blue. A natural source of sodium carbonate (soda ash), it is used to make paper, detergents, soap, glass, food and paper. The world's biggest reserves are in Wyoming (USA) and Lake Magadi (Kenya). In ancient Egypt, salt deposits in the lower Nile delta were exploited for natrum (largely trona), and used in medicine, mummification and to make cement (when mixed with caustic soda from silicate minerals and Nile silt). Notable occurrences include Searles Lake (California), Soda Lake (Nevada, USA) and the Otjiwalundo salt pan (Namibia).

Ma3(CO3)(IICC	З3.2П2 О
Colour:	colourless, grey, white, yellowish, brown
Lustre; opacity:	vitreous to dull, earthy; translucent
Streak:	white
Hardness:	2.5–3.0
Specific gravity:	2.17 .
Cleavage; fracture:	perfect; uneven to subconchoidal
Habit:	crystals prismatic or tabular, fibrous to compact aggregates, earthy crusts or efflorescences
Crystal system:	monoclinic

Na₃(CO₃)(HCO₃).2H₂O

Gaylussite



G aylussite, also known as natrocalcite, is a rare mineral named after the French scientist J.L. Gay-Lussac (1778–1850). It dehydrates when exposed to air, slowly crumbling and leaving calcium carbonate. It is slightly soluble in water, and effervesces in hydrochloric acid; it may luminesce creamy white. It is found in clay, shales and evaporites from soda lakes, along with natron, borax or calcite. Large crystals occur in the desert sands of Namibia and at Lake Amboseli (Kenya). It has also formed as an alteration product at Oldoinyo Lengai (Tanzania), the only volcano erupting natrocarbonatite lava. It occurs at Searles Lake (California, USA), Mt Erebus (Antarctica) and the Gobi Desert (Mongolia).

114204(003)2.51120	
Colour:	white, yellowish-white
Lustre; opacity:	vitreous, dull; transparent to translucent
Streak:	grey–white
Hardness:	2.5–3
Specific gravity:	1.99 (well below average)
Cleavage; fracture:	very good; conchoidal
Habit:	crystals prismatic or wedge-shaped; massive and encrusting
Crystal system:	monoclinic

Na2Ca(CO3)2.5H2O

Borax



 $B_{(Na_2B_4O_7,5H_2O)} b_{(Na_2B_4O_7,5H_2O)} b_{(Na_2B_4O_7,5H_2O$

Colour: colourless, white, tinged with grey, blue or green Lustre; opacity: vitreous, resinous, dull; transparent to opaque Streak: white Hardness: 2-2.5 Specific gravity: 1.7 Cleavage; fracture: perfect; conchoidal Habit: crystals short, prismatic; massive Crystal system: monoclinic

Na2B4O7.10H2O

Ulexite

NaCaR-O- 8H-O

Using the property has made it very popular, and it is sometimes referred to as the leaving to represent the terms of the specimes. The best locality for compact fibres is Boron, Kern County (California, USA). Turkey is the leading producer of boron ore, followed by the USA.

Macab509.011	
Colour:	white or colourless
Lustre; opacity:	vitreous or silky
Streak:	white
Hardness:	2.5
Specific gravity:	1.96
Cleavage; fracture:	perfect; uneven
Habit:	crystals very rare; rounded aggregates with fibrous internal structure, tufted masses
Crystal system:	triclinic

Colemanite

Olemanite is one of the main ores of boron (see also borax, kernite and ulexite). It forms in warm and arid regions in alkaline playa lakes deficient in sodium and carbonate; the boron is probably sourced from geothermal springs. One boron mineral may change to another during diagenesis, e.g. colemanite can form from ulexite through contact with calcium-bearing ground water. Associated minerals include other borates, gypsum, celestine and calcite. More than 80 per cent of the world production of borates comes from the USA (including colemanite from Death Valley, California) and Turkey. Boron is used in the chemical industry and in the manufacture of steel, paints, heat-resistant glass and fluxes for welding.

Colour: colourless, white, yellowish or grey Lustre; opacity: vitreous, adamantine; transparent to translucent Streak: white Hardness: 4-4.5 Specific gravity: 2.44 Cleavage; fracture: perfect; uneven or conchoidal Habit: crystals short, prismatic; compact, granular **Crystal system:** monoclinic

Ca2B6O11.5H2O

Kernite



Kernite is a rare mineral, the last of the four principal ores of boron. Specimens are difficult to keep because they effloresce (lose water), becoming opaque and dull. Kernite forms in intermittent lakes, supplied with boron from geothermal springs. It can form from borax with a rise in temperature and pressure, thus tending to concentrate at the base of stratified borate deposits. It can be associated with colemanite, ulexite and borax. Crystals up to 8.75m (29ft) long have been reported from Boron (Kern County, California, USA), a commercial source of kernite. Crystals up to 2.5m (8ft) occur at Tincalayu Mine (Salta, Argentina). Other important occurrences include Catalonia (Spain) and the Kirka deposit (Eskisehir, Turkey).

Na₂B₄O₇.4H₂O

Colour:	colourless when fresh, turning white
Lustre; opacity:	vitreous, dull, silky; transparent to opaque
Streak:	white
Hardness:	2.5-3
Specific gravity:	1.9
Cleavage; fracture:	perfect; splintery
Habit:	rare short, prismatic crystals; usually cleaved masses with fibrous habit
Crystal system:	monoclinic

Thenardite



Named after the French chemist L.J. Thénard (1777–1857), thenardite is Nanhydrous sodium sulphate. In damp air, samples will absorb water to give the hydrated form, mirabilite or Glauber salt ($Na_2SO_4.10H_2O$). This requires that the mineral be stored in closed containers. It is used in the manufacture of sodium salts and as a desiccant in the chemical industry. Mostly found in salt lakes in especially hot and dry tropical areas and deserts, thenardite is associated with minerals, including mirabilite, gypsum, glauberite, epsomite and halite. It is common in the deserts of the USA (California, Arizona, Nevada), the Sahara, Chile, Central Asia, Egypt and Sudan. Sometimes it is found in fumaroles, such as in Vesuvius (Italy).

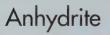
Na ₂ SO ₄	
Colour:	colourless, white or grey-white; yellowish or reddish tinges
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	2.5–3
Specific gravity:	2.7
Cleavage; fracture:	perfect; uneven to hackly
Habit:	crystals, sometimes large, are usually bipyramidal, occasionally prismatic with striated faces; crusts
Crystal system:	orthorhombic

Gypsum



The most common sulphate mineral, gypsum has relatively poor water-solubility and is often the first mineral deposited on evaporation of sea water. Varieties include selenite (colourless and transparent), satin spar (fibrous) and alabaster (massive and granular). Alabaster, usually white or in pastel shades, has been widely used for ornamental carvings for thousands of years. Satin spar is cut into cabochons and polished to give a 'cat's eye' effect. Its low thermal conductivity is utilized to insulate buildings ('drywalling'). Gypsum usually occurs as massive beds among other evaporites. It is often found as a cap rock on salt domes. It is found at Stassfurt (Germany), Paris basin (France) and Nova Scotia (Canada).

CasO ₄ .2H ₂ O	
Colour:	colourless or white; shades of grey, yellow or pink
Lustre; opacity:	vitreous, pearly on cleavage faces; transparent to translucent
Streak:	white
Hardness:	2
Specific gravity:	2.32
Cleavage; fracture:	perfect; conchoidal, splintery
Habit:	simple tabular crystals, often with curved faces; rosette-shaped aggregates and fibrous or granular masses
Crystal system:	monoclinic



A nhydrite is formed by evaporation of sea water above 42°C (108°C); gypsum forms below this. It is mostly produced by dehydration of buried gypsum, which causes the rock to shrink, forming crevices and caverns. Anhydrite and gypsum are also the names of rocks comprising large masses of these minerals. Such rocks are used for producing sulphuric acid and ammonium sulphate (for fertilizer), paper fillers and plasters (notably plaster of Paris). Anhydrite often forms thick beds in the lower strata of evaporite deposits and is abundant as the cap to salt domes. It is found at Stassfurt (Germany), the Paris basin (France) and Louisiana and Texas (USA). Unusual purple crystals are found at Bex (Switzerland).

CaSO ₄	
Colour:	colourless to bluish when transparent, white, pink or mauve when massive; may be further discoloured by impurities
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	white
Hardness:	3–3.5
Specific gravity:	2.9–3
Cleavage; fracture:	good; uneven, splintery
Habit:	usually massive or granular; sometimes fibrous or lamellar; rarely crystals
Crystal system:	orthorhombic

Celestine



Celestine, or celestite, named after the Latin for 'celestial', is the main ore of strontium. The most attractive crystals are a beautiful pale sky blue, but colourless celestine is also common. Large crystals (2–3kg/4½–6½lb) occur at Put-in-Bay (Ohio, USA). Strontium gives a strong crimson colour to a flame, useful in fireworks and flares. It is distinguished from barite by its lower density and very different flame test (barium gives green). Celestine occurs in impure limestones associated with sulphur, and in evaporites with gypsum, anhydrite or halite. Good crystals of celestine come from Lake Erie (USA), Bristol (England) and Sicily (Italy). Celestine, as a strontium ore, is mined in England, Russia and Tunisia.

SrSO₄

4	
Colour:	colourless or pale blue
Lustre; opacity:	vitreous, pearly on cleavage; transparent to translucent
Streak:	white
Hardness:	3–3.5
Specific gravity:	3.9-4
Cleavage; fracture:	perfect; uneven
Habit:	tabular or prismatic crystals; can be fibrous or granular
Crystal system:	orthorhombic

Anglesite



As with most lead minerals, anglesite is notably dense. Anglesite, named after Anglesey, Wales, is an ore of lead formerly much used as a pigment in white paints, but now largely superseded by titanium oxide. Attractive yellow crystals are popular, but are quite soft and fragile. Anglesite occurs in the oxidized zones of galena deposits, associated with cerussite, mimetite and pyromorphite; and rarely as a sublimate in volcanoes. Good crystals come from Tsumeb (Namibia), Anglesey (Wales), Musen (Germany), Phoenixville (USA) and Monteponi (Sardinia). Sulphurcovered crystals are from Los Lamentos (Chihuahua, Mexico). It is mined in Leadhills (Scotland), Derbyshire (England) and Arizona and Utah (USA).

PbSO ₄	
Colour:	colourless or white; can be tinged brown, grey, blue, green, purple or more usually pale yellow
Lustre; opacity:	adamantine, sometimes resinous; transparent to opaque
Streak:	white
Hardness:	2.5-3
Specific gravity :	6.2–6.4
Cleavage; fracture:	perfect; conchoidal
Habit:	crystals prismatic, tabular or pyramidal; massive, compact or granular
Crystal system:	orthorhombic



Barite (also baryte, barytes or heavy spar) is notable for its high density, the Greek baros meaning 'heavy'. Large crystals (up to 1m/3½ft) and roseate clusters of grains called desert roses are popular among collectors. Stalagmitic barite may be sectioned to reveal attractive concentric zoning. Large quantities are mined for paint and paper manufacture, as drilling muds in the oil industry and as a source of metallic barium. Barite is found in lead and zinc mines as a gangue mineral (mining waste) with calcite, fluorite and quartz, occurring in veins in lead, zinc, silver, iron and nickel ores. Good desert roses come from Oklahoma and Kansas (USA); large crystals are to be found in Cumbria and Cornwall (England).

BaSO₄

Barite

Colour:	colourless or white; red, brown, yellow or blue
Lustre; opacity:	vitreous, pearly on cleavage; transparent to translucent
Streak:	white
Hardness:	3–3.5
Specific gravity:	4.3–4.6
Cleavage; fracture:	perfect; uneven
Habit:	crystals tabular, sometimes prismatic; often massive, sometimes fibrous or lamellar clusters, or granular stalagmites
Crystal system:	orthorhombic

Spangolite



A beautiful rare blue-green mineral, spangolite usually occurs as aggregates of small crystals on a matrix; specimens with minerals such as azurite, brochantite and chrysocolla make pleasing displays. Spangolite is named after the engineer and avid mineral collector Norman Spang (1843–1922), who owned the original specimen. Spangolite is widespread, but found only in small quantities. Spangolite occurs in oxidized copper sulphide ores associated with azurite, malachite, adamite and tyrolite. It is found at Blanchard mine (New Mexico, USA), Tintic (Utah, USA), Mujaba Hill (Nevada, USA), St Day (Cornwall, England), Fontana Rossa (Corsica), Broken Hill (NSW, Australia), Laurion (Greece) and Arenas (Sardinia).

0	12010120
Colour:	green or greeny-blue
Lustre; opacity:	vitreous; transparent
Streak:	pale yellow-green
Hardness:	3
Specific gravity:	3.1
Cleavage; fracture:	perfect; conchoidal
Habit:	short prisms and hexagonal plates
Crystal system:	hexagonal

Cu₆AlSO₄(OH)₁₂Cl.3H₂O

Brochantite



Brochantite, named after the French geologist A. Brochant de Villiers (1772–1840), is a bright green mineral popular as a micromount. It is very difficult to distinguish from fibrous forms of antlerite and atacamite, although the latter is softer. Brochantite is formed as a secondary mineral in the oxidized zone of copper deposits, usually in dry climates. It is associated with malachite, atacamite, antlerite, limonite, cuprite, chrysocolla and cyanotrichite. Popular localities are Chuquicamata (Chile), Tsumeb (Namibia), Sardinia, Rezbanya (Romania), Ain Barbar (Algeria) and Broken Hill (NSW, Australia). Artificial brochantite is produced, so collectors should be careful with sources.

Cu ₄ SO ₄ (OH) ₆	
Colour:	bright green
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	light green
Hardness:	3.5–4
Specific gravity:	3.97
Cleavage; fracture:	perfect; uneven to conchoidal
Habit:	small, stubby or acicular, striated crystals, sprays of small needles; fibrous or granular crusts and nodules; rarely massive
Crystal system:	monoclinic

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Alunite



A lunite is the main constituent of the rock alum stone, mined for alum production since the fifteenth century. Alum has the formula $KAl(SO_4)2.12H_2O$ and is used for papermaking and treating skins. Alum and alunite have been used to stem bleeding of small cuts. Most deposits were formed when rocks rich in alkali feldspar (pegmatites, syenites, trachytes) were altered by sulphate-rich water, in turn produced from altered pyrite. Sometimes alunite occurs in veins cutting across schists and in fumaroles. It is often found with halloysite and kaolinite. Alunite is mined at La Tolfa (Italy), Nevada (USA), France, Hungary, Greece, Almeria (Spain) and Bullah Delah (NSW, Australia).

Colour: white; can be grey, yellowish or reddish Lustre; opacity: vitreous; transparent to translucent Streak: white Hardness: 3.5-4 Specific gravity: 2.6-2.9 Cleavage; fracture: good; conchoidal Habit: crystals as rhombohedra or plates; usually granular or earthy masses Crystal system: hexagonal

KAl₃(SO₄)₂(OH)₆



Linarite is a rare mineral usually occurring with copper and lead ores. The name lis derived from Linares, where it was first recognized in 1839. Linarite can be distinguished from the similar mineral azurite, as the latter fizzes with acids. Linarite is found in the oxidized zones of mixed lead and copper deposits, associated with aurichalcite, cerrussite, malachite and hemimorphite. The best crystals, more than 10cm (4in) in length, come from Mammoth Mine (Tiger, Arizona, USA); others come from Kisamori (Akita, Japan), Linares plateau (Jean, Spain), Tsumeb (Namibia), Red Gill (Cumbria, England), Serra de Capitillas (Argentina), Rosas and San Giovani (Sardinia), Leadhills (Scotland).

PbCuSO₄(OH)₂

Colour:	royal blue
Lustre; opacity:	vitreous; transparent to translucent
Streak:	blue
Hardness:	2.5
Specific gravity:	5.3-5.5
Cleavage; fracture:	perfect; conchoidal
Habit:	acicular or tabular crystals, often slender prisms like tourmaline; encrustations, aggregates
Crystal system:	monoclinic

Glauberite



G lauberite is used for the extraction of Glauber salt (Na₂SO₄.10H₂O), after which it is named, as a mordant in textile dying and in pharmaceuticals. Glauberite is very water-soluble and can be dissolved out of rocks and infilled with other mineral pseudomorphs. Glauberite may be found in buried sediments of geological age or in modern salt lakes and other evaporites. Associated minerals include halite, thenardite, calcite and gypsum. Occasionally it is a sublimate in fumaroles. Good crystals of glauberite are found in Villarubia (Spain), Lorraine (France) and salt lakes in California and Arizona (USA). Large masses occur around Saltzburg (Austria), Ruthenia (Russia), Texas (USA) and New Mexico (USA).

Na₂Ca(SO₄)₂ Colour: white, tinged yellow or brick-red Lustre; opacity: vitreous; transparent to translucent Streak: white Hardness: 2.5 - 3Specific gravity : 2.8 Cleavage; fracture: perfect; conchoidal Habit: tabular, prismatic or bipyramidal crystals, sometimes with striated faces and rounded edges; compact masses and crusts Crystal system: monoclinic

Halotrichite



Halotrichite is isomorphous with pickeringite (MgAl₂(SO₄)₄.22H₂O), which is present in most specimens. Masrite is a variety containing manganese and cobalt; apjohnite, dietrichite, redingtonite, wupatkiite and bilinite variously have cobalt, manganese, zinc, nickel and/or chromium replacing the iron or aluminium. Halotrichite, also known as iron alum or feather alum, is named from the Latin for 'hair salt' because of its form as fine fibrous needles. It is found as a pyrite alteration product in aluminous rocks and in fumaroles. It occurs at Sulfatara (Naples, Italy), Falun (Sweden), Recsk (Hungary), Istria (Croatia), Dubnik (Slovakia), Copiapo (Atacama, Chile) and Alum Mountain (New Mexico, USA).

1 2/10 0 4/4/-	
Colour:	colourless, white, yellowish, greenish
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	1.5
Specific gravity:	1.9
Cleavage; fracture:	imperfect; conchoidal
Habit:	acicular, prismatic crystals; crusts and aggregates
Crystal system:	monoclinic

FeAl₂(SO₄)₄.22H₂O

Cyanotrichite



Cyanotrichite is named from the Greek for 'blue hair', after the appearance of felted masses of crystalline needles. Related to but rarer than halotricite, it is also known as lettsomite or velvet copper. Specimens as attractive blue tufts on a matrix are best viewed magnified, as the crystals are usually small. Cyanotrichite occurs in the oxidized zones of copper deposits, associated with minerals such as azurite, malachite and limonite. It is found at Cap Garonne (France), Laurion (Greece), Nemaqualand (South Africa), Mednorudnyansk (Russia), Bisbee (New Mexico, USA), Grandview (Arizona, USA), Moldova Noua (Romania), Traversella (Piedmont) and Rio Marina (Elba, Italy) and St Day (Cornwall, England).

Cu4A12304(0)	11/12.21120
Colour:	orange-red, hyacinth red
Lustre; opacity:	vitreous; transparent to translucent
Streak:	ochre yellow
Hardness:	2.5
Specific gravity:	2.1
Cleavage; fracture:	perfect; conchoidal, irregular
Habit:	elongated prismatic crystals as radiating fibrous aggregates of botryoidal or reniform shape
Crystal system:	monoclinic

Cu4Al2SO4(OH)12.2H2O



Botryogen is a rare dark orange mineral named from the Greek *botrys* for 'grape' and *genos* for 'yield', after its characteristic habit. A popular variety with small aggregates of botryogen on halotrichite needles is found at Smolník (Slovakia). Botryogen, also known as red iron vitriol, is formed by the action of sulphate-rich fluids on mafic minerals. It often occurs near pyrite deposits in arid regions, associated with epsomite, copiapite and voltaite. It is found at Rammelsberg (Germany), Falun (Sweden), Villé Valley (France), Chuquicamata (Chile), Knoxville (Tennessee, USA), San Juan (Argentina), Queensland (Australia), Madeni-Zakli (Iran) and Paracutin (Mexico).

	<u> </u>
Colour:	orange-red, hyacinth red
Lustre; opacity:	vitreous; transparent to translucent
Streak:	ochre yellow
Hardness:	2.5
Specific gravity:	2.1
Cleavage; fracture:	perfect; conchoidal, irregular
Habit:	elongated prismatic crystals as radiating fibrous aggregates of botryoidal or reniform shape
Crystal system:	monoclinic

MgFe(SO₄)₂OH.7H₂O

Kröhnkite



Named after B. Kröhnke, the first person to analyze it, kröhnkite is a rare blue mineral easily mistaken for others such as chalcanthite. Kröhnkite is abundant in large royal blue crystals in the Chuquicamata mine, which is the largest open-pit copper mine in the world. Twinning in crystals can result in attractive heart-shaped forms. It is formed from sulphate-rich fluids in desert regions and found among copper ores often associated with chalcanthite, antlerite and atacamite. Good localities for finding kröhnkite are Chuquicamata and Quetena mines (Chile), Capo Calamita (Elba, Italy), Wheal Hazard (Cornwall, England), Recsk copper deposit (Matra Mountains, Hungary) and Broken Hill (NSW, Australia).

$Na_2Cu(SO_4)_2$.	2H ₂ O
Colour:	blue to light blue
Lustre; opacity:	vitreous; transparent
Streak:	white
Hardness:	2.5–3
Specific gravity:	2.9
Cleavage; fracture:	perfect; conchoidal
Habit:	prismatic crystals; fibrous and granular crusts and aggregates
Crystal system:	monoclinic

Kainite



Kainite is a double salt formed as a mixture of potassium chloride (sylvite) and magnesium sulphate (epsomite). It is named somewhat obscurely after the Greek *kainos*, meaning 'new' or 'contemporary', alluding to its occurrence in rocks of recent origin. Kainite is a common mineral in evaporites, but difficult to identify. Kainite is used in fertilizers and as a source of potassium compounds. It is found among evaporite deposits associated with halite, sylvite and carnallite; and as a volcanic sublimate. Localities where kainite has been mined are Hallstadt (Austria), Eddy County (New Mexico, USA), Kalus (Ukraine) and Sicily (Italy). It is a major mineral found in the large salt deposits of middle and northern Germany.

KMgSO₄Cl.3H₂O

Colour:	white, yellow, greyish
Lustre; opacity:	vitreous
Streak:	white
Hardness:	3
Specific gravity:	2.1–2.2
Cleavage; fracture:	good
Habit:	rare thick, tabular crystals; usually granular or fibrous
	aggregates
Crystal system:	monoclinic

Devilline



Devilline is an attractive blue-green mineral named after the French chemist H. Deville (1818–1881). It is also called devillite or herrengrundite after a locality in Slovakia. A sky-blue zinc-bearing variety is called serpierite or zincian devilline. Devilline is a rare secondary mineral found in copper deposits associated with gypsum, azurite and malachite. It is found in Cornwall (England), Pania Dolina (Slovakia), Tsumeb (Namibia), Montgomery (Pennsylvania, USA) and Vezzani, (Corsica). Serpierite has been found at Kamariza (Laurion, Greece), Ross Island (Ireland) and Broken Hill (NSW, Australia). A light blue manganese-bearing variety campigliaite has been identified at Temerino (Tuscany, Italy).

Colour:	blue-green to emerald green
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	light green
Hardness:	2.5
Specific gravity:	3.1
Cleavage; fracture:	perfect; brittle
Habit:	crystals often lamellar in rosettes or needle-like clusters; crusts, botryoidal aggregates
Crystal system:	monoclinic

CaCu₄(SO₄)₂(OH)₆

Epsomite



Commonly known when pure as Epsom or bitter salts, epsomite is a white watersoluble hydrate of magnesium sulphate. It is used as a mordant in hide tanning and textile dying, and dried for use as a desiccant. It is precipitated from hot springs, fumaroles and saline waters. The thermal waters at Epsom are well known for encrustations of epsomite. It is found on the walls of mines near weathered pyrite deposits. Crystals up to 2–3m (6½–10ft) come from Kruger Hills (Washington, USA). Epsomite is light and fragile, and loses water in dry air, becoming dull in appearance. It is mined at Carlsbad (New Mexico, USA), Saxony (Germany), Sedlec (Czech Republic) and Valle Antrona (Italy).

MgSO₄.7H₂O

white; can be tinged yellow, green, red
vitreous to silky; transparent to translucent
white
2–2.5
1.68
perfect; conchoidal
crystals rare; usually crusts, stalactites and earthy masses
orthorhombic

Römerite



 $\mathbf{R}^{omerite}$ is an attractive yellow-brown mineral named after the German geologist Friedrich Römer (1809–1869). Violet-brown aggregates and dark orange crystals are popular, attractive occurrences. It is formed in oxidized zones of iron sulphide deposits, associated with other sulphates, especially halotrichite, copiapite (Fe₅(SO₄)₆(OH)₂.20H₂O) and melantesite (FeSO₄.7H₂O). It is found in Dresden (Saxony), Rammelsberg (Harz) and Valdsassen (Bavaria, Germany), Rtyne and Pribram (Czech Republic), Madeni Zakh (Iran), Tierra Amarilla and Chuquicamata (Chile), Blyava (Southern Urals) and Kamchatka (Russia).

Colour:	brown to yellow-brown, occasionally violet-tinged
Lustre; opacity:	vitreous, resinous; translucent
Streak:	yellow-brown
Hardness:	3–3.5
Specific gravity:	2.17
Cleavage; fracture:	perfect, good; uneven
Habit:	tabular or pseudocubic crystals; crusts, granular layers and stalactites
Crystal system:	triclinic

Fe₃(SO₄)₄.14H₂O

Crocoite



Crocoite, also called red lead ore, is named from the Greek *krokos*, meaning 'saffron' or 'crocus', after its bright orange-red colour. It is the mineral from which chromium was first extracted, but is not used as an ore commercially. Crocoite is one of the few naturally occurring chromates, which are always strongly coloured. Pure crocoite forms the pigment chrome yellow. It is formed where chromic-acid rich hydrothermal solutions have attacked lead deposits. It is associated with cerrusite, limonite, anglesite, phoenicochroite and vauquelinite. Crystals up to 15cm (6in) have been found at Dundas (Tasmania); other localities include Minas Gerais (Brazil), Nontron (France) and Manmoth Cave (Arizona, USA).

PbCrO ₄	
Colour:	yellow, orange, red
Lustre; opacity:	adamantine; translucent
Streak:	orange
Hardness:	2.5–3
Specific gravity:	6
Cleavage; fracture:	perfect; conchoidal
Habit:	prismatic crystals, often striated, often hollow, acicular crystals; massive, granular
Crystal system:	monoclinic

Scheelite



Scheelite is a major ore of tungsten, used in lightbulb filaments, high-strength Steels and tungsten carbide cutting tools. Named after the Swedish chemist K. Scheele (1742–1786), it forms crystals with a high lustre and fire, and was used as an early synthetic diamond. Fluorescing pale blue under UV light, scheelite is used as a scintilator. The fluorescence becomes greener as the content of powellite (CaMoO₄) increases. Scheelite is found in pegmatites, hydrothermal veins and medium-grade metamorphic rocks. Crystals of a weight up to 500g (1lb 2oz) are reported from Minas Gerais (Brazil) and of a size up to 7.5cm (3in) in Tong Wha (Korea); other localities are Sardinia, Traversella, Val di Flemme and Valsugana (Italy).

CaWO ₄	
Colour:	grey-white, yellow, brown, rèddish, greenish
Lustre; opacity:	vitreous to adamantine; transparent to opaque
Streak:	white
Hardness:	4.5–5
Specific gravity:	5.9–6.1
Cleavage; fracture:	imperfect; conchoidal
Habit:	crystals as bipyramids, pyramids, plates; massive aggregates, granular, crusts, pseudomorphs
Crystal system:	tetragonal

Wolframite



Wolframite is a major ore of tungsten, its name coming from *wolfram*, the German word for tungsten. Its composition lies within the ferberite-heubnerite series (FeWO₄-MnWO₄). The colour darkens as the iron content increases. Tungsten is widely used, but some demand for its use in hardened metals has been taken by depleted uranium. It occurs in pegmatites, hydrothermal veins, pneumatolytic zones and placer deposits. Associated minerals include scheelite, cassiterite, pyrite and sphalerite. Important deposits are in southern China, Queensland (Australia), La Paz (Bolivia), Arizona (USA) and New Mexico (USA); superb crystals at Hualapón (Pasto Bueno, Peru) and Quartz Creek (Idaho, USA).

(Fe,Mn)WO₄

Colour:	yellow to reddish-brown to blackish brown
Lustre; opacity:	submetallic, resinous; opaque
Streak:	brown to black
Hardness:	5–5.5
Specific gravity:	7.1–7.5
Cleavage; fracture:	perfect; uneven
Habit:	vertically striated tabular or lamellar crystals; massive, granular
Crystal system:	monoclinic

Wulfenite



Wulfenite, or yellow lead ore, a minor ore of molybdenum, is named after the Austrian mineralogist Franz Wülfen (1728–1805). Popular specimens comprise clusters of large square red crystals among smaller ones. As with many lead minerals, wulfenite has a high refractive index (2.3–2.4), which enhances its appearance. Wulfenite occurs in the oxidized zones of lead deposits, associated with cerussite, vanadinite, pyromorphite and mimetite. The best localities include: Red Cloud (Arizona, USA), noted for superb red crystals; Rezbanya (Romania) for reddish crystals; Bleiberg (Austria) for yellow crystals; Tsumeb (Namibia) and Phoenixville (Pennsylvania, USA) for colourless crystals.

PbMoO₄ Colour: yellow, orange, red, rarely white Lustre; opacity: resinous; transparent to opaque Streak: yellowish-white Hardness: 2-3 Specific gravity: 6.7-6.9 Cleavage; fracture: imperfect; conchoidal Habit: crystals square tabular or stubby pyramids, often as clusters: massive, granular or earthy aggregates Crystal system: tetragonal

Purpurite



Purpurite is a very rare mineral, much sought-after by collectors. Its name comes from its characteristic colour when fresh, although it may show surface alteration to a darker brown. It forms through the oxidation and leaching of manganese-iron phosphates, particularly lithiophyllite (LiMnPO₄), found in granite pegmatites. It has also been known to form through the reaction of sea water with bat guano, which provides the phosphorus. Associated minerals include lithiophyllite and heterosite. As well as the type locality Faires tin mine (North Carolina, USA), it occurs in the Varuträsk pegmatite (Västerbotten, Sweden), Mangualde (Portugal), La Vilate quarry (Chanteloupe, France) and the Gunong Keriang cave (Malaysia).

(Mn,Fe)PO ₄	
Colour:	reddish purple to deep rose red
Lustre; opacity:	satiny on fresh fractures, or earthy; translucent to opaque
Streak:	pale purple to pale red
Hardness:	4-4.5
Specific gravity:	3.2–3.4
Cleavage; fracture:	good, surfaces may be curved or crinkled; uneven, brittle
Habit:	does not occur as crystals; forms small irregular masses, cleavage fragments may reach 20cm (8in)
Crystal system:	orthorhombic

PHOSPHATES, ARSENATES AND VANADATES

Monazite

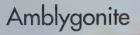


Monazite is a rare earth phosphate, containing up to 12 per cent ThO₂, and is Mfeebly radioactive. It is a common accessory in granites, syenites and metamorphic gneisses, and is also found as a placer deposit, being dense and resistant to chemical weathering. Monazite is the main source of rare earths, used as a fuel-cracking catalyst in the petroleum industry, and of thorium. Thorium is important in the computer, electronic and medical industries. It is also the most efficient nuclear reactor fuel, and could potentially be used instead of uranium. Important commercial sources are Australia (with the largest reserve), Florida (USA), Madagascar, Brazil, Travancore (India), Sri Lanka and South Africa.

(Ce,La,Nd,Th)PO₄

yellow to reddish brown
resinous to waxy, vitreous; transparent to translucent
white
5-5.5
4.6-5.4
good; conchoidal to uneven fracture
crystals tabular or prismatic; usually as grains
monoclinic

PHOSPHATES, ARSENATES AND VANADATES



A mblygonite is a fluorophosphate that can luminesce orange in long-wave Aultraviolet light. It occurs in coarse-grained granitic rocks and in lithium- and phosphate-rich granitic pegmatites – associated, for example, with apatite, pollucite and spodumene. It is used as a source of lithium, but can also be faceted to make gems when transparent or translucent. Amblygonite being relatively soft, these tend to be of interest only to collectors. Gem-quality material occurs in pegmatites at Minas Gerais (Brazil) and at Newry (Maine, USA), and giant crystals weighing several tons have been found at Custer (Dakota, USA); it also occurs in the pegmatites of Pala (California, USA) and in Yauapa County (Arizona, USA).

(Li,Na)Al(PO₄)F

Colour:	milk white, yellow, bluish, greenish, pink or colourless
Lustre; opacity:	vitreous to greasy; transparent to translucent
Streak:	white
Hardness:	5.5–6
Specific gravity:	3.1
Cleavage; fracture:	two good cleavages; uneven
Habit:	crystals equant, rough, and may be twinned; also as cleavable masses, or compact
Crystal system:	triclinic

PHOSPHATES, ARSENATES AND VANADATES

Olivenite



Named after its olive-green colour, olivenite is much in demand by mineral Collectors. Although rare, it is the most common secondary copper arsenate found in the oxidized zone of hydrothermal copper deposits. It is associated with other copper minerals such as azurite, malachite, chrysocolla and chalcopyrite, and sometimes with arsenopyrite and spangolite. The type locality is Carharrak Mine in Cornwall, England; elsewhere in Britain it is found on Alston Moor (Cumbria) and Tavistock (Devon). Other localities include Cap Garonne (France), Clara Mine (near Oberwolfach, Germany), and the porphyry copper deposits of Chuquicamata in Chile. Excellent large crystals occur in Tsumeb (Namibia).

Colour:	shades of olive green, dirty white, straw yellow
Lustre; opacity:	adamantine, vitreous, pearly, silky; transparent to opaque
Streak:	olive green to brown
Hardness:	3
Specific gravity:	4-4.5 ·
Cleavage; fracture:	indistinct; forms small conchoidal fragments, very brittle
Habit:	crystals variable, may be elongated, or of short, prismatic, acicular or fibrous form; also as reniform masses
Crystal system:	monoclinic; pseudo-orthorhombic

Cu₂(AsO₄)(OH)

Adamite

Zn₂(AsO₄)(OH)

Colour:	yellow, brownish yellow to green, pink or violet
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	3.5
Specific gravity:	4.3-4.5
Cleavage; fracture:	good; uneven to subconchoidal
Habit:	crystals elongate tabular, platy or equant; usually forms crusts or spheroidal masses, or roughly radial aggregates
Crystal system:	orthorhombic

Lazulite



Lazulite derives its name from the Arabic *azul*, meaning 'sky' or 'heaven', and the Greek *lithos*, meaning 'stone'. It has sometimes been confused with lazurite, lapis lazuli or azurite. It is a minor semi-precious stone, and may be polished, carved or tumbled to make beads or attractive ornamental stones. The rare transparent form is pleochroic, changing from blue to colourless. Lazulite occurs in high grade, quartz-rich metamorphic rocks, veins and pegmatites, associated with minerals such as andalusite, kyanite, sillimanite, and garnet. Beautiful crystals up to 5cm (2in) long occur in the Graves Mountains (Georgia, USA). Some of the best specimens occur in veins in iron-rich shales at Rapid Creek (Yukon, Canada).

Colour:	deep azure blue to light blue or bluish green; mottled
Lustre; opacity:	vitreous to dull; translucent to opaque
Streak:	white
Hardness:	5.56
Specific gravity:	3.1–3.4 (iron rich)
Cleavage; fracture:	indistinct to good, prismatic; uneven to splintery
Habit:	crystals show steep pyramidal forms, may be tabular; also massive or granular
Crystal system:	monoclinic

MgAl₂(PO₄)₂(OH)₂

Pseudomalachite



Pseudomalachite differs from malachite because it lacks the characteristic colour banding and does not effervesce in warm hydrochloric acid, whereas malachite does. It is a very rare secondary mineral found in the oxidized zone of copper deposits, associated with azurite, malachite, atacamite, limonite or chalcedony. It is sometimes used as a precious stone. Its use in paint has been recorded in an ancient tomb at the prehistoric Mayan site known as Baking Pot, in Belize. It is found at Virneberg Mine (Rheinbreitbach, Germany), Cornwall (England), Libethen (Slovakia), Harquahala Mine (Arizona, USA), Bogolo (Portugal), Manto Cuba and San Salvador Mines (Atacama, Chile) and in Nizhni Tagil (Russia).

Cu5(PO4)2(OH)4

Colour:	emerald green to blackish green; paler when fibrous
Lustre; opacity:	vitreous, greasy; translucent to transparent
Streak:	blue-green, paler than colour
Hardness:	4.5-5
Specific gravity:	4.3
Cleavage; fracture:	perfect; splintery
Habit:	crystals rare, rough and rounded; forms radial clusters, coatings, stalactitic, reniform or botryoidal aggregates
Crystal system:	monoclinic

Brazilianite



Brazilianite is a rare and unusual mineral, named after its discovery at Minas Gerais (Brazil) in 1945. Brazilianite occurs in phosphate-rich pegmatites, and may be associated with clay, lazulite and apatite. A striking yellow or yellowish green, it is sometimes cut for collectors, and is probably the best of the phosphates to be used as a gemstone. Crystals are fragile and easily broken, however, as they have a conchoidal fracture and a very good cleavage parallel to their length. Good crystals occur at Minas Gerais (Brazil) and Palermo Mine (New Hampshire, USA). Other localities include Roľná (Czech Republic), Hagendorf (Germany), Etiro (Namibia) and the Dawson Mining District (Yukon Territory, Canada).

Colour: colourless, light yellow to greenish yellow Lustre; opacity: vitreous; transparent Streak: white Hardness: 5.5 Specific gravity: 3.0 Cleavage; fracture: good; conchoidal fracture Habit: crystals common and well-formed, striated; also as spherical aggregates Crystal system: monoclinic

NaAl₃(PO4)₂(OH)₄

Clinoclase



Clinoclase forms as a secondary mineral in the oxidation zone of arsenic-rich hydrothermal copper deposits. Associated minerals include liroconite, chalcophyllite, olivenite and cornwallite (Cu₅(AsO₄)₂(OH)₄). The type locality is Wheal Gorland (Cornwall, England); now closed, this mine has produced many fine and rare minerals. In common with other unusual secondary copper arsenate minerals, clinoclase occurs at the Majuba Hill Mine (Nevada, USA), also closed. Blueblack crystalline aggregates from this mine have been termed beetle ore. Other localities include Sayda (near Freiberg, Germany), Cap Garonne (France), Novoveská Huta (Slovakia), Stirling Mine (New Jersey, USA) and Tintic (Utah, USA).

Cu₃AsO₄(OH)₃

Colour:	greenish black or bluish green
Lustre; opacity:	vitreous, pearly on cleavage; subtransparent to translucent
Streak:	blue-green
Hardness:	2.5–3
Specific gravity:	4.2-4.4
Cleavage; fracture:	very good parallel to base; uneven
Habit:	crystals small, needle-like, pseudo-orthorhombohedral, or platy; radial, fibrous, spherical aggregates or encrustations
Crystal system:	monoclinic

Descloizite



Descloizite is the zinc end member of a complete series in which copper substitutes for zinc, mottramite being the copper end member. A secondary mineral, it forms in the oxidized zone of lead, zinc and copper ore deposits, and is often associated with lead minerals such as vanadinite, cerussite, wulfenite and pyromorphite. It also forms in sandstone, having been precipitated from circulating mineralized ground water. It is a rare but important ore of vanadium. Large masses are found at Otavi (Namibia), and particularly fine crystals at Bisbee (Arizona, USA) and Lake Valley (New Mexico, USA). It also occurs at Obir (Corinthia, Austria), Bena de Padru (Sardinia, Italy) and Tsumeb (Namibia).

Colour:	orange-red, brown, brownish-red
Lustre; opacity:	resinous to greasy; transparent to translucent
Streak:	orange to reddish brown
Hardness:	3.5
Specific gravity:	5.5–6.2
Cleavage; fracture:	poor; uneven to subconchoidal, brittle
Habit:	crystals prismatic or platy; fibrous, botryoidal masses or crusts
Crystal system:	orthorhombic

PbZn(VO₄)(OH)

Apatite



A patite is the name for a group of minerals including fluorapatite, chlorapatite, and hydroxylapatite, in which fluorine, chlorine and hydroxyl substitute for one another, and carbonate-apatite, $Ca_5(PO_4, CO_3, OH)_3(F, OH)$. Apatite is the main inorganic constituent of bones and teeth. It is a common accessory mineral in igneous rocks, and is found in some contact metamorphic calc-silicate rocks. It also occurs in extensive and economically important bedded deposits called phosphorites. Eighty per cent of world production is used in chemical fertilizers and in the production of phosphoric acid. Major mining areas include Florida and North Carolina (USA), Morocco, and the Kola Peninsula (Russia).

Calaria	The second second second second second
Colour:	usually green, may be violet, red or brown
Lustre; opacity:	vitreous to sub-resinous; transparent to opaque
Streak:	white
Hardness:	5
Specific gravity:	3.1–3.2
Cleavage; fracture:	poor; conchoidal to uneven
Habit:	crystals prismatic or needle-like, sometimes tabular; also massive, concretionary, mammillated or oolitic
Crystal system:	hexagonal

Ca5(PO4)3(F,CI,OH)

Pyromorphite



 $\label{eq:product} Pyromorphite forms one series with mimetite (Pb5(As0_4)_3Cl) and another with vanadinite (Pb5(VO_4)_3Cl). It is named after the Greek pyr ('fire') and morfe ('form'), as melted globules will recrystallize. It can luminesce yellow in ultraviolet light. Pyromorphite often occurs in rounded barrel-shaped forms known as campylite. It forms as a secondary mineral in the oxidized zone of lead veins, and rarely as a volcanic sublimate. Associated minerals include galena, wulfenite, cerussite, vanadinite and descloizite. Pyromorphite is a minor ore of lead known as green lead ore. It occurs in Cornwall (England), Cordoba (Spain), Guang Xi (China), Bunker Hill Mine (Idaho, USA), Broken Hill (NSW, Australia) and Zambia.$

Pb ₅ (PO ₄) ₃ Cl	
Colour:	usually green, can be yellow, orange, brown or colourless
Lustre; opacity:	adamantine, vitreous, resinous; transparent to translucent
Streak:	white
Hardness:	3.5-4
Specific gravity:	7.1; unusually dense for a translucent mineral
Cleavage; fracture:	poor; uneven to subconchoidal
Habit:	crystals sometimes prismatic, occasionally hollow; also reniform or globular
Crystal system:	hexagonal

Mimetite

Mimetite was named from the Greek *mimetes*, the 'imitator', alluding to its resemblance to pyromorphite, with which it forms a series. It is a rare secondary mineral formed in the oxidized zone of arsenic-bearing lead deposits. It is commonly associated with pyromorphite and other lead minerals, such as vanadinite, galena and anglesite, and also with hemimorphite and arsenopyrite. It may fluoresce reddish yellow. It is soluble in hydrochloric acid, producing a very strong smell similar to garlic. Mimetite is a minor ore of lead. Well-crystallized material is found at Johanngeorgenstadt (Germany), Pfilbram (Czech Republic) and particularly at Tsumeb (Namibia).

Pb₅(AsO₄)₃Cl

Colour:	yellow, orange, brown, red, white, or may be colourless
Lustre; opacity:	vitreous, resinous; subtransparent to translucent
Streak:	white
Hardness:	3.5–4
Specific gravity:	7.0–7.3
Cleavage; fracture:	weak; uneven to subconchoidal, brittle
Habit:	crystals slender, acicular, or may be tabular or barrel-shaped; globular, reniform, stalactitic or granular
Crystal system:	hexagonal

Vanadinite



Tanadinite is a rare mineral that darkens on exposure and loses transparency. It shares the same structure as apatite, and hence similar crystal shapes. It forms as a secondary mineral in the oxidation zone of lead deposits, associated with wulfenite, pyromorphite, mimetite, cerussite or anglesite. Vanadinite obtained as a by-product of other mining operations can be used as an ore of vanadium, used in metal alloys such as steel, in dyes and in ceramics. The best crystals are found at Mibladen (Morocco) and Grootfontein (South Africa). The original material from the type locality of Zimapan (Hildalgo, Mexico) was lost at sea. Vanadinite also occurs in Kabwe (Zambia), Obir (Austria) and Tsumeb (Namibia).

$Pb_5(VO_4)_3CI$	
Colour:	bright reddish-orange to yellow, brown
Lustre; opacity:	resinous to adamantine; translucent to opaque
Streak:	yellowish-white to brownish-yellow
Hardness:	2.5–3
Specific gravity:	6.9–7.0
Cleavage; fracture:	none; conchoidal or uneven, brittle
Habit:	crystals stubby, hexagonal, may be hollow; as fibrous radiating masses, crusts, granular, or nodular
Crystal system:	hexagonal

Variscite



Wariscite forms a series with strengite (FePO₄.2H₂O). It typically forms where phosphatic meteoric water has circulated through and reacted with aluminiumrich rocks, such as feldspar-rich igneous rocks. It has also been found in caves, the source of the phosphorus being decomposing bat guano. Associated minerals include apatite, wavellite and limonite. The name comes from Variscia, the old name for Vogtland (Germany), the type locality. Nodules up to 30cm (12in) occur at Fairfield (Utah, USA), and greenish encrustations in Montgomery County (Arkansas, USA). Variscite is occasionally smoothed, polished and sold as turquoise, or it can be cut into cabochons.

AIPO4.2H2O

pale green, bluish green, emerald green; may be colourless
waxy to earthy; transparent to translucent
white
4–5
2.52 (variscite)
good; conchoidal or uneven to splintery
crystals very rare; massive, encrusting or reniform
orthorhombic

Strengite



Strengite is the iron-bearing end member of the strengite-variscite series. It is a secondary mineral that forms in complex granitic pegmatites, as an alteration product of primary phosphate minerals such as triphylite (Li(Fe,Mn)PO₄) – e.g. Bull Moose Mine, South Dakota, USA. It also occurs in limonitic iron ores and gossans, and more rarely in caves where the phosphate from bat guano has been used in the alteration of iron-rich minerals to strengite. It may be associated with dufrenite, cacoxenite, vivianite or apatite. Localities include Hagendorf (Bavaria, Germany), Bomi Hill Caves (Liberia), Indian Mountain (Alabama, USA), Minas Gerais (Brazil) and Iron Monarch Quarry (South Australia).

rep04.2H20	
Colour:	shades of violet, pink or red, may be colourless
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	3.5
Specific gravity:	2.87
Cleavage; fracture:	good; conchoidal
Habit:	crystals of variable habit; botryoidal, radial, or spherical nodules
Crystal system:	orthorhombic

FePO₄.2H₂O

Phosphophyllite



Phosphophyllite is found as a primary precipitate in tin-rich hydrothermal veins in Bolivia. Elsewhere, it occurs as a secondary mineral in complex granitic pegmatites, formed by the alteration of sphalerite and iron-manganese phosphates, and in hydrothermal vein deposits. Associated minerals include vivianite, strengite, sphalerite and apatite. It is a highly prized gemstone when a pale bluish green, and may be cut into facets or cabochons. It is brittle and fragile, however, and is rarely cut, as large crystals are too valuable to break up. It is very rare. Crystals up to 10cm (4in) occur at Potosi (Bolivia); other locations include Hagendorf (Germany), North Groton (New Hampshire, USA) and Kabwe (Zambia).

Colour:	bright blue green, green to colourless
Lustre; opacity:	vitreous; transparent
Streak:	white
Hardness:	3–3.5
Specific gravity:	3.1
Cleavage; fracture:	perfect; conchoidal or uneven
Habit:	crystals equant to prismatic; drusy, forming in cavities
Crystal system:	monoclinic

Zn₂(Fe,Mn)(PO₄)₂.4H₂O



Vivianite was named after J.H. Vivian (1785–1855), the mineralogist who discovered it in St Austell (Cornwall, England). As the pigment blue ochre, it was used in Cologne in the thirteenth and fourteenth centuries, and possibly by the ancient Greeks. It is a common secondary mineral formed by the weathering of metallic ore deposits, and as an alteration product in pegmatites. Associated minerals are siderite, sphalerite and pyrite. It can replace organic matter, forming thin coatings on the damaged surfaces of fossilizing mammoth tusks or skulls (Mexico), whale bones (Richmond, Virginia, USA) or inside bivalves (Kerch iron deposits, Crimea, Russia). It also forms in lake sediments, bog iron ores and peat bogs.

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Colour:	colourless when fresh; darkens on exposure to blue
Lustre; opacity:	vitreous, greasy or pearly; transparent to translucent
Streak:	colourless to bluish-white
Hardness:	1.5-2
Specific gravity:	2.7
Cleavage; fracture:	perfect, thin cleavage plates are flexible; fibrous
Habit:	crystals prismatic to flattened, may appear bent, acicular (needle- like) or fibrous; concretionary, earthy and encrusting masses
Crystal system:	monoclinic
crystar system.	monocimic

Fe₃(PO₄)₂.8H₂O

Erythrite



Erythrite forms a series with annabergite, in which nickel and cobalt substitute for one another. A rare secondary mineral, it typically forms by weathering in the oxidized zone of ore deposits, as at Cobalt (Ontario, Canada). Associated minerals include silver, cobaltite and skutterudite. Its bright colour is a valuable indicator of the presence of cobalt-bearing ore deposits, hence the alternative name cobalt bloom. When heated, the powdered mineral turns lavender blue and smells of garlic. It was developed in 1803 as a natural pale to medium violet pigment, light cobalt violet, but is rarely used now because of its toxicity. Striking specimens come from Daniel Mine (Saxony, Germany) and Bou Azzer (Morocco).

Co₃(AsO₄)₂.8H₂O

Colour:	violet red, light pink, purple red; grey surface alteration
Lustre; opacity:	vitreous, pearly on cleavages; translucent to transparent
Streak:	pinkish red
Hardness:	1.5-2.5
Specific gravity:	3.1–3.2
Cleavage; fracture:	perfect, small, flexible plates; uneven
Habit:	well-formed crystals rare; mostly as crusts or small reniform
	aggregates
Crystal system:	monoclinic

Annabergite



Annabergite is the nickel end member of the series annabergite-erythrite, in which nickel and cobalt substitute for one another. It was named in 1832 after Annaberg in Saxony. A rare secondary mineral, annabergite is also known as nickel bloom because it forms crusts or films through the surface alteration of other nickel minerals, sometimes completely replacing the original mineral. Like erythrite, it has served as a good indicator of the presence of ore, although rarely being of economic importance itself. It may also be generated during mining as part of the milling process. It occurs in Annaberg (Saxony, Germany), Cobalt (Ontario, Canada), Humboldt County (Nevada, USA) and Allemont (France).

3 4.2.	-2-
Colour:	shades of green, white, grey; may be pink
Lustre; opacity:	vitreous, pearly, dull; translucent to opaque
Streak:	light green, paler than its colour
Hardness:	1.5–2.5
Specific gravity:	3.1
Cleavage; Fracture:	perfect; uneven
Habit:	prismatic, striated crystals, small and rare; more usually as earthy crusts or powdery masses
Crystal system:	monoclinic

Ni₃(AsO₄)₂.8H₂O

Cacoxenite



Cacoxenite is a rare secondary mineral formed through the alteration of other phosphates, and associated with hematite, wavellite, strengite and iron oxides. It is named from the Greek *kakos* ('wrong') and *xenos* ('guest') because the phosphorus content of cacoxenite lessened the quality of smelted iron. It is very attractive, forming golden fibrous and radiating crystals within spherical aggregates. It is also found as inclusions in quartz, particularly amethyst, which detracts from the purple of the amethyst and dulls its appearance. Cacoxenite is found at Diamond Hill Quartz Mine (South Carolina), at Ober-Rosbach (Hesse, Germany), in Indian Mountain (Alabama) and at Avant's Claim (Arkansas, USA).

	J. 2
Colour:	yellow, ochre yellow, golden or brown
Lustre; opacity:	vitreous, silky, greasy; translucent to transparent
Streak:	straw yellow
Hardness:	3-3.5
Specific gravity:	2.3
Cleavage; fracture:	poor; fibrous, brittle
Habit:	acicular (needle-like) crystals; as spherical aggregates, internally radiating and giving a stellar appearance
Crystal system:	hexagonal

Fe₄(PO₄)₃(OH₃).12H₂O

Wavellite



Wavellite is named after the English physician W. Wavell, who discovered the mineral in 1805. It forms small globular masses that exhibit a characteristic internally radiating structure of acicular crystals. A common secondary mineral, it forms as a low-temperature hydrothermal alteration product in fissures of aluminium-rich rocks, and also in pegmatites with phosphates. At one locality in Arkansas (USA), bluish-green spherulites form the cement of a phosphoritic breccia. Associated minerals are limonite, hematite, pyrolusite. Wavellite is also found in Dünsberg and Waldgermes (Germany), Cerhovice (Czech Republic), in tin veins in St Austell (Cornwall, England) and Llallagua (Bolivia).

Colour:	white, yellow, greenish, brown, bluish
Lustre; opacity:	vitreous, silky, pearly; translucent
Streak:	white
Hardness:	3.5-4
Specific gravity:	2.3–2.4
Cleavage; fracture:	three good cleavages; uneven to subconchoidal
Habit:	very rare as good prismatic crystals; usually flat to spherical, radiating green to yellow-green fibrous clusters
Crystal system:	orthorhombic

Al₃(PO₄)₂(OH)₃.5H₂O

Turquoise

Turquoise is a rare and valuable ornamental stone. Its name comes from the French *pierre turquoise*, meaning 'Turkish stone', having been brought from Iran to Europe via Turkey. It has been prized for thousands of years, as evidenced from its use in an Egyptian tomb of c. 3000 BC and in the burial mask of Tutankhamun. A secondary mineral, it forms in the alteration zone of hydrothermal porphyry copper deposits, and in veins and pockets through the alteration of volcanic rocks and phosphate-rich sedimentary rocks. It normally occurs in arid regions. The most important source of turquoise is Iran, where it has been mined for 2000 years, many being worked around Nishapur.

Colour:	sky blue to pale blue, greenish blue, green
Lustre; opacity:	vitreous, waxy, greasy or dull; transparent to opaque
Streak:	white or pale greenish
Hardness:	5-6
Specific gravity:	2.6-2.8
Cleavage; fracture:	good; conchoidal to even fracture
Habit:	crystals rare, small, prismatic; usually massive in veins concretions or encrustations
Crystal system:	triclinic

CuAl₆(PO₄)₄(OH)₈.4H₂O

Chalcophyllite



halcophyllite is appealing to collectors because of its attractive colour and high Justre, and because it sometimes forms attractive six-sided crystals arranged into a rosette. A rare secondary mineral, it is found in the zone of oxidation of arsenic-bearing hydrothermal copper deposits. Associated minerals include azurite, chrysocolla, malachite, cuprite, spangolite, limonite and clinoclase. It occurs at Wheal Phoenix (Cornwall, England), Clara Mine (Wolfach, Germany), Bisbee (Arizona, USA) and Nizhni Tagil (Russia). It has also been found in vugs within slags derived from ancient to relatively recent copper refining at Val Varenna (Genova, Italy), which were uncovered in 1993 after severe flooding.

(Cu,Al) ₃ (AsO ₄ ,SO ₄)(OH) ₄ .6H ₂ O	
Colour:	shades of green to blue
Lustre; opacity:	adamantine to vitreous, pearly; transparent to translucent
Streak:	pale green to bluish green
Hardness:	2
Specific gravity:	2.68 (light for copper minerals)
Cleavage; fracture:	perfect, basal; uneven
Habit:	crystals platy, may be striated, in rosettes; foliated, tabular or massive
Crystal system:	trigonal

Liroconite



Liroconite is named from the Greek *liros* ('pale') and *konia* ('powder'). An Lextremely rare secondary mineral, it forms in the oxidized zone of copper deposits, associated with other copper minerals such as clinoclase, chalcophyllite, azurite, malachite and limonite. Liroconite was discovered in the 1780s or 1790s in the old copper mines of Cornwall (England), such as Wheal Gorland. The largest known crystal is 35mm (1½in) long, and is in the Rashleigh Collection held at the Royal Cornwall Museum at Truro. Lesser quality specimens occur in Sayda (near Freiburg, Germany), Herrengrund (Slovakia), Cerro Gordo Mine (California, USA), the Khovu-Aksy deposit (Tuva, Russia), N'Kana (Zambia) and Zaire.

Colour:	sky blue, turquoise blue, verdigris green
Lustre; opacity:	vitreous to resinous; transparent to translucent
Streak:	pale blue, paler than colour
Hardness:	2–2.5
Specific gravity:	2.9–3.0
Cleavage; fracture:	poor; uneven to subconchoidal, slightly curving surfaces
Habit:	well-formed crystals wedge to lens-shaped, frequently appearing bent with rounded edges; also granular
Crystal system:	monoclinic

Cu₂Al(As,P)O₄(OH)₄.4H₂O

Lavendulan



Lavendulan, named after its lavender colour, is a rare secondary mineral formed in the oxidized zones of copper deposits. Associated minerals include erythrite, cuprite, malachite, covellite, brochantite and olivenite. World-class specimens of lavendulan were collected from the important former lead-silver-zinc mining district of Mazarron-Aguilas (Murcia, Spain) in 1992. Although copper mining was never significant, the area is renowned for copper arsenates and associated secondary minerals. Specimens include electric blue bladed crystals up to 4mm (%in), and radial groups up to 10mm (%in). Other localities include Jachymov (Czech Republic), Bou Azzer (Morrocco) and Wheal Owles (Cornwall, England).

CalvaCu5(ASC	Canacu ₅ (AsO ₄) ₄ CI.5H ₂ O	
Colour:	blue, electric blue, greenish blue or lavender	
Lustre; opacity:	vitreous, waxy or satiny in aggregates; translucent	
Streak:	light blue	
Hardness:	2.5	
Specific gravity:	3.54–3.59	
Cleavage; fracture:	good; uneven	
Habit:	crystals rare, generally less than 3mm, twinning common; radiating fibres or rosettes, botryoidal crusts	
Crystal system:	orthorhombic	

CaNaCu₅(AsO₄)₄Cl.5H₂O

Autunite



A utunite is a hydrated calcium uranium phosphate, which loses water over time to form meta-autunite. Autunite is strongly radioactive and fluoresces a striking yellowish green. It forms through the alteration of primary uranium minerals such as uraninite, in pegmatites, hydrothermal veins and in the weathering zone of granite. Associated minerals include torbernite $(Cu(UO_2)_2(PO_4)_2.8-12H_2O)$, which has similar chemistry and properties, although not forming a series with autunite. It is an important uranium ore, widely used in World War II. Named after Autun in France, it also occurs at Hagendorf (Germany), Shinkolobwe (Zaire), Minas Gerais (Brazil), Mt Spokane (Washington, USA) and Colorado (USA).

Colour:	shades of yellow to greenish yellow
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	pale yellow
Hardness:	2-2.5
Specific gravity:	3.1–3.2
Cleavage; fracture:	perfect, thin cleavage sheets are flexible; uneven
Habit:	crystals tabular, may be twinned; crusts, scaly aggregates or earthy masses
Crystal system:	tetragonal

Ca(UO₂)₂(PO₄)₂.10-12H₂O



Carnotite is a hydrated potassium uranium vanadate, the water content and hence specific gravity of which varies with temperature and humidity. It is strongly radioactive, but does not fluoresce. It forms through the alteration of minerals such as uraninite, occurring as an impregnation in sands and sandstones, associated with old channels or near playas. It sometimes occurs in small masses associated with vegetable matter, such as petrified trees, or is disseminated throughout, colouring the rock bright yellow. Used as an ore of vanadium and uranium, carnotite occurs in Colorado's desert regions, Utah's San Rafael Swell, Monument Valley (Arizona), New Mexico (USA), the Fergana desert (Russia) and Radium Hill (South Australia).

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Colour:	canary yellow to greenish yellow
Lustre; opacity:	pearly to dull; semi-opaque to transparent
Streak:	light yellow
Hardness:	very soft, possibly 2
Specific gravity:	4.7-4.9 **
Cleavage; fracture:	perfect, basal
Habit:	crystals flat, pseudo-hexagonal, rare; powdered, earthy aggregates
Crystal system:	monoclinic

K₂(UO₂)₂(VO₄)₂.3H₂O

Willemite



Willemite is a zinc ore that became important after the discovery of huge amounts in marbles at Franklin and Sterling Hill (New Jersey, USA); it was found later in Belgium and named in honour of King Willem I of the Netherlands. Specimens from Franklin show a bright green fluorescence, contrasting beautifully with the red fluorescence of associated calcite. Willemite is a secondary mineral found in zinc-bearing limestones, associated with calcite, franklinite and zincite; specimens displaying all four minerals are popular. Other localities include Tiger (Arizona, USA), Tsumeb (Namibia), Altenberg (Belgium), Mont St Hillaire (Quebec, Canada) and Beltana (South Australia).

bluish
ssive aggregates
1

Forsterite



Forsterite is a member of the forsterite-fayalite series (Mg₂SiO₄-Fe₂SiO₄), which comprises an important group of rock-forming minerals called olivines. Olivine is common in basic and in ultrabasic igneous rocks. Forsterite and quartz do not occur together, as they react to form enstatite (MgSiO₃). Forsterite is found in metamorphosed siliceous dolomites; fayalite, stable in the presence of quartz, is found in quartz-bearing rocks. Nickel-rich forsterites occur in stony meteorites. Forsterite is found on Bheinn-an-Dubhaich (Skye, Scotland), Vesuvius (Italy), Sapat (Pakistan) and Kovdor massif (Kola Peninsula, Russia). Fayalite occurs in Yellowstone Park (Wyoming, USA) and the Mourne Mountains (Ireland).

Mg ₂ SiO ₄	
Colour:	white or yellow (forsterite), brown or black (fayalite)
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	6.5 (forsterite); 7 (fayalite)
Specific gravity:	3.2 (forsterite); 4.4 (fayalite)
Cleavage; fracture:	poor; conchoidal, brittle
Habit:	well-formed crystals rare; granular masses, isolated grains in rock
Crystal system:	orthorhombic

Olivine



Olivine, like mica or serpentine, is not an officially recognized mineral name, but is widely used by petrologists and mineralogists. The gem variety of olivine is called peridot, having a distinctive greasy lustre and an olive- or bottle-green colour. The best peridot stones have about 15 per cent fayalite content to give sufficient green colour, plus possibly a little nickel and/or chromium to provide the best hues. Peridot was used in the Middle Ages to decorate church robes and plates. Gem-quality peridot is found at St John's Island (Zabargad) in the Red Sea, Eiffel (Germany), Mogok (Myanmar), San Carlos (Arizona, USA), Sapat (Pakistan) and Vesuvius (Italy). Olivine can make up half the mass of a stony meteorite.

(Mg,Fe)₂SiO₄ Colour: green, especially olive-coloured Lustre; opacity: vitreous; transparent to translucent Streak: white Hardness: 6.5 - 7Specific gravity: 3.2-4.4 Cleavage; fracture: poor; conchoidal, brittle Habit: well-formed crystals rare; granular masses, isolated grains in igneous rocks Crystal system: orthorhombic

Monticellite



Monticellite is a calcium-rich member of the olivine group, and an end member of the monticellite-kirschsteinite series (CaMgSiO₄-CaFeSiO₄). It also forms a series with glaucochroite (CaMnSiO₄). Consequently, iron and manganese are ubiquitous impurities. It was named after the Italian mineralogist Teodoro Monticelli (1759-1845). Monticellite is formed in contact metamorphic zones between limestones and gabbros, or between granites and dolomites. Associated minerals include gehlenite, spinel, calcite, vesuvianite and apatite. It is found at Vesuvius (Italy), Isle of Muck (Scotland), Magnet Cove (Arkansas, USA), Kovdor (Kola Peninsula, Russia) and Isle Cadieux (Quebec, Canada).

colourless, greenish grey, grey
vitreous; transparent
white
5.5
3.2
poor; subconchoidal to uneven
crystals rare as well-formed prisms; massive, granular
orthorhombic

Garnet: Pyrope



Pyrope, alongside almandine and spessartine, belongs to the pyralspite group of garnets. It is one of the less common garnets, widely used as a gem and noted for its clear red crystals. It forms a series with almandine; a member of which is the red-lavender coloured gemstone rhodolite. Unusual for a garnet, pyrope occurs mostly in igneous rocks, such as eclogites and kimberlites. These are formed at high pressures and can contain diamonds. It also occurs in related serpentinites and placer deposits. It is found at Merunice (Czech Republic), Dora-Meira (Piedmont, Italy), Kimberly (South Africa), the Umba River (Tanzania), Cowee Creek (North Carolina, USA), Buell Park (Arizona, USA) and Bingara (NSW, Australia).

Mg₃Al₂(SiO₄)₃

Calarin	
Colour:	red-orange to deep red
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	7–7.5
Specific gravity:	3.58
Cleavage; fracture:	absent; conchoidal
Habit:	crystals typically rhombic dodecahedra, sometimes
	trapezohedra; massive, granular
Crystal system:	cubic

Garnet: Almandine



Imandine, a pyralspite garnet, is darker red than pyrope and sometimes may appear black. Rare transparent specimens are facetted as gemstones, having a high lustre. Specimens having inclusions such as rutile needles are often cut en cabochon. Like many garnets in metamorphic rocks, their equant, well-formed crystals often stand out as porphyroblasts against a schistose matrix. It is usually found in medium-grade metamorphic rocks; less often in granites and pegmatites; and also as placer deposits. Fine orange-red crystals are found in sands in Minas Novas and Minas Gerais (Brazil) and in Sri Lanka; other locations include Sticken River (Alaska), Adirondacks (New York, USA) and the Zillertal (Austria).

Fe ₃ Al ₂ (SiO ₄) ₃	
Colour:	red-violet to red-brown
Lustre; opacity:	vitreous to resinous; transparent to opaque
Streak:	white
Hardness:	7.5
Specific gravity:	4.4
Cleavage; fracture:	absent; conchoidal
Habit:	crystals typically rhombic dodecahedra, sometimes trapezohedra, embedded as separate crystals in metamorphic rocks; massive
Crystal system:	cubic

Garnet: Spessartine



Spessartine, a pyralspite garnet, is rarely found at gem quality. Forming a series with almandine, it usually contains varying amounts of manganese and iron. The crystal shapes of spessartine – i.e. the 12-sided dodecahdron with rhomb-shaped faces or the 24-sided trapezohedron with trapezium-shaped faces – are ubiquitous in the garnet group. These round-looking crystals are characteristic of garnets. Spessartine occurs in rhyolites, granites and pegmatites; in metasomatized manganous rocks; and as placer deposits. It occurs in Seriphos (Greece), the Spessart Mountains (Germany), Minas Gerais (Brazil), Lieper's Quarry (Pennsylvania, USA), Amelia (Virginia, USA) and Tsilaizina and Anjanabonoina (Madagascar).

Mn₃Al₂(SiO₄)₃

Colour:	yellow, orange, red, brown, black
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	7–7.5
Specific gravity:	4.2
Cleavage; fracture:	absent; conchoidal
Habit:	crystals as the typical rhombic dodecahedra, but more commonly 24-sided trapezohedra; massive, granular
Crystal system:	cubic

Garnet: Grossular



rossular, alongside uvarovite and andradite, belongs to the ugrandite group of J garnets, giving gem-quality specimens in many colours. Massive green grossular, such as Transvaal jade, is popular, often with black inclusions of magnetite. Some has a distinctive gooseberry colour, hence the name derived from the scientific name R. grossularia. Cinnamon orange-red transparent garnets called hessonite were used as gems by the ancient Greeks and Romans. Tsavorite is a transparent green variety. It occurs in metamophosed calcareous rocks and as placer deposits. Fine crystals occur at Chernyshevsk (Russia), Ala Valley (Italy), Asbestos (Canada), Ramona (California, USA), Maharitra (Madagascar) and Telemarken (Norway).

$Ca_3Al_2(SiO_4)_3$	
Colour:	colourless when pure, green, brown, orange, pink, red, black
Lustre; opacity:	vitreous; transparent
Streak:	white
Hardness:	6.5-7.5
Specific gravity:	3.59
Cleavage; fracture:	none; subconchoidal
Habit:	crystals as rhombohedra or trapezohedra; granular, compact, massive
Crystal system:	cubic

Garnet: Uvarovite



Uvarovite, a ugrandite garnet, is emerald-green and much sought-after as a gemstone because of its outstanding brilliance and colour. Uvarovite is mostly found as small crystals, so much jewellery uses small druses. The best clear specimens are found in the Urals, lining cavities and fissures. Uvarovite is named after Count Sergei Semenovitch Uvarov (1765–1855), a Russian statesman and amateur mineral collector. It forms a series with grossular and often contains aluminium. It occurs in chromium-rich serpentinites. Fine crystals are found in Yerkaterinberg and the Saranovskii Mine (Urals, Russia), Outokumpu (Finland), Pico do Posets (Pyrenees, Spain), Quebec (Canada) and the Kop Mountains (Turkey).

Ca₃Cr₂(SiO₄)₃ Colour: bright green to dark green Lustre; opacity: vitreous to adamantine; transparent to translucent Streak: white Hardness: 6.5-7 Specific gravity: 34-38 Cleavage; fracture: absent; conchoidal Habit: crystals as rhombic dodecahedra or trapezohedra; massive, granular Crystal system: cubic

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Garnet: Andradite



Andradite, a ugrandite garnet, has many varieties, the most prized being the chromium-rich, emerald-green demantoid. This has a higher dispersion than diamond and is characterized by inclusions of asbestos fibres known as 'horsetails'. A yellow variety is called topazolite and the black variety melanite. It is named after the Brazilian mineralogist J.B. d'Andrada e Silva (1763–1838). Andradite is found in contact-metamorphosed limestones; schists and serpentinites; and in silica-poor igneous rocks. It occurs in Elba and Livorno (Italy), Arendal (Norway) and Franklin (New Jersey, USA); it is found as dementoid in the Urals among gold-bearing sands, and as melanite at Vesuvius and Lazio (Italy).

Ca₃Fe₂(SiO₄)₃

Colour:	yellow, green, brown, black
Lustre; opacity:	vitreous; transparent to opaque
Streak:	white
Hardness:	6.5-7
Specific gravity:	3.7-4.1
Cleavage; fracture:	absent; conchoidal
Habit:	crystals as rhombic dodecahedra and trapezahedra; massive, crusts showing many rhombic faces
Crystal system:	cubic

Zircon



Zircon has a similar lustre and fire to that of diamond, with colourless zircons having been used as imitation diamonds. Zircon comes in a variety of colours; the yellow variety alluded to in the name comes from the Arabic *zargun* for 'gold colour'. Heat treatment often changes the colour of zircons. Zircon that is not of gem quality is a major ore for zirconium, as well as hafnium and thorium. It occurs as an accessory mineral in acid igneous rocks, their products of metamorphism, and as placer deposits. Excellent crystals are found in river deposits at Matura (Sri Lanka) and the Ilmen Mountains (Russia). Other localities are Renfrew (Canada), Mt Ampanobe (Madagascar) and Teete (Mozambique).

ZrSiO ₄	
Colour:	colourless, yellow, red, brown, grey, green
Lustre; opacity:	vitreous to subadamantine; transparent to opaque
Streak:	white
Hardness:	6.5–7.5
Specific gravity:	3.9–4.8
Cleavage; fracture:	indistinct; conchoidal
Habit:	crystals dipyramidal or prismatic; irregular granules
Crystal system:	tetragonal

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Sillimanite



Sillimanite is a polymorph of aluminium silicate, alongside kyanite and andalusite. The appearance of one of these minerals in metamorphic rocks is a key indicator of the temperatures and pressures undergone by the rock. The presence of sillimanite indicates a high temperature of formation (at least 650°C/1202°F). The slim, prismatic crystals of sillimanite distinguish it, and may be raised on weathered rocks. It is found in metamorphosed pelitic rocks and in pegmatites associated with corundum, tourmaline and topaz. Blue and violet transparent sillimanite is found in Mogok (Myanmar); grey-green varieties are found in Sri Lanka; and other locations are Minas Gerais (Brazil), Maldan (Czech Republic) and Freiberg (Germany).

Al ₂ SiO ₅	
Colour:	grey, brown, påle green
Lustre; opacity:	vitreous to subadamantine; translucent to transparent
Streak:	white
Hardness:	6.5-7.5
Specific gravity:	3.2
Cleavage; fracture:	perfect; uneven
Habit:	crystals long, slender prisms, occasionally acicular, poorly terminated; often silky, fibrous aggregates (fibrolite)
Crystal system:	orthorhombic

Andalusite



Andalusite is a polymorph of aluminium silicate, alongside kyanite and Asillimanite. Its presence in a rock indicates that the rock has undergone low pressures during metamorphism (i.e. has been formed within a few kilometres of the Earth's surface). Greenish-red transparent crystals are used as gems. Twinned crystals called chiastolite form cross shapes (+ shapes) with x-shaped dark lines at the centre. Chiastolite has been used in amulets as a religious symbol. Andalusite occurs in metamophosed pelites, especially contact-metamorphosed slates, and in pegmatites. Large crystals are found at Lisenz (Austria); chiastolite is found in schists at Santiago de Compostela (Spain) and Keiva (Kola Peninsula, Russia).

Al₂SiO₅

Colour:	white, pink, pearl-grey, green, brown
Lustre; opacity:	vitreous, greasy; transparent to opaque
Streak:	white
Hardness:	7.5
Specific gravity:	3.1-3.2
Cleavage; fracture:	good; uneven
Habit:	crystals as square, stubby prisms, often twinned; granular, rod- like aggregates
Crystal system:	orthorhombic

Kyanite



Kyanite is a polymorph alongside andalusite and sillimanite. The presence of kyanite in rocks indicates that they have undergone moderate temperatures and medium to high pressures during metamorphism. Kyanite is named after the Greek kyanos for blue; it is also known as disthene, meaning 'double strength', after the hardness, which is greater across the crystal than lengthwise. Kyanite occurs in pelltic schists and gneisses, associated with garnet, staurolite and micas; and in eclogites. Large blue crystals come from Minas Gerais (Brazil) and Pizzo Forno (Switzerland); green crystals up to 30cm (12in) from Machakos (Kenya); and grey radiating crystals from Bolzano (Italy), the Tyrol (Austria) and Morbihan (France).

Al ₂ SiO ₅	
Colour:	often blue, lighter towards the margins; colourless, grey, green
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	4-4.5 along cleavage planes, 6-7 across cleavage planes
Specific gravity:	3.6-3.7
Cleavage; fracture:	perfect
Habit:	flat, bladed crystals in schists and gneisses, rosettes in quartz; massive aggregates
Crystal system:	triclinic
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Topaz



Topaz has been a highly prized gemstone for millenia. It is found in a variety of colours and is often heat-treated and/or irradiated to give different hues; blue topaz resembling aquamarine is produced thus from colourless stones. Yellow topaz from Brazil turns pink on heating to 300°–450°C (572°–842°F). Crystals in pegmatites can be huge (up to 300kg/661lb). The Brazilian Princess, a pale blue topaz from Teofilo Otoni, is the largest cut topaz, weighing 4266g (9%1b). Topaz is found in pneumatolytic veins in granites and pegmatites, and as a placer deposit. The largest crystals are found at Minas Gerais (Brazil) and Albashka (Siberia, Russia); it is also found in Mino Province (Japan), at Pikes Peak (Colorado, USA) and on Elba (Italy).

Al₂SiO₄(F,OH)₂

Colour:	colourless, yellow, pink, blue, green
Lustre; opacity:	vitreous; transparent to translucent
Streak:	opacity
Hardness:	8
Specific gravity:	3.5–3.6
Cleavage; fracture:	perfect; conchoidal, uneven
Habit:	vertically striated prismatic crystals; columnar, granular
Crystal system:	orthorhombic

Staurolite



Staurolite is a common mineral in pelitic schists and gneisses, occurring often as large porphyroblasts. Its presence is useful when assessing the degree of metamorphism, indicating that the rock has undergone medium temperature conditions. As a gem mineral, the opaque cross-shaped twins characteristic of staurolite are used in jewellery, often as amulets or religious items. The crosses occur at angles of both 90° and 60°. Fine crystals up to about 12cm (5in) are found in Pizzo Forno and Alpe Piona (Switzerland), Finistère and Morbihan (France), Mt Greiner (Austria), Keivy massif (Kola Peninsula, Russia), Franconia (New Hampshire), Blue Ridge (Georgia) and Taos County (New Mexico, USA).

Colour:	reddish-brown, brown, black
Lustre; opacity:	vitreous, resinous, dull; translucent to opaque
Streak:	white
Hardness:	7–7.5
Specific gravity:	3.65–3.83
Cleavage; fracture:	poor; uneven to conchoidal
Habit:	crystals prismatic or tabular, pseudohexagonal truncated diamond shapes, often twinned giving cross shapes
Crystal system:	monoclinic

(Fe,Mg)₂(Al,Fe)₉(Si₄O₂₀)(O,OH)₂

Titanite



Titanite is now the more common name for sphene. Titanite displays interesting colours and strong fire, and is often faceted for display. Impurities such as iron and aluminium are always present, and rare earths such as cerium and yttrium are commonly present. Titanite is a common accessory mineral in intermediate and felsic plutonic rocks, pegmatites and veins, and is also found in gneisses and schists. Large crystals occur at St Gotthard (Switzerland), Bridgewater (Pennsylvania, USA), the Urals (Russia), Kola Peninsula (Russia), Minas Gerais (Brazil), Renfrew (Canada). Gem-quality crystals can be found at Pino Solo and La Huerta (Baja California Norte, Mexico).

CaTiSiO₅

Colour:	yellow, green, red, reddish brown, brown, black
Lustre; opacity:	vitreous to subadamantine; transparent to opaque
Streak:	white
Hardness:	5-5.5
Specific gravity:	3.4–3.6
Cleavage; fracture:	distinct; conchoidal, brittle
Habit:	crystals usually flattened, prismatic, wedge-shaped; compact or lamellar masses, disseminated grains
Crystal system:	monoclinic

Chloritoid



Chloritoid is found as small, platy crystals or occasionally as porphyroblasts in metamorphic rocks. Two related minerals are the magnesium-rich sismondine and the manganese-rich otrellite. Chloritoid indicates that the rock formed under low to medium pressures and temperatures. It shows a strong pleochroism, dark green to light green or yellow in varying orientations. Chloritoid occurs in metamorphic pelites, especially schists and marbles, associated with muscovite, staurolite, garnet, chlorite, kyanite and quartz. It is found at Kosoi Brod (Urals, Russia), Svalbard (Norway) and Natick (Rhode Island, USA). Sismondine is found at Zermatt (Switzerland) and Pregatten (Austria); otrellite at Ottré (Belgium).

(10,1011)2/(14)	2010(011)4
Colour:	dark grey, greenish-grey
Lustre; opacity:	vitreous, pearly
Streak:	white, pale grey, pale green
Hardness:	6.5
Specific gravity:	3.56–3.61
Cleavage; fracture:	perfect; brittle
Habit:	pseudohexagonal tabular crystals; commonly compact, foliated aggregates, massive
Crystal system:	monoclinic or triclinic (but always pseudohexagonal)

(Fe,Mn)₂Al₄Si₂O₁₀(OH)₄

Datolite

Datolite is an ore of boron when available in sufficient quantities. It forms complex crystals that can superficially look like cubic forms such as dodecahedra or trapazohedra, but on closer inspection are revealed as lacking in symmetry. Datolite occurs as a secondary mineral in basalts, serpentinites and in hydrothermal deposits. It is often found in vesicles, associated with zeolites, prehnite and calcite. Datolite can be found in Kratzenberg (Austria), Serra dei Zanchetti and Alpe de Siusi (Italy), St Andreasberg (Harz, Germany), Arendal (Norway), Lane (Massachusetts, USA) and Prospect Park (New Jersey, USA).

CaBSiO₄(OH)

- 1	
Colour:	colourless, white, sometimes light green or yellow
Lustre; opacity:	vitreous, greasy; transparent to translucent
Streak:	white
Hardness:	5–5.5
Specific gravity:	2.9–3
Cleavage; fracture:	imperfect; irregular, subconchoidal
Habit:	crystals as plates or prisms, sometimes large and complex; granular to compact, crusty aggregates with radiating fibres called botryolites
Crystal system:	monoclinic

Gadolinite



G adolinite has varying amounts of other rare-earth elements such as cerium, Ianthanum and neodymium substituting for yttrium; it also contains varying amounts of thorium and uranium, making it radioactive, and the crystals often metamict. Gadolinite is an ore of thorium and the rare-earth elements, which are finding an increasing variety of uses. It is named after the Finnish chemist Johan Gadolin (1760–1852), who discovered yttrium. Gadolinite occurs in granite and syenite pegmatites, and in veins in metamorphic rocks. Crystals (up to 500kg/ 1100lb) occur in Hitterö, Iveland and Hundholmen (Norway). Other localities include Finbo and Ytterby (Sweden), Llano County (Texas, USA) and Novara (Italy).

Be₂FeY₂Si₂O₁₀ Colour: green, greenish-brown, black Lustre; opacity: vitreous; transparent to opaque Streak grey-green Hardness: 6.5 - 7Specific gravity: 4.0-4.5 Cleavage; fracture: none; conchoidal, splintery Habit: rare prismatic crystals of diamond-shape cross-section; microgranular, earthy masses **Crystal system:** monoclinic

Euclase



Euclase forms extremely attractive hard crystals; blue and green varieties are popular. The name is derived from the Greek *eu* for 'well' and *klasis* for 'break', alluding to its facile cleavage. The distinctive crystal shapes can be confused only with barite or celestite, minerals which do not occur in the same environment as euclase. Euclase is formed as an alteration product of beryl in pegmatites and veins, and may be found as a placer deposit. The best crystals are found at Ouro Prêto (Minas Gerais, Brazil) as colourless, blue and green gem-quality stones. Other locations include the Sanarka River (Russia), Park County (Colorado, USA) and Las Cruces (Chivor, Colombia).

Colour:	colourless, white, green, blue
	colouriess, write, green, blue
Lustre; opacity:	vitreous to adamantine; transparent to translucent
Streak:	white
Hardness:	7.5
Specific gravity:	3.1
Cleavage; fracture:	perfect; conchoidal
Habit:	stubby prismatic crystals with non-symmetrical slanted terminations; reniform, stalactitic, mammilliary masses
Crystal system:	monoclinic

AlBeOHSiO₄

Humite



H umite is part of a rare related group of minerals (called the humite group), which also includes norbergite, chondrodite, clinohumite and titanclinohumite. The structure of humites comprises alternating layers of brucite (Mg(OH)₂) and olivine (Mg₂SiO₄). It was named after Sir Abraham Hume (1749–1838), English collector of art, gems and minerals. Humite occurs in hydrothermal veins and contact-metamorphosed limestones and dolomites. Associated minerals include magnetite, diopside, spinel, biotite, serpentine, olivine and calcite. It is found at Vesuvius (Italy), Los Llanos de Januar (Spain), Pargas (Finland), Varmland (Sweden), Franklin (New Jersey, USA) and the Tilly Foster mine (New York, USA).

Colour: often yellow or yellow-green; white, brown, orange Lustre; opacity: vitreous; transparent to translucent Streak: white Hardness: 6 Specific gravity: 3.2–3.3 Cleavage; fracture: poor; conchoidal Habit: small prismatic crystals; granular, grains embedded in matrix Crystal system: orthorhombic

Mg7(OH,F)2(SiO4)3

Braunite



Draunite, named after K. Braun (1790–1872), Adviser of the Chambers, of Gotha D(Germany), is sometimes referred to as braunite-I to differentiate it from the calcium-bearing variant braunite-II (Ca(Mn,Fe)14Si₂O₂₄), Crystals are usually small and rarely well formed, but they can have an attractive metallic grey sheen. Braunite is formed by metamorphism of manganese-rich silicates and by weathering. Associated minerals are calcite, quartz, pyrolusite, hausmannite, rhodonite and spessartine. It is found at Friedrichroda (Thuringia, Germany), Långban and Jacobsberg (Sweden), Sitapar (India), Postmasburg (South Africa), Val d'Aosta (Italy), Batesville (Arkansas, USA) and Mason County (Texas, USA).

black, brownish-black, greyish-black Colour: Lustre: opacity: metallic, greasy; opaque Streak: brownish-black Hardness: 6-6.5 Specific gravity: 4.7-4.8 Cleavage; fracture: perfect; uneven; subconchoidal Habit: crystals pseudooctahedral and bipyramidal up to 5cm (2in), rare; granular or massive aggregates Crystal system: tetragonal

Mn²⁺Mn³⁺₆SiO₁₂

Dumortierite

Dumortierite is a borosilicate second only in abundance to tourmaline. It forms attractive violet to blue masses used for items such as cabochons, beads and sculptures. It was named after Eugène Dumortier (1802–1873), a French palaeontologist. Dumortierite can be mistaken for sodalite, the latter being less dense and usually including more white minerals. Dumortierite is found in aluminium-rich metamorphic rocks, pegmatites and contact metamorphosed rocks. Large deposits are found in Oreana and Rochester (Nevada, USA), Dehesa (California, USA), Arizona (USA); it is also found at the type locality Chaponost (Rhône-Alps, France) and at Minas Gerais (Brazil), Sondria (Italy), Mogra (India) and Madagascar.

1,1003,10102	AI/1003/0104/303	
Colour:	blue or violet crystals, sometimes reddish-brown	
Lustre; opacity:	silky, vitreous; translucent	
Streak:	white	
Hardness:	7–8.5	
Specific gravity:	3.2–3.4	
Cleavage; fracture:	good; uneven	
Habit:	rare prismatic or acicular needles; columnar or fibrous, radiating aggregates	
Crystal system:	orthorhombic	

Al₇(BO₃)(SiO₄)₃O₃

Gehlenite-Åkermanite



The gehlenite-åkermanite group is a solid solution series where magnesium and silicon replace aluminium; melilite is a term applied to intermediate minerals. Specimens can be attractive, varying from transparent white or reddish to opaque grey-green, especially as micromounts. Gehlenite and åkermanite are found in contact-metamorphosed limestones and dolomites, respectively. Melilite occurs in silica-deficient lavas, associated with nepheline or leucite. Åkermanite is found on Vesuvius (Italy) and often in blast furnace slag; gehlenite in Trento (Italy) and Oravita (Banat, Romania); melilite at Mount Monzoni and Canzocoli (Italy), Isle of Muck and Ardnamurchan (Scotland), and Belerberg (Eliffel, Germany).

Ca2A125107-	
Colour:	white, grey, yellow, grey-green, brownish-red
Lustre; opacity:	vitreous, greasy; transparent to opaque
Streak:	white
Hardness:	5
Specific gravity:	2.95–3.05
Cleavage; fracture:	good; conchoidal, uneven
Habit:	crystals as small, stubby, square prisms or plates; granular and massive aggregates
Crystal system:	tetragonal

Ca2Al2SiO7 - Ca2MgSi2O7

Ilvaite



Ivaite, named from Ilva, the old Latin name for Elba, is a blackish, almost opaque mineral. It forms attractive large black crystals and fibrous radiating aggregates, often with calcite, hedenbergite, magnetite, andradite and pyrite. It melts readily in a flame, leaving a magnetic residue. Ilvaite appears translucent on freshly exposed surfaces, but almost opaque on old specimens. It forms in contact metasomatic rocks and metamorphosed iron-rich limestones, and less commonly in syenites. It is found at Livorno and Elba (Italy), Seriphos (Greece), Julianahaab (Greenland), Dal'negorsk (Primorskiy Kray, Russia), Laxey (Idaho, USA), Dragoon Mountains (Arizona, USA), Kamioka (Oita, Japan) and Thyrill (Iceland).

cur c3(5104)2(
Colour:	brownish-black, black
Lustre; opacity:	submetallic, vitreous, greasy; semi-opaque
Streak:	black, greeny-black
Hardness:	5.5-6
Specific gravity:	3.8-4.1
Cleavage; fracture:	good; conchoidal
Habit:	crystals as striated prisms; radiating fibrous, massive, granular
Crystal system:	orthorhombic

CaFe₃(SiO₄)₂(OH)

Hemimorphite



Hemimorphite is a common mineral in all lead, zinc and silver deposits. It forms part of the altered top of sphalerite deposits called 'gossan', or 'iron cap'. It is named from the hemimorphism displayed by its crystals; the prisms usually have different terminations (ends), one being rather blunt and the other being pyramidal. It has long been mined alongside smithsonite as a zinc ore. It forms in the oxidized zones of zinc-bearing veins below 240°C (464°F), above which willemite is formed. It is found at Chihuahua (Mexico), Moresnet (Belgium), Cumberland and Derbyshire (England), Carinthia (Austria), Nerchinsk (Siberia, Russia), Franklin (New Jersey) and Granby (Missouri, USA) and Baita (Romania).

	2
Colour:	colourless, white, tinged yellow, blue or green, grey, brown
Lustre; opacity:	vitreous, pearly, silky; transparent to translucent
Streak:	white
Hardness:	5
Specific gravity:	4.0
Cleavage; fracture:	perfect; conchoidal, splintery
Habit:	crystals hemimorphic prismatic or plates; stalactitic, encrustations, massive granular, fan-shaped aggregates
Crystal system:	orthorhombic

Zn₄Si₂O₇(OH)₂.H₂O

Clinozoisite



Clinozoisite forms a series with the iron-bearing mineral epidote, but tends to be paler. Attractive specimens are transparent rodlike crystals, clusters of randomly orientated prisms and pinkish radiating fibrous aggregates. Clinozoisite occurs in low- to medium-grade metamorphic rocks, in contact-metamorphosed calcium-rich sediments and as altered calcium-rich plagioclase. Associated minerals include amphiboles, plagioclase and quartz. It is found at Goslarwand (Tirol, Austria), Camaderry Mountain (Wicklow, Ireland), Amborompotsy (Madagascar), Chiampernotto (Turin, Italy), Belvidere Mountain (Vermont, USA), Spade Spring Canyon (California, USA), Nightingale (Nevada, USA) and Baja California (Mexico).

Colour:	grey, yellow, greenish, light rose
colour.	grey, yenow, greensn, nghcrose
Lustre; opacity:	vitreous; transparent to opaque
Streak:	white
Hardness:	6.5
Specific gravity:	3.2
Cleavage; fracture:	perfect; uneven
Habit:	prismatic crystals; granular, massive, fibrous aggregates
Crystal system:	monoclinic

Ca2Al3Si3O12(OH)

Epidote



Epidote is noted for its characteristic green colour, sometimes described as 'pistachio'. It is a common secondary mineral in a number of rocks. Epidote is pleochroic, displaying greens, yellows and browns. Rock composed mostly of epidote may be polished or tumbled as unakite. Epidote occurs in low- to mediumgrade metamorphic rocks of mafic composition. Associated minerals depend on the degree of metamorphism and the bulk composition of the rock. It is found at Bourg d'Otsans (France), Arendal (Norway), Traversella (Piedmont, Italy), Knappenwand (Austria), Sulzer (Prince of Wates Island, Alaska, USA), Seven Devils (Idaho, USA) and San Quentin (Baja California, Mexico).

Colour:	dark green to yellow green
Lustre; opacity:	vitreous; translucent
Streak:	grey
Hardness:	6.7
Specific gravity:	3.3-3.5
Cleavage; fracture:	perfect; uneven
Habit:	crystals as columnar prisms, finely striated lengthwise; granular, massive, fibrous aggregates
Crystal system:	monoclinic

Ca2(Al,Fe)3Si3O12(OH)



Piedmontite or piedmontite is a member of the epidote group. Named after the Piedmont area in northwest Italy, it is also referred to as manganiferous epidote. The manganese content gives piemontite a red coloration, which can be difficult to distinguish from red varieties of clinozoisite. Piemontite is widespread and found in low- to medium-grade metamorphic rocks, metasomatized manganese deposits and hydrothermal veins. It is associated with epidote, glaucophane, quartz, orthoclase and calcite. It is found at Saint Marcel (Piedmont, Italy), Ceres (Turin, Italy), Groix (France), Shikoku (Japan), Old Bookoomata (South Australia), Garnet Lake (California, USA), Tucson Mountains (Arizona, USA) and Tachgagalt (Morocco).

Colour:	red, purple, reddish brown, reddish black
Lustre; opacity:	vitreous; translucent to nearly opaque
Streak:	red
Hardness:	6
Specific gravity:	3.4–3.5
Cleavage; fracture:	perfect; splintery
Habit:	blocky, equant, prismatic bladed or acicular crystals; massive, granular
Crystal system:	monoclinic

Ca2(Al,Mn,Fe)3Si3O12(OH)

Allanite



A llanite is a member of the epidote group, calcium being replaced by rare earth elements and thorium. When thorium-rich, allanite can be sufficiently radioactive to be metamict. Allanite–(Ce), –(La) and –(Y) are minerals particularly rich in cerium, lanthanum or yttrium. It is named after the Scottish mineralogist Thomas Allan (1777–1833), its discoverer. It occurs as an accessory in granites and in pegmatites; and rarely in schists, gneisses and contact-metamorphic limestones. Associated minerals are epidote, muscovite and fluorite. Allanite is found at Qáqassuatsiaq (Aluk Island, Greenland), Ytterby and Finbo (Sweden), Miask (Urals, Russia), Franklin (New Jersey, USA) and Barringer Hill (Texas, USA).

(Ce,Ca,Y)₂(Al,Fe)₃Si₃O₁₂(OH)

vitreous, resinous, submetallic; translucent to opaque
virieous, resinous, submetanic, transacent to opaque
grey
5.5–6
3.5-4.2
imperfect; conchoidal, uneven
tabular, prismatic or acicular crystals; massive, granular
monoclinic

Zoisite



Zoisite is an epidote noted for its gem and ornamental use. The violet-blue variety Tanzanite is very popular, having more fire than tourmaline or peridot. It shows a distinct purple to blue to slate-grey pleochroism and can appear more violet in incandescent light. Massive green zoisite containing rubies is a popular rock among collectors and can be polished or carved. A massive manganous pink variety called thulite is also polished and carved. Zoisite occurs in high-grade metamorphic rocks, in hydrothermal veins and as altered calcic plagioclase. Excellent tanzanite is found at Merelani Hills (Letatina Mountains, Tanzania); thulite is found at Telemark (Norway) and in Tennessee and South Carolina (USA).

Colour:	white, blue, pale green, pink, violet-blue
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	6–6.5
Specific gravity:	3.15-3.36
Cleavage; fracture:	perfect; uneven
Habit:	finely striated, elongated, prismatic crystals, usually poorly terminated; grains, granular masses, aggregates
Crystal system:	orthorhombic

Ca2Al3Si3O12(OH)

Vesuvianite



lso known as idocrase, vesuvianite is a popular mineral that forms fine transparent crystals and is often confused with other gemstones. It is commonly cut for collections, but not for wearing. Varieties include green californite or california jade, blue cyprine, yellow-green xanthite and pale-green or whitish wiluite. Vesuvianite is formed by metamorphism of limestones and metasomatism of serpentinized ultrabasic rocks. Associated minerals are grossular, and radite, wollastonite and diopside. Vesuvianite is found at Vesuvius (Italy), California (USA). Morelos and Chiapas (Mexico), Zermatt (Switzerland), Arendal (Norway), the Akhmatovsk mine (Urals, Russia) and Chernyshevsk (Yakutia, Russia).

$Ca_{10}Mg_2AI_4(SiO_4)_5(Si_2O_7)_2(OH)_4$	
Colour:	yellow, green, brown, colourless, white, blue, violet, red , black
Lustre; opacity:	vitreous to resinous; transparent to translucent
Streak:	white
Hardness:	6.5
Specific gravity:	3.27–3.45
Cleavage; fracture:	poor; subconchoidal to irregular
Habit:	stubby, prismatic crystals, rare pyramidal terminations; compact granular masses
Crystal system:	tetragonal

Benitoite



Benitoite is a rare mineral, until recently known only from Diablo Range (San Benito, California). Discovered in 1906, it was mistaken for sapphire. Benitoite shows a high dispersion, similar to diamond, and is strongly pleochroic (i.e. it displays different colours in different orientations), changing from blue to colourless. It gives a pale blue fluorescence under ultraviolet light. Clusters of blue benitoite and black-red neptunite on a matrix of white natrolite are appealing and rare specimens. Benitoite is formed in hydrothermal veins cutting serpentinites associated with natrolite, albite, neptunite and joaquinite. Other localities include Magnet Cove (Arkansas, USA), Ohmi (Niigata, Japan) and Broken Hill (NSW, Australia).

BaTiSi₃O₉

Colour:	sapphire-blue, white, colourless
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	6–6.5
Specific gravity:	3.65
Cleavage; fracture:	poor; conchoidal
Habit:	crystals stubby, prismatic, dipyramidal
Crystal system:	hexagonal

Axinite

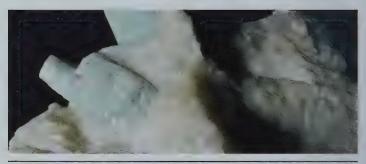


A xinite is a complex mineral in the ferroaxinite-manganaxinite series. Iron-rich axinite is brown to black whereas manganese-rich specimens are yelloworange. Magnesioaxinite is a blue-grey magnesium-rich analogue and tinzenite is a low-calcium variety. Axinite is named after the sharp axe- or spearhead-shaped crystals. Axinite occurs in cavities in granites and adjacent zones, associated with diopside, andradite, quartz, calcite, scheelite and prehnite. It is found at Bourg d'Oisans (France), St Just (Cornwall, England), Obira (Japan) and Luning and Pala (California, USA); manganaxinite is found at Franklin (New Jersey, USA), Tinzon (Switzerland) and Liguria (Italy); and ferroaxinite at Baveno (Italy).

20. 20. 1. 1. 1.	23(
Colour:	yellow to brown, violet, grey
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	6.5-7
Specific gravity:	3.25
Cleavage; fracture:	perfect; conchoidal
Habit:	sharp-edged crystals, varying shape; granular and platy masses
Crystal system:	triclinic

Ca2(Fe,Mn)Al2BO3(OH)Si4O12

Beryl



Beryl is famous for forming a wide variety of gemstones. It has been known since ancient times, the Greeks naming it after the colour of the sea, and druids believing it to aid psychic powers. Beryl is also of great importance as the main ore of beryllium, one of the lightest metals known. Beryl is formed in granites and pegmatites, where crystals can grow to enormous sizes (6m, 18 tonnes). It is also found in metamorphic rocks, hydrothermal veins and placer deposits. The availability varies enormously; non-precious varieties are easily obtained whereas emerald is so expensive that even small crystals are much sought-after. It is mined in Brazil, USA, Madagascar, Germany, Czech Republic, Russia and India.

201201018	
Colour:	colourless, white, blue, green to yellows, rose, peach, red
Lustre; opacity:	vitreous, resinous; transparent to translucent
Streak:	white
Hardness:	7.5–8
Specific gravity:	2.63–2.97
Cleavage; fracture:	imperfect; conchoidal
Habit: 、	hexagonal prismatic crystals, often without clear terminations; rarely in druses or compact masses
Crystal system:	hexagonal

Be₃Al₂Si₆O₁₈

Beryl: Gem varieties



Emerald is the most famous and long-prized variety of beryl, having a beautiful green colour, derived from the presence of chromium and vanadium. It is rarely faultless, making good specimens very expensive, and stones are often oiled to disguise faults. The best emeralds come from Muso and Chivor (Colombia). Aquamarine is usually sky-blue to dark-blue, but in the nineteenth century a sea-green variety was favoured (the name meaning 'sea water'). A dark blue variety comes from Madagascar, often enhanced by heat treatment. The best-quality yellow to golden yellow heliodor comes from the Urals (Russia). Often associated with the Sun, heliodor is found only rarely as gem-quality material.

Be ₃ Al ₂ Si ₆ O ₁₈	
Colour:	colourless, white, blue, green to yellows, rose, peach, red
Lustre; opacity:	vitreous, resinous; transparent to translucent
Streak:	white
Hardness:	7.5–8
Specific gravity :	2.63–2.97
Cleavage; fracture:	imperfect; conchoidal
Habit:	hexagonal prismatic crystals, often without clear terminations; rarely in druses or compact masses
Crystal system:	hexagonal

Cordierite



Cordierite is a common mineral in metamorphic rocks, frequently occurring as porphyroblasts. The most interesting variety is the transparent violet-blue stone often known as iolite or water-sapphire. A black iron-rich variety called sekaninaite occurs only in Dolni Bory (Moravia, Czech Republic). Cordierite is named after the French geologist Pierre Cordier (1777–1861), who first described it, although it had been used as a gem long before this. Associated minerals include sillimanite, potassium feldspar, muscovite, biotite and andalusite. It is found at Bodenmais (Germany), Orijärvi (Finland), Kragerö (Norway), Bity (Madagascar), Tamil Nadu (India) and Thompson (Manitoba, Canada).

Mg2Al4Si5O18

Colour:	grey, rarely blue
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	7
Specific gravity:	2,6–2.66
Cleavage; fracture:	fair; subconchoidal
Habit:	stubby, pseudohexagonal, twinned crystals; granular, compact, massive
Crystal system:	orthorhombic

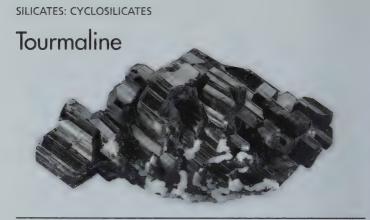
Dioptase



Dioptase is a beautiful deep green mineral, mistaken for emerald when first discovered in the eighteenth century. The strong green colour can hide the fire and transparency of some crystals. The crystal faces are usually very reflective and sparkling in clusters. The name comes from the Greek for 'see through', after clearly visible cleavages in crystals. Dioptase is found in the oxidized zones of copper deposits, associated with chrysocolla, malachite, wulfenite, cerrusite and quartz. It was first found at Altin-Tyube (Kirghiz Steppe, Kazakhstan); other locations include Tsumeb (Namibia), Copiapo and Atacama (Chile), Mindouli (Congo), Mammoth Mine (Tiger, Arizona, USA) and Baita (Romania).

CuSiO₂(OH)₂

A	
Colour:	emerald-green, blue-green
Lustre; opacity:	vitreous; transparent to translucent
Streak:	green
Hardness:	5
Specific gravity:	3.3
Cleavage; fracture:	perfect; conchoidal to uneven
Habit:	crystals short, six-sided prisms with rhombohedral ends; granular, massive
Crystal system:	hexagonal



Tourmalines are a group of aluminoborosilicates, notably elbaite (lithium-rich), dravite (magnesium-rich), schorl (iron- and manganese-rich) and uvite (iron- and magnesium-rich). Tourmalines crystallize as characteristic long prisms of triangular cross-section. All are strongly piezoelectric and pyroelectric, finding use in highpressure gauges. Tourmalines are found in granites, pegmatites and quartz veins, and as an accessory in schists and gneisses. Elbaite occurs in Elba (Italy), the Urals (Russia), Sri Lanka, at Pala and Ramona (California, USA), and Newry (Maine, USA); uvite is found at Franklin (New Jersey, USA) and Gouverneur (New York, USA); and dravite at Yinniethara (Australia) and New York (USA).

	160 18(01)/4
Colour:	colourless, yellow, blue, olive-green, brown, black
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	7-7.5
Specific gravity:	3.01–3.26
Cleavage; fracture:	poor; uneven to conchoida
Habit:	prismatic crystals, often striated and elongated
Crystal system:	trigonal

NaAl₉(BO₃)₃Si₆O₁₈(OH)₄

Tourmaline: Gem Varieties



Of the tourmaline gem varieties, dravite is usually brown, with strong dichroism. Rubellite is a pink or red variety. Fibrous rubellite is cut *en cabochon*, for a cat's-eye effect. Watermelon tourmaline is coloured pink and green, resembling contrasting flesh and rind of the watermelon. Tourmaline may vary throughout the crystal, with up to 10 or more colours or shades. Achroite is a rare colourless variety of elbaite; it is easier to cut than other tourmalines, the strong dichroism of which dictates the direction of cuts. Indicolite is a rare deep blue variety; paraiba tourmaline is deep blue to bluish green. The black opaque iron-rich variety schorl can be found as crystals up to several metres long. Verdelith is a yellow-green.

Colour: colourless, yellow, blue, olive-green, brown, black Lustre; opacity: vitreous; transparent to translucent Streak: white Hardness: 7–7.5 Specific gravity: 3.01–3.26 Cleavage; fracture: por; uneven to conchoida Habit: prismatic crystals, often striated and elongated Crystal system: trigonal

NaAl₉(BO₃)₃Si₆O₁₈(OH)₄

Milarite



Milarite is a rare mineral and the end member of the milarite–osumilite group of minerals. Osumilite contains magnesium and iron, but no beryllium. Crystals of milarite are rarely transparent and usually small, but are excellent as micromounts. Crystals are often well-shaped hexagonal prisms in muted shades of green to yellow. It has been called glufite, but milarite is now the preferred name. Milarite is found in hydrothermal veins, classically in Val Guif and Val Striem (Grissons, Switzerland), as crystals up to 3cm (1¼in); it is also found on the Kola Peninsula (Russia) and in Valencia Guanajuato (Mexico) and Jaguaraçú (Minas Gerais, Brazil). Osumilite is found on the volcano Sakurajima near Osumi (Japan).

Colour:	colourless, green to yellow
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	5.5–6
Specific gravity:	2.4–2.6
Cleavage; fracture:	absent; conchoidal, uneven
Habit:	hexagonal prismatic crystals; radial-fibrous aggregates and intergrowths
Crystal system:	hexagonal

K₂Ca₄Be₄Al₂Si₂₄O₆₀.H₂O



Eudialite is a rare mineral popular for the attractive colours of some specimens. Crystals are commonly red and usually found embedded in a matrix of host rock. The site on the Kola peninsula is well known for pegmatites rich in sodium and some rarer elements giving rise to unusual minerals such as eudialite. The name comes from the Greek, alluding to its ready solubility in acids. Eudialite occurs in nepheline syenites, granites and associated pegmatites, associated with quartz, albite, nepheline, aegerine and natrolite. Notable localties are the Kangerdluarssuk Plateau (Greenland), Magnet Cove (Arkansas, USA), Pajarito Mountain (New Mexico, USA) and the Lovozero and Khibiny massifs (Kola Peninsula, Russia).

(140, Cu, 1 C)621	(01, 0), 0, 0, 0, 0, 0, 0, 0, 0, 0, 0, 0, 0, 0,
Colour:	pink, red, yellow-brown, violet
Lustre; opacity:	vitreous; translucent
Streak:	white
Hardness:	5-5.5
Specific gravity:	2.8–3.0
Cleavage; fracture:	imperfect; conchoidal, uneven, splintery
Habit:	crystals as plates, rarely well formed; granular, massive aggregates
Crystal system:	hexagonal

(Na,Ca,Fe)₆Zr(OH,Cl)(Si₃O₉)₂

Hedenbergite



The pyroxenes are an important group of rock-forming minerals. Their structure is characterized by straight chains of linked silicon oxide (SiO_4) tetrahedra. They can be distinguished from the amphiboles by the angle between their two good cleavages (about 90° rather than 120°). Hedenbergite and diopside $(CaMgSi_2O_6)$ are monoclinic clinopyroxenes, forming a calcium-bearing series in which magnesium and iron substitute for one another. Hedenburgite is common in iron-magnesium skarns at the contact of granitic rocks with limestones, and in contact-metamorphosed iron-rich sediments. Good hedenburgite crystals have been found in the Skardu area (Pakistan), Broken Hill (NSW, Australia) and the Harstig Mine (Varmland, Sweden).

ck; almost opaque
slucent to opaque
ns at about 90°; uneven to subconchoidal
tic; radiating aggregates, granular, massive

Diopside



Diopside is a calcium-bearing clinopyroxene that forms a solid solution series with hedenbergite (CaFeSi₂O₆). Pure diopside is common in contactmetamorphic siliceous magnesian limestones, associated with calcite, dolomite and sometimes forsterite, as on Skye (Scotland). There are several varieties of diopside: chrome diopside is a chromium-rich gem variety found in Burma, Siberia (Russia), Pakistan and South Africa. Some specimens have inclusions (probably of rutile) that may form a 'cat's eye' effect when polished. Violane is a rare blue manganese-bearing variety from Italy. One dark green to black variety from southern India, known as star diopside, shows a four-rayed star when cut *en cabochon*.

CaMgSi₂O₆

Colour:	pale to dark green or black, may be colourless
Lustre; opacity:	vitreous or dull; transparent to opaque
Streak:	white, grey, grey-green
Hardness:	5.5–6.5
Specific gravity:	3.29
Cleavage; fracture:	good in two directions at about 90°; uneven to subconchoidal
Habit:	prismatic crystals; granular, columnar, massive
Crystal system:	monoclinic

Jadeite



J adeite is the characteristic pyroxene of high-pressure metamorphic rocks, in which it may be associated with glaucophane, aragonite and quartz, and also of eclogites, along with garnet. It occurs in the high-pressure metamorphic belt in California (USA) and at Shibukawa (Japan). Jadeite is used as an ornamental stone for carving, and as a precious stone in jewellery. Of the two minerals known as jade, (the other being nephrite, an amphibole), it is less common. The emerald green chromium-bearing variety is most prized, and known as imperial jade. For more than two centuries, China has been supplied with such jade from Burma, whereas the Central American Indians used jadeite from Guatemala.

0
shades of light and dark green; rarely white or violet
subvitreous, pearly on cleavages; translucent
white
6-7
3.2-3.4
good in two directions at about 90°; splintery
crystals rare, prismatic; commonly massive, granular, compact felty masses
monoclinic

Na(Al,Fe)Si₂O₆

Spodumene



The pyroxene spodumene was named from the Greek for 'ash', in reference to its colour. Spodumene comes in other colours: gem varieties include the lilac pink kunzite, coloured by manganese, and bright emerald green hiddenite, coloured by chromium. Well-cut stones display strong pleochroism from colourless to two shades of body colour. Crystals can weigh up to 65 tonnes (72 tons). Spodumene is mined as a raw material for lithium compounds and ceramics. It is commonly found in lithium-rich granitic pegmatites and also occurs in gneisses. Associated minerals include quartz, albite, petalite, lepidolite and beryl. Notable localities include the Black Hills (South Dakota, USA) and Sterling and Chesterfield (Massachussetts, USA).

LiAlSi2O6

~ ~	
Colour:	commonly yellowish-grey, but shows a range of colours
Lustre; opacity:	vitreous, dull; transparent to translucent
Streak:	white
Hardness:	6.5–7
Specific gravity:	3.1–3.2
Cleavage; fracture:	good in two directions at about 90°; uneven to subconchoidal
Habit:	crystals prismatic, may be striated; commonly massive
Crystal system:	monoclinic

Aegirine



First described from Norway, aegirine is named after Aegir, a Scandinavian god of the sea. It is also known as acmite, from the Greek for 'point', in reference to its unusually sharply pointed crystals. This helps to distinguish it from augite, which is very similar. Aegerine occurs in alkaline rocks such as sodium-rich nepheline syenites, carbonatites and pegmatites, and rarely in regionally metamorphosed schists, gneisses and iron formations. Associated minerals include analcime, nepheline, clinochlore, eudialyte and rhodochrosite. Good crystals occur at Malosa (Zomba District, Malawi), Mont Saint-Hilaire (Quebec, Canada), Khibiny Massif (Kola Peninsula, Russia) and Magnet Cove, Arkansas (USA).

NaFeSi ₂ O ₆	
Colour:	dark green, greenish-black, reddish-brown
Lustre; opacity:	vitreous to resinous; translucent to opaque
Streak:	yellowish-grey
Hardness:	6
Specific gravity:	3.5 .
Cleavage; fracture:	good in two directions at about 90°; uneven, brittle
Habit:	acicular crystals terminated by a steep pyramid, striated; disseminated grains, fibrous, in radiating aggregates
Crystal system:	monoclinic

Augite

A ugite is a calcium-bearing clinopyroxene, similar to the diopside-hedenbergite structure. The commonest pyroxene, it can be distinguished from hornblende by the angle between the cleavages. It is an important rock-forming mineral in basic igneous rocks (basalt and gabbro) and some syenite, and in ultrabasic rocks (pyroxenites and peridotites). It also forms in high-temperature metamorphic rocks such as pyroxene gneisses and pyroxene granulites. It is commonly associated with plagioclase feldspars, amphiboles, olivine and orthopyroxenes. Rare well-formed crystals occur at Vesuvius and Stromboli (Italy), and at Eifel (Germany).

dark green, brown, black, rarely cream
vitreous, resinous to dull; transparent to opaque
greyish-green
5.5-6
3.2–3.6
good in two directions at about 90°; uneven to subconchoidal
short prismatic crystals of four- or eight-sided cross-section, often twinned; compact, granular
monoclinic

(Ca,Na)(Mg,Fe,Al,Ti)(Si,Al)₂O₆

Enstatite

Enstatite and ferrosilite (Fe₂Si₂O₆) form a series in which iron and magnesium Substitute for each other. The intermediate composition, MgFeSi₂O₆, is called hypersthene. Minerals of this series are calcium-poor and are mostly orthorhombic. These orthopyroxenes are important rock-forming minerals, and occur worldwide in basic and ultrabasic igneous rocks, and high-temperature metamorphic rocks such as pyroxene gneisses. Enstatite is also found in meteorites. Crystals are quite rare, but rolled pebbles are sometimes faceted as gems. The variety chrome-enstatite is emerald green; a dark brown six-rayed star enstatite occurs in Mysore (India); and unique gem-quality colourless crystals are found at Embilipitiya (Sri Lanka).

Mg ₂ Si ₂ O ₆	
Colour:	pale to dark brownish-green, white, greyish, yellowish
Lustre; opacity:	vitreous; transparent to opaque
Streak:	white to greyish
Hardness:	5-6
Specific gravity:	3.2–3.4
Cleavage; fracture:	good in two directions at about 90°; uneven
Habit:	crystals stumpy, may be twinned; as grains, lamellar or massive
Crystal system:	orthorhombic

Bronzite



The orthopyroxene bronzite is characterized by a bronze sheen on its cleavage surface; however, it is now regarded as an iron-bearing enstatite, and is no longer recognized as a separate species. It is affected by partial alteration of a type called 'schillerization', which gives rise to the bronzelike submetallic lustre. Bronzite is sometimes cut and polished for ornaments, and may have a fibrous structure. It occurs in the basic igneous rock norite in the Fichtelgebirge (Germany) and in the serpentine of Kraubat (Styria, Austria). Relict crystals of more highly altered bronzite or enstatite occurring in serpentinite are known as bastite, after the locality Baste (Harz, Germany); it also occurs in the Lizard (Cornwall, England).

(Mg,Fe)₂Si₂O₆

Colour:	grey-green or bronze-brown
Lustre; opacity:	vitreous, pearly, bronze sheen on cleavage; translucent to opaque
Streak:	white to greyish
Hardness:	5-6
Specific gravity:	3.2-3.3
Cleavage; fracture:	good in two directions at about 90°; scaly; uneven
Habit:	coarse prismatic crystals; massive, granular
Crystal system:	orthorhombic

Cummingtonite



Cummingtonite belongs to the amphibole group, whose structure is characterized by double chains of silicon oxide (SiO₄) tetrahedra. It is in the middle compositional range of a magnesium-iron series, the end members of which are magnesiocummingtonite and grunerite. Cummingtonite cannot be distinguished from anthophyllite, its polymorph, with the naked eye. It forms in medium-grade regionally metamorphosed rocks. The only commercial source of brown asbestiform amosite was the Precambrian ironstone formations of Transvaal (South Africa). Grunerite is found in the iron deposits of Lake Superior (USA); cummingtonite is found at Cummington (Massachusetts, USA) and Val d'Ossola (Italy).

Colour:	dark green, pale to dark brown depending on iron content
Lustre; opacity:	vitreous to silky; translucent to transparent
Streak:	white
Hardness:	5-6
Specific gravity:	3.1–3.6 (increasing with iron content)
Cleavage; fracture:	two good cleavages at about 120°; splintery
Habit:	individual crystals rare; aggregates of rodlike or fibrous crystals, often radiating
Crystal system:	monoclinic

(Mg, Fe)₇Si₈O₂₂(OH)₂

Glaucophane



G laucophane is named after the Greek *glaucos* ('blue') and *fanos* ('appearing'). It is an alkali-bearing amphibole which, like riebeckite, has a characteristic bluegrey colour, and typical 120° prismatic amphibole cleavage. It characteristically forms in subduction zones under high-pressure, low-temperature conditions, and may eventually be exposed at the surface. Glaucophane-rich schists are known as blueschists. Glaucophane may be associated with the high-pressure minerals lawsonite, jadeite and aragonite, and also with garnet, epidote or pumpellyite. Glaucophane occurs in the Franciscan belt of California (USA), in Shikoku (Japan), Anglesey (Wales), Zermatt (Switzerland) and Euboea Island (Greece).

grey-blue to lavender blue
vitreous, silky in fibrous varieties; translucent
blue-grey
6
3.08–3.22
good in two directions at about 120°; uneven to subconchoidal
good crystals rare, slender prismatic to acicular; sometimes massive, fibrous, columnar or granular
monoclinic

Na2(Mg,Fe)3Al2Si8O22(OH)2

Riebeckite



Named after the German traveller E. Riebeck, riebeckite is a blue-grey alkaliamphibole formed in alkali granites, syenites, more rarely in granite pegmatites and in some volcanic rocks. The asbestiform variety crocidolite is found in iron formations and was formerly mined in South Africa and Australia. The associated health risk means that specimens should be professionally sealed in boxes or bags. Associated minerals include nepheline, albite, aegerine, tremolite, magnetite, hematite or siderite. Crocidolite inclusions in quartz, and their pseudomorphing by quartz, have given rise to the gems tiger's eye and hawk's eye. The type locality for riebeckite is Socotra Island (Indian Ocean, Yemen).

N8022(01)2
dark blue to black
vitreous or silky; translucent
none determined
5
3.32-3.38
two good cleavages at about 120°; conchoidal to uneven
long, prismatic and striated crystals; massive, fibrous or asbestiform
monoclinic

Na2(Mg,Fe)5Si8O22(OH)2

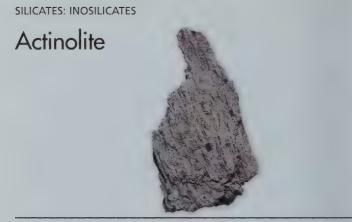
Tremolite



Tremolite is a calcic amphibole that forms a series with the iron-bearing ferroactinolite. Pure tremolite is creamy white, but grades to green with increasing iron content. It may fluoresce yellow or pink. It forms through the contact metamorphism of calcium- and magnesium-bearing siliceous sediments, impure dolomitic limestones, or ultramafic rocks. It can be associated with calcite, dolomite, garnet, wollastonite, talc, diopside or forsterite. A fibrous variety of tremolite has been mined for use as asbestos. It is named after Tremola (Italy). Other notable localities include St Marcel (Piedmont, Italy), Franklin (New Jersey, USA), Wilberforce (Ontario, Canda) and Brumada Mine (Bahia, Brazil).

Cu21119551802	2011/2
Colour:	colourless, white, grey, can be green, pink or brown
Lustre; opacity:	vitreous or silky to dull; transparent to translucent
Streak:	white
Hardness:	5–6
Specific gravity:	2.9–3.2
Cleavage; fracture:	two good cleavages at about 120°; uneven to subconchoidal
Habit:	crystals long, bladed, often twinned; columnar, fibrous, plumose aggregates, radiating or granular
Crystal system:	monoclinic

Ca2Mg5Si8O22(OH)2



ctinolite is a fairly common calcic amphibole within the series tremolite-Aferroactinolite. It occurs in relatively low-grade metamorphosed mafic, ultramafic or magnesian carbonate rocks, and also in glaucophane-bearing blue schists. It is a common alteration product of primary pyroxenes in gabbro or dolerite, and used to be called uralite. Byssolite is an asbestiform type. The variety nephrite is the more common of the two types of green gemstone known as jade; it is tough, and takes a good polish. Beautiful nephrites from New Zealand are known as Maori stone because of their widespread use in traditional Maori art. Actinolite is found in the Zillertal (Austria), Val Malenco (Italy) and the Urals (Russia).

Ca ₂ (Mg,Fe) ₅ Si ₈ O ₂₂ (OH) ₂	
Colour:	bright to greyish green
Lustre; opacity:	vitreous, silky when fibrous; transparent to translucent
Streak:	white
Hardness:	5-6
Specific gravity:	3.0–3.24
Cleavage; fracture:	two good cleavages at about 120°; uneven or splintery
Habit:	long bladed or prismatic crystals, may be bent; columnar, radiating fibrous to asbestiform, granular, massive
Crystal system:	monoclinic



Hornblende is a calcic amphibole and a very important rock-forming mineral. It is widespread in igneous rocks (granodiorites, diorites, syenites and some gabbros) and in medium-temperature metamorphosed basalts (amphibolites) accompanied by plagloclase feldspar and sometimes garnet. It is occasionally found in metamorphosed impure dolomitic limestones and ironstones. It forms complex solid-solution series with several other amphiboles. The variety edenite is pale green, iron-poor and found at Edenville (New York, USA); pargasite is dark green, iron-rich and found in Pargas (Finland); hastingsite is richer in sodium and aluminium, and occurs in gabbros in Ontario (Canada).

(Ca,Na)2-3(Mg,Fe,Al)5(Al,Si)8022(OH)2

Colour:	black to dark green
Lustre; opacity:	vitreous to dull; opaque
Streak:	brown to grey
Hardness:	5-6
Specific gravity:	3.28–3.41
Cleavage; fracture:	two good cleavages at about 120°; uneven
Habit:	crystals long and thin, or short and stumpy, generally six-sided; granular, may form radiating aggregates
Crystal system:	monoclinic

silicates: inosilicates Anthophyllite



A nthophyllite is the orthorhombic polymorph of monoclinic cummingtonite, from which it can be distinguished only through optical, density or x-ray study. It occurs in medium-grade metamorphic rocks, derived from mafic or ultramafic igneous rocks, or dolomitic sedimentary rocks. It can form through the hydration of olivine. It is associated with talc, cordierite, chlorite, mica, hornblende or olivine. Anthophyllite has been commercially used in asbestos, with the associated health hazards. It is named from the Latin anthophyllum, meaning 'clove', in reference to its distinctive colour. Good crystals occur at Köngsberg (Norway); it is also found at Fahlan (Sweden), Orijarvi (Finland) and the Isle of Elba (Italy).

Colour:	grey-brown, cinnamon or clove brown, green
Lustre; opacity:	vitreous, silky when fibrous; translucent to translucent
Streak:	white
Hardness:	5.5–6
Specific gravity:	2.85–3.57
Cleavage; fracture:	two good cleavages at about 120°; uneven, brittle
Habit:	individual crystals rare; usually aggregates of prismatic crystals, fibrous, asbestiform
Crystal system:	orthorhombic

(Mg,Fe)₇Si₈O₂₂(OH)₂

Wollastonite



Wollastonite, formerly known as table spar, is found in thermally metamorphosed siliceous limestones at igneous contacts, or in regionally metamorphosed rocks, associated with calcium-rich garnets, diopside, epidote and tremolite. It also occurs in xenoliths in igneous rocks. There are many different polymorphs, generally called wollastonite. Although common, commercially viable deposits are unusual, occurring in New York (USA), Finland, Mexico, China, India and Africa. Wollastonite has many applications, e.g. in refractory ceramics, plastics, and sealants. It was named after the British mineralogist and chemist William H. Wollaston (1766–1828), who also discovered palladium and rhodium.

CaSiO₃

Colour:	white or greyish
Lustre; opacity:	vitreous, may be silky; subtransparent to translucent
Streak:	white
Hardness:	4.5–5
Specific gravity:	2.87-3.09
Cleavage; fracture:	perfect; splintery, uneven
Habit:	crystals tabular to short prismatic, twinning common; fibrous masses, radial, granular, compact
Crystal system:	triclinic (or monoclinic)

Pectolite



Pectolite must be handled carefully, as the fine white needles are sharp and easily puncture the skin and can become embedded. Some specimens are triboluminescent. A variety called larimar is a fine pale blue and green, and comes from the Dominican Republic. Pectolite occurs as a primary mineral in nepheline syenites, but is also formed by hydrothermal processes, filling cavities in basalts, associated with zeolites, datolite, prehnite and calcite. It is also found filling fractures in serpentinites and peridotites, and in contact-metamorphosed limestones. Notable occurrences include Lake County (California, USA), Franklin (New Jersey, USA), Bahamas, Mt Baldo (Trento Province, Italy) and England.

NaCa251308(OH)	
colourless, white or grey, larimar is pale blue	
vitreous to silky; translucent to opaque	
white	
4.5-5	
2.84–2.90	
perfect in two directions close to 90°; uneven, splintery	
crystals tabular or acicular; fibrous tufts or fibrous-radiating spherical aggregates or compact masses	
triclinic	

NaCa₂Si₃O₈(OH)

Rhodonite



Rhodonite has a similar structure to wollastonite and pectolite, built of chains of silicon oxide (SiO_4) tetrahedra with bases that do not lie in a plane. It has a distinctive pink colour and can be distinguished from rhodochrosite (which is also pink) because it is harder and does not dissolve in dilute hydrochloric acid. It often occurs in association with manganese ore deposits in hydrothermal veins, or in contact and regional metamorphosed manganese-bearing sedimentary rocks. Rhodonite from Sverdlovsk in the Urals (Russia) has been used as an ornamental stone. It also occurs at Långban (Sweden) in iron ore, at Broken Hill (NSW, Australia) and at Franklin and Sterling Hill (New Jersey, USA).

(Mn,Fe,Mg)SiO₃

	3
Colour:	deep pink, with brown or black surface oxidation
Lustre; opacity:	vitreous, pearly on cleavages; transparent to translucent
Streak:	white
Hardness:	5.5-6.5
Specific gravity:	3.57-3.76
Cleavage; fracture:	perfect in two directions close to 90°, with a third good cleavage; conchoidal to uneven
Habit:	crystals rare, tabular or prismatic; massive or granular
Crystal system:	triclinic

Babingtonite



 $\label{eq:basic} Babingtonite is a rare mineral named after the Irish physicist and mineralogist BW. Babington (1757–1833). Relatively recently discovered, it contains both ferrous iron (Fe²⁺) and ferric iron (Fe³⁺) (which takes the place of aluminium typical of many silicates), and is weakly magnetic. It tends to grow in cavities in mafic volcanic rocks and gneisses, which enables crystals to grow freely; it is also found in veins cross-cutting granite and diorite, and in skarns. It is commonly associated with prehnite, epidote, pyrite and quartz, and with zeolites. Notable occurrences include Poona (India), Devon (England), Herbornseelbach (Germany), Baveno (Italy), Yakubi Mine (Japan) and Massachusetts (USA), where it is the official mineral emblem.$

k to brownish-black opaque to translucent
opaque to translucent
ions at about 90°; uneven to subconchoidal
mnar, often striated; as platey, radial or jates

Ca₂(Fe,Mn)FeSi₅O₁₄(OH)

Neptunite



Neptunite is a rare mineral that forms as an accessory in intermediate plutonic igneous rocks such as nepheline syenite, and its pegmatites. It is also found in serpentinites associated with benitoite and natrolite. Its name is derived from Roman mythology, Neptune being the god of the sea, and alludes to its close association at its type locality with aegerine, named after Aegir, the Scandinavian sea god. Excellent crystals occur at San Benito (California, USA), where it is found in natrolite veins in a serpentinite body, along with blue benitoite; other localities include Barnavave (Ireland), Kola Peninsula (Russia), Igaliko (Greenland) and Mont St Hilaire (Ouebec, Canada).

MazikLi(10,1011	1/21/25/80/24
Colour:	black with deep reddish-brown internal reflections
Lustre; opacity:	vitreous to submetallic; opaque to translucent
Streak:	dark reddish-brown
Hardness:	5-6
Specific gravity:	3.19–3.23
Cleavage; fracture:	perfect; conchoidal
Habit:	elongate prismatic crystals of square cross-section and pointed terminations
Crystal system:	monoclínic

Na2KLi(Fe,Mn)2Ti2Si8O24

Bavenite



Bavenite is an extremely rare mineral, named in 1901 after Baveno (Italy) where it was discovered. It is formed by the alteration of beryl and other berylliumbearing minerals, and occurs as druses in cavities in granite and associated pegmatites. It is also found in hydrothermal veins and skarns. Associated minerals include orthoclase, fluorite, albite and beryl. It has a distinctive habit, with very thin tabular crystals forming diverging groups like the pages of a book. It can also form fibrous crystals, as in the granitic pegmatite bodies of Bustarviejo (Madrid, Spain). Other localities include Shap Pink Quarry (Cumbria, England), Mont St Hilaire (Quebec, Canada), Strzegom (Poland) and Londonderry (Australia).

4 2 2	12 3 20
Colour:	colourless, white, green, pink, brown
Lustre; opacity:	silky, vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	5.5–6
Specific gravity:	2.7 .
Cleavage; fracture:	perfect; uneven
Habit:	very thin tabular crystals, may be twinned; lamellar, rose-shaped aggregates, radiating tufts of needle-like crystals
Crystal system:	orthorhombic

Ca₄Al₂Be₂(OH)₂Si₉O₂₆

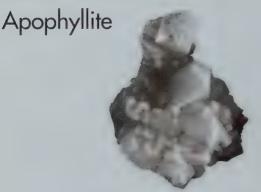
Prehnite



Prehnite is found in hollows in basaltic lavas, often associated with zeolites; it also occurs in very low grade metamorphic rocks, and through the decomposition of plagioclase feldspar. Associated minerals include datolite, epidote or calcite. It was the first mineral to be named after a person, the Dutchman Colonel Hendrick von Prehn (1733–1785), who discovered it at the Cape of Good Hope. It occasionally forms epimorphs (crystal growth over another mineral surface) on laumontite, which may subsequently dissolve, leaving the prehnite as a crust. Yellowish brown prehnite is occasionally cut *en cabochon*. Pale green masses are found in Scotland, and dark green or greenish-brown prehnite in Australia.

Colour:	pale watery or oily green, sometimes white or yellowish
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	6-6.5
Specific gravity:	2.8–2.9
Cleavage; fracture:	distinct; uneven, rather brittle
Habit:	crystals rare, prismatic, tabular or pyramidal; usually botryoidal or globular masses with a radiating structure
Crystal system:	orthorhombic

Ca2Al2Si3O10(OH)2



A pophyllite is the name given to a group of minerals including fluoroapophyllite and hydroxyapophyllite. Its name, from the Greek, roughly means 'to flake off' because it peels apart in leafs on heating as water is driven off. Apophyllite is quite abundant and popular, forming some attactive clear, colourless crystals and some of pale pastel shades; the most famous is an emerald-green variety from India. Apophylite occurs in cavities in basalts, associated with stilbite, scolecite, calcite, prehnite and analcime. Excellent crystals occur at Bolzano (Italy), Poona (India), Mont St Hilaire (Canada), St Andreasburg (Harz, Germany), Paterson (New Jersey, USA) and Rio Grande do Sul (Brazil).

Colour:	colourless, white, pale pink, green or yellow
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	4.5–5
Specific gravity:	2.33–2.37
Cleavage; fracture:	perfect; irregular
Habit:	tabular pseudocubic or pseudooctahedral crystals, sometimes dipyramidal or platy; aggregates
Crystal system:	tetragonal

KCa4Si8O20(F,OH).8H2O

Pyrophyllite



Pyrophyllite is named after the Greek for 'fire' and' leaf', after its behaviour on heating, when water is driven off and the mineral peels into flakes. Breaking into thin flakes like this is typical of phyllosilicates and reflects a sheetlike structure. Pyrophyllite has a greasy feel and is used as a dry lubricant like talc, from which it is almost indistinguishable. The variety agalmatolite has been used for carved ornaments in China. Pyrophyllite occurs in hydrothermal veins and schists, associated with kyanite, andalusite, topaz, mica and quartz. It is found in Orange County (North Carolina), Graves Mountain (Georgia) and La Paz County (Arkanas, USA), Ibitiara (Bahia, Brazil), Zermatt (Switzerland) and Krassik (Urals, Russia).

Al2Si4O10(OH)2

2-4-10	12
Colour:	yellowish-white, grey, green
Lustre; opacity:	pearly, dull; translucent to opaque
Streak:	white
Hardness:	1-1.5
Specific gravity:	2.8
Cleavage; fracture:	perfect; uneven or splintery
Habit:	lamellar or radiating foliated aggregates, granular, massive
Crystal system:	monoclinic

Talc



Talc, a member of the mica group, is noted for being soft, by definition 1 on Moh's scale. It is used in many industries, including paper, paints, personal care and roofing. Massive steatite, or soapstone (after its greasy feel), is easily carved or turned on a lathe. Talc is a good electrical and heat insulator, and repels water. The name comes from the Persian or Arabic *talq*. Talc occurs in schists and hydrothermally altered mafic rocks, associated with serpentine, actinolite, chlorite, vermiculite, dolomite and calcite. It is found at Mt Greiner (Austria), Zermatt (Switzerland), Pfitschal (Italy), Trimouns (Ariège, France), Onotosk (Siberia, Russia), Yellowstone mine (Montana, USA) and Delta (Pennsylvania, USA).

Mg₃Si₄O₁₀(OH)₂

Colour:	white, brown, green
Lustre; opacity:	greasy when massive, pearly on cleavage; translucent
Streak:	white
Hardness:	1
Specific gravity:	2.58–2.83
Cleavage; fracture:	perfect; uneven, lamellar
Habit:	very rare crystals as pseudohexagonal plates; usually scaly foliated aggregates
Crystal system:	monoclinic



Muscovite is the most common mica, well known for its cleavage into thin, Miflexible lamellae – a manifestation of its underlying two-dimensional sheetlike structure. The name comes from muscovy glass, after its former use in windows. Muscovite is used for electrical and heat insulation, as a dry lubricant and in paper, rubber, paints and plastics. The variety fuchsite is emerald green and popular among collectors. Muscovite is a very common rock-forming mineral in granites, pegmatites, phyllites, schists and gneisses. Crystals up to 30–50 m² (323–538 sq ft) are found in pegamatites in Methuen and Calvin (Ontario, Canada), Nellore (Andhra Pradesh, India), Custer (South Dakota, USA) and Minas Gerais (Brazil).

KAI2(AISI3)010(0H)2	
Colour:	white, grey, yellow, brown, greenish; bright green
Lustre; opacity:	vitreous, pearly, silky; transparent to translucent
Streak:	white
Hardness:	2.5
Specific gravity:	2.76-2.88
Cleavage; fracture:	perfect; uneven, lamellar
Habit:	pseudo-hexagonal tabular crystals, deep striations on prism faces; foliated, scaly, lamellar masses
Crystal system:	monoclinic

KAl₂(AlSi₃)O₁₀(OH)₂

Phlogopite



Phlogopite forms a series with the more common mica biotite, which is iron-rich and generally of darker colour. It is used in electronics for its excellent heat- and electrical-insulating properties. Phlogopite forms tabular crystals up to 2m (6%/ft) across, some strikingly transparent. It decomposes in concentrated sulphuric acid, distinguishing it from muscovite. The name comes from the Greek flogopos, meaning 'to resemble fire'. It occurs in metamorphosed dolomites and in ultramafic rocks, associated with dolomite, calcite, diopside and epidote. It is found at Campolungo (Switzerland), Frontenac (Ontario, Canada), Ødegården (Norway), Franklin (New Jersey, USA), Saharakara (Madagascar) and Anxiety Point (New Zealand).

Colour:	brown, reddish-brown, yellow, green
Lustre; opacity:	pearly; transparent to translucent
Streak:	white
Hardness:	2–3
Specific gravity:	2.86
Cleavage; fracture:	perfect; uneven, lamellar
Habit:	crystals as six-sided plates; foliated, granular aggregates
Crystal system:	monoclinic

KMg₃(Al,Fe)Si₃O₁₀(F,OH)₂

Biotite



Biotite, named after the French mineralogist J.-B. Biot (1774–1862), is also called black mica, as opposed to white mica (muscovite). It can form enormous crystals in pegmatites, but is most common as small dark plates in granites, contrasting with larger, paler quartz and feldspar. Biotite in exposed surfaces of rocks can sparkle in sunlight, occasionally with a golden sheen. It is an important mineral in schists, gneisses, granites, nepheline syenites and contact-metamorphosed rocks. Associated minerals include muscovite, pyroxenes, amphiboles, andalusite and cordierite. Good crystals occur at Vesuvius (Italy), Miass (Ilmen Mountains, Russia), Arendal (Norway), Franklin (New Jersey, USA) and Bancroft (Ontario, Canada).

Colour: brown, black, dark green, greyish-yellow vitreous, pearly on cleavage; transparent to translucent Lustre; opacity: Streak: grey-white Hardness: 2.5 - 3Specific gravity: 28-3.2 Cleavage; fracture: perfect; irregular, lamellar rare tabular or short prismatic pseudohexagonal crystals; Habit: foliated masses or aggregates, disseminated grains **Crystal system:** monoclinic

K(Mg,Fe)₃(AlSi₃)O₁₀(F,OH)₂

Lepidolite



Lepidolite, or lithium mica, named after the Greek for 'scale', is an ore of lithium land a source of rubidium and caesium. The pink colour is fairly characteristic, but may be confused with pink muscovite. Like other micas, lepidolite can be found as 'books' of crystal sheets. Large forms are carved, shaped or used as polished stones. Attractive specimens from Brazil comprise lepidolite with red tourmaline and colourless quartz. Lepidolite occurs in lithium-rich pegmatites and altered granites, associated with quartz, feldspars, spodumene and tourmaline. It is found at Pala (California, USA), Mount Mica (Maine, USA), Rozna (Czech Republic), Alabashka (Urals, Russia) and Virgem de Lape (Minas Gerais, Brazil).

pink to lilac
pearly; transparent to translucent
white
2.5-4
2.8–2.9
perfect; uneven, lamellar giving flexible sheets
six-sided tabular crystals; fine, platy aggregates, massive
monoclinic

K(Li,Al)₃(Si,Al)₄O₁₀(F,OH)₂

Zinnwaldite



Zinnwaldite is a widespread but rare mica, difficult to distinguish from other micas but for the environment in which it occurs. Its colour is usually darker than muscovite, but lighter than biotite. Zinnwaldite forms six-sided crystals up to 20cm (8in). As with other micas, zinnwaldite occurs as 'books' of pseudohexagonal crystals. Zinnwaldite occurs in tin-bearing pneumatolytic deposits in greisens and rarely in pegmatites and quartz veins, associated with topaz, cassitrerite, lepidolite, wolframite and spodumene. It is found at Cinovec (Zinnwald, Czech Republic), Altenberg and Waldstein (Germany), St Just (Cornwall, England), Amelia (Virginia, USA), Pala (California, USA) and Antaboaka (Madagascar).

KEII CAI(AIZ)		
Colour:	grey-brown, yellow-brown, pale violet, dark green	
Lustre; opacity:	vitreous, pearly; transparent to translucent	
Streak:	white	
Hardness:	2.5-4	
Specific gravity:	2,9–3.2	
Cleavage; fracture:	perfect; uneven, lamellar giving flexible, elastic scales	
Habit:	short prismatic or tabular crystals; rosettes, lamellar or scaly aggregates, disseminated	
Crystal system:	monoclinic	

KLiFeAI(Al₂Si₂)O₁₀(F,OH)₂

Margarite



A argarite is a brittle mica named from the Greek for 'pearl', after its lustre. WBrittle micas differ from common micas in that the thin platy crystals are inflexible and readily broken. Beryllian margarite is a related brittle mica, rich in beryllium. Brittle micas have doubly charged ions positioned between the aluminosilicate sheets, whereas common micas have singly charged ions. It occurs in low- to medium-grade metamorphic rocks; commonly in emery deposits, associated with corundum, diaspore, tourmaline, chlorite, and alusite, calcite and guartz, It is found at Mt Greiner (Austria), Naxos (Greece), Smyrna (Turkey), Chester (Massachusetts, USA), Glen Esk (Scotland) and Sverdlovsk (Russia).

CaAl ₂ (Al ₂ Si ₂)O ₁₀ (OH) ₂	
Colour:	pale pink, white, grey, green
Lustre; opacity:	pearly
Streak:	white
Hardness:	3.5–4.5
Specific gravity:	2.99–3.1
Cleavage; fracture:	perfect; uneven, giving brittle lamellae
Habit:	rare, tabular, pseudohexagonal crystals; usually foliated aggregates
Crystal system:	monoclinic



Vermiculite, when heated to 300°C (572°F), quickly loses water and expands 18 to 25 times its original volume, giving long, twisted forms of a golden yellow colour; hence the name, from the Greek for breeding worms. The expanded material is used extensively as packing and insulating material, in soil conditioners, hydroponics, fireproof fillings, and many others. It is formed by hydrothermal alteration of biotite or phlogopite, especially at the contact between felsic intrusions in ultramafic rocks. Associated minerals include corundum, serpentine and talc. Large masses are found at Bulong (Western Australia), Palabora (South Africa), Libby (Montana, USA), Macon (North Carolina, USA) and Ajmer (Rajasthan, India).

(Ivig, re, AI) ₃ (AI, SI) ₄ U ₁₀ (UR) ₂ .4R ₂ U	
Colour:	white, yellow, green, brown
Lustre; opacity:	vitreous, pearly; transparent to opaque
Streak:	white
Hardness:	1.5
Specific gravity:	2.2–2.6
Cleavage; fracture:	perfect; uneven
Habit:	large crystalline plates, possibly pseudohexagonal; scaly aggregates
Crystal system:	monoclinic

Mg,Fe,Al)₃(Al,Si)₄O₁₀(OH)₂.4H₂O

Pennine



Pennine, or penninite, is a pseudotrigonal variety of clinochore. It is a silica-rich member of the chlorite group and occurs as complex admixtures with chromium-rich kammererite. Pennine forms hexagonal platy crystals and it flakes on heating. It is named after the Pennine Alps in Switzerland, where it was first found. Massive pennine is easily carved and is used for ornaments. Pennine occurs in low-grade metamorphic and hydrothermally altered rocks, associated with actinolite, epidote and garnet. It is an essential component of some massive chlorite schists. Pennine is found at Rimpfischwäge and Zermatt (Switzerland), Zillertal (Austria), Val d'Ala (Turin, Italy) and Harstigen (Värmland, Sweden).

Colour:	white, yellow, green
Lustre; opacity:	vitreous, pearly; translucent
Streak:	white
Hardness:	2-2.5
Specific gravity:	2.5–2.6
Cleavage; fracture:	perfect; uneven
Habit:	crystals as plates or rhombohedra; scaly, massive, granular aggregates
Crystal system:	monoclinic

(Mg,Fe)₅Al(Si,Al)₄O₁₀(OH)₈

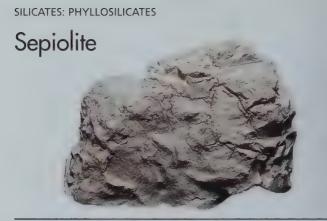
Kämmererite



Kämmererite is a rare chromium-rich variety of clinochlore, displaying a variety of shades of red. The red colours contrast with the green colours produced by iron substitution in clinochlore. It is usually a mixture with pennine, which is regarded by some as a green variety of kämmererite. Particularly good clusters of deep magenta crystals come from Kop Daglari (Erzerum Province, Turkey), many of the crystals being clear, well-formed rhombohedra. Kämmererite occurs in metamorphic rocks, especially serpentinized rocks containing chromium-rich olivines. It is associated with serpentine, chromite and uravorite. It is found at Miass (Ilmen Mountains, Russia), Texas (USA) and the Shetland Islands (UK).

(Mg,Cr)₃(OH)₂AlSi₃O₁₀Mg₃(OH)₆

Colour:	red to purple
Lustre; opacity:	vitreous; transparent to translucent
Streak:	reddish
Hardness:	2-2.5
Specific gravity:	2.64
Cleavage; fracture:	perfect; uneven
Habit:	pseudohexagonal crystals; scaly aggregates
Crystal system:	monoclinic



Sepiolite, also called meerschaum or sea foam, is a light and porous clay mineral named after the Greek for 'cuttle-fish'. It has been traditionally used in tobacco pipes, and is now used to absorb oil on workshop floors or as a filler, binder or free-flow agent in formulations. Thousands of tons per year are imported to the United Kingdom from Spain. Possible asbestos contamination requires monitoring of its use. It is a sedimentary clay mineral formed by surface alteration of magnesite or serpentine. It is associated with opal and dolomite. Sepiolite is found at Eskiflhahir (Turkey), Vallecas and Cabañas (Spain), Middletown (Pennsylvania, USA), Amarillo (Texas, USA), Hobbs (New Mexico, USA) and Cerro Mercado (Mexico).

Colour:	white, greyish-, yellowish- or reddish-white
Lustre; opacity:	dull; opaque
Streak:	white
Hardness:	<mark>2–2.5</mark>
Specific gravity:	1-2 .
Cleavage; fracture:	unknown; conchoidal
Habit:	compact, nodular, earthy, massive
Crystal system:	orthorhombic

Mg₄(OH)₂Si₆O₁₅.2-4H₂O

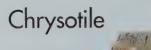
Clinochlore



Clinochlore is a common mineral of the chlorite group, the members being difficult to distinguish and usually referred to simply as chlorite by petrologists. Clinochlore is magnesium-rich and forms a series with the iron-rich chamosite. Massive green clinochlore, often banded with a silvery iridescence, is used for carving and polishing as seraphinite. It is an essential mineral in chlorite and talc schists and is found in serpentinites and marbles, associated with serpentine, biotite, actinolite, plagioclase, calcite and dolomite. It is found at Val d'Ala (Turin, Italy), Val Malenco (Sondrio, Italy), Chester (Massachusetts, USA,) Chester County (Pennsylvania, USA), the Zillertal (Austria) and Akhmatovsk (Urals, Russia).

Colour:	grass-green, olive-green, white, pink, yellowish
Lustre; opacity:	pearly, greasy, dull; transparent to translucent
Streak:	greenish-white to white
Hardness:	2-2.5
Specific gravity:	2.6–2.8
Cleavage; fracture:	perfect; uneven, giving inelastic lamellae
Habit:	pseudohexagonal crystals with tapering prismatic faces; foliated, granular, earthy, massive
Crystal system:	monoclinic

(Mg,Fe)₅Al(Si,Al)₄O₁₀(OH)₈



hrysotile is a fibrous variety of serpentine and the main type of asbestos. Formerly widely used for thermal and electrical insulation, it is now being removed because of the carcinogenic properties of some fibrous forms. The fibres are flexible enough to form woven products, such as fire blankets. It is produced by low-grade metamorphism of ultrabasic rocks in water-rich environments (especially the ocean floor). It is associated with antigorite and lizardite in serpentinites, often filling veins, with the fibres aligned in the direction of the vein. Large masses are found in Asbestos (Quebec, Canada), Eden Mills (Vermont, USA), Brewster (New York, USA), the Urals (Russia), the Alps and Cyprus.

WIG351205(OH)4	
Colour:	greyish-white, green, yellow, brown
Lustre; opacity:	silky; translucent to opaque
Streak:	white
Hardness:	2.5–4
Specific gravity:	2.55
Cleavage; fracture:	none; splintery
Habit:	microcrystalline; aggregates of flexible fibres
Crystal system:	monoclinic (clinochrysotile) and orthorhombic (orthochrysotile)

Chrysocolla



Chrysocolla is an attractive blue-green copper silicate used as an ornamental stone. It is softer than tourquoise or chalcedony, with which it can be confused. Specimens of chrysocolla with quartz not only go well together but also tend to be harder ('agatized') and are more suitable for use in jewellery. It is named from the Greek for 'gold' and 'glue', after its use in soldering gold in ancient times. It occurs in the oxidized zones of copper deposits, associated with malachite, tenorite, quartz and cuprite. Chrysocolla is found at Chuquicamata (Chile), Nizhni Tagil (Urals, Russia), Lubumbashi (Katanga, Congo), Timna (King Solomon's) mine (Israel), Santa Rita (New Mexico, USA) and Clifton (Arizona, USA).

CuSiO₃.nH₂O

green, blue-green
vitreous, greasy, earthy; translucent to opaque
light green
2-4
2.0–2.4
: none; conchoidal
rare acicular crystals in radiating clusters; concretions, porcellanous masses, earthy
monoclinic
-

Serpentine



Serpentine is a group of closely related minerals: antigorite, lizardite and chrysotile, the latter most likely to be fibrous and form asbestos. Fibrous serpentine is now regarded as a health hazard. Antigorite and lizardite are the main components of the rock serpentinite, which is formed under oceans by low-grade metamorphism of ultrabasic rocks. Such rocks may be cut by veins of fibrous chrysotile. Attractive, multi-hued and easily worked, serpentine is popular for ornaments; some varieties (bowenite and williamsite) are used for jewellery. The name refers to the snakelike appearance of massive forms. It is found widely, including at Stillwater (Montana, USA), Vai Malenco (Italy), Lizard Peninsula (England) and Troodos Mountains (Cyprus).

(119,193)205(01)4	
mostly green, also yellow, brown, black, red	
greasy, dull; translucent to opaque	
white, grey	
2.5-4	
2.5–2.6	
none; conchoidal, splintery	
microcrystalline; lamellar (antigorite), fibrous (chrysotile), scaly (lizardite)	
monoclinic/orthorhombic/hexagonal	

(Mg,Fe)₃Si₂O₅(OH)₄

SILICATES: TECTOSILICATES

Nepheline



Nepheline is a major rock-forming mineral found in igneous rocks that are low in silica and contain no quartz, such as nepheline syenites. It is a feldspathoid, a group of minerals like alkali feldspars, but poorer in silica. Although abundant, nepheline rarely forms attractive specimens. Nepheline is especially liable to alteration; it becomes cloudy when treated with acids, hence the name from the Greek for 'cloud'. Nepheline is often altered to zeolites, often natrolite. It occurs in alkaline rocks and pegmatites associated with feldspar, leucite, augite and olivine. Some of the many locations include Vesuvius (Italy), Kola Peninsula and Urals (Russia), Bancroft (Canada) and Litchfield (Maine, USA).

(Na,K)AlSiO₄

Colour:	off-white, grey, brown, other tints
Lustre; opacity:	greasy; translucent
Streak:	white
Hardness:	5.5–6
Specific gravity:	2.56-2.66
Cleavage; fracture:	poor; conchoidal
Habit:	prismatic crystals; compact, granular aggregates
Crystal system:	hexagonal

SILICATES: TECTOSILICATES

Petalite



Petalite is one of the less common feldspathoids and an ore of lithium. It is too fragile and too rarely of high quality to be commonly used in jewellery, but is sometimes cut for collectors. When heated, petalite slowly gives a blue phosphorescence; it is easily melted and colours a flame crimson (due to lithium). Petalite is named from the Greek for 'leaf', after its perfect cleavage. Petalite occurs in lithium-rich granite pegmatites associated with spodumene, lepidolite, tourmaline, topaz, albite, microcline and quartz. Good localities for petalite include Varaträsk (Sweden), Bikita (Zimbabwe), Norwich (Massachusetts, USA), Oxford County (Maine, USA) and Minas Gerais (Brazil).

LiAlSi ₄ O ₁₀	
Colour:	colourless or white, grey to reddish-grey
Lustre; opacity:	vitreous to pearly; transparent to translucent
Streak:	white
Hardness:	6-6.5
Specific gravity:	2.4
Cleavage; fracture:	perfect; conchoidal
Habit:	tabular or columnar crystals; usually massive
Crystal system:	monoclinic

Leucite



Leucite is a feldspathoid found abundantly in potassium-rich, silica-poor volcanic trocks. The usual trapezohedral crystal form of leucite is pseudomorphic after the cubic,-leucite, which forms above 605°C (1121°F). Leucite usually occurs as single crystal grains embedded in host rock. The garnets and analcime are the only other common minerals to crystallize in the form of a trapezohedron. The name is derived from the Greek for 'white', and leucite has been known as white garnet. Good crystals occur in many localities, including Alban Hills and Vesuvius (Italy), Leucite Hills (Wyoming, USA), Bear Paw Mountains (Montana, USA), Mount Nviragongo (Congo) and Kaiserstuhl (Germany).

KAISi ₂ O ₆		
Colour:	white	
Lustre; opacity:	vitreous; transparent to translucent	
Streak:	white	
Hardness:	5-5.5	
Specific gravity:	2.24-2.29	
Cleavage; fracture:	absent; conchoidal, brittle	
Habit:	trapezohedral crystals; round grains	
Crystal system:	tetragonal	

Sanidine



Sanidine is a potassium-feldspar formed at high temperatures, above 900°C (1652°F) (orthoclase and microcline are polymorphs formed at medium/high and low temperatures, respectively). It forms a partial series with albite, NaAlSi₃O₈, which can form up to 20 per cent of its composition. Sanidine is found in felsic igneous rocks, potassium-rich contact metamorphic rocks and hydrothermal veins. Associated minerals include quartz, sodic plagioclase, biotite, muscovite, hornblende and magnetite. Locations include Vesuvius and Monte Cimine (Italy), Drachenfels and Hohenfels (Germany), Daichi (Japan), Tooele (Utah), Cottonwood Canyon (Arizona) and Rabb Canyon (New Mexico, USA), and Kanchin-do (Korea).

KAISi ₃ O ₈	
Colour:	colourless, white, greyish, yellowish
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	6
Specific gravity:	2.5
Cleavage; fracture:	perfect; uneven,conchoidal
Habit:	tabular, prismatic crystals, often twinned
Crystal system:	monoclinic

Orthoclase



Orthoclase is abundant as an essential mineral in granitic rocks, forming between about 500° and 900°C (932° and 1652°F). It also occurs in pegmatites and syenites; cavities in basalts; high-grade metamorphic rocks; and hydrothermal veins. Adularia is a transparent colourless gem variety with a bluish sheen. Yellow orthoclase is often faceted when transparent or cut *en cabochon* when displaying a cat's-eye effect. Moonstone is opalescent with a blue or white sheen resembling moonlight. Locations include Madagascar, Sri Lanka and Burma for gem varieties; fine crystals are found at Baveno and Elba (Italy), Carlsbad (Czech Republic), Kirkpatrick (Scotland) and Sverdlovsk (Urals, Russia).

KAISi₃O₈

Colour:	colourless or white, pale yellow, pink, blue or grey
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness;	6
Specific gravity:	2.55–2.63
Cleavage; fracture:	perfect; conchoidal
Habit:	prismatic, columnar or tabular crystals; compact, granular masses
Crystal system:	monoclinic

Microcline



Microcline is a potassium feldspar that forms at low temperatures (usually less than 400°C/752°F), abundant in granites and related rocks. Amazonite is a blue-green semi-opaque stone resembling jade or turquoise. Perthite is microcline or orthoclase containing undulating layers or intergrowths of plagioclase feldspar, which have separated out on cooling. Microcline is found in felsic plutonic igneous rocks, metamorphic rocks and hydrothermal veins; it is associated with quartz, sodic plagioclase, biotite, muscovite and hornblende. Localities include Arendal (Norway), Ilmen Mountains (Russia), Cala Francese (Sardinia, Italy); amazonite is found at Pikes Peak (Colorado, USA), Bancroft (Canada) and Minas Gerais (Brazil).

KAlSi ₃ O ₈	
Colour:	white, pink, red, yellowish, blue-green
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	6–6.5
Specific gravity:	2.55-2.63
Cleavage; fracture:	perfect; conchoidal, uneven
Habit:	prismatic crystals, often twinned; compact aggregates
Crystal system:	triclinic

Hyalophane



Hyalophane is an unusual barium-rich feldspar forming a series with orthoclase and celsian (BaAl₂Si₂O₈). The clarity of the crystals, which may be found up to 20cm (8in) long, inspires the name from the Greek for 'glassy'. Hyalophane often shows a weak violet fluorescence under ultraviolet light. Found in metamorphosed manganese-rich rocks and manganiferous deposits, it is associated with epidote, rhodonite, plagioclase and analcime. Some excellent crystals have come from the Zagradski Potok region of Bosnia. Other locations include Lengenbach (Switzerland), Långban and Värmland (Sweden), Slyudyanka (Baikal, Russia), Broken Hill (NSW, Australia) and Kaso (Japan).

(K,Ba)Al(Al,Si)Si₂O₆

Colour:	colourless, white, yellow
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	6-6.5
Specific gravity:	2.6–2.8
Cleavage; fracture:	perfect; uneven, conchoidal
Habit:	crystals as prisms or rhombohedra; compact, massive
Crystal system:	monoclinic

Albite



A lbite is a plagioclase feldspar in the series albite-anorthite (NaAlSi₃O₈-CaAl₂Si₂O₈) containing less than 10 per cent anorthite. It forms a hightemperature series with sanidine, but separates at lower temperatures to give intergrowths (perthites). A thin, platy and often transparent variety often associated with tourmaline is called cleavelandite. Albite is abundant in many acid igneous rocks; it is also found in mica schists and gneisses, and in rocks altered ('albitized') by sodium-rich fluids (sea-floor metamorphism). It is associated with orthoclase, quartz, muscovite and biotite. Good crystals are found at St Gotthard (Switzerland), Baveno (Italy), Cazadero (California, USA) and Rio Grande do Sul (Brazil).

NaAlSi₃O₈ colourless or white Colour: vitreous; transparent to translucent Lustre; opacity: Streak: white 6-6.5 Hardness: Specific gravity: 2.62 Cleavage; fracture: perfect; conchoidal crystals as prisms or plates, often twinned; granular and Habit: massive aggregates Crystal system: triclinic

Anorthite



Plagioclase feldspars range from sodium-rich albite, through oligoclase, andesine, labradorite and bytownite, to calcium-rich anorthite (contains up to 10 per cent albite). Sunstone is a variety of oligoclase containing reflective platy inclusions, red, orange or green in colour, which give a metallic glitter. Labradorite (50–70 per cent anorthite) is used as a gem or ornamental stone, displaying a play of colours produced by internal interference of light. Plagioclase occurs in igneous and metamorphic rocks, becoming more calcium-rich as the rocks become more mafic. Labradorite occurs in Larvik-Tvedalan (Norway) and Labrador (Canada). Anorthite is found at Vesuvius (Italy), Hokkaido (Japan) and Tunaberg (Sweden).

CaAl₂Si₂O₈

Colour:	white to grey, yellowish, greenish, reddish
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	6
Specific gravity:	2.62–2.76
Cleavage; fracture:	good; uneven
Habit:	crystals as prisms or plates, often twinned; granular or massive aggregates
Crystal system:	triclinic

Hauyne



Hauyne, or hauynite, is a feldspathoid and a member of the sodalite group, where calcium partially substitutes for sodium, and sulphate groups replace chlorine atoms. It is popularly collected as bright blue grains in a matrix and occasionally faceted, but is difficult to cut because of its perfect cleavage. Hauyne is found in phonolites and related silica-poor rocks, associated with nepheline, leucite, augite, sanidine, biotite and apatite. It is named after the French crystallographer Abbé René Haüy (1743-1822), who discovered it on Vesuvius (Italy); other localities are Laacher See and Niedermendig (Germany), Winnett (Montana, USA), Edwards Mine (New York, USA), Tasmania (Australia) and Nanjing (Jiangsu, China).

$(Na, Ca)_{4-8}(A1_6S1_6)U_{24}(SU_4, C1)_{1-2}$		
Colour:	bright blue to greenish-blue, white, grey, brown	
Lustre; opacity:	vitreous, greasy; transparent to translucent	
Streak:	white	
Hardness:	5.5	
Specific gravity:	2.44–2.5	
Cleavage; fracture:	perfect; conchoidal	
Habit:	rare octahedral or dodecahedral crystals; usually rounded grains	
Crystal system:	cubic	

(ALC:)0 100

Sodalite

Sodalite is a feldspathoid mostly notable for its occurrence in a bright blue, massive form, used as a semi-precious stone for carvings and jewellery. It is a component of lapis lazuli, but, unlike it, sodalite rarely contains contrasting golden yellow pyrite and has a lower density. Crystals of sodalite are rare and usually small. Sodalite is found in nepheline syenites, phonolites and related rocks, and in metasomatic calcareous rocks. Associated minerals include nepheline, microcline and sanidine. Excellent massive sodalite is found at Bancroft (Canada), Litchfield (Maine, USA), Magnet Cove (Arkansas, USA), Ilimaussaq (Greenland) and Cerro Sapo (Bolivia); clear crystals are found in the calcium-rich lavas of Vesuvius (Italy).

Na8Cl2(AlSiO4)6

Colour:	bright blue, white, grey, green
Lustre; opacity:	vitreous, greasy; transparent to opaque
Streak:	white
Hardness:	5-6
Specific gravity:	2.3
Cleavage; fracture:	perfect; conchoidal, uneven
Habit:	very rare dodecahedral crystals; compact masses
Crystal system:	cubic

Lazurite



azurite is most commonly found as a component of lapis lazuli, the semi-precious massive stone used for jewellery and carved ornaments. Lapis lazuli is mostly lazurite, but contains blue sodalite and hauyne, white calcite and brassy pyrite. The deposits in Afghanistan were worked 6000 years ago for high-quality lapis lazuli used, for example, for the mask of Tutunkhamun. Lazurite is found in contact metamorphosed limestones. Exceptional crystals and fine lapis lazuli are found at Sar-e-Sang (Kokscha Valley, Afghanistan); other locations include Vesuvius and the Alban Hills (Italy), Sayan Mountains (Russia), Baffin Island (Canada) and California and Colorado (USA).

Colour:	deep blue, blue-green
Lustre; opacity:	vitreous; translucent to opaque
Streak:	light blue
Hardness:	5-5.5
Specific gravity:	2.38–2.42
Cleavage; fracture:	imperfect; uneven
Habit:	very rare octahedral crystals; compact masses
Crystal system:	cubic

(Na,Ca)8(Al,Si)12O24(S,SO4)

Scapolite

 $\label{eq:scapelite} \begin{aligned} & \textbf{S}_{\text{Ca}_4(\text{Si},\text{Al})_{12}\text{O}_{24}(\text{CO}_3,\text{SO}_4)) \text{ and marialite (Na}_4(\text{Si},\text{Al})_{12}\text{O}_{24}\text{Cl}). It is formed by alteration of plagioclase feldspars, its composition generally reflecting the sodium/calcium ratio of the parent mineral. Clear crystals may be faceted; those containing inclusions are cut$ *en cabochon.* $Scapolite is named from the Greek for 'rod', alluding to the crystal shape. Scapolite is found in metamorphosed limestones and hydrothermally altered basic rocks. Crystals up to 50cm (20in) are found at Rossie and Pierrepoint (New York, USA); other localities are Renfrew (Ontario) and Grenville (Quebec, Canada), Lake Tremorgio (Switzerland) and Minas Gerais (Brazil). \end{aligned}$

(Na,Ca)8(Cl2,SO4,CO3)(AlSi3O8)6

Colour:	white, bluish, grey, pink
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	56.5
Specific gravity:	2.54–2.77
Cleavage; fracture:	poor; conchoidal
Habit:	columnar prismatic crystals; fibrous or massive aggregates
Crystal system:	tetragonal

Natrolite



Natrolite is a zeolite forming attractive sprays of radiating needles. It is named from the Greek *natron* for 'soda', alluding to its sodium content. Being a typical molecular sieve, it holds water in voids in its structure. On heating to 300°C (572°F), it loses water without changing crystal structure, which it will reabsorb from the atmosphere when cooled. Natrolite may also exhibit orange fluorescence under ultraviolet light. It is found in hydrothermal veins in basalts, associated with nepheline, sodalite, quartz and other zeolites. Good crystals come from Nova Scotia and British Columbia, and specimens up to 1m (3½ft) long from Asbestos (Canada); other localities are Puy-de-Dôme (France) and White Head (Antrim, Northern Ireland).

Colour:	colourless, white, pink, yellowish
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	5.5–6
Specific gravity:	2.2–2.4
Cleavage; fracture:	good; conchoidal
Habit:	acicular needles; globular aggregates of radiating needles rarely compact
Crystal system:	orthorhombic

Na2(Al2Si3)O10.2H2O

Mesolite



Mesolite forms characteristic sprays of transparent acicular crystals. A typical Miscolite, it forms only as a secondary mineral from the breakdown of feldspars and fills voids in altered igneous rocks. It is a molecular sieve and easily loses and absorbs water. It is named from the Greek for 'middle', having a composition between natrolite and scolecite. Mesolite is found in cavities in volcanic rocks, especially basalt, and hydrothermal veins. It is associated with natrolite and scolecite (from which it is difficult to distinguish), and other zeolites and calcite. Locations include Nova Scotia (Canada), Grant County (Oregon, USA), Faeroe Islands (Norway), Puy-de-Dôme (France) and Poona (Maharashtra, India).

white, grey, yellow
vitreous, silky; transparent to translucent
white
5-5.5
2.2–2.4
perfect; conchoidal, uneven
acicular needles; massive, spherulitic aggregates, earthy
monoclinic

Na2Ca2(Al2Si3O10)3.8H2O

Thomsonite



Thomsonite is a rare zeolite, difficult to distinguish from natrolite. The name, after the Scottish chemist Thomas Thomson, is applied to the series thomsonite. Ca to thomsonite-Sr. The strontium-rich minerals are visually indistinguishable and rarer. Thomsonite is found filling voids in mafic igneous rocks, especially basalts, and in hydrothermal veins in other igneous rocks. Associated minerals are calcite, prehnite, quartz and other zeolites. Localities include Mount Monzoni (Italy), Faeroe Islands (Norway), Old Kilpatrick and Bishopton (Scotland), Disko Island (Greenland) and Springfield (Oregon, USA). Attractively banded specimens of massive nodules are found on the southern shore of Lake Superior.

Colour:	white, brown
ustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	5–5.5
specific gravity:	2.2–2.4
leavage; fracture:	perfect; uneven
Habit:	individual crystals rare; radiating globular clusters
Crystal system:	monoclinic

NaCa₂(Al₅Si₅)O₂₀.6H₂O

Scolecite



Scolecite is a rare zeolite occurring as sprays of white needles. It is closely related to natrolite and mesolite, and difficult to distinguish from these. The name, from the Greek *skolec* for 'worm', is an allusion to its behaviour in a blowpipe flame, when it curls into wormlike forms before melting. Scolecite is found lining cavities in lavas, especially basalts; and also in veins in contact-metamorphosed pelites and limestones. Associated minerals are calcite, prehnite and other zeolites. Excellent crystals are found at Teigarhorn (Iceland), Ben More, Isle of Mull and Talisker Bay, Isle of Skye (Scotland) and the Faeroe Islands (Norway); other locations include the Deccan Traps (Poona, India) and Vesuvius (Italy).

Colour:	colourless or white
Lustre; opacity:	vitreous, silky; transparent
Streak:	white
Hardness:	5-5.5
Specific gravity:	2.2–2.4
Cleavage; fracture:	perfect; fragile
Habit:	prismatic crystals; fibrous, radiating masses
Crystal system:	orthorhombic

Ca(Al₂Si₃)O₁₀.3H₂O

Mordenite



Mordenite is a rare but widespread zeolite, found often as clusters of white or pinkish needles. It is used in catalysis, petrochemicals and fine chemicals, and is synthetically produced for such applications. Mordenite, like other zeolites, has holes in its molecular framework of exact sizes, making it specific in interactions for absorption and catalysis. It is a secondary mineral, found in veins and cavities in igneous rocks, associated with calcite, kaolinite, glauconite and other zeolites; it is also deposited among some sediments. Locations include Morden (Nova Scotia, Canada), Berufjord (Iceland), Elba (Italy), Isle of Mull (Scotland), Custer County (Colorado, USA) and Hoodoo Mountains (Wyoming, USA).

Colour:	colourless, white, yellowish, pinkish
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	4.5
Specific gravity:	2.1
Cleavage; fracture:	perfect; uneven
Habit:	crystals as striated needles; fibrous, reniform aggregates
Crystal system:	orthorhombic

(Ca,K2,Na2)(AlSi5012)2.6H2O

Laumontite



Laumontite crystals can be impressive when large, almost acicular prisms. It is a Lzeolite named after the Frenchman François de Laumont (1747–1834), who found the mineral. Specimens exposed to dry air become opaque and powdery due to partial dehydration; this can be avoided by using a sealant or airtight containment. Laumontite is found in veins and cavities in most rock types and as a cement in sandstones. Associated minerals include datolite, calcite, chlorite and other zeolites. Crystals up to 30cm (12in) are found at Bishop (California, USA); good crystals are also found in the eastern Pyrenees (France), St Gotthard (Switzerland), the Tirol (Austria), Baveno (Italy) and Poona and Khandivali (Maharashtra, India).

Ca(A12514)012	.4120
Colour:	white to grey, yellowish, brownish, pink
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	3-4
Specific gravity:	2.2–2.3
Cleavage; fracture:	perfect; uneven
Habit:	square, prismatic crystals often elongated; columnar, fibrous and radiating aggregates
Crystal system:	monoclinic

Ca(Al₂Si₄)O₁₂.4H₂O

Heulandite



Heulandite is a common zeolite forming attractive aggregates of tabular crystals in a range of different hues. Distinguished from stilbite in 1818, it was first named euzeolite, meaning 'beautiful zeolite', but eventually named after the English mineral collector John Heuland (1778-1856). Heulandite is now a term referring to a series of calcium-, sodium-, potassium- or strontium-rich rocks. It is found in cavities in volcanic rocks, in veins in schists and gneisses, and disseminated in sedimentary rocks; it is usually associated with stilbite and chabazite. Good crystals are found at Paterson (New Jersey, USA), Poona (Maharashtra, India), Berufjord (Iceland), Stirlingshire (Scotland), Nova Scotia (Canada) and Hawaii (USA).

(Na,Ca)4-6A16	(AI, SI)4SI26072.24H20
Colour:	colourless, yellow, green, reddish-orange
Lustre; opacity:	vitreous, pearly; transparent, usually translucent
Streak:	white
Hardness:	3.5-4
Specific gravity:	2.2
Cleavage; fracture:	perfect; uneven
Habit:	tabular crystals; aggregates
Crystal system:	monoclinic

Stilbite

Stilbite is a common mineral and one of the most popularly collected zeolites. It is named from the Greek for 'shine', after the pearly effect on the cleavage plane. The structure of stilbite contains channels that hold certain sizes of molecules and ions. This is put to use in the separation of hydrocarbons in petroleum refining and in ion-exchange processes. Stilbite is soluble in acids and melts easily, giving a white glass. It is found in hydrothermal veins in basalts, associated with calcite and other zeolites, especially heulandite. Fine crystals are found at Paterson (New Jersey, USA), Kilpatrick and Isle of Skye (Scotland), Nova Scotia (Canada), Rio Grande do Sul (Brazil), Teigarhorn (Iceland) and Poona (India).

NaCa2(A155113)036.141120	
Colour:	white, grey, reddish-brown
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	3.5–4
Specific gravity:	2.1
Cleavage; fracture:	perfect; conchoidal, fragile
Habit:	prismatic crystals, usually in sheaflike aggregates, cruciform twins; radiating fibrous masses
Crystal system:	monoclinic

NaCa2(Al5Si13)O36.14H2O

Harmotome



H armotome is a rare zeolite, but popular for its twinned crystals. Single twinning gives crosses of interpenetrating prisms, with the ends resembling blunt Phillips head screwdrivers; more uncommonly, double twinning can result in three prisms interpenetrating at 90° to each other. Harmotome is named from the Greek for 'joint' and 'cut', after the easily separated twinned crystals. Harmotome occurs in hydrothermal veins in basalts and other volcanic rocks, in gneisses and in some ore veins; it is associated with calcite, barite and quartz. Good crystals are found in St Andreasburg (Germany), Argyll (Scotland), Pribram (Czech Republic), Thunder Bay (Ontario, Canada) and Manhattan (New York City, USA).

Colour:	white, grey, yellow, red, brown
Lustre; opacity:	vitreous, pearly; transparent to translucent
Streak:	white
Hardness:	4.5
Specific gravity:	2.44-2.5
Cleavage; fracture:	distinct; uneven, subconchoidal
Habit:	prisms, plates, often twins; aggregates
Crystal system:	monoclinic

BaAl₂Si₆O₁₆.6H₂O

Gismondine



G ismondine is a rare zeolite, often forming clear colourless pseudotetragonal crystals. It was named after Professor Carlo Gismondi (1762–1824), the Italian mineralogist who first examined it. A similar mineral gismondine-Ba has been found on lead-rich slags. Gismondine is easily melted in a flame and soluble in hydrochloric acid. It occurs in hydrothermal veins in nepheline and olivine basalts, associated with calcite, quartz, chlorite and other zeolites. It is found at Capo di Bove (Lazio, Italy), Bühne (Westphalia), Fulde (Hesse) and Arensberg (Eiffel, Germany), Round Top (Oahu) and Alexander Dam (Kauai, Hawaii), Antrim (Northern Ireland) and Reydarfjord and Fáskrúdsfjord (Iceland).

Colour: white, grey, bluish, reddish Lustre; opacity: vitreous; transparent to translucent Streak: white Hardness: 4.5 Specific gravity: 2.26 Cleavage; fracture: imperfect; conchoidal Habit: bipyramidal crystals; stellate or radiating spherulitic aggregates Crystal system: monoclinic

CaAl₂Si₂O₈.4H₂O

Chabazite



Chabazite is a lesser known zeolite forming rhombohedral crystals that appear Calmost like cubes, with angles close to 90°. Chabazite-Na, -K and -Sr are varieties rich in sodium, potassium or strontium. It is used as an acid-resistant absorbant in natural gas production and to remove heavy metals from waste streams. Chabazite occurs in hydrothermal veins in basalts and andesites; and rarely in limestones, schists and ore veins. Associated minerals include nepheline, olivine, pyroxenes, tridymite, calcite and dolomite. Chabazite is found at Idar-Oberstein (Germany), Repcice (Czech Republic), Kilmalcolm (Scotland), Breidhdalsheidhi (Iceland), Bowie (Arizona, USA) and Paterson (New Jersey, USA).

Colour:	colourless, white, yellow, pink, red
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	4.5
Specific gravity:	2.08
Cleavage; fracture:	distinct; uneven
Habit:	crystals as rhombohedra, twins; druses, granular, massive
Crystal system:	trigonal

Ca(Al₂Si₄)O₁₂.6H₂O

Gmelinite



G melinite is a rare zeolite that forms characteristic crystals in the form of shallow six-sided double pyramids, which have been described as 'flying saucers'. It can also occur as thin plates and attractive rosettes of such crystals. Varieties are rich in calcium, sodium or potassium. It is named after Christian Gmelin (1792–1860), a German chemist. Gmelinite occurs in basalts, related rocks and pegmatites, associated with calcite, aragonite, quartz and other zeolites, especially chabazite and analcime. Locations include Montecchio Maggiore (Vicenza, Italy), Glenarm (Antrim, Northern Ireland), Pyrgos (Cyprus), Paterson (New Jersey, USA), Sarbay-Sokolov (Kazakhstan) and Bekiady (Madagascar).

(
Colour:	white, yellow, pink, reddish	
Lustre; opacity:	vitreous; transparent to opaque	
Streak:	white	
Hardness:	4.5	
Specific gravity:	2.03	
Cleavage; fracture:	imperfect; uneven	
Habit:	crystals as plates, bipyramids, rhombohedra, twins; druses, rarely radiating aggregates or granular	
Crystal system:	hexagonal	
Association and a second secon		

(Na,Ca)Al₂Si₄O₁₂.6H₂O

Analcime

nalcime, or analcite, is a popular zeolite forming distinctive trapezohedral crystals, a shape commonly found only in garnets (harder and strongly coloured) and leucite (lower specific gravity). It has been classified as a feldspathoid, but has an open structure, typical of zeolites. It is named from the Greek for 'weak', after the weak static charge produced on heating or rubbing. Analcime occurs in basalts and phonolites, associated with prehnite, glauconite, quartz and other zeolites. It is found at Kotchechovmo (Krasnoyarski, Russia), St Keverne (Cornwall, England), Breidhdalsheidhi (Iceland), Lake Superior (Michigan, USA), Bergen Hill (New Jersey, USA) and Mt Saint Hilaire (Quebec, Canada).

NaAlSi ₂ O ₆ .2H ₂ O	
Colour:	white, colourless, grey, pink, greenish
Lustre; opacity:	vitreous, dull; transparent to translucent
Streak:	white
Hardness:	5.5
Specific gravity:	2.2–2.3
Cleavage; fracture:	perfect; uneven, conchoidal
Habit:	trapezohedral crystals; granular, compact, massive, showing concentric structure
Crystal system:	cubic

Pollucite



Pollucite is a zeolite and the major ore of caesium. Typically pollucite contains about 28 per cent caesium, and it may also be a source of rubidium, which is present at about 1–2 per cent. Caesium compounds are used in catalyst promoters, special glasses and radiation monitoring equipment. The name comes from its association with petalite (castorite) from classical mythology, Pollux being the brother of Castor. Pollucite occurs in kilotonne quantities in lithium-rich granite pegmatites, associated with spodumene, lepidolite, potassium feldspar, microcline and quartz. Pollucite is found at Varutrask (Sweden), Elba (Italy), Bikita (Zimbabwe), Bernic Lake (Manitoba, Canada) and Shengus, Skardu and Gilgit (Pakistan).

(0,10,715120	6.1.20
Colour:	colourless, white, grey, pale pink, pale blue
Lustre; opacity:	vitreous; transparent to translucent
Streak:	white
Hardness:	6.5
Specific gravity:	2.44-2.5
Cleavage; fracture:	absent; brittle, uneven
Habit:	cubic crystals; grains, granular and massive aggregates
Crystal system:	cubic

(Cs,Na)AlSi₂O₆.H₂O

OTHER MINERALS

Amber



A mber is a fossil tree resin that, although not mineralized, is widely used in plant remains, moss, pine needles, spiders or even frogs and lizards. Amber will become electrically charged when rubbed. Unlike most minerals, amber will melt and burn on heating, being chiefly of polymerized hydrocarbons called 'terpenes'. Most amber is *c.* 30–90 million years old; semi-fossilized resin is called copal. The best known amber deposits, known as succinite, are in the Baltic region, along the coasts of Poland and Russia, especially the Sambia Peninsula near Kaliningrad. Burmese amber is called burmite and amber from Sicily is known as simetite.

Formula

Colour:	golden yellow/orange, brown, reddish, green, violet, black
Lustre; opacity:	resinous; transparent to translucent
Streak:	white
Hardness:	1.5-3
Specific gravity :	1.05–1.1
Cleavage; fracture:	absent; conchoidal
Habit: •	irregular nodules in marine sands and clays; often a placer deposit washed up on shore
Crystal system:	amorphous

Petrified wood



Petrified wood is really a fossil, but is included here, as it is very popular as ornaments and polished articles. Like opal, it is usually composed of hydrated silica, which has slowly replaced the lignin and cellulose of the wood as they have decomposed. The original structure of the wood is often preserved in great detail, down to microscopic levels. Various colours can be produced by small amounts of impurities such as iron, manganese and copper, which can help to pick out the original structure. The most famous petrified forests are those of Santa Cruz (Patagonia, Argentina), Navá Paka (Czech Republic), Lesbos (Greece) and the Petrified Forest National Park (Arizona, USA).

SIU2.NH2U	
Colour:	various
Lustre; opacity:	translucent to opaque
Streak:	white
Hardness:	5-7
Specific gravity :	1.9–2.5
Cleavage; fracture:	absent; conchoidal
Habit:	best examples are intact tracts of forest, with trees up to 3m (10ft) diameter and 30m (98ft) long and complete root systems
Crystal system:	amorphous

SiO₂.nH₂O



J et is a hard gem variety of lignite (a type of coal), which is often carved or faceted, and takes a good polish. Occasionally it has pyrite inclusions displaying a brassy, metallic lustre. Used in jewellery since ancient times, jet has been traditionally made into rosaries for monks. As part of her mourning attire, Queen Victoria wore jet from Whitby (England), famous at the time for the mining and crafting of jet. Unlike higher grade coals, jet retains a microscopic, woody structure. Jet is found at Pictou (Nova Scotia, Canada), Cabo Mondego (Coimbra, Portugal), Wet Mountain Valley (Colorado, USA), Acoma (New Mexico, USA), Anne Arundel County (Maryland, USA) and Henry Mountains and Coaly Basin (Utah, USA).

Formula		
Colour:	dark brown, black	
Lustre; opacity:	velvety, waxy; opaque	
Streak:	brown	
Hardness:	2.5	
Specific gravity :	1.33	**
Cleavage; fracture:	none; conchoidal	
Habit:	masses in bituminous shales	
Crystal system:	amorphous	

OTHER MINERALS

Pearl



Once thought to be the tears of the gods, pearls are used widely in jewellery and decorations. They form in modern shellfish, such as mussels, as protection against a natural or emplaced irritant (e.g. a piece of sand) within the soft parts. Layers of nacre are secreted around the foreign body at a rate of up to 0.6mm per year. The nacre consists of aragonite crystals held together by conchiolin, a hornlike substance, and water. Light reflecting from overlapping layers produces the characteristic iridescent lustre. The most important source of creamy white pearls is the Persian Gulf, attracting the highest prices; they have also been collected from the Gulf of Manaar (Sri Lanka) and the Red Sea for thousands of years.

Formula

Colour:	pale yellow, white, salmon pink to grey, brown or black
Lustre; opacity:	pearly, iridescent; translucent to opaque
Streak:	generally white
Hardness:	3
Specific gravity :	2.71
Cleavage; fracture:	pearls are soft and easily damaged
Habit:	roughly spherical, but a variety of shapes depending on original irritant and whether the pearl has moved
Crystal system:	orthorhombic (aragonite)

Glossary

accessory mineral	minor component of a rock; not necessary to define the rock type
acid	(of igneous rock) containing at least 10% quartz and chemically, more than $65\%~{\rm SiO}_2$
amorphous	non-crystalline, having no regular microscopic structure
aqua regia	a mixture of 3:1 nitric and hydrochloric acids
basic	(of igneous rock) containing 45–55% SiO_2
cation	a positively charged ion such as Na+ or Fe ₂ +
columnar	elongated prismatic form of a crystal
druse	fine crystalline coating on a matrix or filling a cavity
en cabochon	a rounded, convex cut of a stone
essential mineral	one that is necessary for the definition of a rock
euhedral	(of a mineral) having well-formed crystal faces
evaporite	sedimentary rock formed by evaporation of an igneous rock
felsic	(of rocks) composed mostly of light-coloured minerals, especially feldspar and quartz or feldspathoid
fluorescence	the emission of visible light on absorption of invisible ultraviolet (uv) light, giving an apparent 'glow-in-the-dark' effect
foliated	made up of thin aligned plates that flake easily
fumarole	a vent for volcanic gasses
gangue	the waste component of a mineral deposit
grade, metamorphic	degree of metamorphism, mostly related to temperatures and pressures experienced by a rock
hydrothermal vein	a vein produced by hot, mineral-rich waters of igneous origin
igneous rock	a rock produced from magma or lava
iridescence	a play of colours on a surface produced by light interference
lamellar	made of crystals mostly developed in parallel to each other
mafic	(of rocks) composed mostly of dark, ferromagnesian minerals, especially olivine, pyroxene and amphibole
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magma	hot, molten rock material, called lava on reaching the surface
matrix	fine-grained background mass of a rock
metamict	pertaining to a mineral whose structure has been broken down by radiation damage, typically powdery
metamorphic rock	rock produced by increased temperature and/or pressure when minerals are altered without substantial melting
metasomatism	the alteration of a mineral or rock by hot fluids, either removing or adding chemical elements
oxidation	chemical process involving addition of oxygen
pelitic	(of a rock or sediment) of high aluminium and low calcium content, especially of mudstones and clays
phosphorescence	the emission of light after illumination (by visible or ultaviolet light) has ceased
placer deposit	concentration of residual resistant, heavy minerals found in river gravels after removal of other components of a weathered and eroded rock
pneumatolysis	the reaction of magma-derived hot gasses with the surrounding rock
porphyroblast	large, well-developed crystal found in a metamorphic rocks
schist	metamorphic rock containing platy minerals, usually mica, in a roughly parallel arrangement
secondary	(of minerals) one produced by alteration of pre-existing minerals
sedimentary rock	rock formed by sedimentary processes such as weathering, erosion, transport, deposition, compaction and cementation
solid solution	(of minerals) a product of two or more minerals mixing at a molecular level while retaining the same structure to give intermediate compositions
sublimate	a material condensing as a solid from a gas
thermoluminescence	luminescence produced by heating
triboluminescence	luminescence produced by rubbing
ultrabasic	(of rocks) containing <45% SiO ₂
ultramafic	(of rocks) containing >90% mafic minerals
vein	an irregular but essentially tabular intrusion, of a width between a millimetre and tens of centimetres

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