

Ferromerrillite

$\text{Ca}_9\text{NaFe}^{2+}(\text{PO}_4)_7$

Crystal Data: Hexagonal. *Point Group:* $3m$. As anhedral grains to $50 \mu\text{m}$.

Physical Properties: *Cleavage:* None. *Fracture:* n.d. *Tenacity:* n.d. Hardness = ~ 5
 $D(\text{meas.}) = 3.14$ (Los Angeles) $D(\text{calc.}) = 3.11$

Optical Properties: Transparent. *Color:* Colorless. *Streak:* n.d. *Luster:* Vitreous.
Optical Class: Uniaxial (−). $\omega = 1.623$ $\varepsilon = 1.621$ Also anomalously biaxial with $2V$ up to 20° .

Cell Data: *Space Group:* $R\bar{3}c$. $a = 10.372(2)$ $c = 37.217(13)$ $Z = 6$

X-ray Powder Pattern: Shergotty meteorite/Los Angles meteorite.

2.861 (100)/2.86 (100), 3.191 (81)/3.19 (60), 2.594 (73)/2.594 (50), 6.46 (33)/6.42 (20),
2.741 (21)/2.747 (20), 1.710 (21)/1.713 (20), 5.196 (20)/5.24 (10)

Chemistry:	(1)	(2)
Na ₂ O	1.7	1.4
CaO	46.8	47.0
MgO	1.5	0.9
FeO	3.5	5.2
P ₂ O ₅	46.2	45.7
Total	99.7	100.2

(1) Shergotty meteorite; average of 8 electron microprobe analyses; corresponds to $\text{Ca}_{9.00}(\text{Na}_{0.60}\text{Ca}_{0.07})_{\Sigma=0.67}(\text{Fe}^{2+}_{0.53}\text{Mg}_{0.40})_{\Sigma=0.93}\text{P}_{7.08}\text{O}_{28}$. (2) Los Angeles meteorite; average of 10 electron microprobe analyses; corresponds to $\text{Ca}_{9.00}(\text{Na}_{0.49}\text{Ca}_{0.15})_{\Sigma=0.64}(\text{Fe}^{2+}_{0.78}\text{Mg}_{0.23})_{\Sigma=1.02}\text{P}_{7.03}\text{O}_{28}$.

Occurrence: An accessory phase in the basaltic and olivine-phyric subgroups of shergottite meteorites.

Association: Clinopyroxene, maskelynite (impact-melted plagioclase glass).

Distribution: From the Shergotty and Los Angeles shergottite meteorites.

Name: The Fe²⁺-dominant analogue of *merrillite*.

Type Material: A.E. Fersman Mineralogical Museum, Russian Academy of Sciences, Moscow, Russia (3514/1).

References: (1) Britvin, S.N., S.V. Krivovichev, and T. Armbruster (2016) Ferromerrillite, $\text{Ca}_9\text{NaFe}^{2+}(\text{PO}_4)_7$, a new mineral from the Martian meteorites, and some insights into merrillite-tuite transformation in shergottites. *Eur. J. Mineral.*, 28, 125-136. (2) (2016) Amer. Mineral., 101, 2357 (abs. ref. 1, with comment on the oxidation state of iron).