

## Hydropyrochlore

(H<sub>2</sub>O, □)<sub>2</sub>Nb<sub>2</sub>(O, OH)<sub>6</sub>(H<sub>2</sub>O)

**Crystal Data:** Cubic. *Point Group:* 4/m  $\bar{3}$  2/m. As octahedra, to 1 cm, which may be corroded.

**Physical Properties:** *Fracture:* [Uneven] (by analogy to the pyrochlore supergroup).  
*Tenacity:* [Brittle.] Hardness = 4-4.5 D(meas.) = 3.40-3.48 D(calc.) = 3.40-3.44

**Optical Properties:** Transparent. *Color:* Greenish. *Luster:* [Vitreous to resinous.]  
*Optical Class:* Isotropic.  $n = 1.93\text{-}1.99$

**Cell Data:** *Space Group:* Fd $\bar{3}$  m.  $a = 10.604(1)$  Z = 8

**X-ray Powder Pattern:** n.d.

Chemistry:	(1)	(2)	(3)		(1)	(2)	(3)
U <sub>3</sub> O <sub>8</sub>	0.10	0.08		PbO	0.01	0.02	
Nb <sub>2</sub> O <sub>5</sub>	78.60	80.05	75.69	MgO	0.11	0.11	
Ta <sub>2</sub> O <sub>5</sub>	0.06	0.11		MnO	0.08	0.06	
V <sub>2</sub> O <sub>5</sub>	0.02	0.03		CaO	0.41	0.13	0.12
TiO <sub>2</sub>	4.10	4.12	5.01	BaO	0.28	0.38	
ZrO <sub>2</sub>	0.66	0.37		SrO	2.60	1.73	1.75
SnO <sub>2</sub>		0.06		Na <sub>2</sub> O	0.35	0.58	
ThO <sub>2</sub>	0.12	0.17		K <sub>2</sub> O	2.73	2.76	2.12
Al <sub>2</sub> O <sub>3</sub>	0.18			F	0.38	0.11	
RE <sub>2</sub> O <sub>3</sub>	0.43	0.50		H <sub>2</sub> O <sup>+</sup>	8.54	8.37	[16.05]
Fe <sub>2</sub> O <sub>3</sub>	0.20	0.13		$\text{--O} \equiv \text{F}_2$	[0.16]	[0.05]	
FeO	0.06	0.07		Total	[99.86]	[99.89]	100.74

(1) Lueshe, Congo; original total given as 100.192%, corrected for goyazite 7.8%, ilmenite 2.0%, (rutile, brookite, anatase) 1.3%, calcite 0.9%, kaolinite 0.55%, goethite 0.3%, H<sub>2</sub>O<sup>+130°C</sup>. (2) Do.; original total given as 100.066%, corrected for goyazite 7.5%, ilmenite 1.9%, (rutile, anatase) 1.1%, calcite 1.6%, kaolinite 0.4%, goethite 0.2%, H<sub>2</sub>O<sup>+130°C</sup>. (3) Do.; by electron microprobe, H<sub>2</sub>O from structure analysis; corresponds to [(H<sub>2</sub>O)<sub>0.99</sub>Sr<sub>0.05</sub>Ca<sub>0.01</sub>]<sub>Σ=1.05</sub>(Nb<sub>1.80</sub>Ti<sub>0.20</sub>)<sub>Σ=2.00</sub>[O<sub>4.06</sub>(OH)<sub>1.94</sub>]<sub>Σ=6.00</sub>[(H<sub>2</sub>O)<sub>0.86</sub>K<sub>0.14</sub>]<sub>Σ=1.00</sub>.

**Mineral Group:** Pyrochlore supergroup (general formula - A<sub>2</sub>B<sub>2</sub>X<sub>6</sub>Y); pyrochlore group ( $B = \text{Nb}^{3+}$ ).

**Occurrence:** In alluvial deposits and residual soils from a carbonatite deposit, formed from pyrochlore by the leaching of Na, Ca, and F in waters rich in K ions.

**Association:** Na-Ca pyrochlores, lueshite, columbite, fersmite, ilmenite, rutile, barian goyazite.

**Distribution:** In the Lueshe carbonatite, 150 km north of Goma, Kivu Province, Congo (Zaire).

**Name:** For a member of the *pyrochlore* group with a prefix to indicate dominant H<sub>2</sub>O (*hydro*) in the Y site and in the A site. Formerly ‘kalipyrochlore’, now redefined.

**Type Material:** National Museum of Natural History, Washington, D.C., USA (136440).

**References:** (1) Van Wambeke, L. (1978) Kalipyrochlore, a new mineral of the pyrochlore group. Amer. Mineral., 63, 528-530. (2) Ercit, T.S., F.C. Hawthorne, and P. Černý (1994) The structural chemistry of kalipyrochlore, a “hydropyrochlore”. Can. Mineral., 32, 415-420. (3) Atencio, D., M.B. Andrade, A.G. Christy, R. Gieré, and P.M. Kartashov (2010) The pyrochlore supergroup of minerals: nomenclature. Can. Mineral., 48, 673-698.