THERMAL ENERGY STORAGE

Thermal storage is a method of storing heat for future use. Most often, thermal storage is used when future heat sources or sinks are either not available at a sufficient rate or will be more expensive. A common application of thermal storage is the domestic hot water heater. Most household water heaters use an electric resistance element or a natural gas burner as a heat source. A resistance element may supply heat at a rate of 6 kW, whereas a hot shower requires 20 kW to heat 6 L/min of water from 16° C to 40° C. Some amount of hot water must be stored in the tank because the element cannot heat the water at a rate sufficient for a shower or most other uses.

Heat is stored in the hot water tank to meet the demand for future use. The number or duration of showers possible depends upon the quantity of water stored and the tempera-

ture at which it is stored. To meet the desired water tempera- Phase change materials (PCMs) used in thermal storage ture of $40^{\circ}\mathrm{C}$ with 60° ter can be mixed 2.6 L/min with 16° C cold water. If the water is stored at 70° C, only 2.8 L/min of hot water is required.

ties of any thermal storage system are the amount of energy changes would be highly impractical to construct. However, stored and the temperature at which it is stored. If the tem- compressed air energy storage has been used at a few electriperature of the storage medium is below ambient conditions, cal plants where underground caverns are used to contain the it is a lack of heat (a heat sink) that is stored. Storage of a compressed air. The quantity of heat necessary to change heat sink rather than a heat source can alternatively be phase from solid to liquid (or from liquid to solid) is commonly called cool or cold storage. In either case, the potential to called the latent heat of fusion. For most substances, their transfer heat to or from the sink or source depends upon the latent heat of fusion is equivalent to the quantity of heat retemperature and mass of the storage medium. quired for a large change in temperature of the substance in

or changing its phase (e.g., from solid to liquid) or both. Cer- pressure (335 kJ/kg) is approximately 80 times greater than tain storage media may be better suited for a particular application depending upon their thermodynamic properties. The kJ/kg). specific heat, *c*, of a substance is a property relating the quan-
Chilled water storage systems (a heat sink or alternatively tity of energy stored or removed due to an increase or de- called a cooling source) have seen extensive use in building crease in temperature (without a change in phase.) Specific cooling systems. Usually, the cooling storage is used because ever, most practical thermal storage media are used in a solid (i.e., chillers) can be purchased at a cheaper rate during ceror liquid state where the variance of specific heat with pres- tain hours of the day. Chilled water storage is also employed sure is negligible. Also, the change in specific heat over mod- when the building's cooling requirements sometime exceed erate temperature ranges can be neglected in many applica- the capacity of the installed chillers (1). tions.

Materials with high specific heats are advantageous for use as thermal storage media. In many cases the volume oc- **FUNDAMENTALS** cupied by the storage medium is of more concern than the weight. In these cases the specific heat-density product pro- To illustrate the application of thermal storage, consider the heat capacity (2,454 kJ/m³-K vs. 3,935 kJ/m³ kJ/m^3-K .

applications are almost exclusively from solid to liquid or vice versa. Changing a liquid to the gas phase is accompanied by order of magnitude changes in volume (or pressure if the vol-As in the hot water tank example, the fundamental proper- ume is held fixed) and storage facilities to handle such Heat is stored by raising the storage medium temperature a single phase. Water's latent heat of fusion at atmospheric the heat associated with changing liquid water by 1° C (4.18)

heat is a temperature and pressure dependent property. How- the electricity used to operate the water chilling equipment

vides a better indicator in comparing alternative media transport of energy, whether thermal, mechanical or electrichoices. For example, aluminum has a specific heat nearly cal, from a source to a building cooling system as shown in twice that of steel (0.9 kJ/kg-K vs. 0.5 kJ/kg-K), but because Fig. 1. There may be a number of energy conversions (e.g., of their different densities, steel has a 60% larger volumetric from thermal to mechanical to electrical) as the energy moves from source to the end use. The original source could be from standard conditions, has a volumetric heat capacity of 4,175 the combustion of coal or natural gas, or from heat liberated in the fission of radioactive materials. It should be noted that

Figure 1. Without any intervening energy storage, the required energy must be simultaneously supplied through a chain of energy conversion processes.

the energy flows in the diagram are of ''utilizable'' energy which decreases in quantity at each conversion and transport process. The energy transferred from source to end use is conserved and, although not shown, significant quantities are transported to the ambient environment in each conversion and at the end process.

Without any intervening energy storage along the path shown in Fig. 1, the rate of energy transferred from the source and the rate of all energy conversions is directly linked to the rate of energy consumed by the refrigeration cycle. As the cooling load met by the refrigeration cycle increases or decreases, the source and all the conversions must respond immediately by increasing or decreasing their rate of energy

ergy generation, transfer and consumption can be decoupled. There are great economic advantages to decoupling the link. Without storage, all generation, conversion, and distribution equipment and facilities must be of sufficient capacity to meet

of the heat sources that contribute to the cooling load include heat conducted through exterior walls, heat from lighting inside the building, heat from building occupants as well as heat from personal computers and monitors. Also, fresh air used to ventilate the building must be cooled and dehumidified.

A building's exterior and interior mass constitute a thermal storage system. The rate of heat added or removed from the building mass depends primarily upon the weather (i.e., outdoor air temperature, incident solar radiation) and the building's internal air temperature. The interior temperature can vary over a relatively small range of comfortable interior temperatures during occupied hours and over a larger range during unoccupied hours. Thus the building mass can be employed as an active thermal storage system by controlling the interior temperature. Braun (2) discusses the control of building mass thermal storage as a means to reduce energy costs 6a.m 10 2p.m. 6 10 2a.m. 6 10 2p.m. 6 10 2a.m. 6 $\frac{1}{2}$ and peak electrical demand. Morris et al. (3) present results from experimental and simulated optimal control of building **Figure 3.** An example building cooling load profile over a 48-h thermal energy storage. The period.

transfer. (In any real energy conversion and transport system, e.g., an electrical generation and distribution network,
there is some energy storage; however, the amount is very
small in comparison with the quantity of ene

the largest expected load (e.g., peak load). With storage some-
where between the source ad back (e.g., peak load). With storage some where between the source and the end use, not all equipment
and facilities must be size

experience peak demands that heavily task their generating drawn from the other end. and distribution resources. Accordingly, utilities have a time of day rate structure with relatively high costs during high-
load hours and lower costs during low-load hours. With ther-
mal storage, ice storage uses the heat trans-
mal storage, the chiller can run either solely or mor during the lower rate periods charging the storage tank to meet future loads that occur during the higher rate period.

In addition to time of day rates, utilities often assess a charge for the customer's largest electrical demand during ice is formed and stored (5). In external ice-on-coil systems, some billing period (e.g., during one month.) Using thermal ice forms on coils submerged in a water storage tank. Alstorage can also reduce electrical demand peaks. Usually, the though the source of cooling in most ice storage systems is a storage is charged by running the chiller overnight when refrigeration cycle, a secondary fluid rather than the refrigerother major electrical systems (e.g., lighting, personal com- ant is circulated inside the coils. Typically an ethylene glycol puters, etc.) are off, resulting in a reduction in the total build- and water mixture is used as the secondary fluid. In external
ing electrical demand. Also, the longer the time available to ice-on-coil systems, ice is bu ing electrical demand. Also, the longer the time available to charge the storage tank, the lower the required chiller capac- thick as 7 cm. The water in the tank is aerated or otherwise ity (rate of charging) and therefore the lower the chiller agitated so that the ice forms on the coils uniformly. Excespower demand. sively thick ice formation creates ice bridges, restricting wa-

ing loads by chilled water. The chilled water is usually in a the charging mode, the secondary fluid circulates between the closed loop system where a refrigeration cycle provided the coils in the tanks and a heat exchanger coupled with the recooling source. A simple storage method is to use the chilled frigeration system evaporator. In discharging the store, the water as the storage medium. Typically, large water tanks secondary fluid is rerouted through another heat exchanger
are used to store the water (4). In some facilities the on-site coupled with the cooling water loop that are used to store the water (4) . In some facilities the on-site water storage for the fire suppression system is used for load. The ice melts from the inside (at the coil external sur-

As chilled water is drawn from a storage system, such as the tank in Fig. 2, an equivalent amount of warmer water advantageous over the external melt system because the refrom the load is added to the system. To have an effective frigeration cycle does not have to operate at the lower temperchilled water storage system the colder chilled water must be atures needed to freeze water on the outside of the formation. separated from the warmer water it displaces. When a single Ice harvester systems use a separate tank to store the ice tank is used for chilled water storage, the natural stratifica- (6). Ice is built up in cycles on the external surface of a vertition of colder water at the bottom of the tank and the warmer cally positioned refrigerant ev tion of colder water at the bottom of the tank and the warmer cally positioned refrigerant evaporator. After building up a water at the top can be employed as a separation method. In certain thickness of ice on the coils (approximately 0.5 to 1.0) order to maintain a sharp vertical gradient in temperature cm), the refrigerant is valved such th order to maintain a sharp vertical gradient in temperature (thermocline) between the warmer and colder water masses, compressor enters the evaporator for a short period of time. steps must be taken to minimize mixing in the tank. Baffles The hot gas melts a thin film of ice at the evaporator external may be placed in the tank to minimize horizontal mixing. surface causing the ice build-up to fall into an ice storage tank Also, using many distributed inlet and outlet nozzles helps to below. Alternatively, in some systems the ice is scraped away minimize mixing as water is added or removed from the tank. mechanically. Chilled water is pumped from the storage tank Since water's density at standard atmospheric pressure is to the heat exchangers serving the cooling load, then back greatest at about 4° C, care must be taken to not charge the over the evaporator coils. tank with water at or below this temperature. Other systems use small containers of water (e.g., plastic

phragm) to keep the cold and warm water masses separate. and melting the ice is similar to the internal melt on coil stor-The membrane can rise and fall as the tank is charged and age systems. A secondary heat transfer fluid is circulated discharged. The cold and warm water masses can also be sep- around the water-filled containers and the refrigerant evapoarated using a series of smaller tanks or compartments. The rator when freezing the ice. During the melting cycle, the sec-

Usually, because almost all buildings will experience peak colder chilled water is always drawn or added to the system late afternoon loads in the summer, electrical utilities also at one end of the series and the warmer water added to or

mal storage, the chiller can run either solely or more often As with chilled water storage, ice storage uses the heat trans-
during the lower rate periods charging the storage tank to fer fluid as the storage medium. Howev the relatively high latent heat of fusion of water (335 kJ/k $^{\circ}$ C at 0° C) is utilized. Ice storage systems differ in the way the To charge the storage tank adequately for future load re- ter circulation around the coils and impeding heat transfer.

quirements, a predicted load profile must be assumed. If it is When building ice externally on the coils, a thickness limit desired that the storage tank serve all of the load in Fig. 3 is imposed, as the increasing thickness of the ice tends to infrom 6 a.m. to 10 p.m. on the second day, the storage tank sulate the coil. A colder coolant temperature inside the coils would need to be charged with a total capacity equal to or is required to continue forming ice as the thickness increases. greater than the integrated load over the 6 a.m. to 10 p.m. The lower coolant temperature requires the refrigeration systime period. tem evaporator to operate at a lower temperature and consequently reduces cycle efficiency.

In discharging the ice storage, warmer water is circulated about the coils, melting the ice from the outside of the forma-Chilled Water Storage The Storage Energy in the Chilled Water Storage In internal melt ice-on-coil systems, the secondary heat

In many large cooling systems, heat is transported from cool- transfer fluid is used in charging and discharging modes. In chilled water storage.

As chilled water is drawn from a storage system, such as water to the ice storage tank. The internal melt system is
 $\frac{1}{100}$

Thermal storage tanks may also use a membrane (dia- balls filled with water) to store ice. The method of freezing

ondary fluid is circulated around the containers and another gions with long periods of subzero weather where air source heat exchanger coupled with the chilled water loop serving heat pumps are not advantageous over resistance heat. Also, the cooling load. in the extreme cold climates, electrical utility peaks occur

be used in cool storage systems (7). Typical building air condi- rockbed is charged during lower rate off-peak periods in the tioning systems use approximately 7°C water in coils to cool air flows from approximately 24°C to 13°C . Using a PCM with a phase-change temperature of approximately 6° C rather than water at 0° C allows the refrigeration cycle to operate at than water at 0° C allows the refrigeration cycle to operate at
a warmer evaporator temperature and therefore use less en-
ergy input per kilojoule of cooling. Typically mixtures of inor-
ganic salts and water having

age (at 0°C) take advantage of the cooler thermal sink and
circulate smaller amounts of relatively low temperature air
through the building (1). The lower temperature air condi-
through the building (1). The lower temperat

with heating requirements. Solar heating systems are gener- ments during the day. The use of PCMs in the building mate-
ally classified as passive or active systems. Passive solar heat-
rials, such as wallboard, is a metho ally classified as passive or active systems. Passive solar heat-
ing systems rely on natural circulation in transporting heat
storage canacity while limiting temperature changes to storage whereas active systems use forced circulation (i.e.,

utilizing pumps or fans).

Solar water heating systems, active or passive, typically

use water storage tanks for thermal storage Simple domestic Many electric generating utilities use gas turbines to help use water storage tanks for thermal storage. Simple domestic Many electric generating utilities use gas turbines to help
solar water heating systems may use existing electric or naturalization existelling a gas turbine pea ral gas fired water heaters for storage. When solar energy is insufficient to meet the water heating load, the electrical ele-
larger base load plant capacities even though the gas turbine ply. With an air/water heat exchanger, solar water heating systems can also serve air heating loads.
Various passive solar heating systems may be employed to \overline{A} gas turbine can lose 25% of its generating capacity as

Various passive solar heating systems may be employed to store some amount of incident solar energy during the day for the air inlet temperature increases from 15^oC to 38^oC. During
use overnight (8.9) A large area of glazing (windows) on the the peak demand periods, it is t use overnight (8,9). A large area of glazing (windows) on the the peak demand periods, it is the plant's capacity and not
south face of a huilding (in the porthern hemisphere) can be the cost of energy that is of the great south face of a building (in the northern hemisphere) can be the cost of energy that is of the greatest value. Ice storage
combined with an interior wall facing the glazing During the systems have been employed to precool combined with an interior wall facing the glazing. During the systems have been employed to precool turbine inlet air dur-
daylight hours incident solar energy warms the wall to the ing peak load periods (11). Electrical e daylight hours, incident solar energy warms the wall to the ing peak load periods (11). Electrical energy from the same
extent that the wall cools and warms the air in the house turbine or any other source on the grid is u extent that the wall cools and warms the air in the house during the hours of darkness. Heat transfer from the wall to peak hours to build the ice store. Although the overall cost the air in the house is primarily by convection and in some per kilowatt-hour may be increased with the investment and cases the air flow can be controlled to minimize losses during operation of an ice storage/inlet air co cases the air flow can be controlled to minimize losses during operation of an ice storage/inlet air cooling system, the in-
the storage period and to enhance heat transfer during the crease in available capacity is usuall the storage period and to enhance heat transfer during the crease in available
discharge period. discharge period.

Electric Heat—Rockbed Storage DESIGN

Rockbed thermal storage systems are usually employed in conjunction with electrical resistance household heating sys- Thermal storage systems may be designed to meet all or part

Phase-change materials (PCMs) other than water may also during the early morning hours of extremely cold days. The daylight hours and discharged at night.

Building Mass Thermal Storage

start and much having phase enargie competitions with the building's air conditioning system. During the night,
some building air conditioning systems that use ice stor-
age (at 0°C) take advantage of the cooler thermal si

Solar Heating Systems is a possibility of condensation on surfaces when ventilation air is brought into the building. However, because of the large Some means of thermal storage is required in solar heating building mass Some means of thermal storage is required in solar heating building mass, even a few degrees of decreased building mass
systems as the availability of solar energy is seldom in synch temperature can contribute to decreasin temperature can contribute to decreasing cooling requirestorage capacity while limiting temperature changes.

solar water heating systems may use existing electric or natu-
ral gas fired water heaters for storage. When solar energy is ten more economical for a utility than investing in increasing ment or natural gas burner will maintain the hot water sup-
now have a higher cost-per-kilowatt-hour rate. Warmer
now With an air/water heat exchanger solar water heating weather, which contributes to higher electrical dem

C to 38°C. During

tems. With time-of-day electrical rates (on-peak and off-peak of a given load over some period of time. For example, in pricing), the thermal storage system can heat the rock mass chilled water storage systems, the water storage tank may be during the lower rate period for discharge during the high- sized to meet all the building's cooling load during occupied rate period. The primary advantage of the electrical resis- hours with the chiller system charging the entire store overtance heater/rockbed combination is the ability to use off- night. Alternatively, in a situation where the building's chillpeak electricity. The rockbed storage is typically used in re- ing system is undersized, a thermal storage system may be

sized only to meet the difference between the building's peak cient rate or at relatively low cost) and discharge it over an-

that must be met and the capacity of the thermal energy controlled. To maximize the benefits of thermal storage, an source. An example cooling load profile over a 48-h period is optimal control strategy that minimizes cost or some other shown in Fig. 3. One possible profile for operating the chiller desired criterion should be employed (15). is also shown. The quantity of energy storage required is indi- In some cases, when the heat source or sink either will not cated by the integrated area below the required load curve be available in the future or will be available only at great and above the available chiller load curve. In order to charge cost, the optimal strategy is obvious. In a solar heating systhe store, the integrated area where the chiller capacity ex- tem, the incident solar energy is free, and, if the heating receeds the required load must exceed the area indicating quirement exceeds the maximum amount of heat storage, the storage. store should be charged as much as possible. If the on- and

load and can have almost any profile of operation so long as of magnitude, a chilled water system should be charged fully the integrated area associated with charging the store ex- (or at least to some amount greater than the expected next ceeds the area associated with storage discharge. However, day's cooling load) when the lower rate is available. there is at least one profile of chiller operation that minimizes In most applications, the best or optimal control strategy energy costs over the subject time period. The storage indi- is not entirely obvious, but may be determined through simucated by this optimal chiller operation profile is the optimal lation or other analysis. A general outline of an approach usenergy storage requirement over the subject period. But for a ing simulation follows: valid economic analysis, operation over a longer period, such as the anticipated life of the system, along with equipment Develop a model of the system including the storage unit costs must be considered. and the load served by the store. Often weather-depen-

predicted load profiles can be used to estimate operating costs using statistical models. over a given period of time. Weather data used in the simula- Identify controlled variables such as the mass flow rate of tion can be real measured data or meteorological data which the fluid entering the store. Identify the external time simulates a typical year. Various storage charging and dis- varying uncontrolled variables such as the predicted charging schedules along with various storage sizes and heat-transfer load to be served by the storage system.

chiller capacities can be simulated to determine an optimal and heat-transfer load to be served by the storag

Electrical rates that differ depending on the time of day clude the outdoor weather conditions.
are usually the primary influence in determining the opti-
magnetic process with are usually the primary influence in determining the optimation of the model should reasonably agree with the actual pro-
number and discharging and discharging theremining the best operating scheedule (or profile) for ch

rate structure. For example, if RTP rates are significantly Create a computational model of the combined system,
lower over a weekend it may be economically advantageous storage unit, and load. The computational model shou lower over a weekend, it may be economically advantageous storage unit, and load. The computational model should
to have a storage system canable of meeting loads over more take as input, predictions of the load profile as to have a storage system capable of meeting loads over more take as input, predictions of the load profile as well as
then one day Darvanian and Bohn (12) investigate and com-
time varying values of the controlled variable time varying values of the controlled variables. Model
than one day. Daryanian and Bohn (12) investigate and com-
output should include calculated values of the objective
output should include calculated values of the obje pare optimal sizing of storage systems under both RTP and output should include calculated values of day electrical rates time of day electrical rates.

signed for use by electrical generation plants. Somasundaram gramming that can determine a trajectory of control et al. (13,14) investigate the application of thermal storage in variable values that minimizes the objective function power plants as well as applications in cogeneration systems. (e.g., energy costs) over a particular time period (e.g., 24

period (when the heat source or sink is available at a suffi- ing the results (15).

cooling load and the undersized chiller's capacity. other. However, in most thermal storage systems, the rate Storage size can be determined by assuming a load profile and duration of charging and discharging the store can be

With thermal storage, the chiller is decoupled from the off-peak rates of an electrical rate structure differ by orders

- Models of the thermal storage system, chiller system, and dent or otherwise time-varying loads can be predicted
- In some cases the thermal loads may be calculated from storage size. a building model where the external variables will in-
	-
	-
	-
	-
- High temperature thermal storage systems may also be de-
Use an optimal control algorithm such as dynamic proh). Although a study period of 30 days may produce a particular minimum energy cost, shortening the time **OPERATION** period (i.e., horizon) to 24 h, for example in a building cooling storage system, could greatly decrease the diffi-A simple operating strategy is to charge the store over one culty of the problem solution while only slightly degrad-

In the United States as well as other industrialized nations, 12. B. Daryanian and R. E. Bohn, Sizing of electric thermal storage electric utilities have been converted from government- or in-
under real time pricing, IEEE electric utilities have been converted from government- or in-

<u>under real time pricing</u>, *m* (1993, *8* (1993, *8* vestor-owned regional monopolies to open market competi-
tors. In an open market, electrical energy is a commodity 13. S. Somasundaram et al. Integrating thermal energy storage in tors. In an open market, electrical energy is a commodity 13. S. Somasundaram et al., Integrating thermal energy storage in the sumply and demand However unlike $\frac{1}{2}$ power plants, Mech. Eng., 115 (9): 84–90, 1993. priced according to its supply and demand. However, unlike most other commodities, marketed electrical energy is not 14. S. Somasundaram et al., Cost evaluation of diurnal thermal en-

Although electrical energy is not normally storable by elec-
city consumers, with thermal storage, relatively lower cost. 15. G. P. Henze, R. H. Dodier, and Moncef Krarti, Development of a tricity consumers, with thermal storage, relatively lower cost 15. G. P. Henze, R. H. Dodier, and Moncef Krarti, Development of a
periodic consumer of a storage, HVAC electricity available during low-demand periods can be used
to generate or transport heat into sources or sinks for use R . Res., 3 (3): 233–264, 1997.
during periods of bigbor demand and sect. As the number and 16. J. S during periods of higher demand and cost. As the number and and S of thermal storage systems increase, the large varia-
capacity of thermal storage systems increase, the large varia-
tions in electrical demand will decre

by Caldwell and Bahnfleth (16).
This article has presented the basic uses and operation of Air Force Institute of Technology thermal energy storage systems. In addition to the applica-
Air Force Institute of Technology tions listed, there are other unique thermal storage systems including those used in storing high-temperature heat sources used in power generating cycles (17). Although the media (typically molten salts) and temperatures in power cycle thermal storage are significantly different than chilled water or ice storage, the fundamental use and operation of the thermal storage is the same.

BIBLIOGRAPHY

- 1. C. E. Dorgan and J. S. Elleson, ASHRAE design guide for cool thermal storage, *ASHRAE Trans.,* **100** (1): 33–38, 1994.
- 2. J. E. Braun, Reducing energy costs and peak electrical demand through optimal control of building thermal storage, *ASHRAE Trans.,* **96** (2): 876–888, 1990.
- 3. F. B. Morris et al., Experimental and simulated performance of optimal control of building thermal storage, *ASHRAE Trans.,* **100** (1): 402–414, 1994.
- 4. K. O. Homa, C. W. Sohn, and S. L. Soo, Thermal performance of stratified chilled water storage tanks, *HVAC R. Res.,* **2** (2): 158–170, 1996.
- 5. *ASHRAE Handbook: Heating, Ventilating, and Air-Conditioning Applications.* Atlanta: ASHRAE, 1995.
- 6. D. E. Knebel, Optimal design and control of ice harvesting thermal energy storage systems, *Proc. ASME Natl Heat Transfer Conf.,* Minneapolis, MN, 1991, pp. 1–9.
- 7. I. O. Salyer and A. K. Sircar, Review of phase change materials research for thermal energy storage in heating and cooling applications at the University of Dayton from 1982 to 1996, *Int. J. Global Energy Issues,* **9** (3): 183–198, 1997.
- 8. J. A. Duffie and W. A. Beckman, *Solar Engineering of Thermal Processes,* 2nd ed., New York: Wiley, 1991.
- 9. I. Dincer and S. Dost, Perspective on thermal energy storage systems for solar energy applications, *Int. J. Energy Res.,* **20** (6): 547–557, 1996.
- 10. I. Andersen and M. J. Brandemuehl, Heat storage in building thermal mass: A parametric study, *ASHRAE Trans.,* **98** (1): 910– 918, 1992.
- 11. K. J. Cross et al., Evaluation of thermal storage options for combustion turbine inlet air cooling, in *Thermodynamics and the De-*

CONCLUSION *sign, Analysis and Improvement of Energy Systems,* New York: ASME, 1991.

-
-
- ergy storage for cogeneration applications, *Energy Eng.*, **89** (4):
 $8-22$ 1992
	-
	-
	-