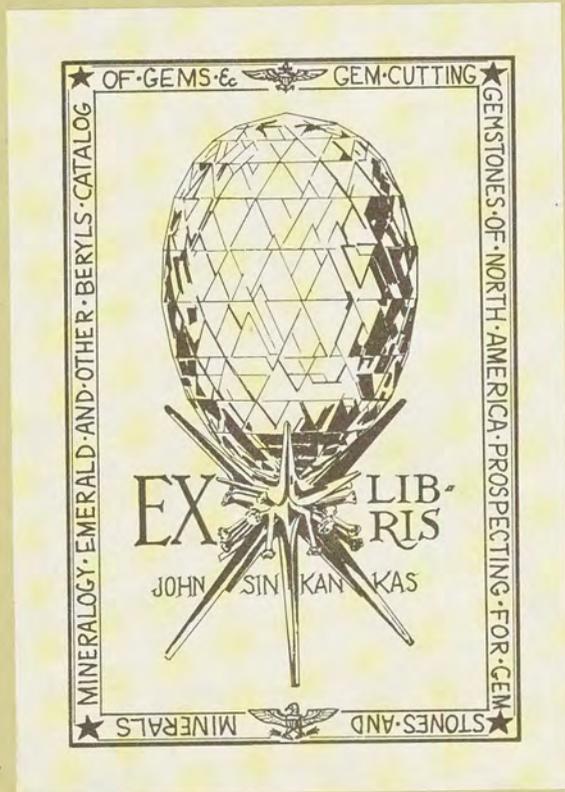


Tafeln

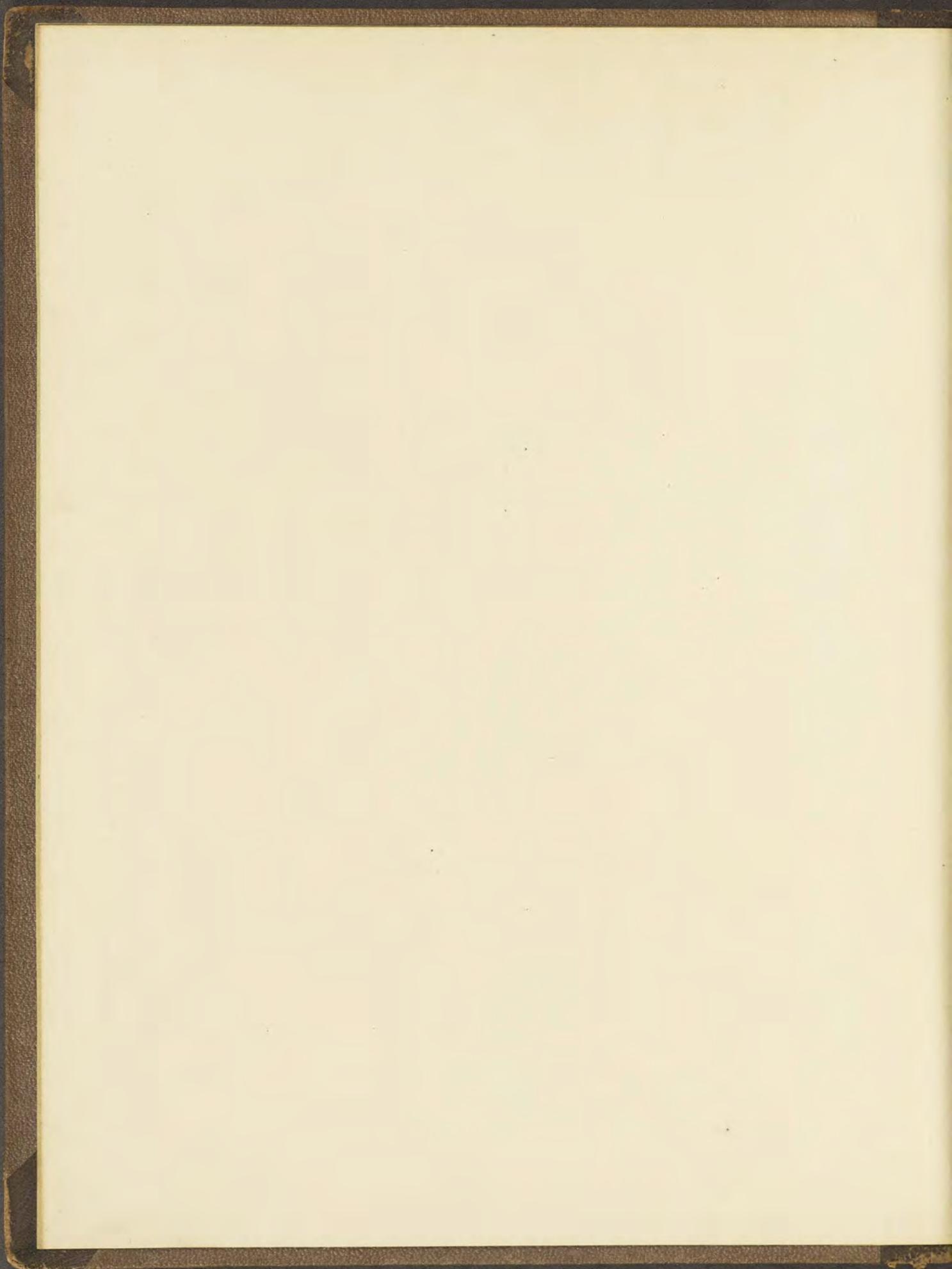
ZU W. H. MILLER'S

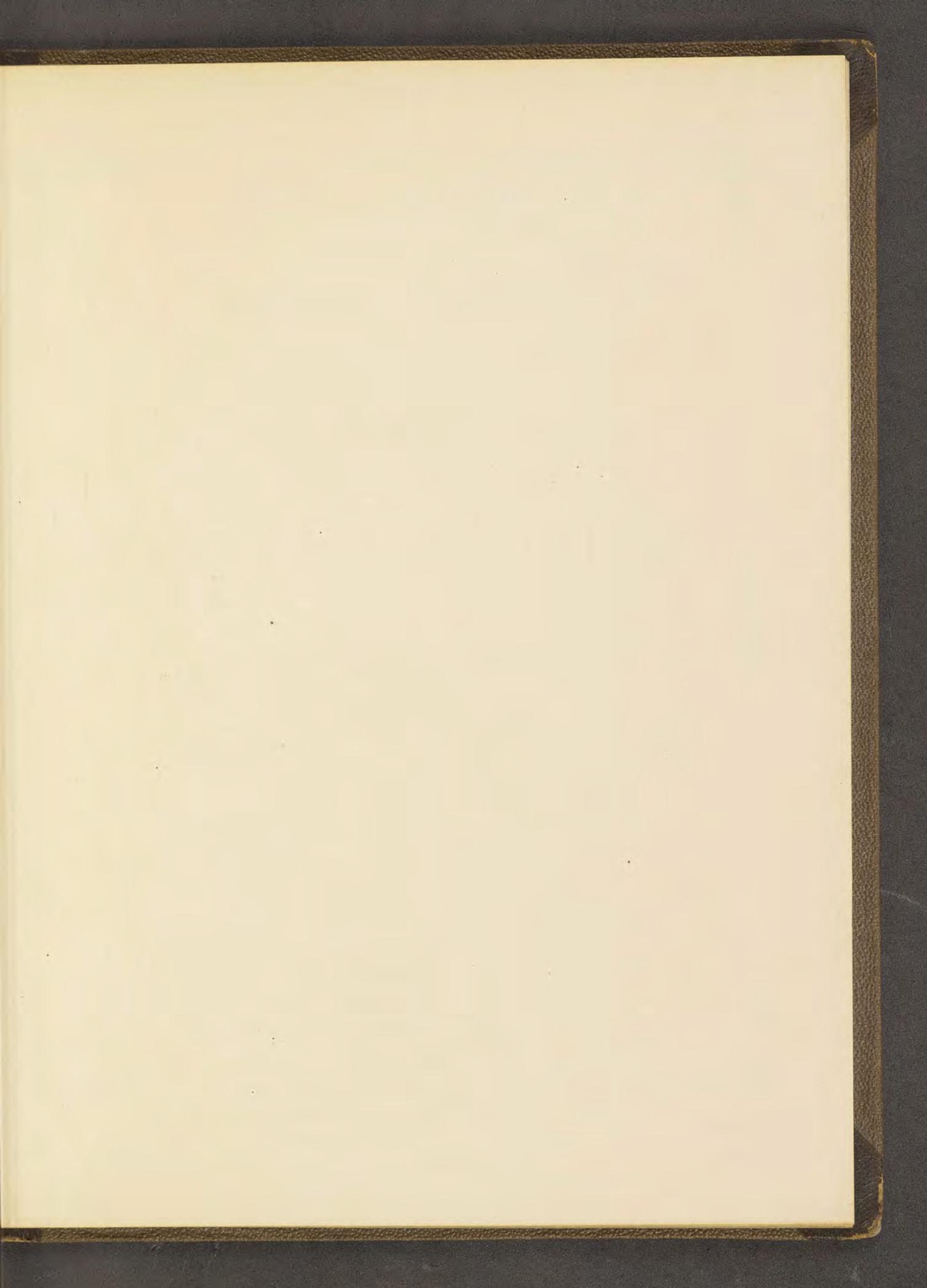
KRISTALLOGRAPHIE

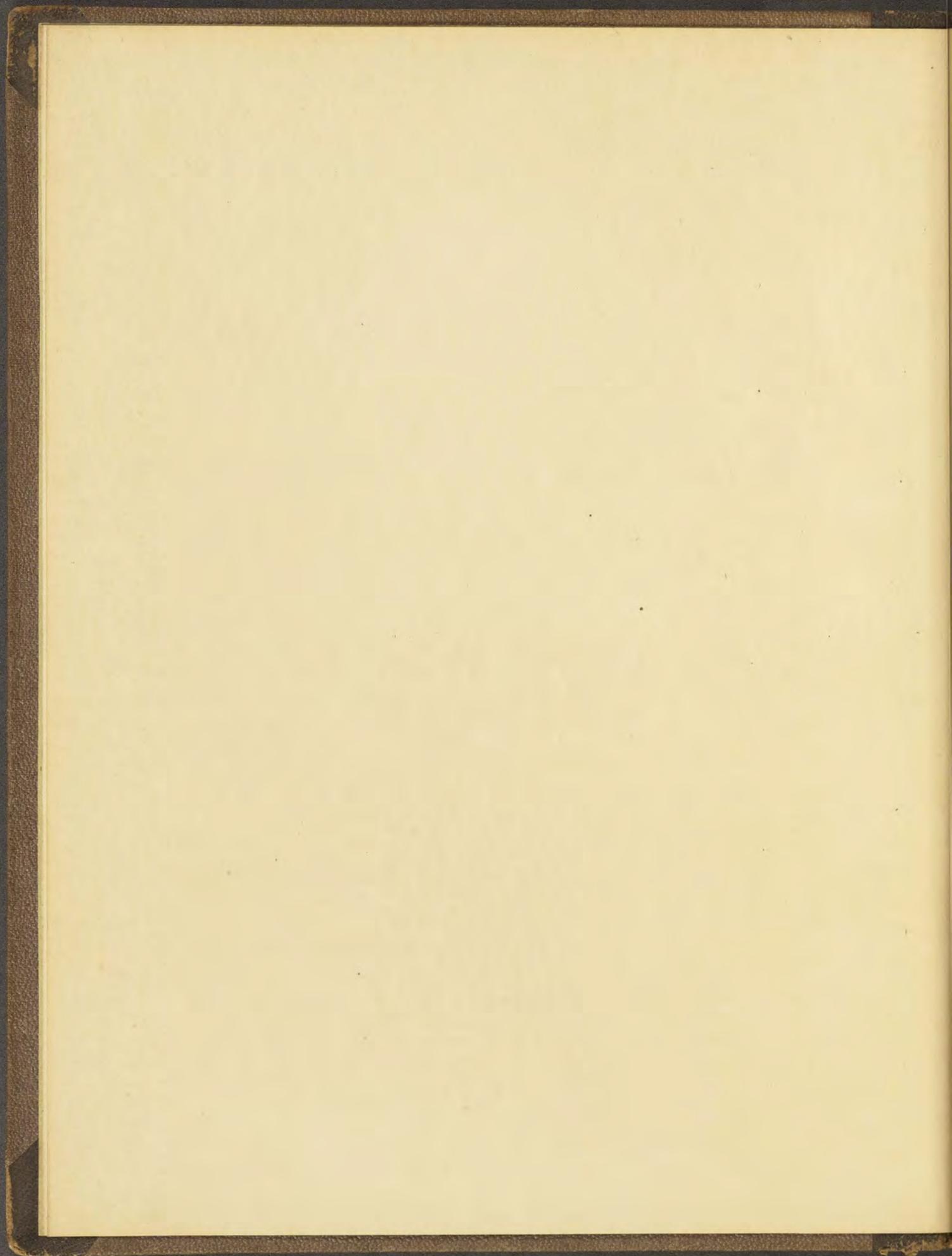
ea 5/11/85
PHB 4



H







00047970

M 3 a.

Kupfer - Tafeln

zum

Lehrbuch

der

KRYSTALLOGRAPHIE

von

Professor W. H. MILLER.

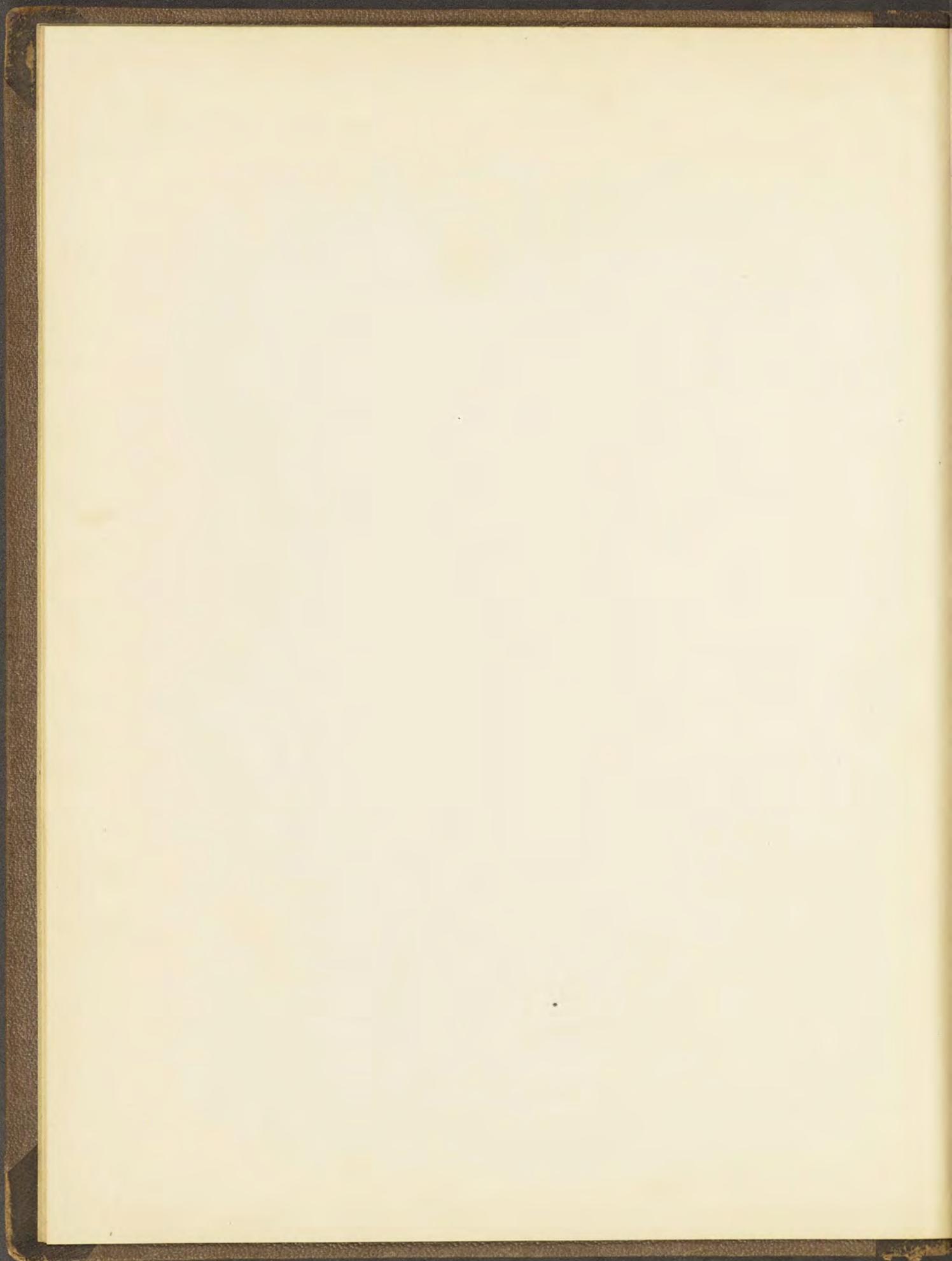
Uebersetzt von Dr. J. Grailich.

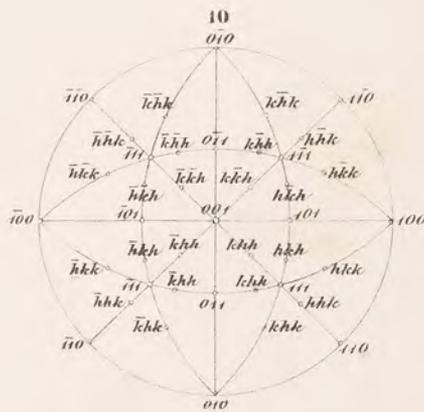
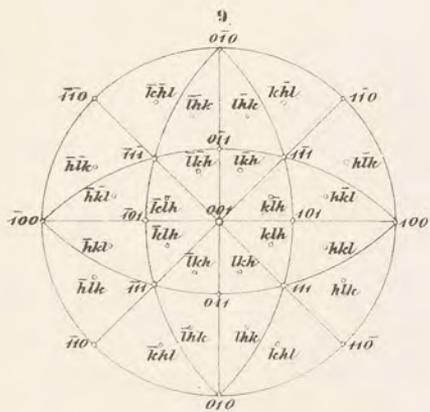
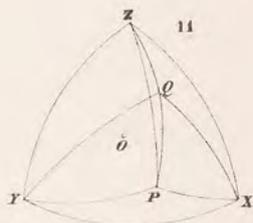
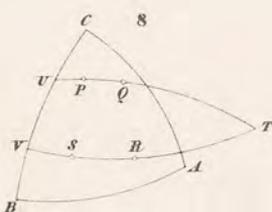
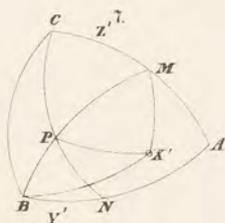
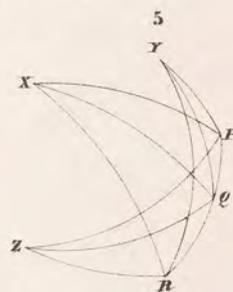
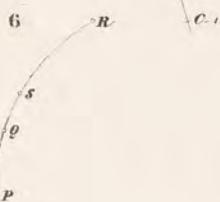
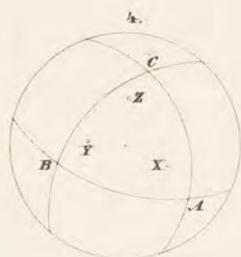
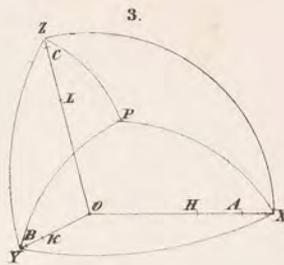
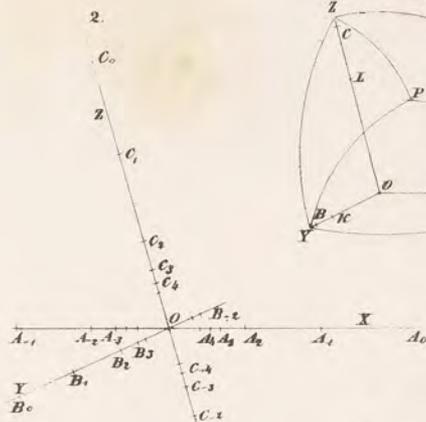
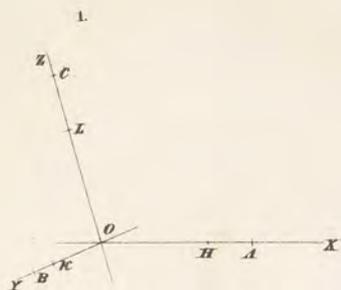
Enthaltend: Tafel I — XIX.

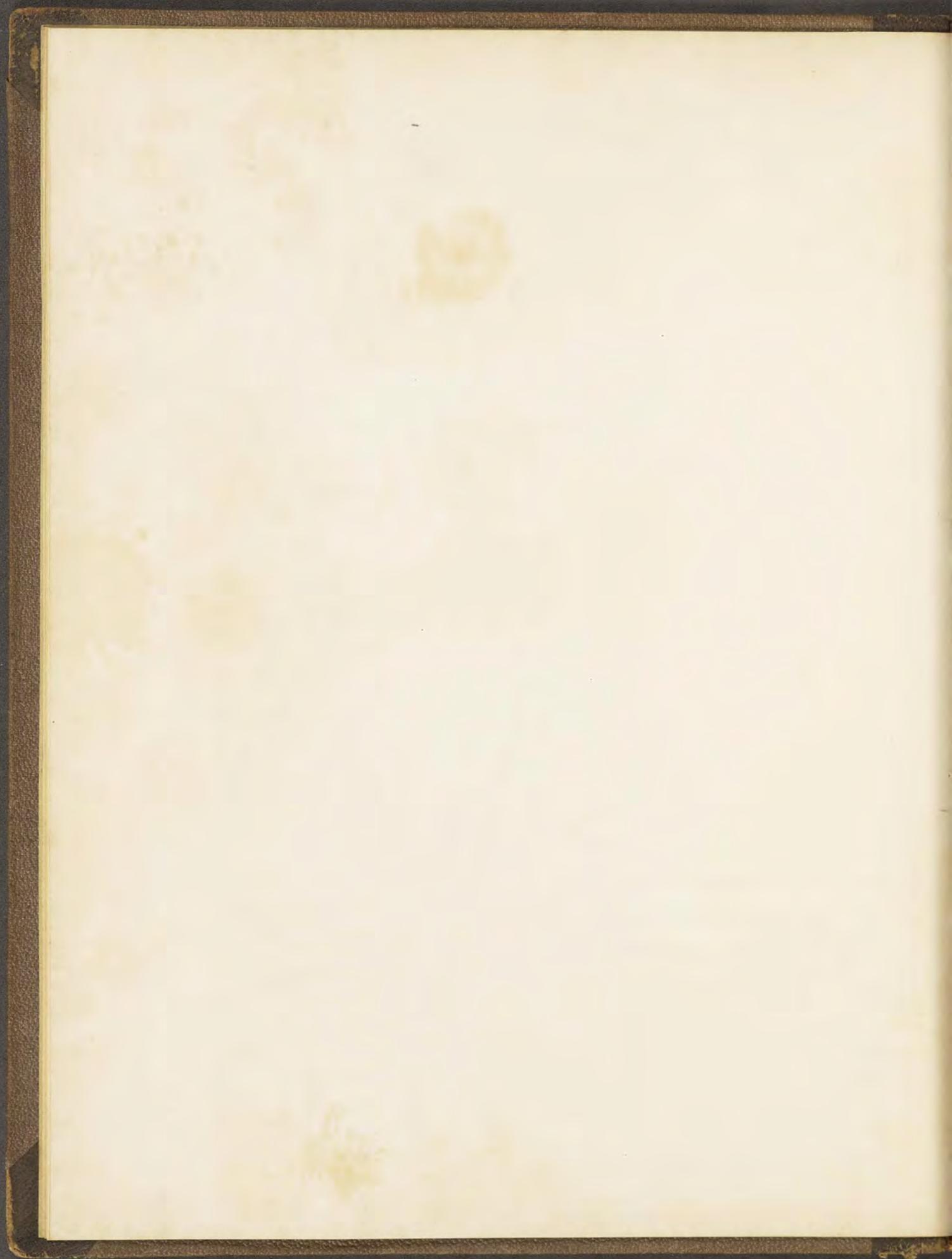
WIEN.

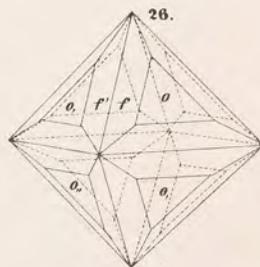
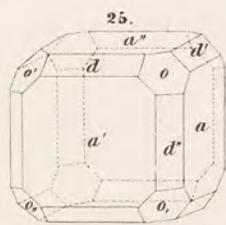
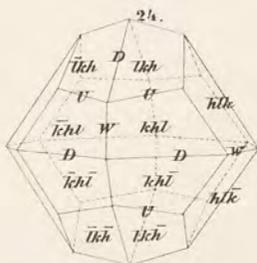
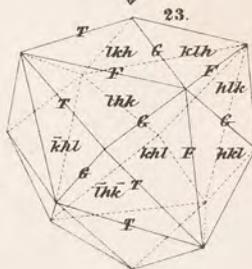
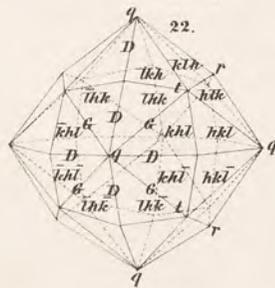
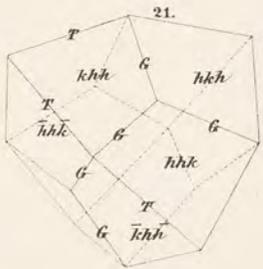
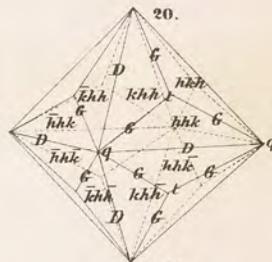
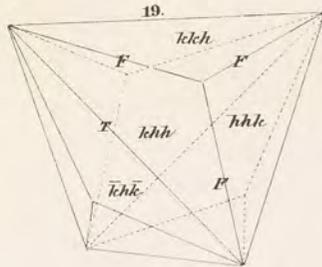
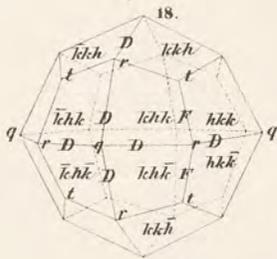
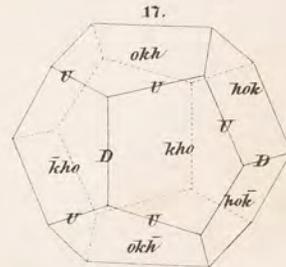
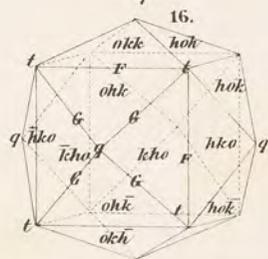
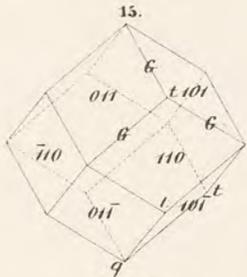
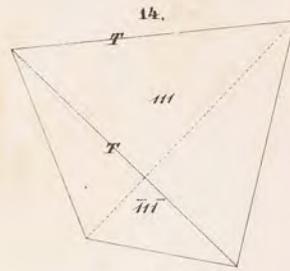
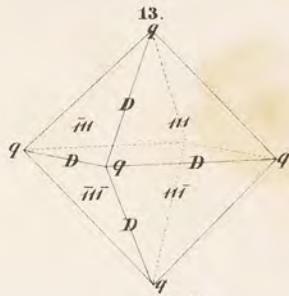
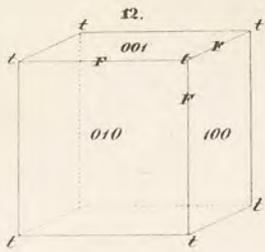
Verlag von CARL GEROLD'S SOHN.

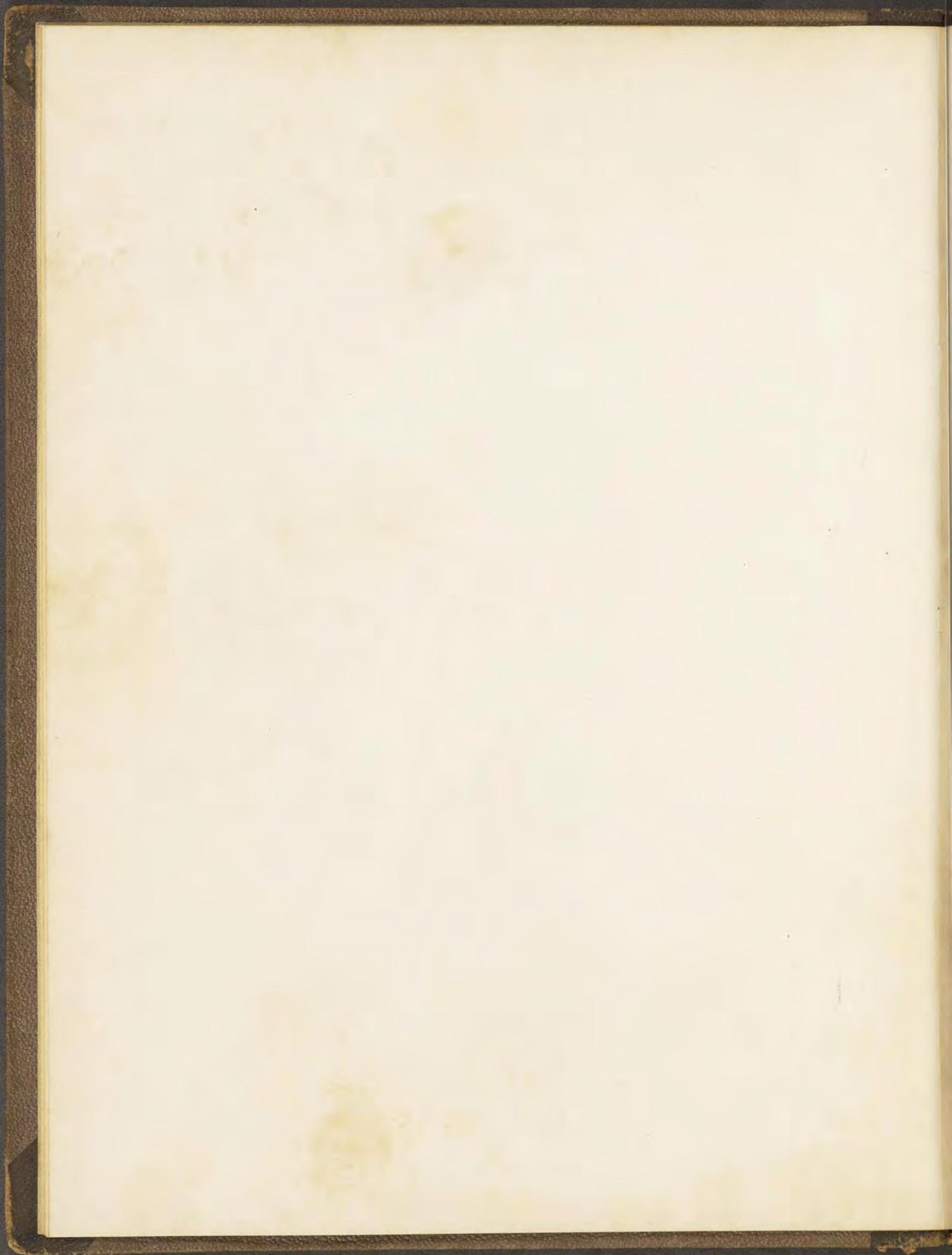
1856.

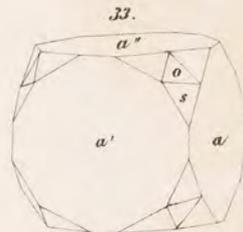
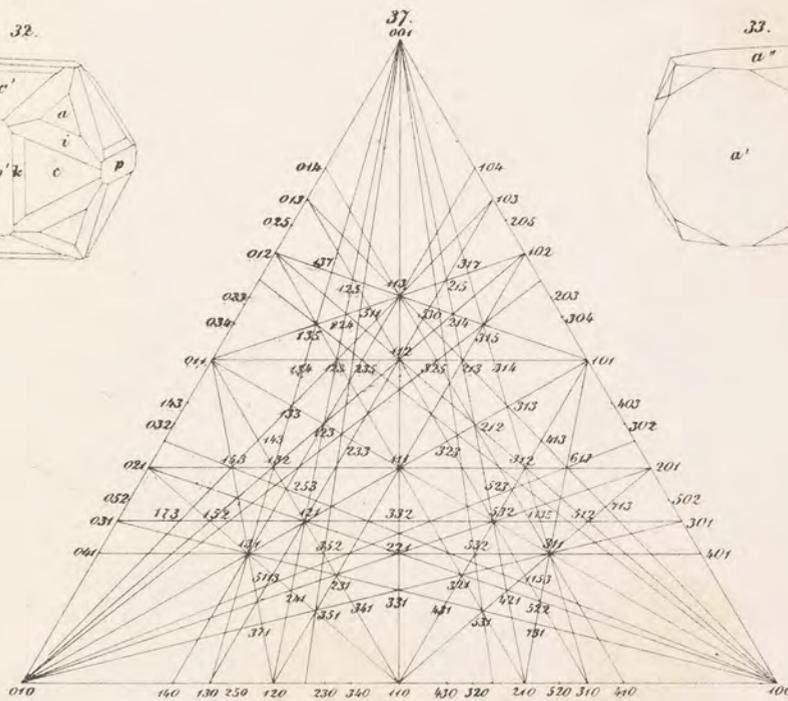
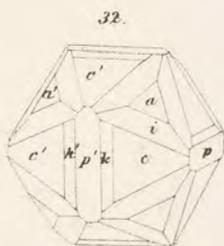
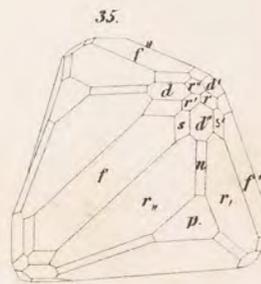
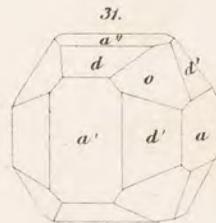
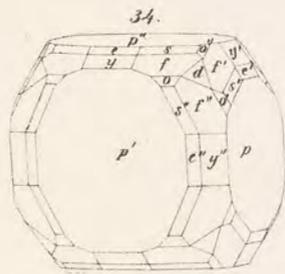
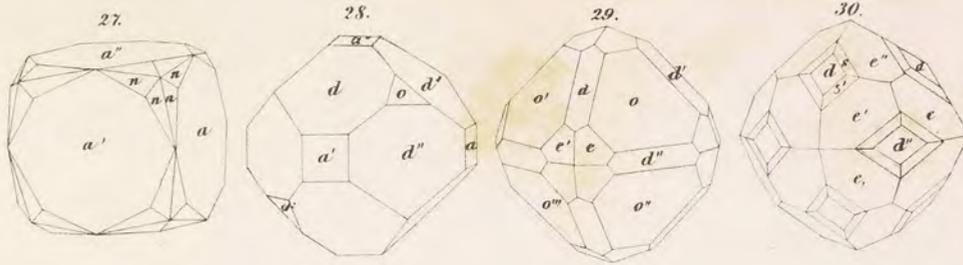


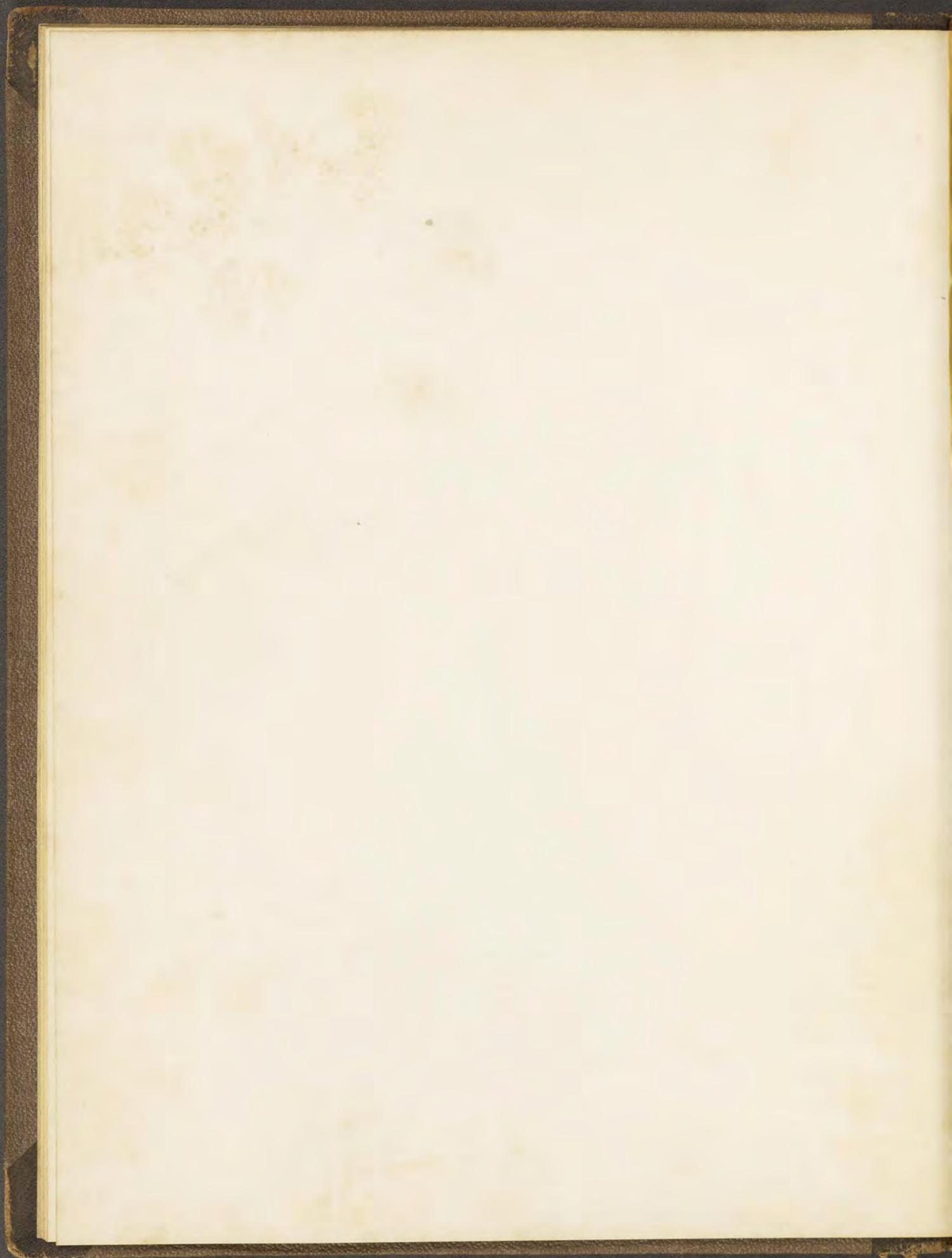


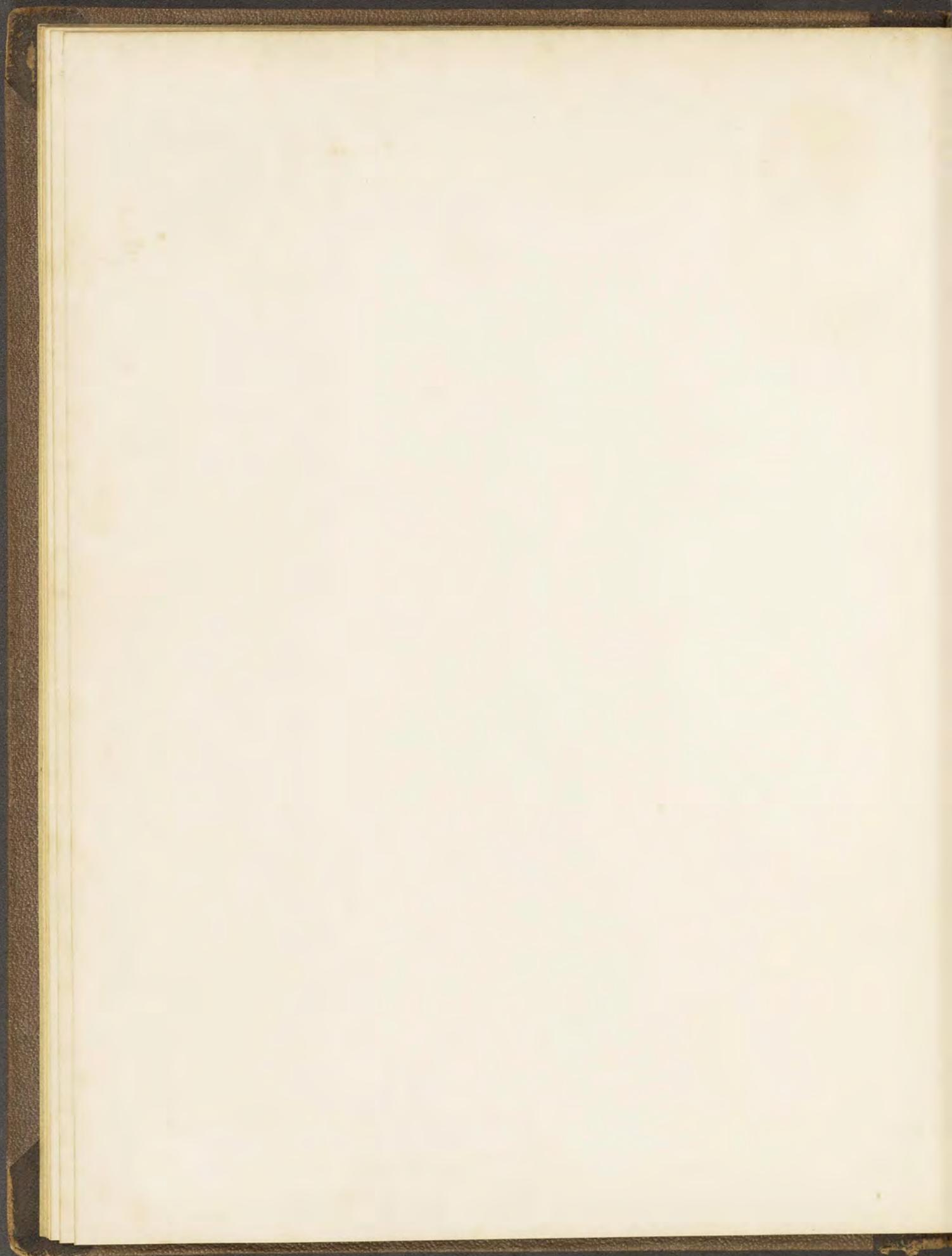


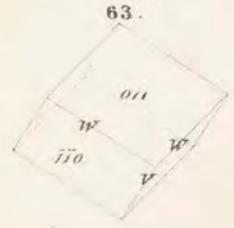
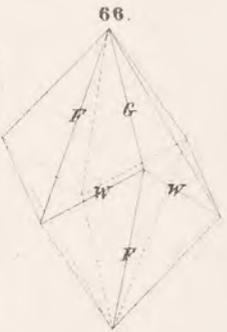
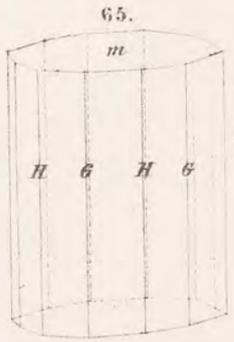
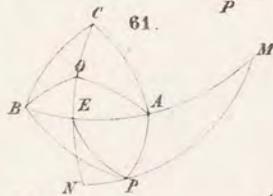
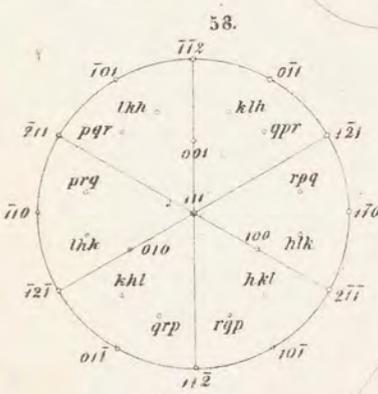
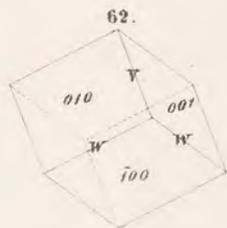
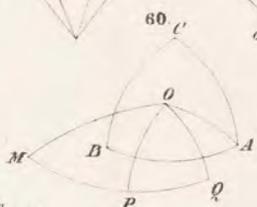
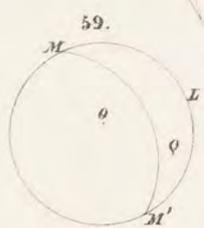
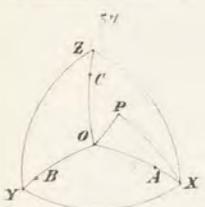
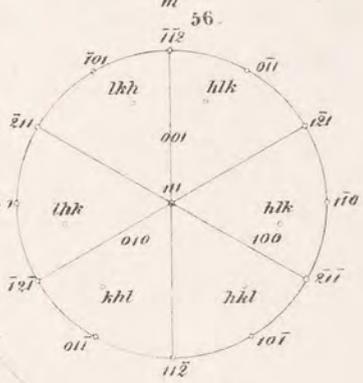
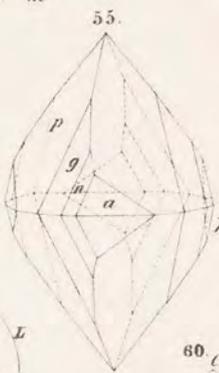
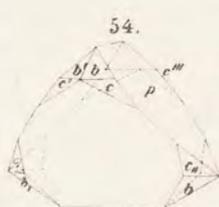
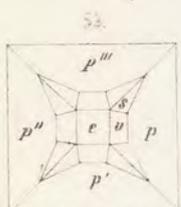
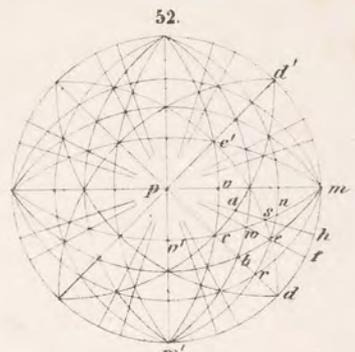
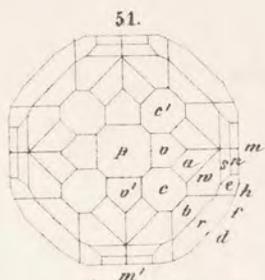
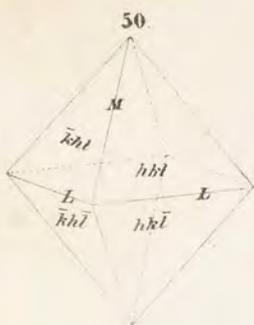


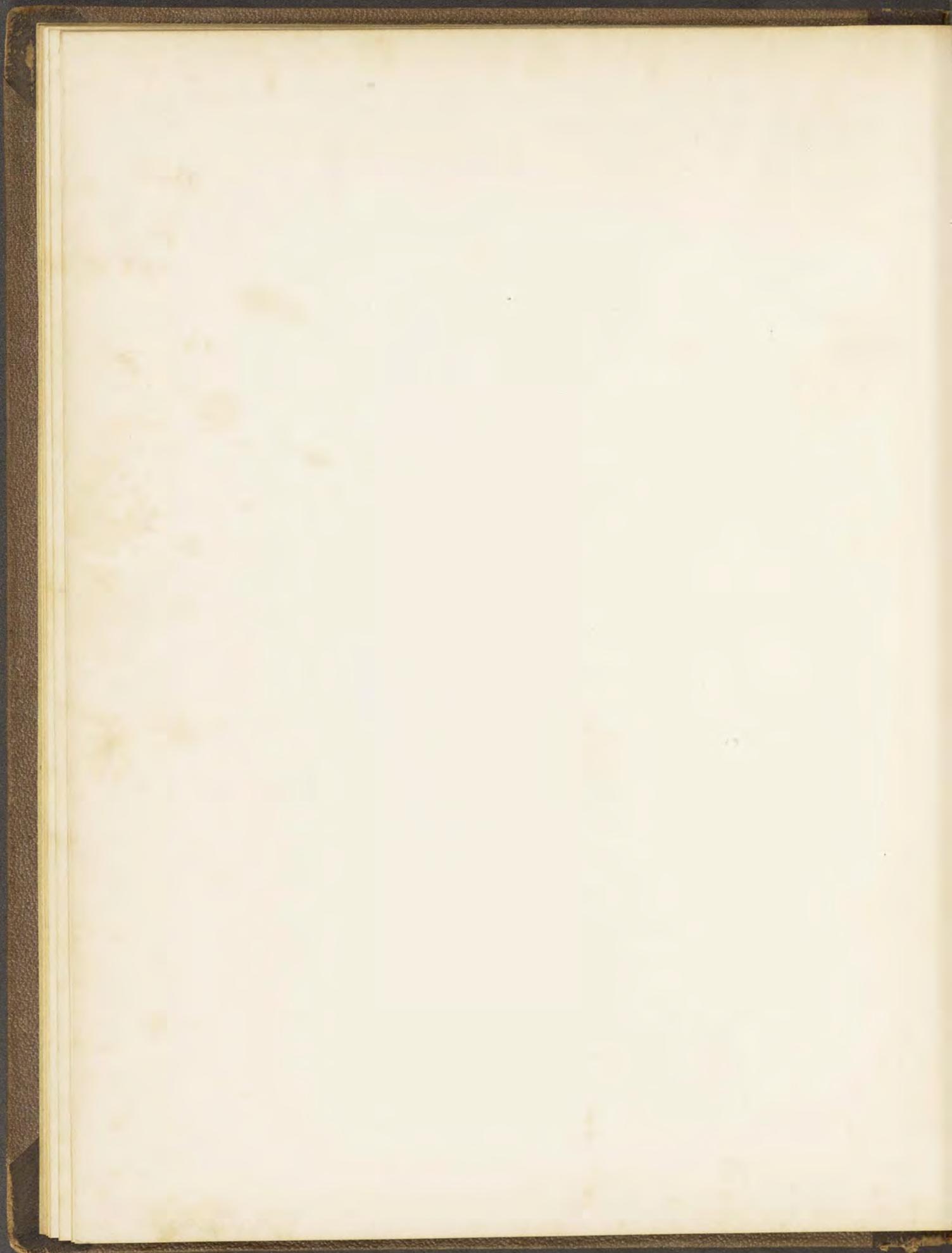


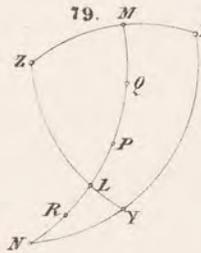
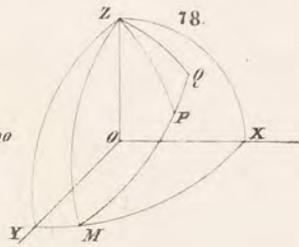
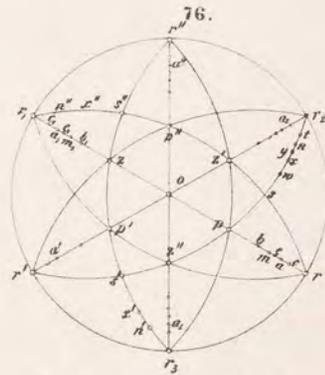
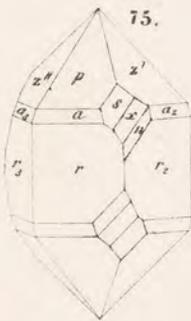
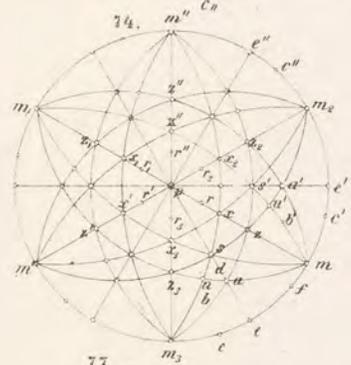
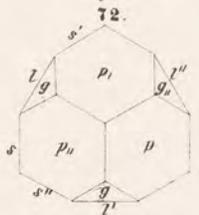
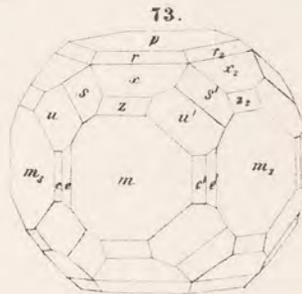
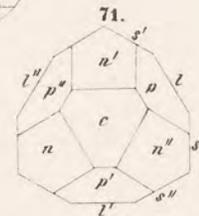
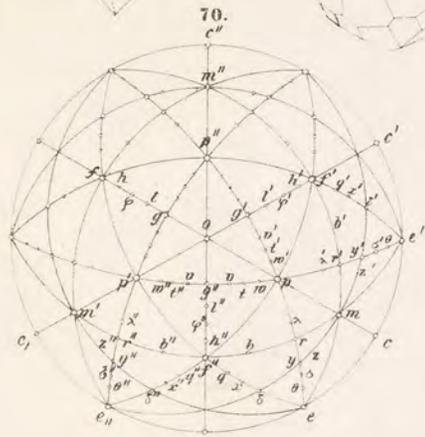
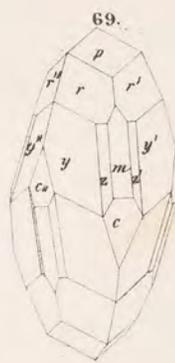
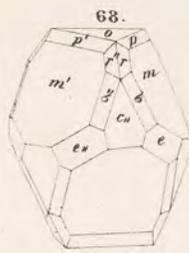
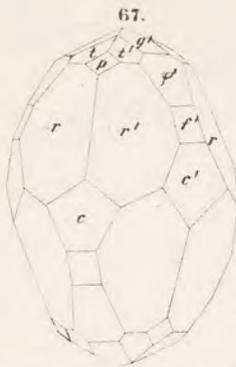
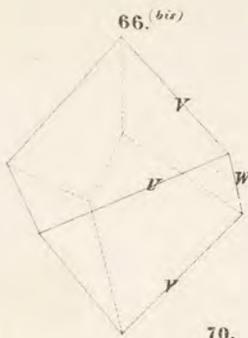


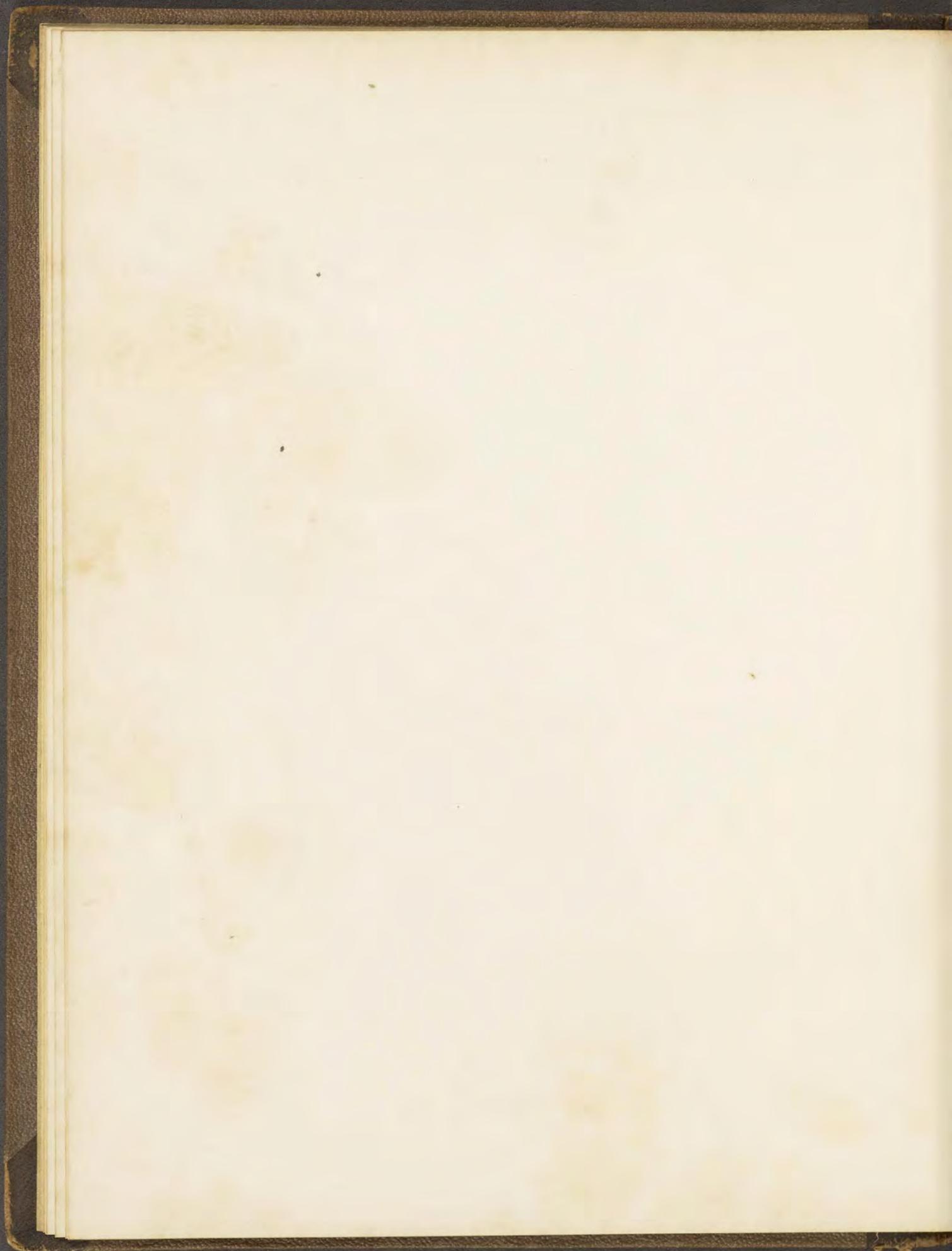


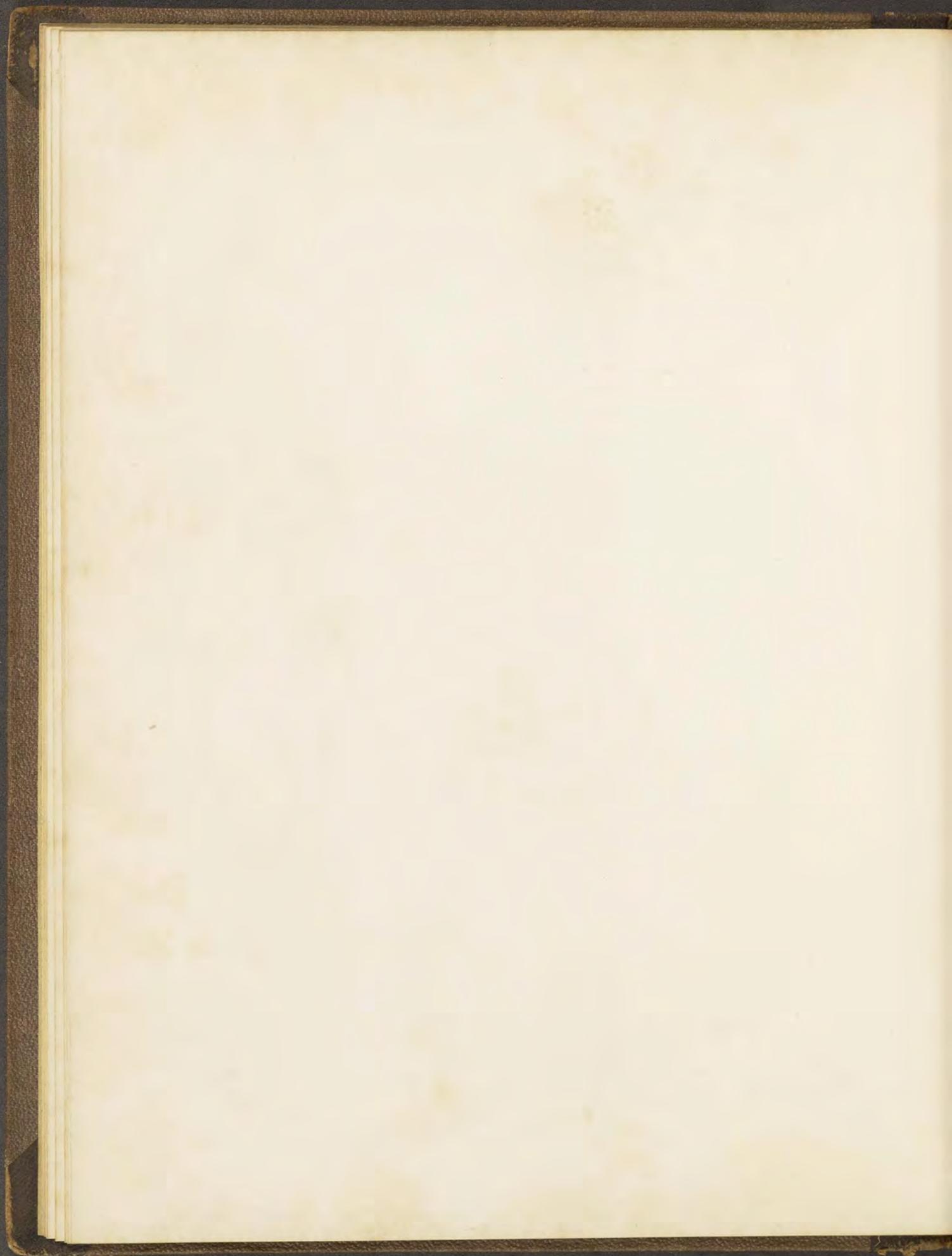


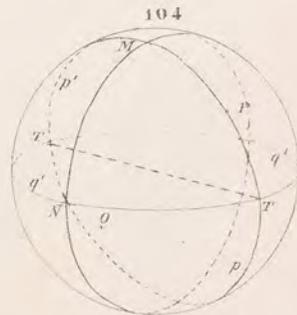
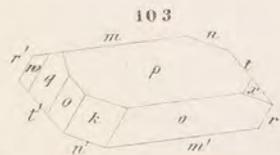
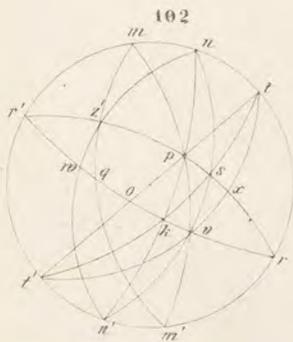
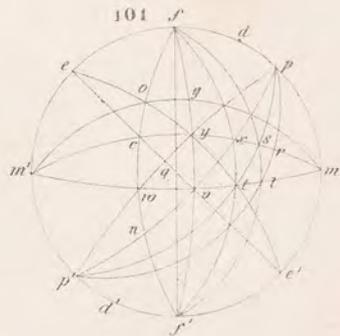
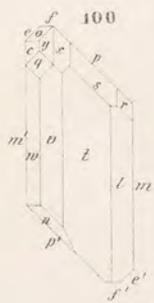
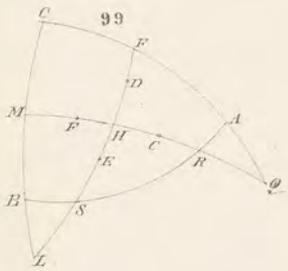
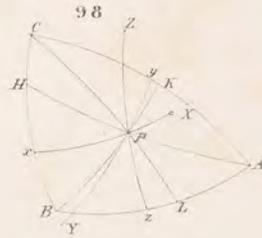
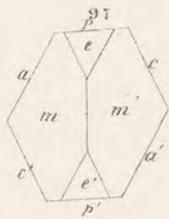
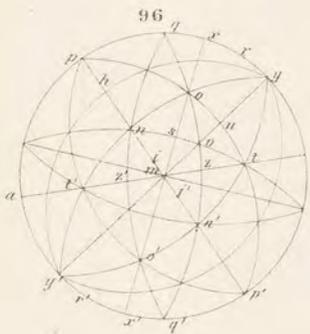
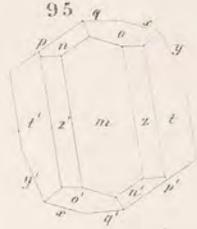
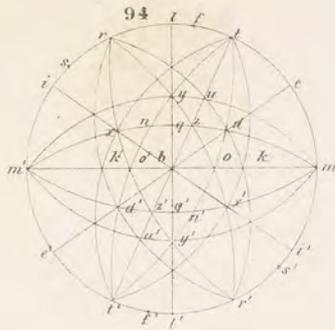
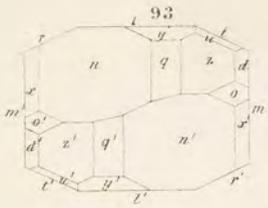


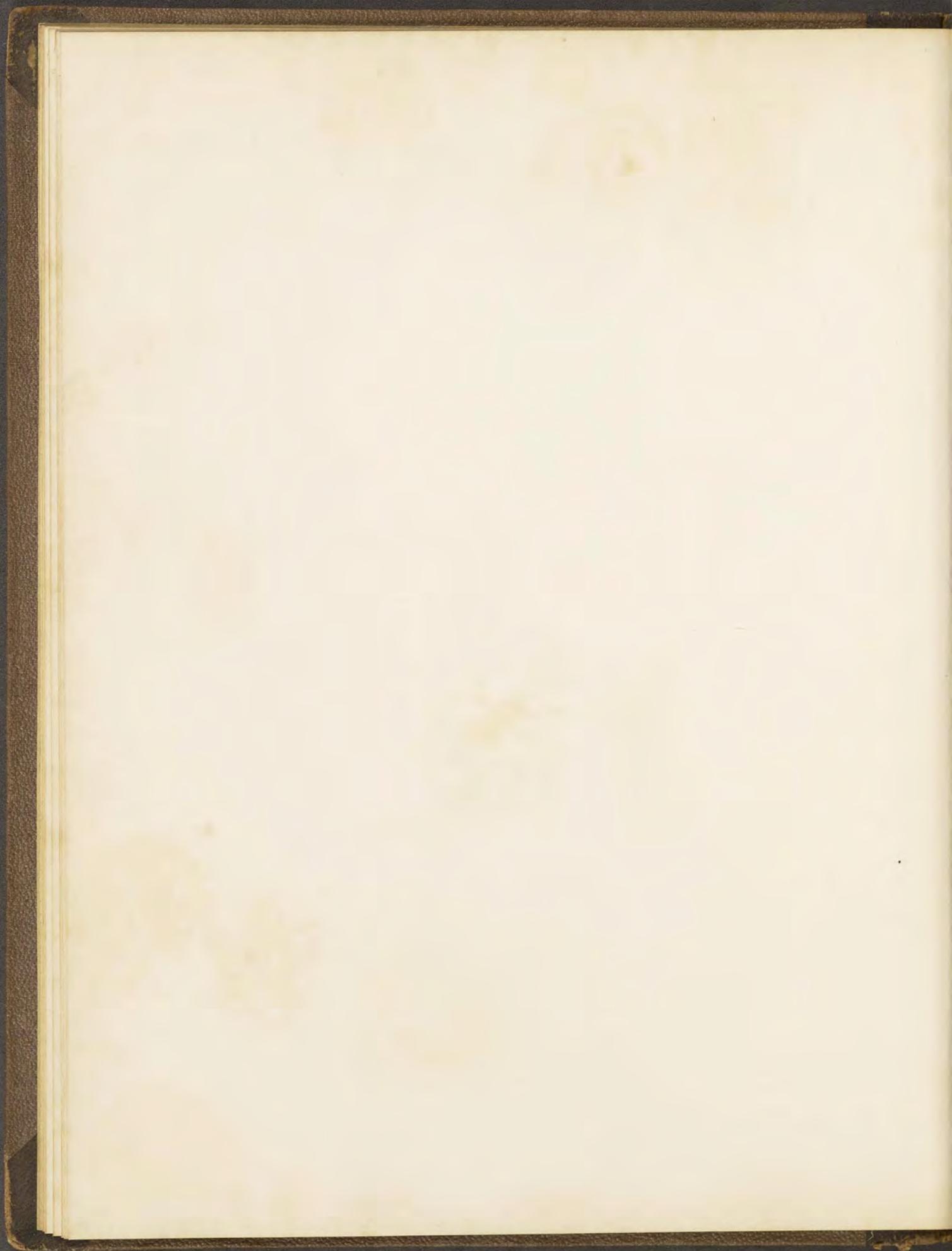




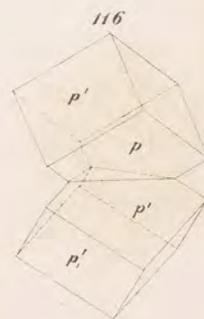
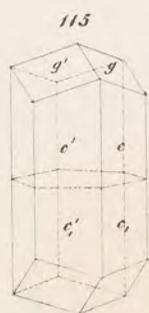
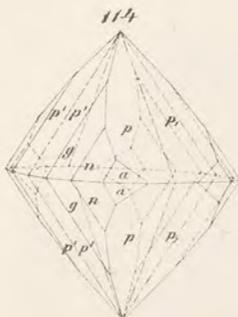
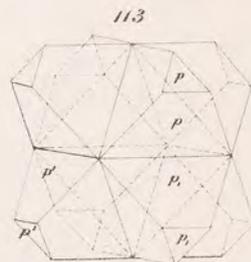
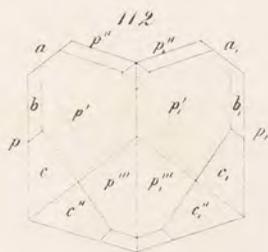
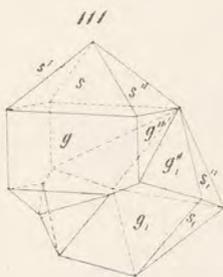
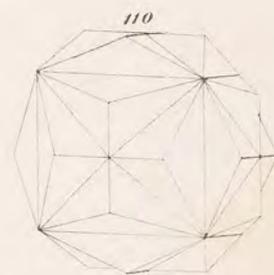
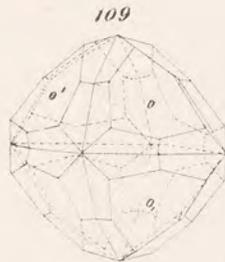
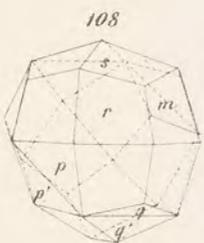
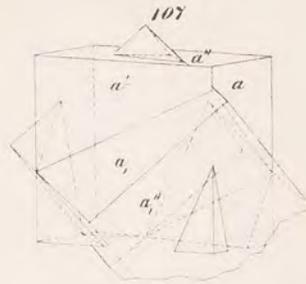
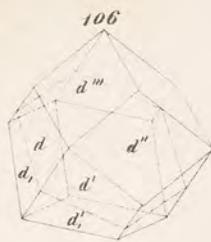
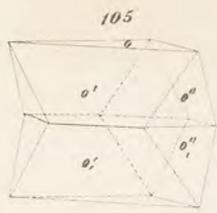






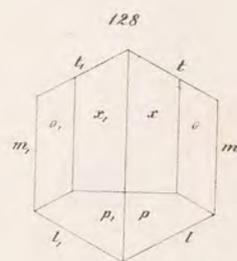
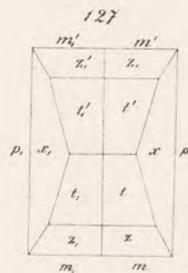
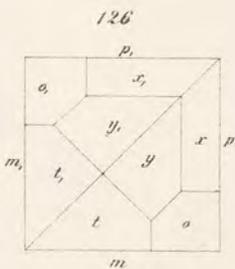
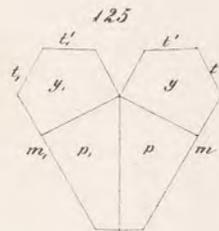
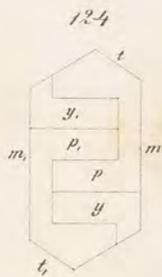
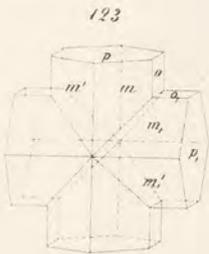
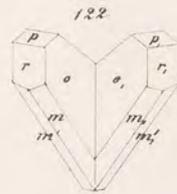
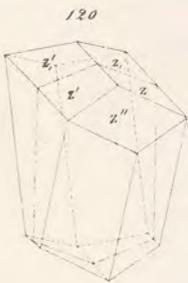
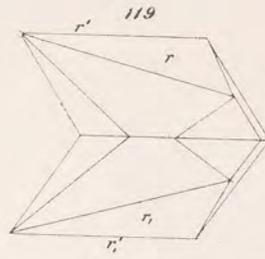
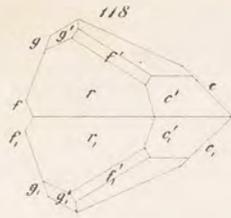
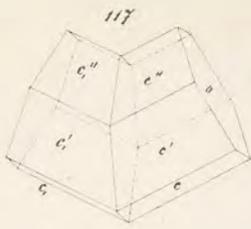


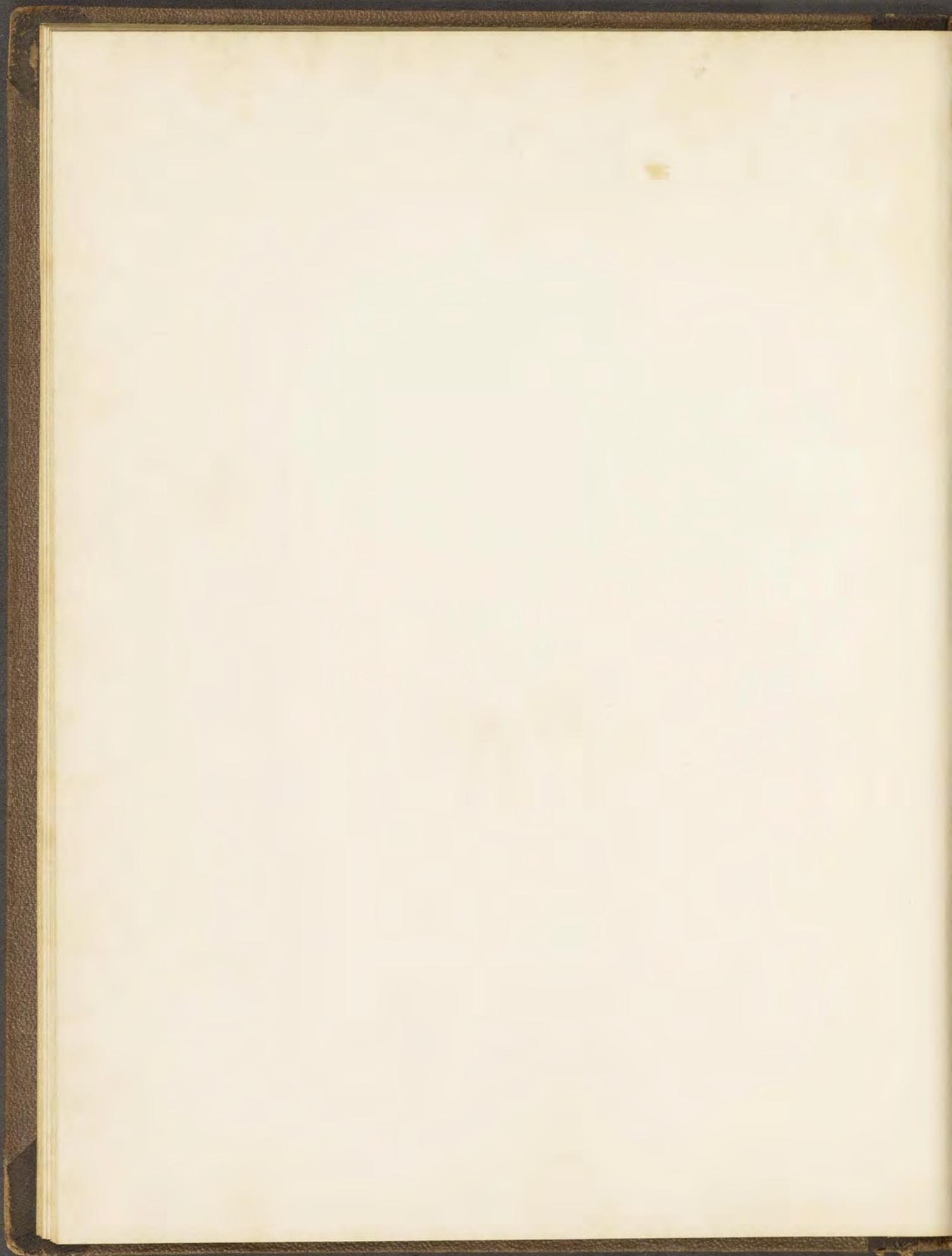
TAF. IX

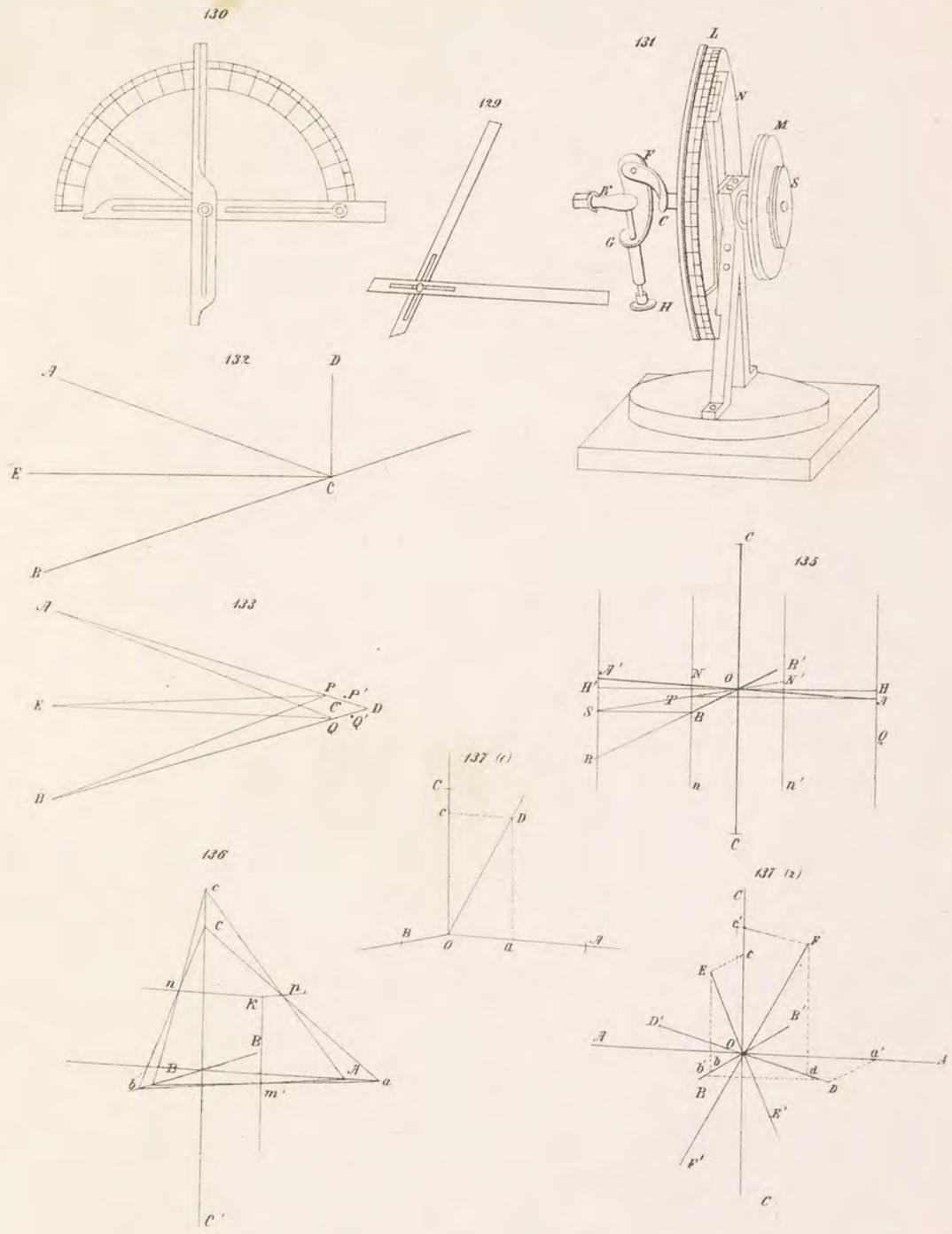


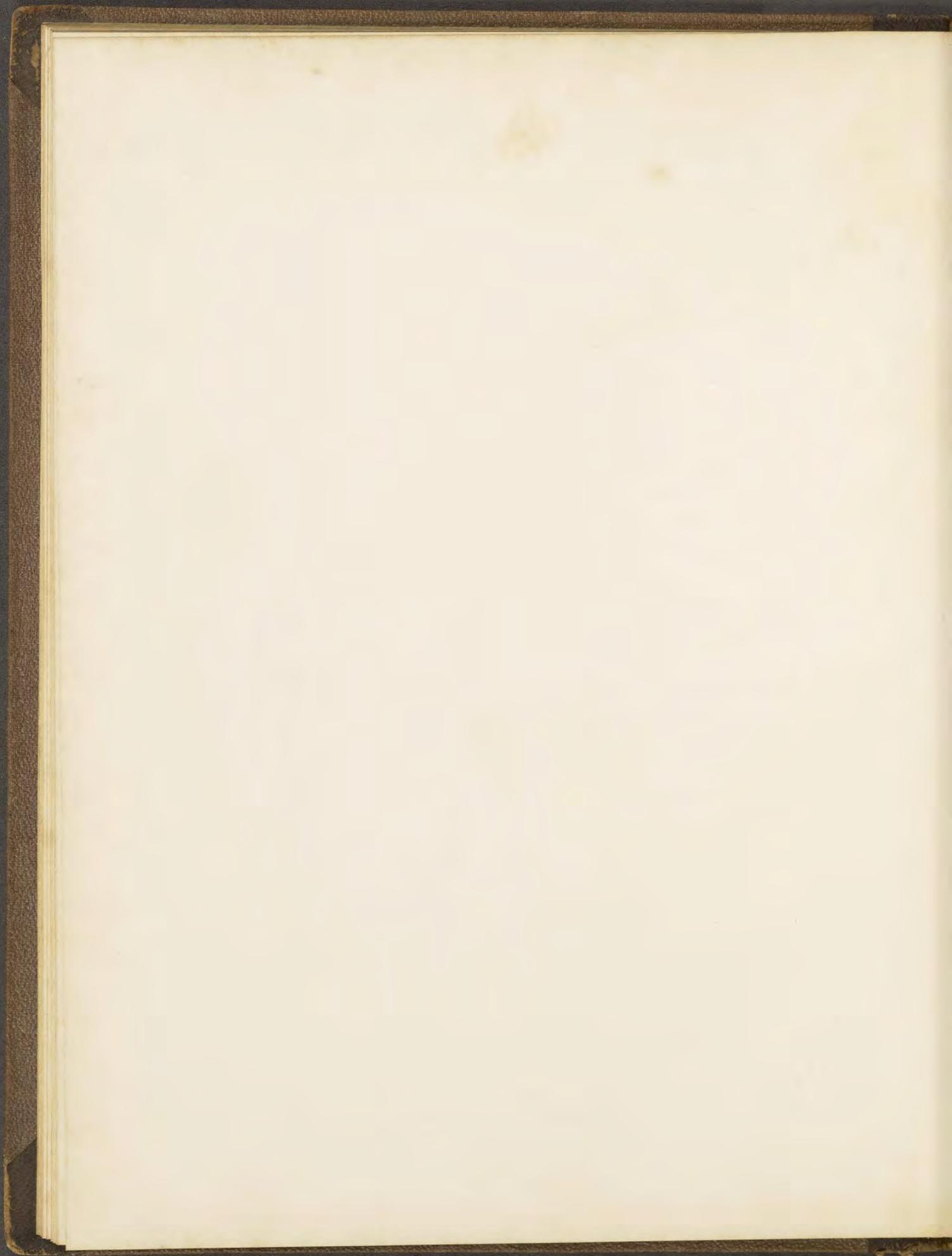


TAF. X



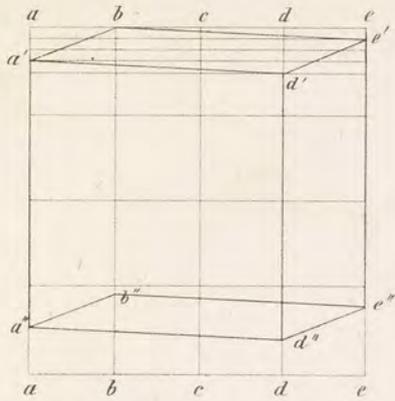




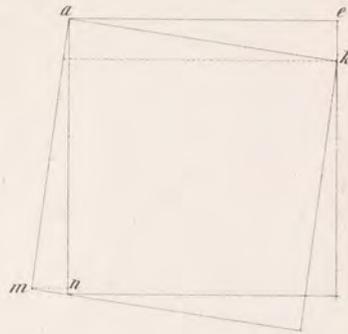


TAF. XII.

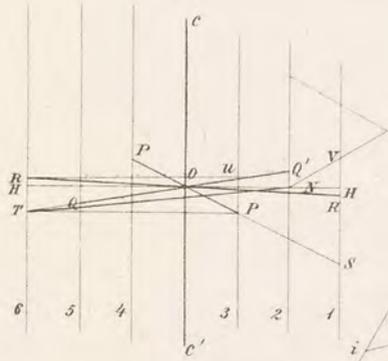
135 bis



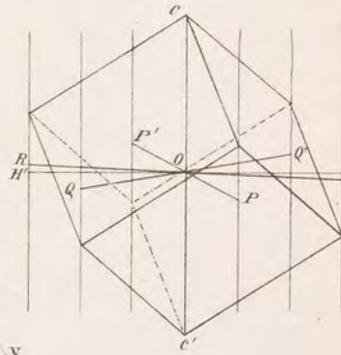
135 bis



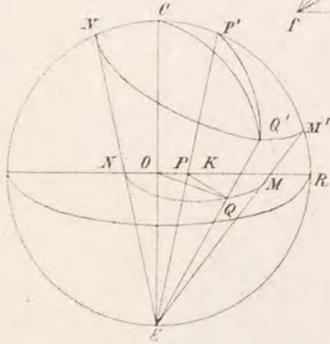
138



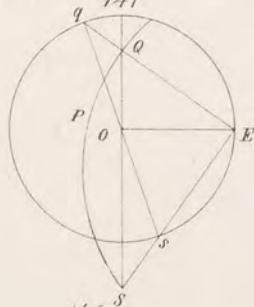
139



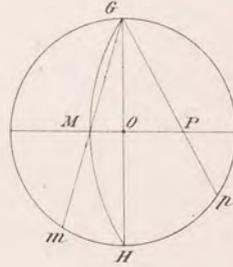
140



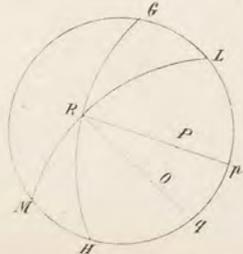
141



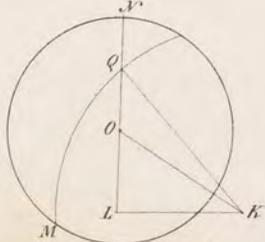
142



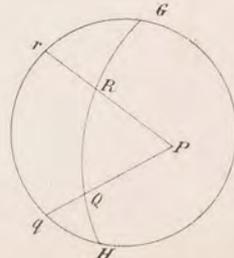
144

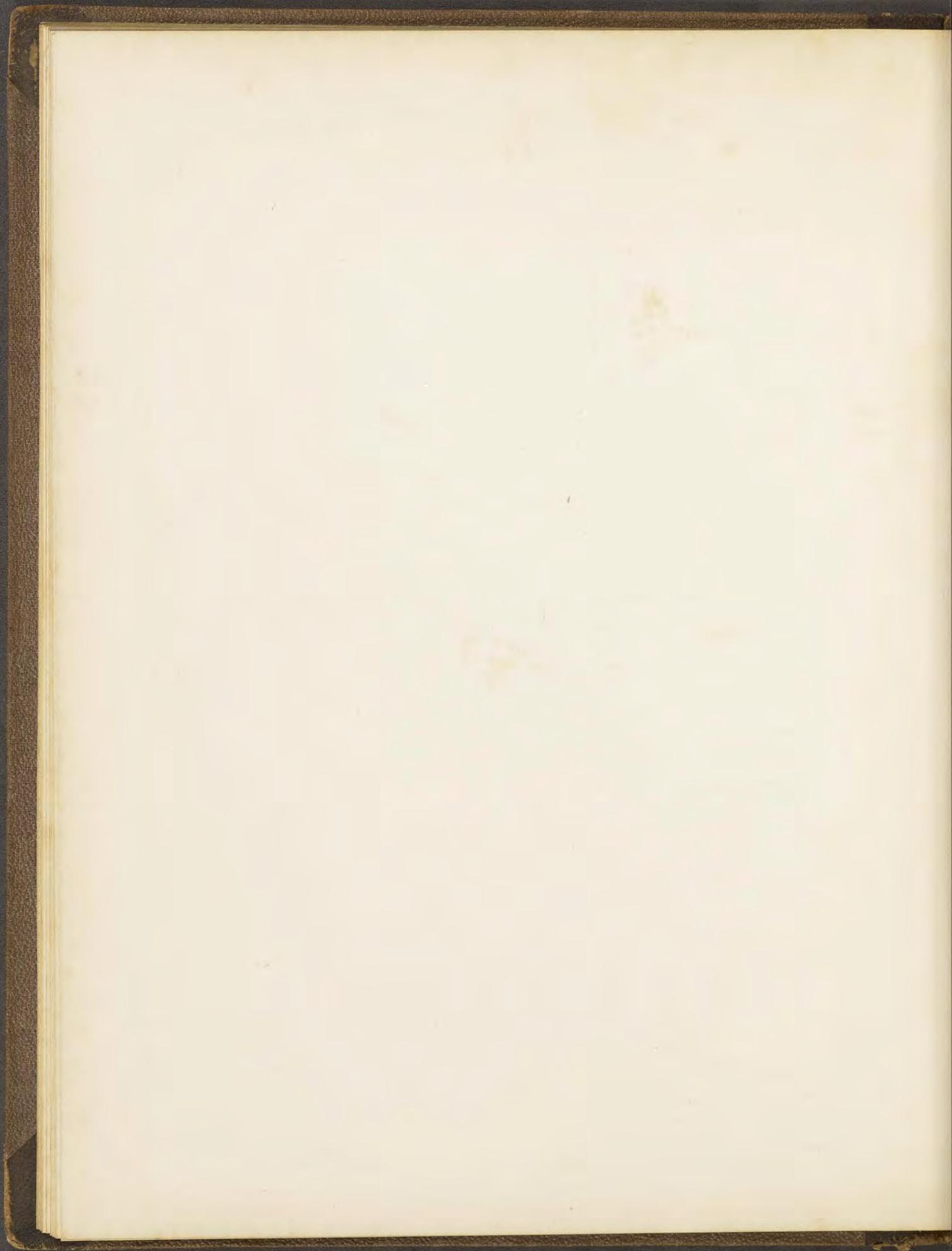


145

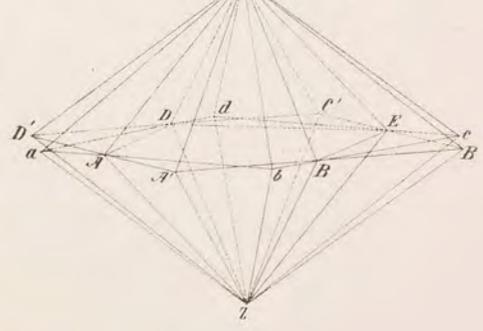
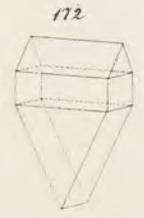
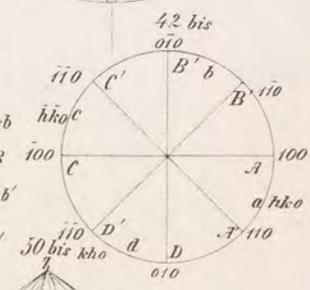
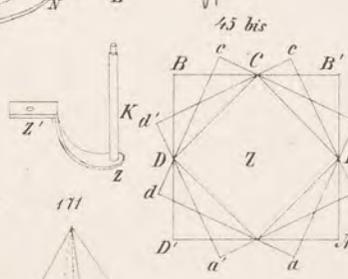
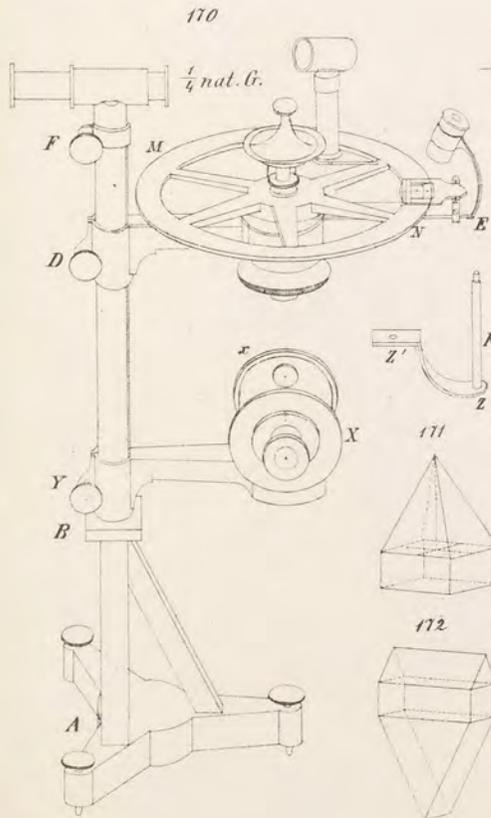
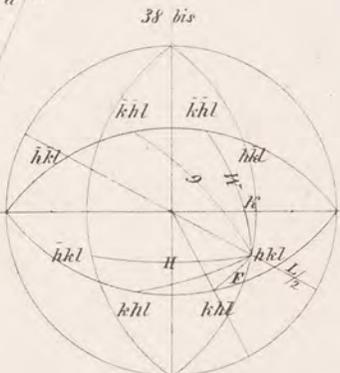
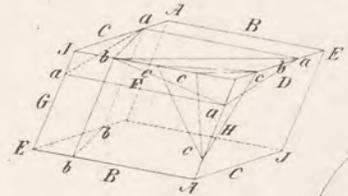
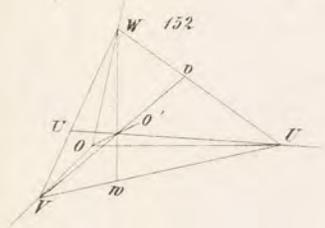
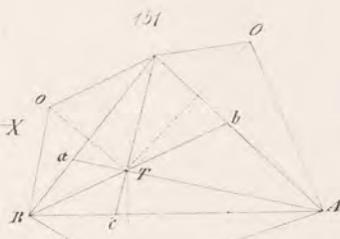
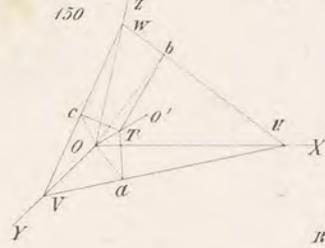
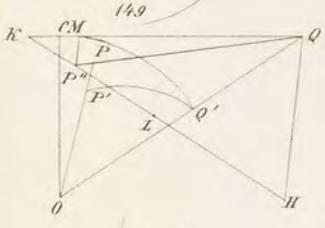
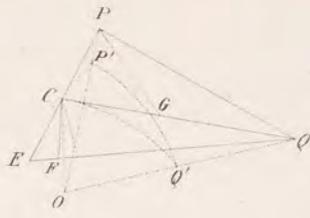
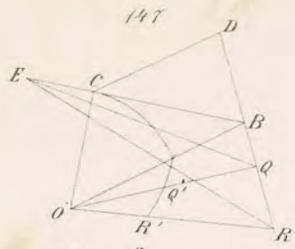
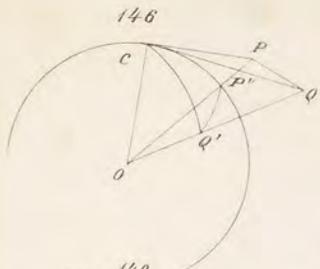


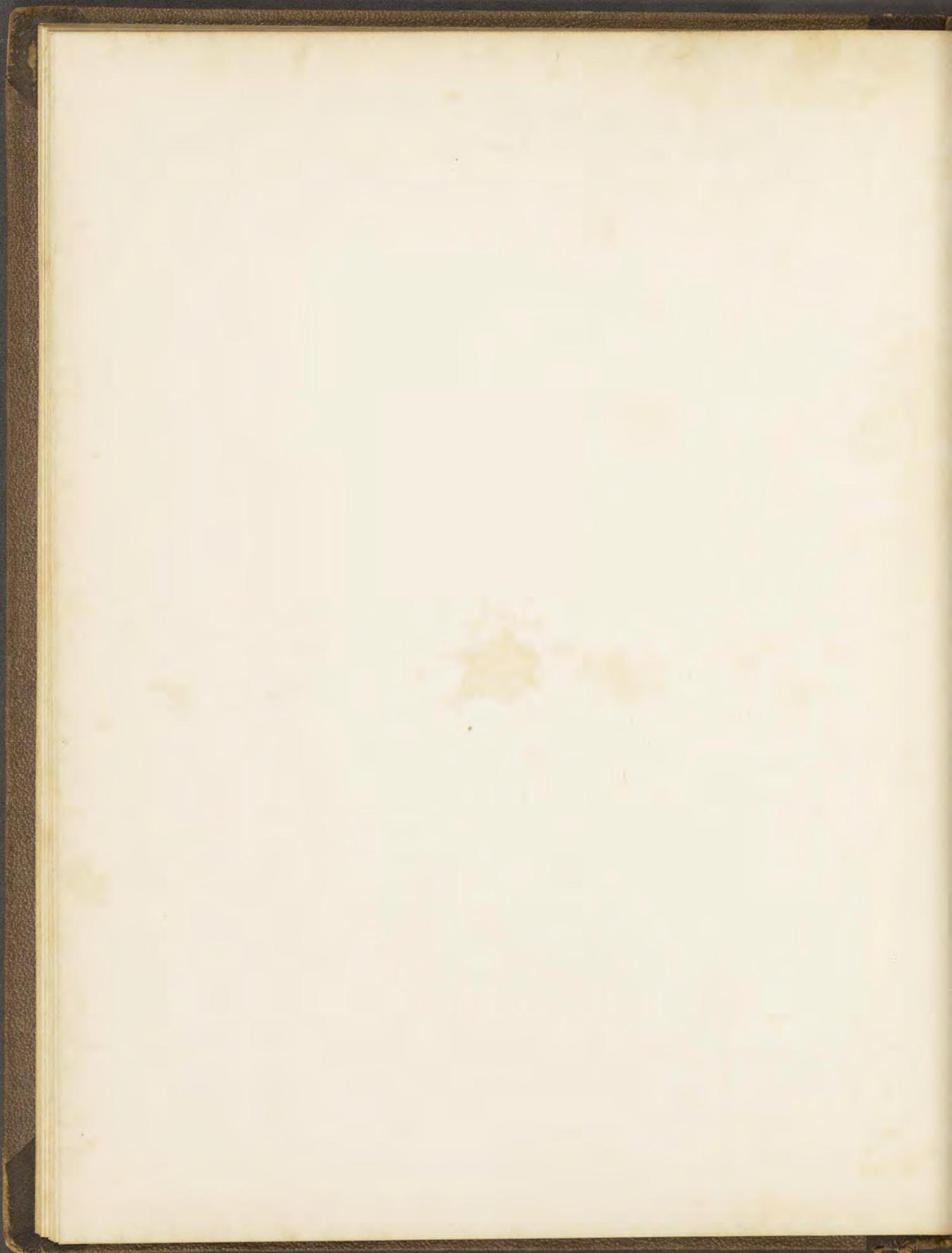
143





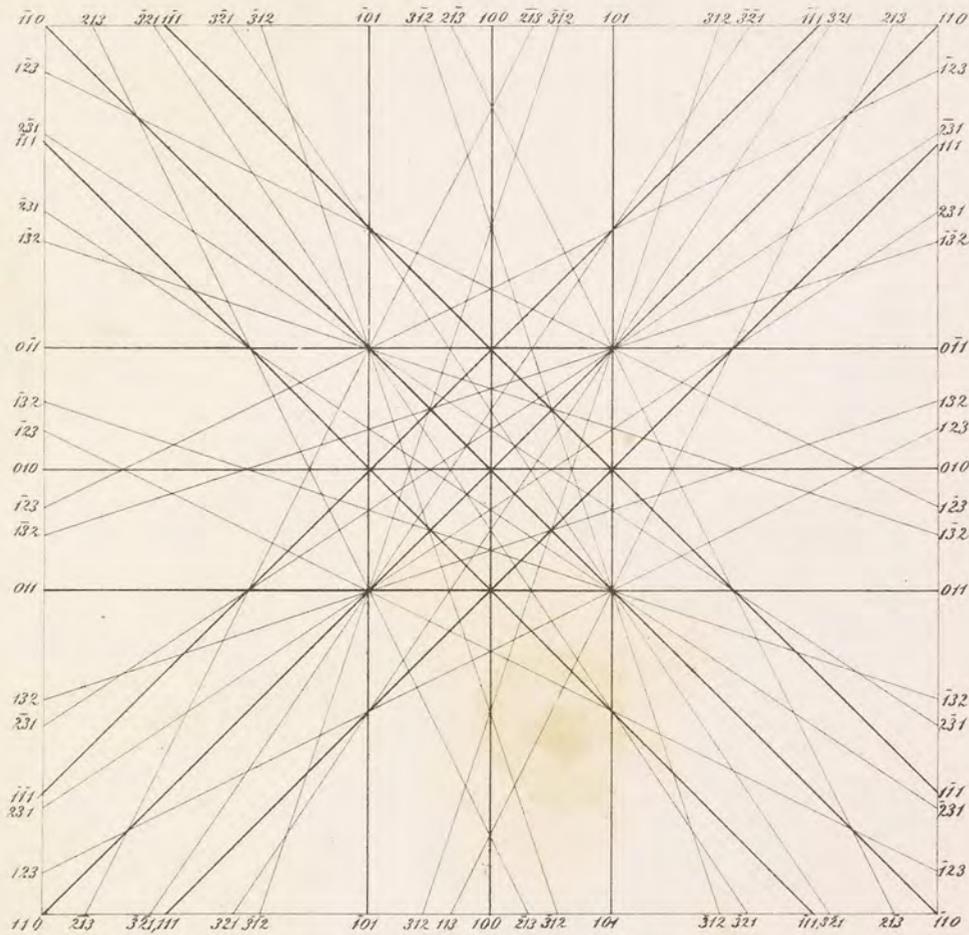
TAF. XIII.



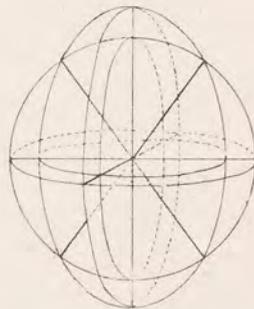




149 (ter)



159



162



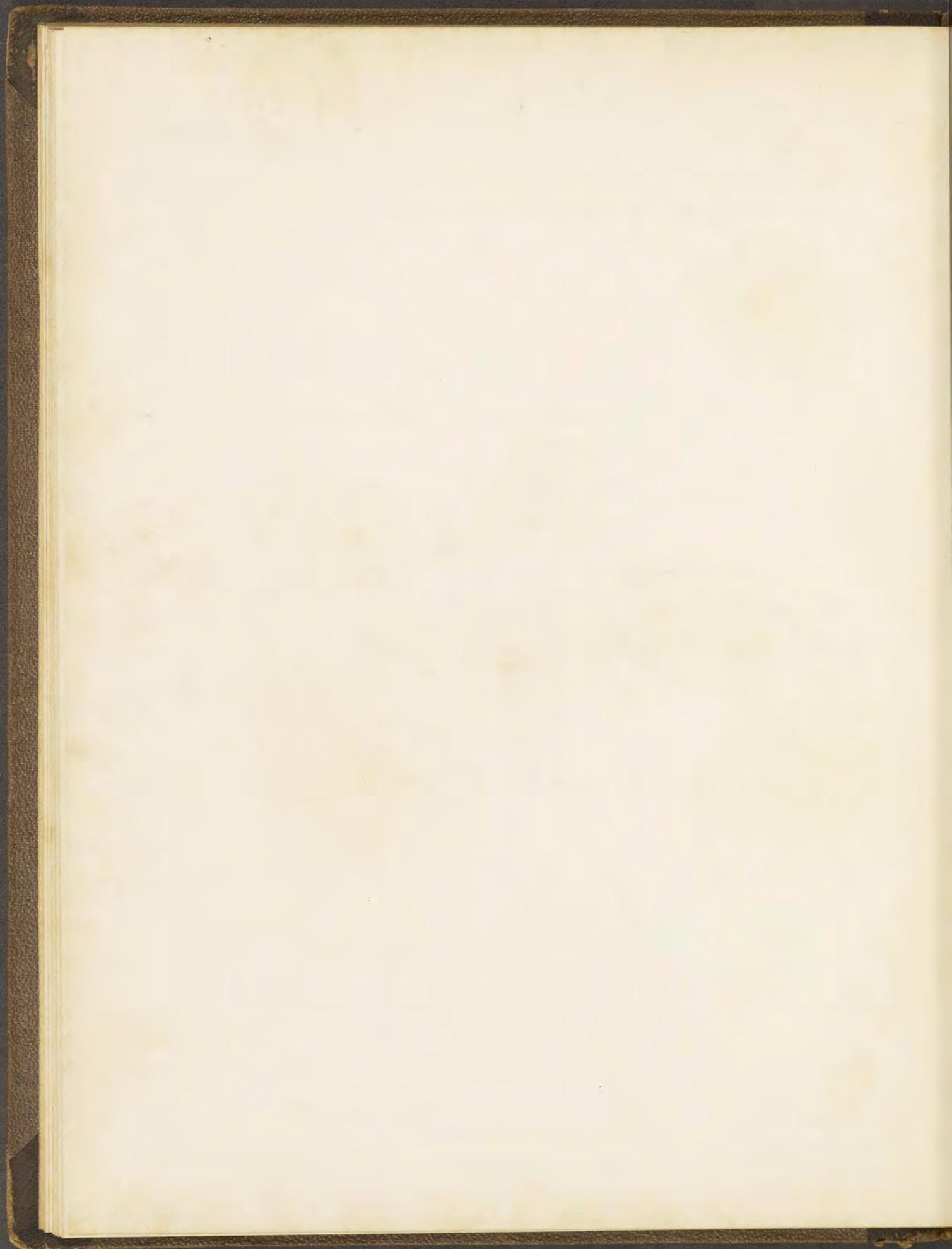
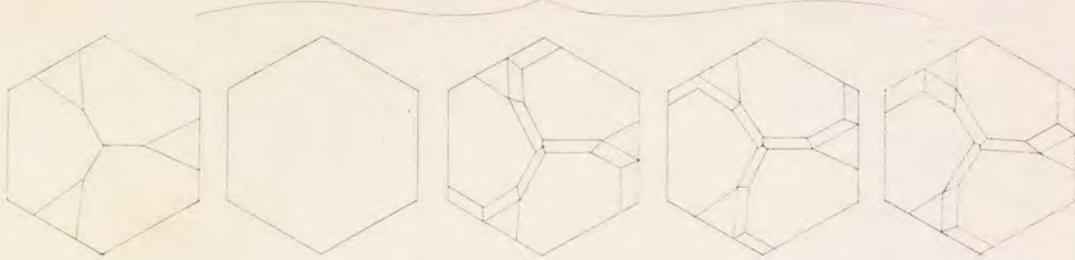
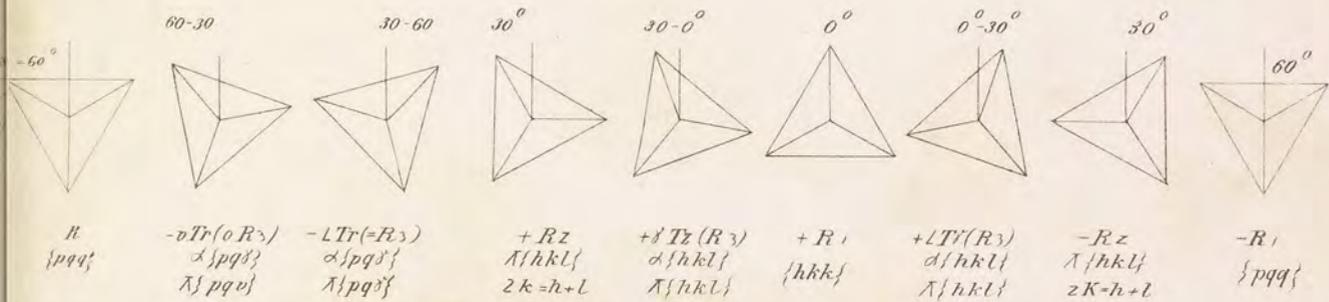


Fig. 1.



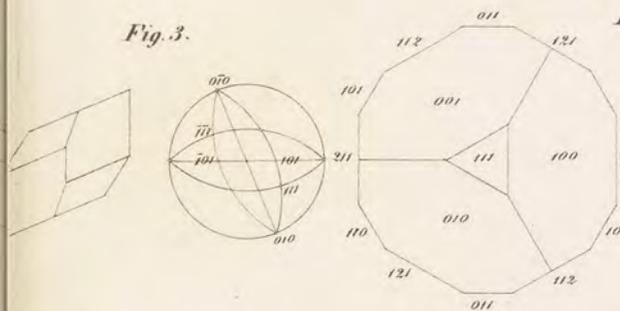
Cohäsion

Fig. 155.



Härte.

Fig. 3.



Glauberit

Wärme

Fig. 154.

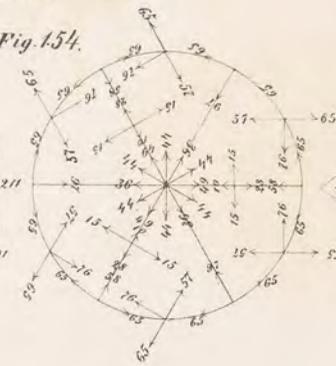
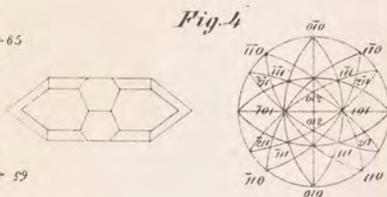
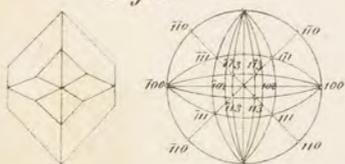


Fig. 4.



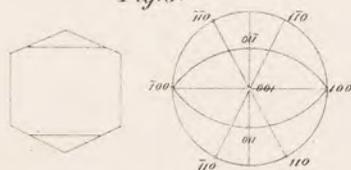
Wolfram

Fig. 5.



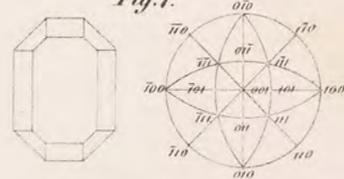
Antimonglanz

Fig. 6.

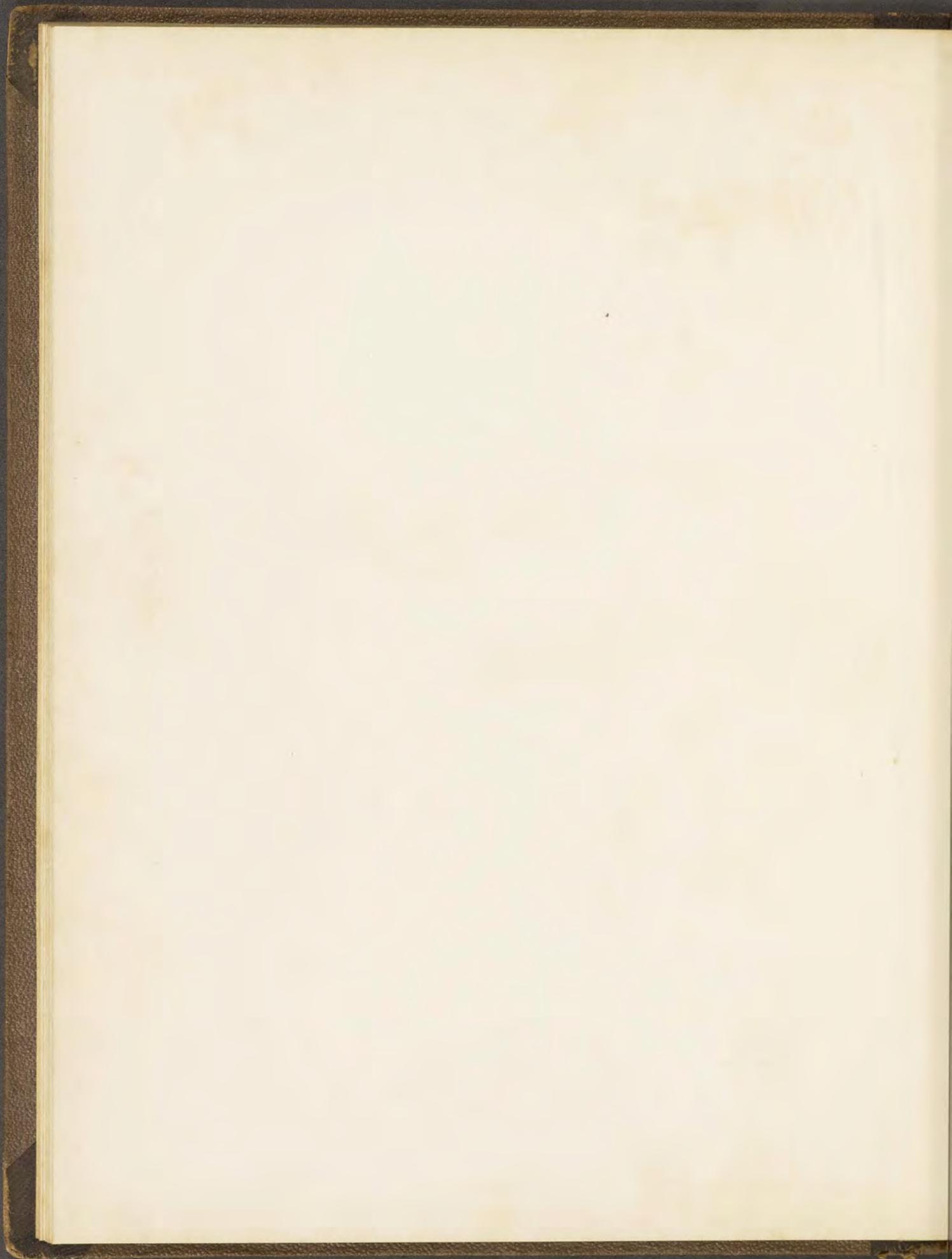


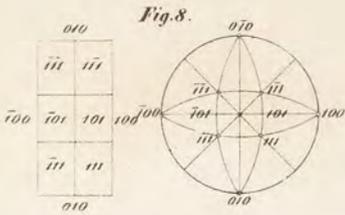
Staurolit

Fig. 7.

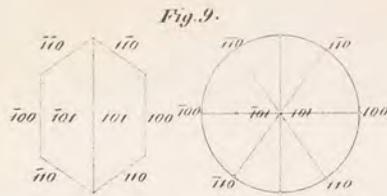


Bournonit

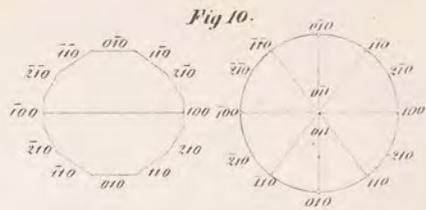




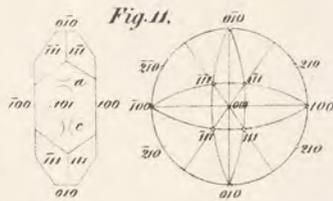
Salpetersaures Uranoxyd



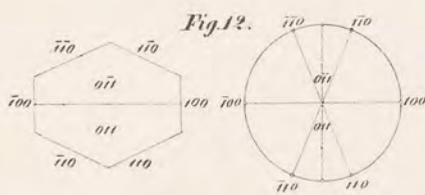
Lithium Kalium Platincyanur



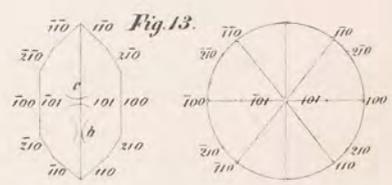
Essigsaaures Uranoxyd



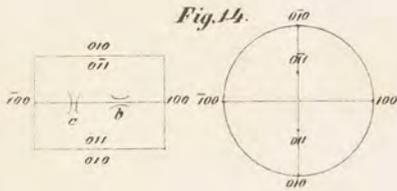
Salzaures Azophenylamin



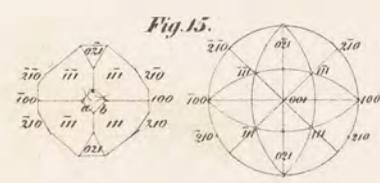
Isatin



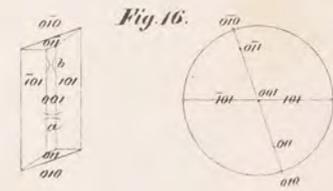
Weissbleierz



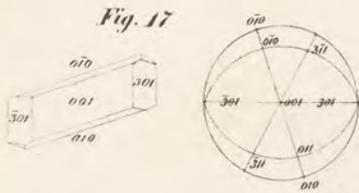
Caledonit



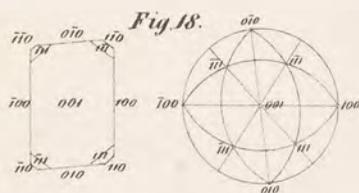
Skorodit



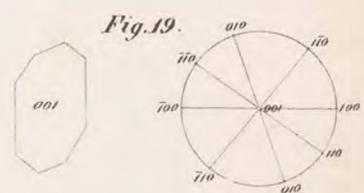
Gregorin



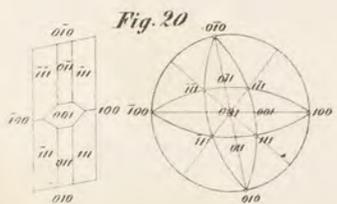
Oxalsaaures Eisenoxydkali



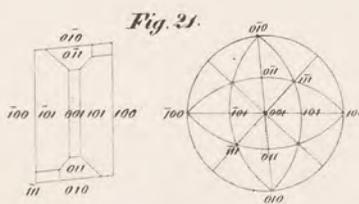
Schwefelsaaures Ceroxyd Ammoniak



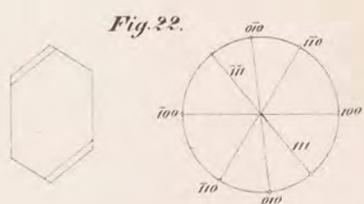
Rotheres oxalsaaures Chromoxyd Kali



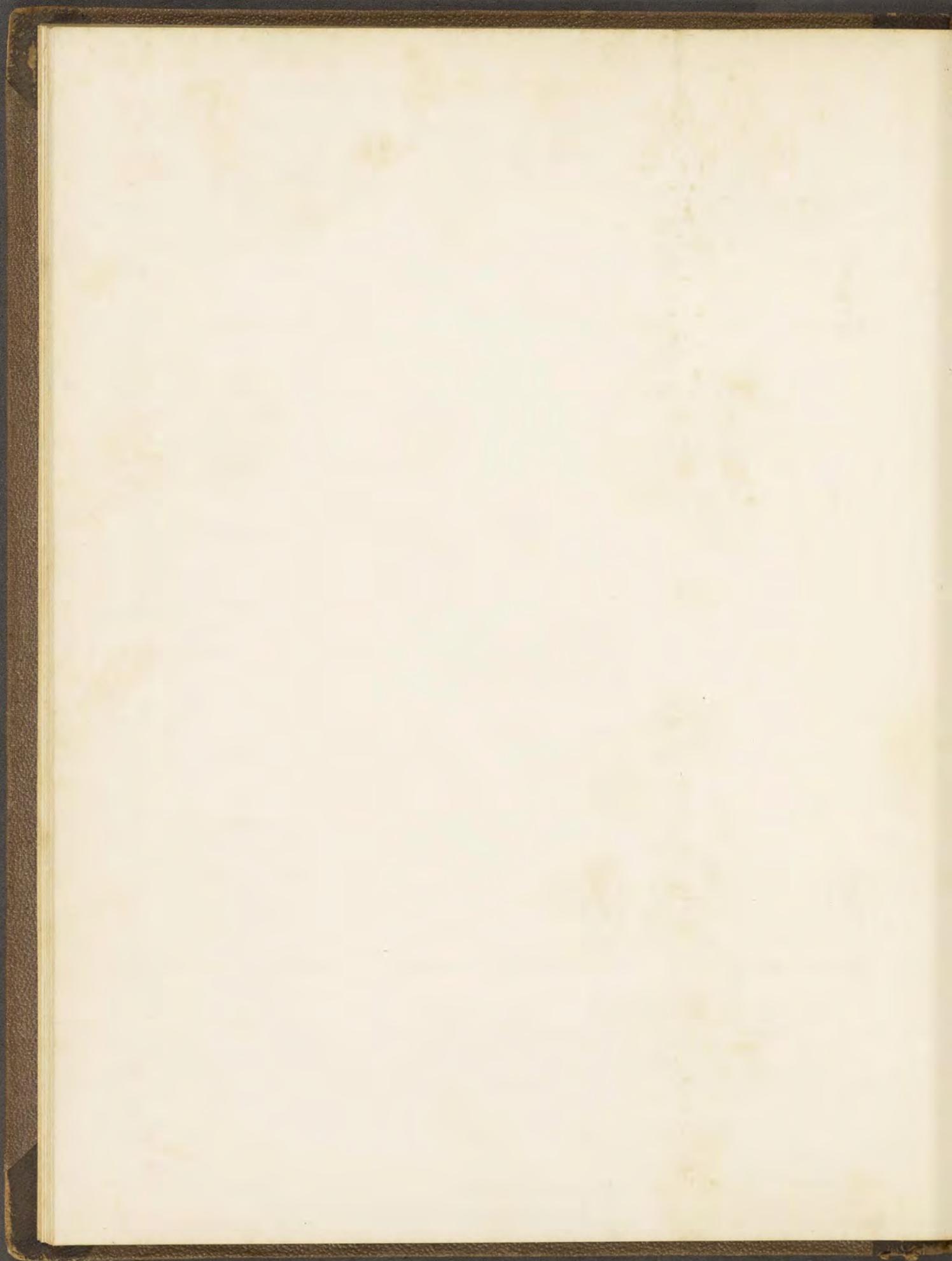
Blaues oxalsaaures Chromoxyd Natron

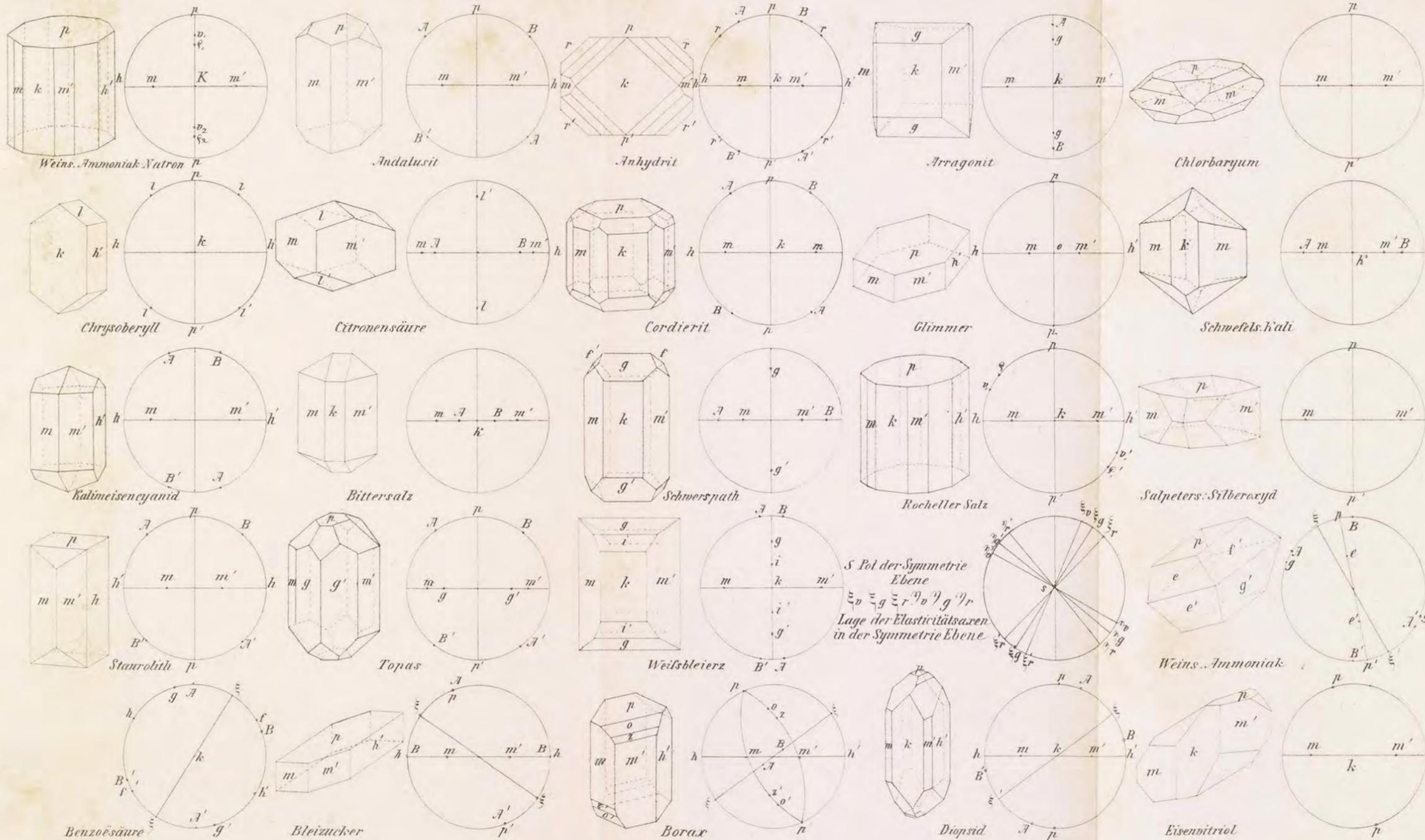


Propionsaaures Kupferoxyd

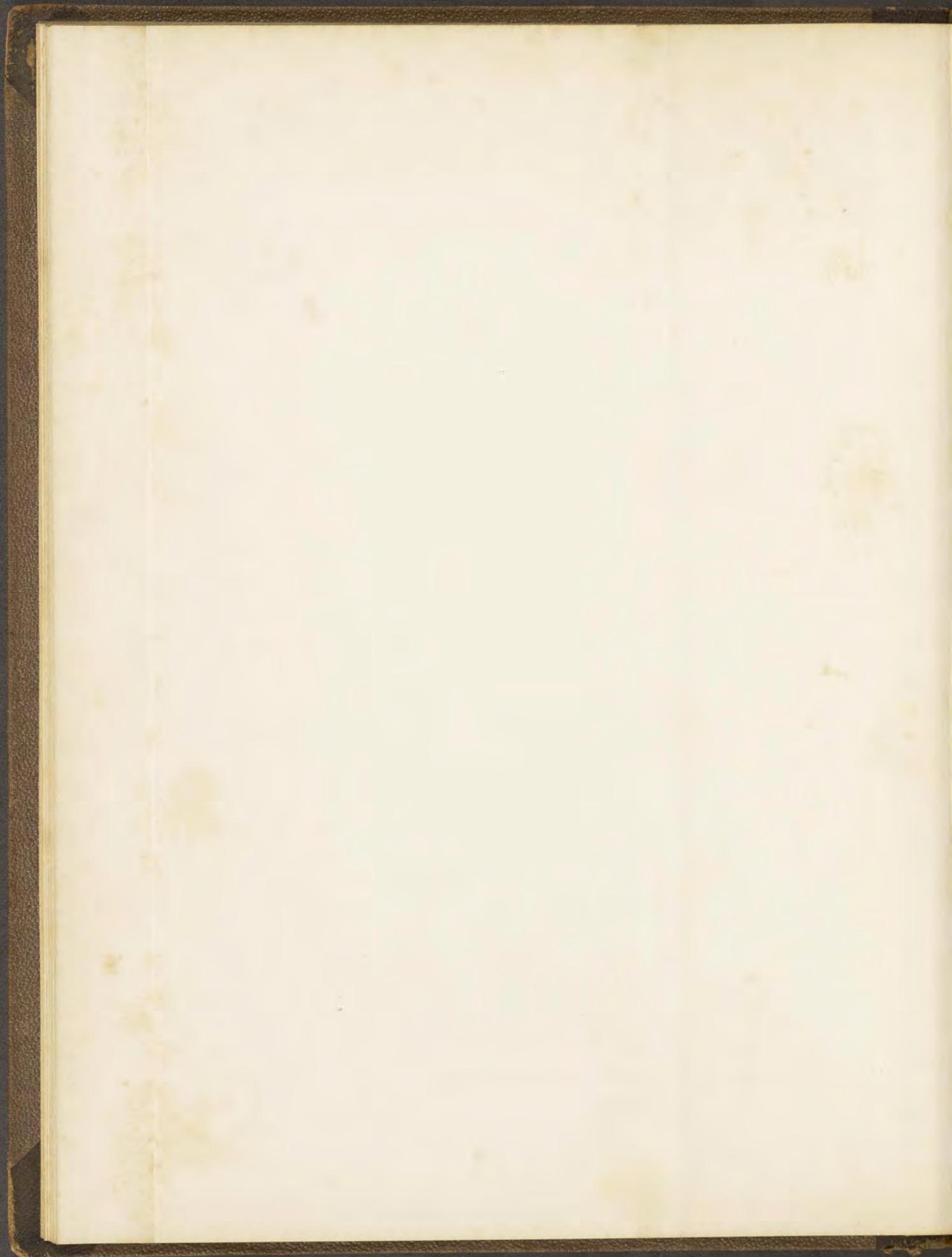


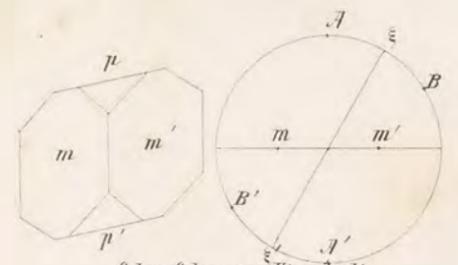
Dinitranilin



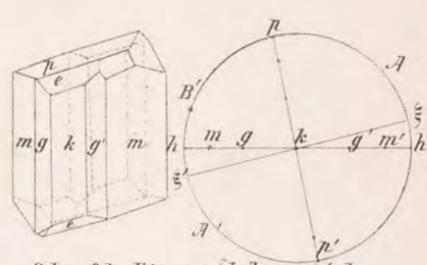


Nach Beer's hoh. Optik.

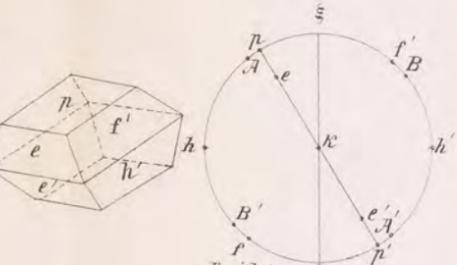




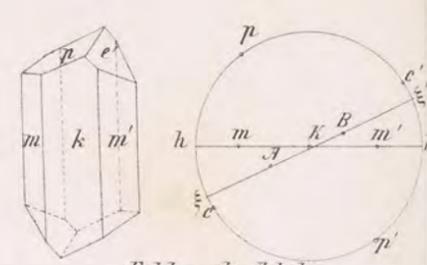
Schwefelsaures Eisenkali



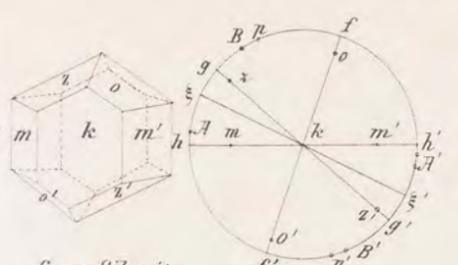
Schwefels. Eisenoxyd-Ammoniak



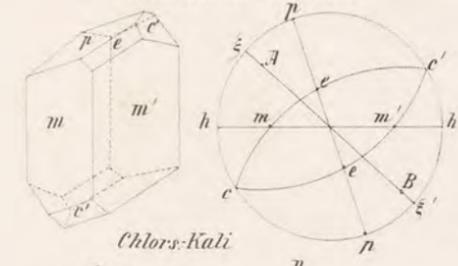
Epidot



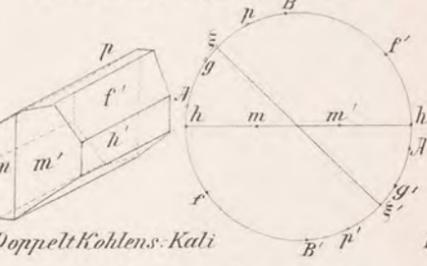
Feldspath, Adular



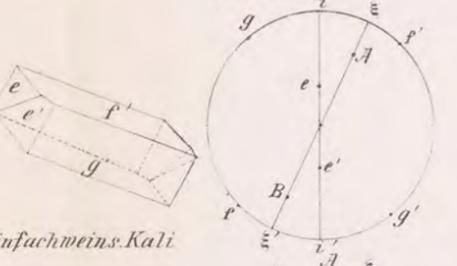
Gyps, Selenit



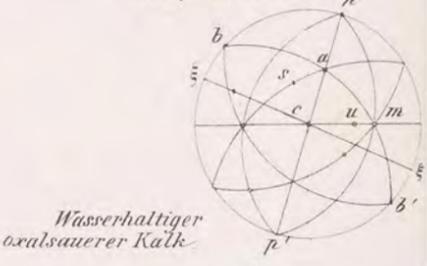
Chlor-Kali



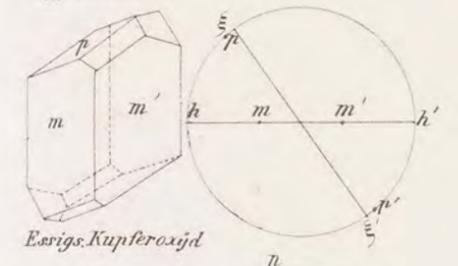
Doppelt Kohlens. Kali



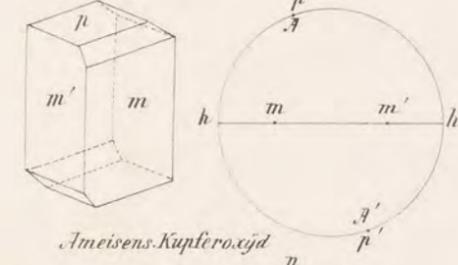
Einfachweins. Kali



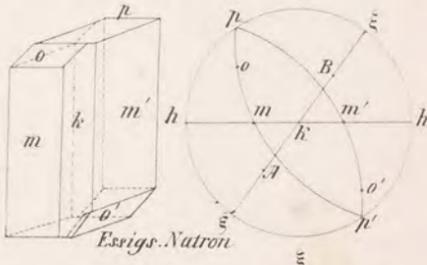
Wasserhaltiger oxalsaurer Kalk



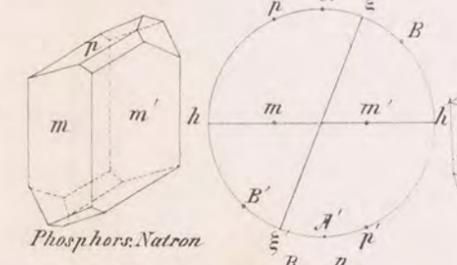
Essigs. Kupferoxyd



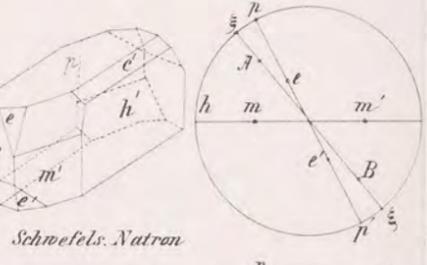
Ameisens. Kupferoxyd



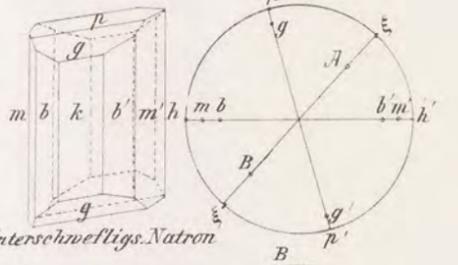
Essigs. Natron



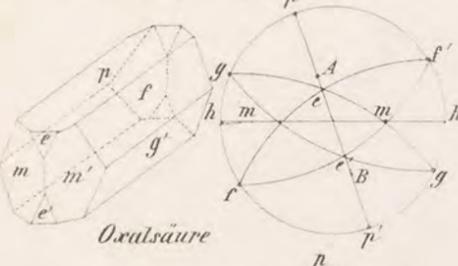
Phosphors. Natron



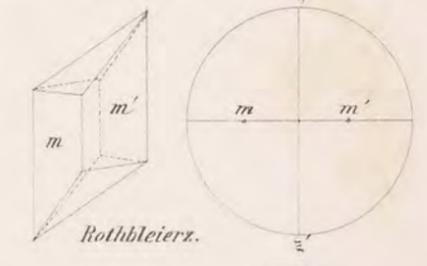
Schwefels. Natron



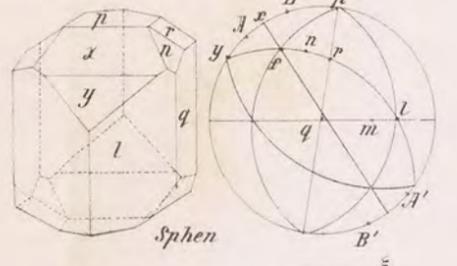
Unterschweifligs. Natron



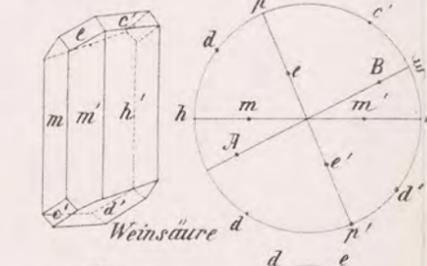
Oxalsäure



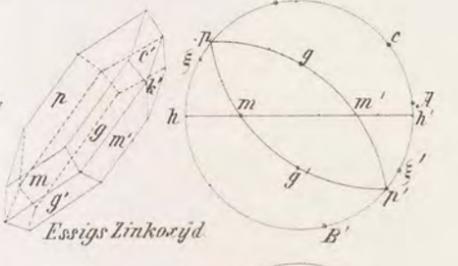
Rothbleierz.



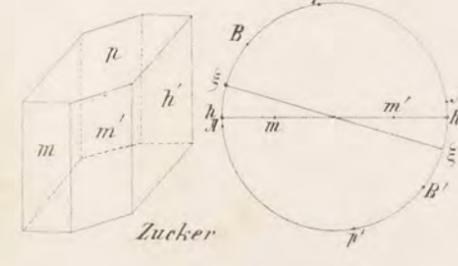
Sphen



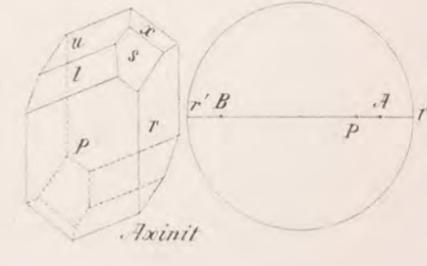
Weinsäure



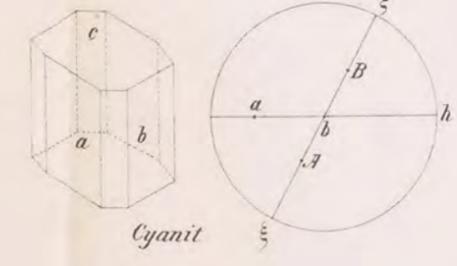
Essigs. Zinkoxyd



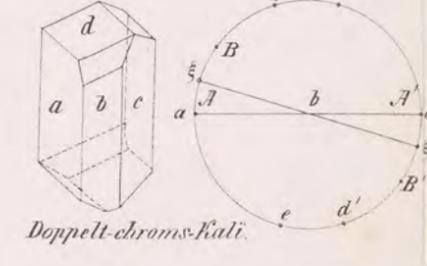
Zucker



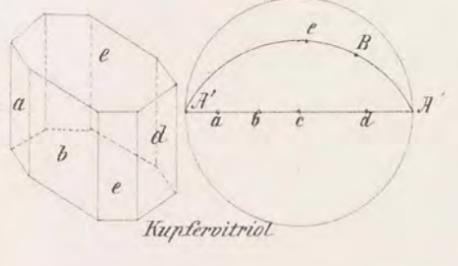
Axinit



Cyanit



Doppelt-chroms. Kali.



Kupfervitriol

Nach Berz. höh. Optik.

